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(54) Title: AROMATIC SULFONE HYDROXAMIC ACID METALLOPROTEASE INHIBITOR

(57) Abstract

A treatment process is disclosed that comprises administering an effective amount of an aromatic sulfone hydroxamic acid that exhibits excellent inhibitory activity of one or more matrix metalloprotease (MMP) enzymes, such as MMP-2, MMP-9 and MMP-13, while exhibiting substantially less inhibition at least of MMP-1 to a host having a condition associated with pathological matrix metalloprotease activity. Also disclosed are metalloprotease inhibitor compounds having those selective activities, processes for manufacture of such compounds and pharmaceutical compositions using an inhibitor. A contemplated compound corresponds in structure to formula (B).

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AROMATIC SULFONE HYDROXAMIC ACID METALLOPROTEASE INHIBITOR

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Description

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Technical Field

This invention is directed to proteinase (protease) inhibitors, and more particularly to the use of aromatic sulfone hydroxamic acid compounds that, inter alia, are selective inhibitors of matrix metalloproteinases in a process for treating conditions associated with pathological matrix metalloproteinase activity, the selective inhibitors themselves, compositions of proteinase inhibitors, 20 intermediates for the syntheses of proteinase inhibitors, and processes for the preparation of proteinase inhibitors.

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Background of the Invention

Connective tissue, extracellular matrix constituents and basement membranes are required components of all mammals. These components are the biological materials that provide rigidity, differentiation, attachments and, in some cases, elasticity to biological systems including human beings and other mammals. Connective tissues components include, for example, collagen, elastin, proteoglycans, fibronectin and laminin. These

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biochemicals makeup, or are components of structures, such as skin, bone, teeth, tendon, cartilage, basement membrane, blood vessels, cornea and vitreous humor.

Under normal conditions, connective tissue turnover and/or repair processes are controlled and in equilibrium. The loss of this balance for whatever reason leads to a number of disease states. Inhibition of the enzymes responsible loss of equilibrium provides a control mechanism for this tissue decomposition and, therefore, a treatment for these diseases.

Degradation of connective tissue or connective tissue components is carried out by the action of proteinase enzymes released from resident tissue cells and/or invading inflammatory or tumor cells. A major class of enzymes involved in this function are the zinc metalloproteinases (metalloproteases).

The metalloprotease enzymes are divided 20 into classes with some members having several different names in common use. Examples are: collagenase I (MMP-1, fibroblast collagenase; EC 3.4.24.3); collagenase II (MMP-8, neutrophil collagenase; EC 3.4.24.34), collagenase III (MMP-13), 25 stromelysin 1 (MMP-3; EC 3.4.24.17), stromelysin 2 (MMP-10; EC 3.4.24.22), proteoglycanase, matrilysin (MMP-7), gelatinase A (MMP-2, 72 kDa gelatinase, basement membrane collagenase; EC 3.4.24.24), gelatinase B (MMP-9, 92 kDa gelatinase; EC 3.4.24.35), stromelysin 3 (MMP-11), metalloelastase (MMP-12, HME, human macrophage elastase) and membrane MMP (MMP-14). MMP is an abbreviation or acronym representing the term Matrix Metalloprotease with the attached numerals providing differentiation between 35 specific members of the MMP group.

The uncontrolled breakdown of connective tissue by metalloproteases is a feature of many pathological conditions. Examples include rheumatoid arthritis, osteoarthritis, septic arthritis; corneal, epidermal or gastric ulceration; tumor metastasis, invasion or angiogenesis; periodontal disease; proteinuria; Alzheimers Disease; coronary thrombosis and bone disease. Defective injury repair processes also occur. This can produce improper wound healing leading to weak repairs, adhesions and scarring. These latter defects can lead to disfigurement and/or permanent disabilities as with post-surgical adhesions.

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Metalloproteases are also involved in the 15 biosynthesis of tumor necrosis factor (TNF), and inhibition of the production or action of TNF and related compounds is an important clinical disease treatment mechanism. TNF- α , for example, is a cytokine that at present is thought to be produced 20 initially as a 28 kD cell-associated molecule. It is released as an active, 17 kD form that can mediate a large number of deleterious effects in vitro and in vivo. For example, TNF can cause and/or contribute to the effects of inflammation, rheumatoid arthritis, autoimmune disease, multiple sclerosis, graft 25 rejection, fibrotic disease, cancer, infectious diseases, malaria, mycobacterial infection, meningitis, fever, psoriasis, cardiovascular/ pulmonary effects such as post-ischemic reperfusion 30 injury, congestive heart failure, hemorrhage, coagulation, hyperoxic alveolar injury, radiation damage and acute phase responses like those seen with infections and sepsis and during shock such as septic shock and hemodynamic shock. Chronic release of active TNF can cause cachexia and anorexia. 35 be lethal, and TNF can help control the growth of tumor cells.

TNF-α convertase is a metalloprotease
involved in the formation of soluble TNF-α.
Inhibition of TNF-α convertase (TACE) inhibits
production of active TNF-α. Compounds that inhibit

5 both MMPs activity and TNF-α production have been disclosed in WIPO International Publication Nos. WO 94/24140, WO 94/02466 and WO 97/20824. Compounds that inhibit MMPs such as collagenase, stromelysin and gelatinase have been shown to inhibit the release of TNF (Gearing et al. Nature 376, 555-557 (1994), McGeehan et al., Nature 376, 558-561 (1994)). There remains a need for effective MMP inhibitors. There also remains a need for effective TNF-α convertase inhibiting agents.

15 MMPs are involved in other biochemical processes in mammals as well. Included is the control of ovulation, post-partum uterine involution, possibly implantation, cleavage of APP (β -Amyloid Precursor Protein) to the amyloid plaque and inactivation of α_1 -protease inhibitor (α_1 -PI). Inhibition of these metalloproteases permits the control of fertility and the treatment or prevention

25 administered serine protease inhibitor drug or biochemical such as α_1 -PI supports the treatment and prevention of diseases such as emphysema, pulmonary diseases, inflammatory diseases and diseases of aging such as loss of skin or organ stretch and resiliency.

of Alzheimers Disease. In addition, increasing and

maintaining the levels of an endogenous or

Inhibition of selected MMPs can also be desirable in other instances. Treatment of cancer and/or inhibition of metastasis and/or inhibition of angiogenesis are examples of approaches to the treatment of diseases wherein the selective inhibition of stromelysin, gelatinase A or B, or

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collagenase III appear to be the relatively most important enzyme or enzymes to inhibit especially when compared with collagenase I (MMP-1). A drug that does not inhibit collagenase I can have a superior therapeutic profile. Osteoarthritis, another prevalent disease wherein it is believed that cartilage degradation of inflamed joints is at least partially caused by MMP-13 released from cells such as stimulated chrondrocytes, may be best treated by administration of drugs one of whose modes of action is inhibition of MMP-13. See, for example, Mitchell et al., J. Clin. Invest., 97:761-768 (1996) and Reboul et al., J. Clin. Invest., 97:2011-2019 (1996).

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Inhibitors of metalloproteases are known. Examples include natural biochemicals such as tissue 15 inhibitors of metalloproteinases (TIMPs), α_2 macroglobulin and their analogs or derivatives. These endogenous inhibitors are high molecular weight protein molecules that form inactive complexes with 20 metalloproteases. A number of smaller peptide-like compounds that inhibit metalloproteases have been described. Mercaptoamide peptidyl derivatives have shown ACE inhibition in vitro and in vivo. Angiotensin converting enzyme (ACE) aids in the production of angiotensin II, a potent pressor 25 substance in mammals and inhibition of this enzyme leads to the lowering of blood pressure.

Thiol group-containing amide or peptidyl amide-based metalloprotease (MMP) inhibitors are known as is shown in, for example, W095/12389, W096/11209 and U.S. 4,595,700. Hydroxamate group-containing MMP inhibitors are disclosed in a number of published patent applications such as WO 95/29892, WO 97/24117, WO 97/49679 and EP 0 780 386 that disclose carbon back-boned compounds, and WO 90/05719, WO 93/20047, WO 95/09841 and WO 96/06074 that disclose hydroxamates that have a peptidyl back-

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bones or peptidomimetic back-bones, as does the article by Schwartz et al., Progr. Med. Chem., 29:271-334(1992) and those of Rasmussen et al., Pharmacol. Ther., 75(1): 69-75 (1997) and Denis et al., Invest. New Drugs, 15(3): 175-185 (1997).

One possible problem associated with known MMP inhibitors is that such compounds often exhibit the same or similar inhibitory effects against each of the MMP enzymes. For example, the peptidomimetic hydroxamate known as batimastat is reported to exhibit IC50 values of about 1 to about 20 nanomolar (nM) against each of MMP-1, MMP-2, MMP-3, MMP-7, and MMP-9. Marimastat, another peptidomimetic hydroxamate was reported to be another broad-spectrum MMP inhibitor with an enzyme inhibitory spectrum very similar to batimastat, except that marimastat exhibited an IC50 value against MMP-3 of 230 nM. Rasmussen et al., Pharmacol. Ther., 75(1): 69-75 (1997).

20 Meta analysis of data from Phase I/II studies using marimastat in patients with advanced, rapidly progressive, treatment-refractory solid tumor cancers (colorectal, pancreatic, ovarian, prostate) indicated a dose-related reduction in the rise of 25 cancer-specific antigens used as surrogate markers for biological activity. Although marimastat exhibited some measure of efficacy via these markers, toxic side effects were noted. The most common drugrelated toxicity of marimastat in those clinical trials was musculoskeletal pain and stiffness, often 30 commencing in the small joints in the hands, spreading to the arms and shoulder. A short dosing holiday of 1-3 weeks followed by dosage reduction

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permits treatment to continue. Rasmussen et al., Pharmacol. Ther., 75(1): 69-75 (1997). It is thought that the lack of specificity of inhibitory effect among the MMPs may be the cause of that effect.

5 International application WO 98/38163, published on September 3, 1998 disclose a large group of hydroxamate inhibitors of MMPs and TACE. The compounds of WO 98/38163 contain one or two substituents adjacent to the hydroxamate

10 functionality and a substituent that can be an aromatic sulfonyl group adjacent to those one or two substituents.

International application WO 98/37877, published on September 3, 1998 discloses compounds that contain a 5- to 7-membered heterocyclic ring adjacent to the hydroxamate functionality and can contain an aromatic sulfonyl group adjacent to the heterocyclic ring.

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Although many of the known MMP inhibitors 20 such as batimastat, marimastat and the hydroxamates of WO 98/37877 and WO 98/38163 exhibit a broad spectrum of activity against MMPs, those compounds are not particularly selective in their inhibitory activity. This lack of selectivity may be the cause of the musculoskeletal pain and stiffness observed 25 with their use. In addition, it can be therapeutically advantageous to utilize a medicament that is selective in its activity as compared to a generally active material so that treatment can be more closely tailored to the pathological condition 30 presented by the host mammal. The disclosure that follows describes a process for treating a host mammal having a condition associated with

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pathological matrix metalloprotease activity that utilizes a compound that selectively inhibits one or more MMPs, while exhibiting less activity against at least MMP-1.

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Summary of the Invention

The present invention is directed to a treatment process that comprises administering a contemplated aromatic sulfone hydroxamic acid metalloprotease inhibitor in an effective amount to a 10 host mammal having a condition associated with pathological metalloprotease activity. A contemplated molecule, inter alia, exhibits excellent inhibitory activity of one or more matrix metalloprotease (MMP) enzymes, such as MMP-2, MMP-9 15 and MMP-13, while exhibiting substantially less inhibition at least of MMP-1. By "substantially less" it is meant that a contemplated compound exhibits an IC50 value ratio against one or more of MMP-2, MMP-9 or MMP-13 as compared to its IC50 value 20 against MMP-1, e.g., IC50 MMP-2:IC50 MMP-1, that is less than about 1:10, preferably less than about 1:100, and most preferably less than about 1:1000 in the in vitro inhibition assay utilized hereinafter. The invention also contemplates particular compounds 25 that selectively inhibit the activity of one or more of MMP-2, MMP-9 and MMP-13, while exhibiting substantially less inhibition at least of MMP-1, as well as a composition containing such a MMP inhibitor as active ingredient. Similarly contemplated are 30 particular compounds such as those of Examples 16, 498, 667, 672 and 684 that selectively inhibit the activity of one or more of MMP-2, MMP-9 and MMP-13, while exhibiting substantially less inhibition at least of MMP-7, as well as a composition containing 35 such a MMP inhibitor as active ingredient.

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invention further contemplates intermediates in the preparation of a contemplated aromatic sulfone hydroxamic acid molecule and a process for preparing an aromatic sulfone hydroxamic acid molecule.

Briefly, one embodiment of the present invention is directed to a treatment process that comprises administering a contemplated aromatic sulfone hydroxamic acid metalloprotease inhibitor that selectively inhibits matrix metalloprotease 10 activity as above in an effective amount to a host mammal having a condition associated with pathological metalloprotease activity. The administered enzyme inhibitor corresponds in structure to formula I, below, or a pharmaceutically acceptable salt thereof:

HONH—
$$C$$
 R^1
 R^2
 R^3

wherein

 ${\tt R}^1$ and ${\tt R}^2$ are both hydrido or ${\tt R}^1$ and ${\tt R}^2$ together with the atoms to which they are bonded form 20 a 5- to 8-membered ring containing one, two or three heteroatoms in the ring that are oxygen, sulfur or nitrogen.

R³ in formula I is an optionally substituted aryl or optionally substituted heteroaryl 25 radical. When R³ is a substituted aryl or heteroaryl radical, a contemplated substituent is selected from the group consisting of an aryl, heteroaryl, aralkyl, heteroaralkyl, aryloxy, arylthio, aralkoxy, 30 heteroaralkoxy, aralkoxyalkyl, aryloxyalkyl,

aralkanoylalkyl, arylcarbonylalkyl, aralkylaryl, aryloxyalkylaryl, aralkoxyaryl, arylazoaryl, arylhydrazinoaryl, alkylthioaryl, arylthioalkyl, alkylthioaralkyl, aralkylthioalkyl, an aralkylthioaryl radical, the sulfoxide or sulfone of any of the thio substituents, and a fused ring structure comprising two or more 5- or 6-membered rings selected from the group consisting of aryl, heteroaryl, carbocyclic and heterocyclic.

heteroaryl radical of which the R³ radical is comprised itself can be substituted with one or more substituents; i.e., the substituting substituent is optionally substituted. When that aryl or heteroaryl radical is substituted, and the substituting moiety (group, substitutent, or radical) is itself substituted, the last-named substitutent is independently selected from the group consisting of a cyano, perfluoroalkyl, trifluoromethylalkyl, trifluoromethylalkyl, trifluoromethylalkyl,

- 20 trifluoromethylthio, haloalkyl, trifluoromethylalkyl, aralkoxycarbonyl, aryloxycarbonyl, hydroxy, halo, alkyl, alkoxy, nitro, thiol, hydroxycarbonyl, aryloxy, arylthio, aralkyl, aryl, arylcarbonylamino, heteroaryloxy, heteroarylthio, heteroaralkyl,
- 25 cycloalkyl, heterocyclooxy, heterocyclothio, heterocycloamino, cycloalkyloxy, cycloalkylthio, heteroaralkoxy, heteroaralkylthio, aralkoxy, aralkylthio, aralkylamino, heterocyclo, heteroaryl, arylazo, hydroxycarbonylalkoxy, alkoxycarbonylalkoxy,
- alkanoyl, arylcarbonyl, aralkanoyl, alkanoyloxy, aralkanoyloxy, hydroxyalkyl, hydroxyalkoxy, alkylthio, alkoxyalkylthio, alkoxycarbonyl,

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aryloxyalkoxyaryl, arylthioalkylthioaryl, aryloxyalkylthioaryl, arylthioalkoxyaryl, hydroxycarbonylalkoxy, hydroxycarbonylalkylthio, alkoxycarbonylalkoxy, alkoxycarbonylalkylthio, amino, 5 wherein the amino nitrogen is (i) unsubstituted, or (ii) substituted with one or two substituents that are independently selected from the group consisting of an alkyl, aryl, heteroaryl, aralkyl, cycloalkyl, aralkoxycarbonyl, alkoxycarbonyl, arylcarbonyl, aralkanoyl, 10 heteroarylcarbonyl, heteroaralkanoyl and an alkanoyl group, or (iii) wherein the amino nitrogen and two substituents attached thereto form a 5- to 8-membered heterocyclo or 15 heteroaryl ring containing zero to two additional heteroatoms that are nitrogen, oxygen or sulfur and which ring itself is (a) unsubstituted or (b) substituted with one or two groups independently selected from the group 20 consisting of an aryl, alkyl, heteroaryl, aralkyl, heteroaralkyl, hydroxy, alkoxy, alkanoyl, cycloalkyl, heterocycloalkyl, alkoxycarbonyl, hydroxyalkyl, trifluoromethyl, benzofused heterocycloalkyl, hydroxyalkoxyalkyl, 25 aralkoxycarbonyl, hydroxycarbonyl, aryloxycarbonyl, benzofused heterocycloalkoxy, benzofused cycloalkylcarbonyl, heterocycloalkylcarbonyl, and a cycloalkylcarbonyl group, carbonylamino 30 wherein the carbonylamino nitrogen is (i) unsubstituted, or (ii) is the reacted amine of an amino acid, or (iii) substituted with one or two radicals selected from the group consisting of an alkyl, hydroxyalkyl, hydroxyheteroaralkyl,

cycloalkyl, aralkyl, trifluoromethylalkyl,

heterocycloalkyl, benzofused heterocycloalkyl,

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benzofused heterocycloalkyl, benzofused cycloalkyl, and an N,N-dialkylsubstituted alkylamino-alkyl group, or (iv) the carboxamido nitrogen and two substituents bonded thereto together form a 5- to 8-membered heterocyclo, heteroaryl or benzofused heterocycloalkyl ring that is itself unsubstituted or substituted with one or two radicals independently selected from the group consisting of an alkyl, alkoxycarbonyl, nitro, heterocycloalkyl,

hydroxy, hydroxycarbonyl, aryl, aralkyl, heteroaralkyl and an amino group,

wherein the amino nitrogen is (i) unsubstituted, or (ii) substituted with one or two substituents that are independently selected from the group consisting of alkyl, aryl, and heteroaryl, or (iii) wherein the amino nitrogen and two substituents attached thereto form a 5- to 8-membered heterocyclo or heteroaryl ring, and an aminoalkyl group

wherein the aminoalkyl nitrogen is (i) unsubstituted, or (ii) substituted with one or two substituents independently selected from the group consisting of an alkyl, aryl, aralkyl, cycloalkyl, aralkoxycarbonyl, alkoxycarbonyl, and an alkanoyl group, or (iii) wherein the aminoalkyl nitrogen and two substituents attached thereto form a 5- to 8membered heterocyclo or heteroaryl ring.

Preferably, the R³ substituent is Ph-Q-A-R-E-Y wherein Ph is phenyl substituted at the 4position relative to the depicted SO2 group, and -Q-A-R-E-Y is a substituent in which Q is a 5- to 7membered heterocyclic ring containing one or two nitrogen atoms, one of which is bonded the depicted

phenyl group, and whose remaining members are defined hereinafter for the substituent G-A-R-E-Y.

A compound of formula I is a compound of more general formula A, wherein \mathbb{R}^3 , \mathbb{R}^1 and \mathbb{R}^2 are as defined before and \mathbb{R}^{20} is defined below.

$$\begin{array}{c|c}
O \\
R^{20} - C \\
\hline
R^1 & R^2
\end{array}$$

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The substituent R^{20} is (a) $-0-R^{21}$, where R^{21} is selected from the group consisting of a hydrido, C_1 - C_6 -alkyl, aryl, ar- C_1 - C_6 -alkyl group and a pharmaceutically acceptable cation, (b) $-NH-O-R^{22}$ wherein R^{22} is a selectively removable protecting group such as a 2-tetrahydropyranyl, benzyl, pmethoxybenzyl (MOZ), carbonyl-C₁-C₆-alkoxy, 15 trisubstituted silyl group or o-nitrophenyl group, peptide synthesis resin and the like, wherein the trisubstituted silyl group is substituted with C1-C6alkyl, aryl, or ar-C₁-C₆-alkyl or a mixture thereof, (c) $-NH-O-R^{14}$, where R^{14} is hydrido, a 20 pharmaceutically acceptable cation or $C(W)R^{25}$ where W is 0 (oxo) or S (thioxo) and \mathbb{R}^{25} is selected from the group consisting of an C_1-C_6 -alkyl, aryl, C_1-C_6 -

alkoxy, heteroaryl-C₁-C₆-alkyl, C₃-C₈-cycloalkyl-C₁-

 C_6 -alkyl, aryloxy, ar- C_1 - C_6 -alkoxy, ar- C_1 - C_6 -alkyl,

heteroaryl and amino C1-C6-alkyl group wherein the

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amino C₁-C₆-alkyl nitrogen is (i) unsubstituted or (ii) substituted with one or two substituents independently selected from the group consisting of an C_1-C_6 -alkyl, aryl, ar- C_1-C_6 -alkyl, C_3-C_8 -5 cycloalkyl-C₁-C₆-alkyl, ar-C₁-C₆-alkoxycarbonyl, C₁- C_6 -alkoxycarbonyl, and C_1 - C_6 -alkanoyl radical, or (iii) wherein the amino C₁-C₆-alkyl nitrogen and two substituents attached thereto form a 5- to 8-membered heterocyclo or heteroaryl ring, or (d) $-NR^{26}R^{27}$, where R^{26} and R^{27} are independently selected from the group consisting of a hydrido, C_1-C_6 -alkyl, amino C_1 - C_6 -alkyl, hydroxy C_1 - C_6 -alkyl, aryl, ar- C_1 - C_6 -alkyl group, or R^{26} and R^{27} together with the depicted nitrogen atom form a 5- to 8-membered ring containing zero or one additional heteroatom that is oxygen, 15 nitrogen or sulfur. When used in a contemplated process or method, R^{20} is $-NH-O-R^{22}$, as defined above.

In preferred practice, R¹ and R² together
with the atoms to which they are bonded form a
6-membered ring.

An R^3 radical preferably has a length that is greater than that of a pentyl group [a -(CH₂)₄CH₃ chain], more preferably greater than about that of a hexyl group [a -(CH₂)₅CH₃ chain], and most preferably greater than an octyl group [a -(CH₂)₇CH₃ chain]. An R^3 radical preferably has a length that is less than that of an icosyl group [a -(CH₂)₁₉CH₃ chain], and more preferably a length that is less than that of a

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stearyl group [a -(CH₂)₁₇CH₃ chain). A preferred R³ group contains two or more 5- or 6-membered rings. A contemplated R³ group, when rotated about an axis drawn through the SO₂-bonded 1-position and the substituent-bonded 4-position of a 6-membered ring or the SO₂-bonded 1-position and substituent-bonded 3- or 4-position of a 5-membered ring, defines a three-dimensional volume whose widest dimension has the width in a direction transverse to that axis to rotation of about one furanyl ring to about two phenyl rings.

It is also preferred that a R³ radical be a single-ringed aryl or heteroaryl group that is 5- or 6-membered, and is itself substituted at its own 415 position when a 6-membered ring or at its own 3- or 4-position when a 5-membered ring with an optionally substituted substituent selected from the group consisting of one other single-ringed aryl or heteroaryl group, a C₃-C₁₄ alkyl group, a N-piperidyl group, a N-piperazyl group, a phenoxy group, a thiophenoxy group, a 4-thiopyridyl group, a phenylazo group and a benzamido group. The substituent of the 5- or 6-membered aryl or heteroaryl group can itself be substituted as discussed before.

A preferred compound for use in a contemplated process has a structure that corresponds to formula II, below, or a pharmaceutically acceptable salt thereof:

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$$(CH_2)_n - Z$$
 $(CH_2)_n - Z$
 $(CH_2)_p$
 $(CH_2)_p$

wherein

R¹⁴ is hydrido, a pharmaceutically

- acceptable cation or C(W)R¹⁵ where W is O or S and R¹⁵ is selected from the group consisting of an C₁-C₆-alkyl, aryl, C₁-C₆-alkoxy, heteroaryl-C₁-C₆-alkyl, C₃-C₈-cycloalkyl-C₁-C₆-alkyl, aryloxy, ar-C₁-C₆-alkoxy, ar-C₁-C₆-alkyl, heteroaryl and amino C₁-C₆-alkyl group wherein the aminoalkyl nitrogen is (i)
- alkyl group wherein the aminoalkyl nitrogen is (i) unsubstituted or (ii) substituted with one or two substituents independently selected from the group consisting of an C₁-C₆-alkyl, aryl, ar-C₁-C₆-alkyl, C₃-C₈-cycloalkyl-C₁-C₆-alkyl, ar-C₁-C₆-
- alkoxycarbonyl, C_1 - C_6 -alkoxycarbonyl, and C_1 - C_6 -alkanoyl radical, or (iii) wherein the amino C_1 - C_6 -alkyl nitrogen and two substituents attached thereto form a 5- to 8-membered heterocyclo or heteroaryl ring;

20 m is zero, 1 or 2;

n is zero, 1 or 2;

p is zero, 1 or 2;

the sum of m + n + p = 1, 2, 3 or 4;

- (a) one of X, Y and Z is selected from the
- 25 group consisting of C(0), NR^6 , O, S, S(0), $S(0)_2$ and

 $NS(0)_2R^7$, and the remaining two of X, Y and Z are CR^8R^9 , and $CR^{10}R^{11}$, or

(b) X and Z or Z and Y together constitute a moiety that is selected from the group consisting of $NR^6C(0)$, $NR^6S(0)$, $NR^6S(0)_2$, NR^6S , NR^6O , SS, NR^6NR^6 and OC(0), with the remaining one of X, Y and Z being CR^8R^9 , or

(c) n is zero and X, Y and Z together
constitute a moiety selected from the group
10 consisting of

wherein wavy lines are bonds to the atoms of the depicted ring;

 R^6 and R^6 are independently selected from the 5 group consisting of hydrido, formyl, sulfonic- C_1 - C_6 alkyl, C₁-C₆-alkoxycarbonyl-C₁-C₆-alkyl, $hydroxycarbonyl-C_1-C_6-alkyl, C_1-C_6-alkylcarbonyl-C_1-alkylca$ C_6 -alkyl, R^8R^9 -aminocarbonyl- C_1 - C_6 -alkyl, C_1 - C_6 alkoxycarbonyl-C₁-C₆-alkylcarbonyl, hydroxycarbonyl-10 C_1-C_6 -alkylcarbonyl, C_1-C_6 -alkylcarbonyl- C_1-C_6 alkylcarbonyl, C₁-C₆-alkoxycarbonylcarbonyl, hydroxycarbonylcarbonyl, C₁-C₆-alkylcarbonylcarbonyl, R^8R^9 -aminocarbonylcarbonyl, C_1 - C_6 -alkanoyl, aryl- C_1 - C_6 -alkyl, aroyl, bis(C_1 - C_6 -alkoxy- C_1 - C_6 -alkyl)- C_1 - C_6 -15 alkyl, C_1-C_6 -alkyl, C_1-C_6 -haloalkyl, C_1-C_6 perfluoroalkyl, C₁-C₆-trifluoromethylalkyl, C₁-C₆perfluoroalkoxy- C_1 - C_6 -alkyl, C_1 - C_6 -alkoxy- C_1 - C_6 alkyl, C3-C6-cycloalkyl, heteroarycarbonyl,

heterocyclocarbonyl, C₃-C₈-heterocycloalkyl, C₃-C₈-heterocycloalkylcarbonyl, aryl, C₅-C₆-heterocyclo, C₅-C₆-heteroaryl, C₃-C₈-cycloalkyl-C₁-C₆-alkyl, aryloxy-C₁-C₆-alkyl, heteroaryloxy-C₁-C₆-alkyl, heteroarylcy-C₁-C₆-alkyl, heteroarylthio-C₁-C₆-alkyl, arylsulfonyl, C₁-C₆-alkylsulfonyl, C₅-C₆-heteroarylsulfonyl, carboxy-C₁-C₆-alkyl, C₁-C₄-alkoxycarbonyl-C₁-C₆-alkyl, aminocarbonyl, C₁-C₆-alkyl(R⁸N)iminocarbonyl, aryl(R⁸N)iminocarbonyl, C₅-C₆-alkyl, C₁-C₆-alkylthio-C₁-C₆-alkyl, arylthio-C₃-C₆-alkyl, C₁-C₆-alkylthio-C₃-C₆-alkenyl, C₅-C₆-heteroaryl-C₁-C₆-alkyl, halo-C₁-C₆-alkanoyl, hydroxy-

15 $C_3-C_6-alkynyl$, $C_1-C_4-alkoxy-C_1-C_4-alkyl$, $C_1-C_5-alkoxycarbonyl$, aryloxycarbonyl, $NR^8R^9-(R^8)$ iminomethyl, $NR^8R^9-C_1-C_5-alkylcarbonyl$, hydroxy- $C_1-C_5-alkyl$, $R^8R^9-aminocarbonyl$, $R^8R^9-aminocarbonyl-C_1-C_6-alkylcarbonyl$, hydroxyaminocarbonyl, $R^8R^9-aminocarbonyl$, $R^8R^9-aminocarbonyl$,

 C_1-C_6 -alkanoyl, thiol- C_1-C_6 -alkanoyl, C_3-C_6 -alkenyl,

amino- C_1 - C_6 -alkylsulfonyl and an R^8R^9 -amino- C_1 - C_6 -alkyl group;

 R^7 is selected from the group consisting of a arylalkyl, aryl, heteroaryl, heterocyclo, C_1 - C_6 -alkyl, C_3 - C_6 -alkynyl, C_3 - C_6 -alkenyl, C_1 - C_6 -carboxyalkyl and a C_1 - C_6 -hydroxyalkyl group;

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 R^8 and R^9 and R^{10} and R^{11} are independently selected from the group consisting of a hydrido, hydroxy, C_1-C_6 -alkyl, C_1-C_6 -alkanoyl, aroyl, aryl, ar-C₁-C₆-alkyl, heteroaryl, heteroar-C₁-C₆-alkyl, C₂-5 C_6 -alkynyl, C_2 - C_6 -alkenyl, thiol- C_1 - C_6 -alkyl, C_1 - C_6 alkylthio- C_1 - C_6 -alkyl, cycloalkyl, cycloalkyl- C_1 - C_6 alkyl, heterocycloalkyl-C₁-C₆-alkyl, C₁-C₆-alkoxy-C₁- C_6 -alkyl, aralkoxy- C_1 - C_6 -alkyl, C_1 - C_6 -alkoxy- C_1 - C_6 alkoxy-C₁-C₆-alkyl, hydroxy-C₁-C₆-alkyl, $hydroxycarbonyl-C_1-C_6-alkyl, hydroxycarbonylar-C_1-C_6-alkyl, hydroxycarbonylar-C_1-C_6-alkyl,$ alkyl, aminocarbonyl- C_1 - C_6 -alkyl, aryloxy- C_1 - C_6 alkyl, heteroaryloxy-C₁-C₆-alkyl, arylthio-C₁-C₆alkyl, heteroarylthio-C₁-C₆-alkyl, the sulfoxide or sulfone of any said thio substituents, perfluoro-C1-C₆-alkyl, trifluoromethyl-C₁-C₆-alkyl, halo-C₁-C₆-15 alkyl, alkoxycarbonylamino-C1-C6-alkyl and an amino- C_1 - C_6 -alkyl group wherein the aminoalkyl nitrogen is (i) unsubstituted or (ii) substituted with one or two radicals independently selected from the group 20 consisting of C₁-C₆-alkyl, ar-C₁-C₆-alkyl, cycloalkyl and C_1-C_6 -alkanoyl, or wherein R^8 and R^9 or R^{10} and R^{11} and the carbon to which they are bonded form a carbonyl group, or wherein R^8 and R^9 or R^{10} and R^{11} , or ${\tt R}^{8}$ and ${\tt R}^{10}$ together with the atoms to which they are bonded form a 5- to 8-membered carbocyclic ring, 25 or a 5- to 8-membered heterocyclic or heteroaryl ring containing one or two heteroatoms that are nitrogen,

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oxygen, or sulfur, with the proviso that only one of R^8 and R^9 or R^{10} and R^{11} is hydroxy;

 R^{12} and R^{12} are independently selected from the group consisting of a hydrido, C_1-C_6 -alkyl, 5 aryl, ar- C_1 - C_6 -alkyl, heteroaryl, heteroaralkyl, C_2 - C_6 -alkynyl, C_2 - C_6 -alkenyl, thiol- C_1 - C_6 -alkyl, cycloalkyl, cycloalkyl-C1-C6-alkyl, heterocycloalkyl- $C_1-C_6-alkyl$, $C_1-C_6-alkoxy-C_1-C_6-alkyl$, aryloxy- $C_1-C_6-alkyl$ alkyl, amino- C_1 - C_6 -alkyl, C_1 - C_6 -alkoxy- C_1 - C_6 -alkoxy- $C_1-C_6-alkyl$, hydroxy- $C_1-C_6-alkyl$, hydroxycarbonyl- $C_1-alkyl$ C₆-alkyl, hydroxycarbonylar-C₁-C₆-alkyl, aminocarbonyl-C₁-C₆-alkyl, aryloxy-C₁-C₆-alkyl, heteroaryloxy-C₁-C₆-alkyl, C₁-C₆-alkylthio-C₁-C₆alkyl, arylthio-C₁-C₆-alkyl, heteroarylthio-C₁-C₆alkyl, the sulfoxide or sulfone of any said thio 15 substituents, perfluoro-C₁-C₆-alkyl, trifluoromethyl-C₁-C₆-alkyl, halo-C₁-C₆-alkyl, alkoxycarbonylamino- C_1-C_6 -alkyl and an amino- C_1-C_6 -alkyl group wherein the aminoalkyl nitrogen is (i) unsubstituted or (ii) 20 substituted with one or two radicals independently selected from the group consisting of C1-C6-alkyl, ar-C1-C6-alkyl, cycloalkyl and C1-C6-alkanoyl;

 $\rm R^{13}$ is selected from the group consisting of a hydrido, benzyl, phenyl, $\rm C_1\text{-}C_6\text{-}alkyl,\ C_2\text{-}C_6\text{-}$ alkynyl, $\rm C_2\text{-}C_6\text{-}alkenyl$ and a $\rm C_1\text{-}C_6\text{-}hydroxyalkyl$ group; and

G-A-R-E-Y is a substituent that preferably has a length greater than that of a pentyl group, and more preferably has a length greater than that of a

hexyl group. The substituent G-A-R-E-Y preferably has a length that is less than that of an icosyl group, and is more preferably less than that of a stearyl group. In this substituent:

G is an aryl or heteroaryl group;

A is selected from the group consisting of

- (1) -0-;
- (2) -S-;
- $(3) NR^{17} ;$
- 10 (4) $-\text{CO-N}(\mathbb{R}^{17})$ or $-\text{N}(\mathbb{R}^{17})$ -CO-, wherein \mathbb{R}^{17} is hydrogen, C_1-C_4 -alkyl, or phenyl;
 - (5) -CO-O- or -O-CO-;
 - (6) -0-C0-0-;
 - (7) -HC=CH-;
- 15 (8) —NH-CO-NH-;
 - (9) -C=C-;
 - (10) -NH-CO-O- or -O-CO-NH-;
 - (11) -N=N-;
 - (12) -NH-NH-; and
- 20 (13) $-CS-N(R^{18})-$ or $-N(R^{18})-CS-$, wherein R^{18} is hydrogen C_1-C_4 -alkyl, or phenyl; or
 - (14) A is absent and G is bonded directly to R;
- R is a moiety selected from the group consisting of alkyl, alkoxyalkyl, aryl, heteroaryl, cycloalkyl, heterocycloalkyl, aralkyl, heteroaralkyl, heterocycloalkylalkyl, cycloalkylalkyl, cycloalkoxyalkyl,
- 30 aryloxyalkyl, heteroaryloxyalkyl, arylthioalkyl, heteroarylthioalkyl, cycloalkylthioalkyl, and a

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heterocycloalkylthioalkyl group wherein the aryl or heteroaryl or cycloalkyl or heterocycloalkyl substituent is (i) unsubstituted or (ii) substituted with one or two radicals selected from the group consisting of a halo, alkyl, perfluoroalkyl, perfluoroalkoxy, perfluoroalkylthio, trifluoromethylalkyl, amino, alkoxycarbonylalkyl, alkoxy, C1-C2-alkylene-dioxy, hydroxycarbonylalkyl, hydroxycarbonylalkylamino, nitro, hydroxy, hydroxycarbonyl group, and R is other than alkyl or alkoxyalkyl when A is -O- or -S-;

E is selected from the group consisting of

- (1) $-CO(R^{19})$ or $-(R^{19})CO$, wherein R^{19} is a heterocycloalkyl, or a cycloalkyl group;
 - (2) -CONH- or -HNCO-; and
 - (3) -CO-;
 - (4) $-SO_2-R^{19}$ or $-R^{19}-SO_2$;
- 20 (5) -SO₂-;
 - (6) $-NH-SO_2-$ or $-SO_2-NH-$;
 - (7) -S-;
 - (8) -NH-CO-O- or -O-CO-NH-; or
 - (9) E is absent and R is bonded directly
- 25 to Y; and

the moiety Y is absent or is selected from the group consisting of a hydrido, alkyl, alkoxy, haloalkyl, aryl, aralkyl, cycloalkyl, heteroaryl, hydroxy, aryloxy, aralkoxy, heteroaryloxy, heteroaralkyl, perfluoroalkoxy, perfluoroalkylthio, trifluoromethylalkyl, alkenyl, heterocycloalkyl,

cycloalkyl, trifluoromethyl, alkoxycarbonyl, and a

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aminoalkyl group, wherein the aryl, heteroaryl, aralkyl or heterocycloalkyl group is (i) unsubstituted or (ii) substituted with one or two radicals independently selected from the group 5 consisting of an alkanoyl, halo, nitro, aralkyl, aryl, alkoxy, trifluoroalkyl, trifluoroalkoxy and an amino group wherein the amino nitrogen is (i) unsubstituted or (ii) substituted with one or two groups independently selected from hydrido, alkyl, and an aralkyl group.

A particularly preferred compound for use in a contemplated process corresponds in structure to formula III, below, or a pharmaceutically acceptable salt thereof:

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$$R^{14}O$$
— HN
 $(CH_2)_m$
 $(CH_2)_m$
 $(CH_2)_p$
 $(CH_2$

wherein

m, n, p, X, Z, Y and R^{14} are as defined 20 above for formula II, and the R³ radical that is defined below is a sub-set of the previously discussed G-A-R-E-Y substituents.

Thus, R^3 is a radical that is comprised of a single-ringed aryl or heteroaryl group that is 5-25 or 6-membered, and is itself substituted at its own 4-position when a 6-membered ring and at its own 3-

-25-

or 4-position when a 5-membered ring with a substituent selected from the group consisting of a thiophenoxy, 4-chlorophenoxy, 3-chlorophenoxy, 4methoxyphenoxy, 3-benzodioxol-5-yloxy, 3,4dimethylphenoxy, 4-fluorophenoxy, 4fluorothiophenoxy, phenoxy, 4-trifluoromethoxyphenoxy, 4-trifluoromethylphenoxy, 4-(trifluoromethylthio)-phenoxy, 4-(trifluoromethylthio)-thiophenoxy, 4-chloro-3fluorophenoxy, 4-isopropoxyphenoxy, 4-10 isopropylphenoxy, (2-methyl-1,3-benzothiazol-5yl)oxy, 4-(1H-imidazol-1-yl)phenoxy, 4-chloro-3methylphenoxy, 3-methylphenoxy, 4-ethoxyphenoxy, 3,4difluorophenoxy, 4-chloro-3-methylphenoxy, 4-fluoro-3-chlorophenoxy, 4-(1H-1,2,4-triazol-1-yl)phenoxy, 15 3,5-difluorophenoxy, 3,4-dichlorophenoxy, 4cyclopentylphenoxy, 4-bromo-3-methylphenoxy, 4bromophenoxy, 4-methylthiophenoxy, 4-phenylphenoxy, 4-benzylphenoxy, 6-quinolinyloxy, 4-amino-3methylphenoxy, 3-methoxyphenoxy, 5,6,7,8-tetrahydro-20 2-naphthalenyloxy, 3-hydroxymethylphenoxy, Npiperidyl, N-piperazinyl and a 4-benzyloxyphenoxy group.

A more particularly preferred compound for 25 use in a contemplated process has a structure that corresponds to formula IV, below, or a pharmaceutically acceptable salt thereof:

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wherein R^3 is as defined above for formula I, more preferably as defined for formula II (wherein this R^3 group is the G-A-R-E-Y substituent), and more preferably still as defined for formula III, and Z is selected group the group consisting of O, S, NR^6 , SO, SO₂, and NSO_2R^7 ,

wherein R⁶ is selected from the group consisting of hydrido, C₁-C₅-alkyl, C₁-C₅-alkanoyl, benzyl, benzoyl, C₃-C₅-alkynyl, C₃-C₅-alkenyl, C₁-C₃-alkoxy-C₁-C₄-alkyl, C₃-C₆-cycloalkyl, heteroaryl-C₁-C₆-alkyl, C₁-C₅-hydroxyalkyl, C₁-C₅-carboxyalkyl, C₁-C₅-alkoxy C₁-C₅-alkylcarbonyl, and NR⁸R⁹-C₁-C₅-alkylcarbonyl or NR⁸R⁹-C₁-C₅-alkyl wherein R⁸ and R⁹ are independently hydrido, C₁-C₅-alkyl, C₁-C₅-alkoxycarbonyl or aryl-C₁-C₅-alkoxycarbonyl, or NR⁸R⁹ together form a heterocyclic ring containing 5- to 8-atoms in the ring; and

20 R⁷ is selected from the group consisting of an arylalkyl, aryl, heteroaryl, heterocyclo, C₁-C₆-alkyl, C₃-C₆-alkynyl, C₃-C₆-alkenyl, C₁-C₆-carboxyalkyl and a C₁-C₆-hydroxyalkyl group.

A still more preferred group of compounds for use in a contemplated process correspond in structure to formula V, below, or a pharmaceutically acceptable salt thereof:

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wherein

Z is as previously defined in formula IV;

W and Q are independently oxygen (O), NR⁶ or sulfur (S), and R⁶ is as defined in formula IV; and q is zero or one such that when q is zero, the trifluoromethyl group is bonded directly to the depicted phenyl ring.

particularly preferred. One group of these compounds corresponds in structure to formula B (including formulas B, B-A, B-1, B-1A, B-2, B-2A, B-3 and B-3A), formula VIC, and more still particularly to formula VIC-1 and formula VIC-2, and formula VIII, below. In those formulas, ring structure Q is a substituent of the depicted phenyl ring and can itself be substituted. Substituent Q including the depicted nitrogen atom is a heterocylic ring that contains 5- or 7-members, preferably 6-members, and can contain zero or one additional nitrogen atom. The substituents of Q such as A-R-E-Y, R-E-Y and E-Y are

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as defined before, and such a substituent is bonded at the 4-position relative to that depicted nitrogen atom when Q is a 6- or 7-membered ring and at the 3- or 4-position relative to that depicted nitrogen when Q is a 5-membered ring. The remaining members of such a Q-beraing substituent (e.g., A-R-E-Y) are defined herein for the substituent G-A-R-E-Y. In addition, R²⁰, X, Y, Z, m, n, and p of the ring system and g are as before described, with Z preferably being O or NR⁶.

$$R^{20} \xrightarrow{(CH_2)_0(CH_2)_p} Q \xrightarrow{A} R \xrightarrow{E} Y$$

$$R^{20} \xrightarrow{(CH_2)_0(CH_2)_p} B \xrightarrow{Z} Q \xrightarrow{A} R \xrightarrow{E} Y$$

$$HONH \xrightarrow{S(O)_2} B \xrightarrow{B-A}$$

$$\begin{array}{c} (CH_2)_{\widehat{\mathbf{n}}} & Z \\ X \\ (CH_2)_{\widehat{\mathbf{n}}} & (CH_2)_{\widehat{\mathbf{p}}} \\ S(O)_{\widehat{\mathbf{g}}} & VIC \end{array}$$

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$$\begin{array}{c|c} (CH_2)_{\overline{n}} & Z \\ X & & \\ X$$

The compounds of formulas IX, IX-1, IX-2, X, XI, XI-1, XI-2 and XII, below, are more 5 particularly preferred among the compounds of formula VIC, formula VIC-1, formula VIC-2, and formula VIII. In those latter formulas, Z is as before described, with 2 preferably being O or NR6, and substituent Q is a 6-membered ring, as is shown. The A moiety of 10 the Q ring substituent -A-R-E-Y (e.g. of formula B or B-1) is preferably absent in some embodiments, as in the compounds of formulas XI through XII, whereas both moieties A and R of that substituent group are absent in compounds of formulas IX through X. 15 moieties A, R, E and Y of the substituent group -A-R-E-Y are as defined for the substituent group -G-A-R-E-Y.

X

When used in a contemplated in a beforedescribed process, a compound of formulas A, B, and
I-VI, VI VIC, VIC-1, VIC-2, VIII, IX, IX-1, IX-2, X,
XI, XI-1, XI-2 and XII, a R²⁰ group is preferably
-NH-O-R²² as defined above, and such a compound can
also be present as a pharmaceutically acceptable
salt. In addition, when so used, g is 2 in formulas
B, VIC, VIC-1, VIC-2 and VII. The compounds of
formulas A, B, and I-VI, VI VIC, VIC-1, VIC-2, VIII,
IX, IX-1, IX-2, X, XI, XI-1, XI-2 and XII and their
pharmaceutically acceptable salts are contemplated
compounds of this invention.

The present invention also contemplates a precursor or intermediate compound that is useful in preparing a compound of formulas I-X. Such an intermediate compound corresponds in structure to formula VI, below:

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$$(CH_2)_n - Z$$
 $(CH_2)_m (CH_2)_p$
 $S(O)_g$
 R^{24}
 O
 VI

wherein m, n, p, X, Z and Y are as defined

5 above for formula II, g is zero, 1 or 2 and R²⁴ is R³
as defined in formulas I, III or IV, is the
substituent G-A-R-E-Y of formula II (formula VIA) or
is R³, an aryl or heteroaryl group that is
substituted with a coupling substituent reactive for
10 coupling with another moiety (formula VIB), such as a
nucleophilically displaceable leaving group, D.

$$(CH_2)_n - Z$$
 $(CH_2)_n - Z$
 $(CH_$

Exemplary nucleophilically displaceable leaving groups, D, include a halo (fluoro, chloro, bromo, or iodo) nitro, azido, phenylsulfoxido, aryloxy, C₁-C₆-alkoxy, a C₁-C₆-alkylsulfonate or arylsulfonate group and a trisubstituted ammonium group in which the three substituents are independently aryl, ar- C₁-C₆-alkyl or C₁-C₆-alkyl.

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 R^{20} is (a) $-0-R^{21}$, where R^{21} is selected from the group consisting of a hydrido, C_1-C_6 -alkyl, aryl, ar-C₁-C₆-alkyl group and a pharmaceutically acceptable cation, (b) -NH-O-R²² wherein R²² is a selectively removable protecting group such as a 2-5 tetrahydropyranyl, benzyl, p-methoxybenzyl (MOZ), carbonyl-C₁-C₆-alkoxy, trisubstituted silyl group or o-nitrophenyl group, peptide synthesis resin and the like, wherein the trisubstituted silyl group is substituted with C₁-C₆-alkyl, aryl, or ar-C₁-C₆-alkyl 10 or a mixture thereof, (c) $-NH-O-R^{14}$, where R^{14} is hydrido, a pharmaceutically acceptable cation or $C(W)R^{25}$ where W is O (oxo) or S (thioxo) and R^{25} is selected from the group consisting of an C1-C6-alkyl, aryl, C_1 - C_6 -alkoxy, heteroaryl- C_1 - C_6 -alkyl, C_3 - C_8 cycloalkyl-C₁-C₆-alkyl, aryloxy, ar-C₁-C₆-alkoxy, ar-C₁-C₆-alkyl, heteroaryl and amino C₁-C₆-alkyl group wherein the amino C_1-C_6 -alkyl nitrogen is (i) unsubstituted or (ii) substituted with one or two 20 substituents independently selected from the group consisting of an C_1 - C_6 -alkyl, aryl, ar- C_1 - C_6 -alkyl, C_3-C_8 -cycloalkyl- C_1-C_6 -alkyl, ar- C_1-C_6 alkoxycarbonyl, C₁-C₆-alkoxycarbonyl, and C₁-C₆alkanoyl radical, or (iii) wherein the amino C_1-C_6 alkyl nitrogen and two substituents attached thereto 25 form a 5- to 8-membered heterocyclo or heteroaryl ring, or (d) $-NR^{26}R^{27}$, where R^{26} and R^{27} are independently selected from the group consisting of a hydrido, C_1-C_6 -alkyl, amino C_1-C_6 -alkyl, hydroxy C_1 - 5

 C_6 -alkyl, aryl, ar- C_1 - C_6 -alkyl group, or R^{26} and R^{27} together with the depicted nitrogen atom form a 5- to 8-membered ring containing zero or one additional heteroatom that is oxygen, nitrogen or sulfur.

A particularly preferred precursor intermediate to an intermediate compound of formula VI is an intermediate compound of formula VII

$$R^{20} \xrightarrow{(CH_2)_m} \xrightarrow{(CH_2)_p} S(O)_g$$
VII

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wherein m, n, p, g, X, Z, Y, D and \mathbb{R}^{20} are as defined above for formula VI.

Among the several benefits and advantages of the present invention are the provision of compounds and compositions effective as inhibitors of matrix metalloproteinase activity, the provision of such compounds and compositions that are effective for the inhibition of metalloproteinases implicated in diseases and disorders involving uncontrolled breakdown of connective tissue.

More particularly, a benefit of this invention is the provision of a compound and composition effective for selectively inhibiting certain metalloproteinases, such as one or more of MMP-2, MMP-9 and MMP-13, associated with pathological conditions such as, for example, rheumatoid arthritis, osteoarthritis, septic arthritis, corneal,

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epidermal or gastric ulceration, tumor metastasis, invasion or angiogenesis, periodontal disease, proteinuria, Alzheimer's Disease, coronary thrombosis and bone disease.

An advantage of the invention is the provision of compounds, compositions and methods effective for treating such pathological conditions by selective inhibition of a metalloproteinase such as MMP-2, MMP-9 or MMP-13 associated with such 10 conditions with minimal side effects resulting from inhibition of other metalloproteinases, such as MMP-1, whose activity is necessary or desirable for normal body function.

Yet another advantage of the invention is 15 the provision of a process for preparing such compounds.

Another benefit is the provision of a method for treating a pathological condition associated with abnormal matrix metalloproteinase activity.

A further advantage of the invention is the provision of a process for preparing such compositions.

Still further benefits and advantages of 25 the invention will be apparent to the skilled worker from the disclosure that follows.

Detailed Description of the Invention

In accordance with the present invention, it has 30 been discovered that certain aromatic sulfone hydroxamic acids (hydroxamates) are effective for inhibition of matrix metalloproteinases ("MMPs") believed to be associated with uncontrolled or otherwise pathological breakdown of connective In particular, it has been found that these 35 tissue. certain aromatic sulfone hydroxamates are effective

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for inhibition of one or more enzymes such as MMP-2, MMP-9 and MMP-13, which can be particularly destructive to tissue if present or generated in abnormal quantities or concentrations, and thus exhibit a pathological activity. Included in that pathological activity is the assistance of tumors and tumor cells in the process of penetrating basement membrane, and developing a new or improved blood supply; i.e., angiogenesis.

Moreover, it has been discovered that these 10 aromatic sulfone hydroxamates are selective in the inhibition of one or more of MMP-2, MMP-9 and MMP-13 without excessive inhibition of other collagenases essential to normal bodily function such as tissue turnover and repair. More particularly, it has been 15 found that a contemplated aromatic sulfone hydroxamate of the invention, or a pharmaceutically acceptable salt thereof, is particularly active in inhibiting of one or more of MMP-2, MMP-9 and MMP-13 20 in an in vitro assay that is predictive of in vivo activity. In addition, while being selective for one or more of MMP-2, MMP-9 and MMP-13, a contemplated aromatic sulfone hydroxamate, or its salt, has a limited or minimal in vitro inhibitory effect on MMP-25 1.

There is thus a substantial difference in the activity of a compound used in a contemplated process toward one or more of MMP-2, MMP-9 and MMP-13 and MMP-1. This substantial difference is assayed using the *in vitro* inhibition assay discussed in the examples. A substantial difference in activity corresponds to a compound exhibiting an IC₅₀ value against one or more of MMP-2, MMP-9 and MMP-13 that

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is about 0.1 times that of the compound against MMP-1, and more preferably 0.01 times that against MMP-1 and most preferably 0.001 times that against MMP-1, or more. Indeed, some compounds exhibit selectivity differences measured by IC50 values that exceed the bounds of the assay at the number 100,000-fold. These selectivities are illustrated in the Inhibition Tables hereinafter.

Put differently, a contemplated compound

10 can inhibit the activity of MMP-2 compared to MMP-9

or MMP-13 and MMP-1. Similarly, a contemplated

compound can inhibit the activity of MMP-13 and MMP
2, while exhibiting less inhibition against MMP-1 and

MMP-9. In addition, a contemplated compound can

inhibit the activity of a MMP enzyme, while having

less of an effect on tumor necrosis factor release.

The advantages of the selectivity of a contemplated compound can be appreciated, without wishing to be bound by theory, by considering the therapeutic uses the compounds. For example, inhibition of MMP-1 is suggested to be undesirable due to its role as a housekeeping enzyme, helping to maintain normal connective tissue turnover and repair. Inhibition of MMP-1 can lead to toxicities or side effects such as such as joint or connective tissue deterioration and pain. On the other hand, MMP-13 has been suggested to be intimately involved in the destruction of joint components in diseases such as osteoarthritis. Thus, potent and selective inhibition of MMP-13 compared with inhibition MMP-1 is highly desirable because a MMP-13 inhibitor can have a positive effect on disease progression in a

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patient in addition to having an anti-inflammatory effect.

Inhibition of MMP-2 and MMP-9 can be desirable for inhibition of tumor growth, metastasis, invasion and/or angiogenesis. A profile of selective inhibition of MMP-2 and MMP-9 relative to MMP-1 can provide a therapeutic advantage.

Yet another advantage of a contemplated compound is the selectivity with respect to tumor necrosis factor release and/or tumor necrosis factor receptor release that provides the physician with another factor to help select the best drug for a particular patient. While not wishing to be bound by theory, it is believed that there are several factors to this 15 type of selectivity to be considered.

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The first is that presence of tumor necrosis factor can be desirable for the control of cancer in the organism, so long as TNF is not present in a toxic excess. Thus, uncontrolled inhibition of 20 release of TNF can be counterproductive and actually can be considered an adverse side effect even in cancer patients. In addition, selectivity with respect to inhibition of the release of the tumor necrosis factor receptor can also be desirable. 25 presence of that receptor can be desirable for maintaining a controlled tumor necrosis level in the mammal by binding excess TNF.

A contemplated selective MMP inhibitor compound useful in a contemplated process can be administered to by various routes and provide adequate therapeutic blood levels of enzymatically active inhibitor. A compound can be administered, for example, by the oral (IG, PO) or intravenous (IV) routes. Oral

administration is advantageous if the patient is ambulatory, not hospitalized, physically able and sufficiently responsible to take drug at the required intervals. This is true even if the person is being treated with more than one drug for one or more diseases. On the other hand, IV drug administration is an advantage in a hospital setting wherein the dose and thus the blood levels can well controlled. A contemplated inhibitor can also be formulated for IM administration if desired. This route of 10 administration can be desirable for the administration of prodrugs or regular drug delivery to patients that are either physically weak or have a poor compliance record or require constant drug blood 15 levels.

Thus, in one embodiment, the present invention is directed to a treatment process that comprises administering a contemplated aromatic sulfone hydroxamic acid metalloprotease inhibitor, or a pharmaceutically acceptable salt thereof, in an 20 effective amount to a host mammal having a condition associated with pathological matrix metalloprotease activity. A contemplated aromatic sulfone hydroxamate inhibitor compound useful in such a 25 process inhibits the activity of one or more of MMP-2, MMP-9 and MMP-13, and exhibits substantially less inhibitory activity against at least MMP-1 in the in vitro assay noted above and discussed in detail hereinbelow. An aromatic sulfone hydroxamate inhibitor compound for use in a contemplated process 30 corresponds in structure to formula I, below:

-40-

HONH—
$$C$$
 R^1
 R^2
 R^3

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wherein

In one embodiment, R¹ and R² are both hydrido. In another embodiment, R¹ and R² together with the atoms to which they are bonded form a 5- to 8-membered ring containing one, two or three heteroatoms in the ring that are oxygen, sulfur or nitrogen.

10 It is preferred that R¹ and R² together with the atoms to which they are bonded form a five-to eight-membered ring that contains one or two heteroatoms in the ring, although R¹ and R² together with the atoms to which they are bonded form a 5- to 8-membered ring containing one, two or three heteroatoms. The heterocyclic ring can itself also be substituted with up to six C₁-C₆-alkyl groups or groups that comprise a another 5- to 8-membered carbocyclic or heterocyclic ring, an amino group, or contain one or two oxo (carbonyl) groups.

R³ in formula I is an optionally substituted aryl or optionally substituted heteroaryl radical. That R3 radical is selected from the group consisting of an aryl, heteroaryl, aralkyl, heteroaralkyl, aralkoxy, heteroaralkoxy, aralkoxyalkyl, aryloxyalkyl, aralkanoylalkyl, arylcarbonylalkyl, aralkylaryl, aryloxyalkylaryl, aralkoxyaryl, arylazoaryl, arylhydrazinoaryl,

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alkylthioaryl, arylthioalkyl, alkylthioaralkyl, aralkylthioalkyl, an aralkylthioaryl radical, the sulfoxide or sulfone of any of the thio substituents, and a fused ring structure comprising two or more 5-or 6-membered rings selected from the group consisting of aryl, heteroaryl, carbocyclic and heterocyclic.

The substituent of which R³ is comprised itself is unsubstituted or substituted with one or more substituents independently selected from the 10 group consisting of a cyano, perfluoroalkyl, trifluoromethylalkyl, hydroxy, halo, alkyl, alkoxy, nitro, thiol, hydroxycarbonyl, aryloxy, arylthio, aralkyl, aryl, heteroaryloxy, heteroarylthio, heteroaralkyl, cycloalkyl, heterocyclooxy, 15 heterocyclothio, heterocycloamino, cycloalkyloxy, cycloalkylthio, heteroaralkoxy, heteroaralkylthio, aralkoxy, aralkylthio, aralkylamino, heterocyclo, heteroaryl, arylazo, hydroxycarbonylalkoxy, 20 alkoxycarbonylalkoxy, alkanoyl, arylcarbonyl, aralkanoyl, alkanoyloxy, aralkanoyloxy, hydroxyalkyl, hydroxyalkoxy, alkylthio, alkoxyalkylthio, alkoxycarbonyl, aryloxyalkoxyaryl, arylthioalkylthioaryl, aryloxyalkylthioaryl, arylthioalkoxyaryl, hydroxycarbonylalkoxy, 25 hydroxycarbonylalkylthio, alkoxycarbonylalkoxy, alkoxycarbonylalkylthio, amino,

wherein the amino nitrogen is (i) unsubstituted, or (ii) substituted with one or two substituents that are independently selected from the group consisting of an alkyl, aryl, heteroaryl, aralkyl, cycloalkyl, aralkoxycarbonyl, alkoxycarbonyl, arylcarbonyl, aralkanoyl,

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heteroarylcarbonyl, heteroaralkanoyl and an alkanoyl group, or (iii) wherein the amino nitrogen and two substituents attached thereto form a 5- to 8-membered heterocyclo or heteroaryl ring containing zero to two additional heteroatoms that are nitrogen, oxygen or sulfur and which ring itself is (a) unsubstituted or (b) substituted with one or two groups independently selected from the group consisting of an aryl, alkyl, heteroaryl, aralkyl, heteroaralkyl, hydroxy, alkoxy, alkanoyl, cycloalkyl, heterocycloalkyl, alkoxycarbonyl, hydroxyalkyl, trifluoromethyl, benzofused heterocycloalkyl, hydroxyalkoxyalkyl, aralkoxycarbonyl, hydroxycarbonyl, aryloxycarbonyl, benzofused heterocycloalkoxy, benzofused cycloalkylcarbonyl, heterocycloalkylcarbonyl, and a cycloalkylcarbonyl group, carbonylamino wherein the carboxamido nitrogen is (i) unsubstituted, or (ii) is the reacted amine of an amino acid, or (iii) substituted with one or two radicals selected from the group consisting of an alkyl, hydroxyalkyl, hydroxyheteroaralkyl,

unsubstituted, or (ii) is the reacted amine of an amino acid, or (iii) substituted with one or two radicals selected from the group consisting of an alkyl, hydroxyalkyl, hydroxyheteroaralkyl, cycloalkyl, aralkyl, trifluoromethylalkyl, heterocycloalkyl, benzofused heterocycloalkyl, benzofused cycloalkyl, and an N,N-dialkylsubstituted alkylamino-alkyl group, or (iv) the carboxamido nitrogen and two substituents bonded thereto together form a 5- to 8-membered heterocyclo, heteroaryl or benzofused heterocycloalkyl ring that is itself unsubstituted or substituted with one or two radicals independently selected from the group consisting of an alkyl, alkoxycarbonyl, nitro, heterocycloalkyl,

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hydroxy, hydroxycarbonyl, aryl, aralkyl, heteroaralkyl and an amino group,

wherein the amino nitrogen is

(i) unsubstituted, or (ii) substituted with one or two substituents that are independently selected from the group consisting of alkyl, aryl, and heteroaryl, or (iii) wherein the amino nitrogen and two substituents attached thereto form a 5- to 8-membered heterocyclo or heteroaryl ring,

and an aminoalkyl group

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wherein the aminoalkyl nitrogen is (i) unsubstituted, or (ii) substituted with one or two substituents independently selected from the group consisting of an alkyl, aryl, aralkyl, cycloalkyl, aralkoxycarbonyl, alkoxycarbonyl, and an alkanoyl group, or (iii) wherein the aminoalkyl nitrogen and

group, or (iii) wherein the aminoalkyl nitrogen and two substituents attached thereto form a 5- to 8-membered heterocyclo or heteroaryl ring. A compound of formula I can also be used in the form of a pharmaceutically acceptable salt.

The R³ radical has a length that is greater than that of a pentyl group [a -(CH₂)₄CH₃ chain], is more preferably greater than about the length of a hexyl group [a -(CH₂)₅CH₃ chain], and most preferably is greater than about the length of an octyl group [a -(CH₂)₇CH₃ chain]. A R³ group has a length that is less than that of an icosyl group [eicosyl; a - (CH₂)₁₉CH₃ chain), and more preferably, a length that is less than that of a stearyl group [a -(CH₂)₁₇CH₃ chain). When rotated about an axis drawn through the SO₂-bonded 1-position and the substituent-bonded 4-position of a 6-membered ring or the SO₂-bonded 1-

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position and substituent-bonded 3- or 4-position of a 5-membered ring, a contemplated R³ radical defines a three-dimensional volume whose widest dimension has the width of about one furanyl ring to about two phenyl rings in a direction transverse to that axis to rotation.

A compound of formula I is a compound of more general formula A, wherein ${\sf R}^3$, ${\sf R}^1$ and ${\sf R}^2$ are as defined before and ${\sf R}^{20}$ is defined below.

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$$R^{20}$$
 C R^{1} R^{2} R^{2}

Α

The substituent R^{20} is (a) $-0-R^{21}$, where \mathbf{R}^{21} is selected from the group consisting of a hydrido, C_1-C_6 -alkyl, aryl, ar- C_1-C_6 -alkyl group and a pharmaceutically acceptable cation, (b) $-NH-O-R^{22}$ 15 wherein R²² is a selectively removable protecting group such as a 2-tetrahydropyranyl, benzyl, pmethoxybenzyl (MOZ), carbonyl-C₁-C₆-alkoxy, trisubstituted silyl group or o-nitrophenyl group, peptide synthesis resin and the like, wherein the 20 trisubstituted silyl group is substituted with C1-C6alkyl, aryl, or ar-C1-C6-alkyl or a mixture thereof, (c) $-NH-O-R^{14}$, where R^{14} is hydrido, a pharmaceutically acceptable cation or $C(W)R^{25}$ where W is O (oxo) or S (thioxo) and R²⁵ is selected from the 25 group consisting of an C_1 - C_6 -alkyl, aryl, C_1 - C_6 -

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alkoxy, heteroaryl-C₁-C₆-alkyl, C₃-C₈-cycloalkyl-C₁-C₆-alkyl, aryloxy, ar-C₁-C₆-alkoxy, ar-C₁-C₆-alkyl, heteroaryl and amino C1-C6-alkyl group wherein the amino C₁-C₆-alkyl nitrogen is (i) unsubstituted or (ii) substituted with one or two substituents 5 independently selected from the group consisting of an C_1 - C_6 -alkyl, aryl, ar- C_1 - C_6 -alkyl, C_3 - C_8 cycloalkyl-C₁-C₆-alkyl, ar-C₁-C₆-alkoxycarbonyl, C₁-C₆-alkoxycarbonyl, and C₁-C₆-alkanoyl radical, or (iii) wherein the amino C₁-C₆-alkyl nitrogen and two substituents attached thereto form a 5- to 8-membered heterocyclo or heteroaryl ring, or (d) -NR²⁶R²⁷, where R^{26} and R^{27} are independently selected from the group consisting of a hydrido, C1-C6-alkyl, amino C1- C_6 -alkyl, hydroxy C_1 - C_6 -alkyl, aryl, ar- C_1 - C_6 -alkyl group, or R²⁶ and R²⁷ together with the depicted nitrogen atom form a 5- to 8-membered ring containing zero or one additional heteroatom that is oxygen, nitrogen or sulfur.

Several exemplary R¹ and R² groups that together form a contemplated heterocyclic ring are shown in the Tables that follow hereinafter, as well as in the descriptions of those 5- to 8-membered rings and the specific Examples, as are several contemplated

25 aromatic sulfone hydroxamic acid compounds.

In more preferred practice, R¹ and R² of formula I or formula A together with the atom to which they are bonded form a 5- to 8-membered ring that contains one, two or three heteroatoms. Most preferably, that ring is a 6-membered ring that contains one

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heteroatom located at the 4-position relative to the position at which the SO₂ group is bonded. Other preferred compounds for use in a contemplated process correspond in structure to one or more of formulas II, III, IV or V, which are discussed hereinafter.

In one embodiment, a preferred compound used in a contemplated process has a structure that corresponds to formula II, below:

$$(CH_2)_n - Z$$
 $(CH_2)_m (CH_2)_p$
 $G - A - R - E - Y$
 $O - HN - G - A - R - E - Y$

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wherein

R¹⁴ is hydrido, a pharmaceutically acceptable cation or C(W)R¹⁵ where W is O or S and R¹⁵ is selected from the group consisting of an C₁-C₆-alkyl, aryl, C₁-C₆-alkoxy, heteroaryl-C₁-C₆-alkyl, C₃-C₈-cycloalkyl-C₁-C₆-alkyl, aryloxy, ar-C₁-C₆-alkyl, heteroaryl and amino C₁-C₆-alkyl group wherein the aminoalkyl nitrogen is (i) unsubstituted or (ii) substituted with one or two substituents independently selected from the group consisting of an C₁-C₆-alkyl, aryl, ar-C₁-C₆-alkyl, C₃-C₈-cycloalkyl-C₁-C₆-alkyl, aryl, ar-C₁-C₆-alkyl, arkly araklyl and C₁-C₆-alkoxycarbonyl, and C₁-C₆-alkoxycarbonyl, and C₁-C₆-alkoxycarbonyl and C₁-C₆-alkoxycarbon

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alkyl nitrogen and two substituents attached thereto form a 5- to 8-membered heterocyclo or heteroaryl ring;

m is zero, 1 or 2;

n is zero, 1 or 2;

p is zero, 1 or 2;

the sum of m + n + p = 1, 2, 3 or 4;

- (a) one of X, Y and Z is selected from the group consisting of C(O), NR^6 , O, S, S(O), $S(O)_2$ and
- 10 $NS(0)_2R^7$, and the remaining two of X, Y and Z are CR^8R^9 , and $CR^{10}R^{11}$, or
 - (b) X and Z or Z and Y together constitute a moiety that is selected from the group consisting of $NR^6C(0)$, $NR^6S(0)$, $NR^6S(0)_2$, NR^6S , NR^6O , SS, NR^6NR^6 and OC(0), with the remaining one of X, Y and Z being CR^8R^9 , or
 - (c) n is zero and X, Y and Z together constitute a moiety selected from the group consisting of

$$R^{6}$$
 R^{6}
 R^{12}
 R^{12}

$$R^{6}$$
 R^{6}
 R^{6}
 R^{12}
 R^{12}
 R^{12}
 R^{13}
 R^{13}

wherein wavy lines are bonds to the atoms of the depicted ring;

 ${\tt R}^6$ and ${\tt R}^6{'}$ are independently selected from the 5 group consisting of hydrido, formyl, sulfonic- C_1 - C_6 alkyl, C_1-C_6 -alkoxycarbonyl- C_1-C_6 -alkyl, hydroxycarbonyl-C₁-C₆-alkyl, C₁-C₆-alkylcarbonyl-C₁- C_6 -alkyl, R^8R^9 -aminocarbonyl- C_1 - C_6 -alkyl, C_1 - C_6 $alkoxycarbonyl-C_1-C_6-alkylcarbonyl, hydroxycarbonyl-$ 10 C₁-C₆-alkylcarbonyl, C₁-C₆-alkylcarbonyl-C₁-C₆alkylcarbonyl, C₁-C₆-alkoxycarbonylcarbonyl, hydroxycarbonylcarbonyl, C1-C6-alkylcarbonylcarbonyl, R^8R^9 -aminocarbonylcarbonyl, C_1 - C_6 -alkanoyl, aryl- C_1 - C_6 -alkyl, aroyl, bis(C_1 - C_6 -alkoxy- C_1 - C_6 -alkyl)- C_1 - C_6 -15 alkyl, C_1 - C_6 -alkyl, C_1 - C_6 -haloalkyl, C_1 - C_6 perfluoroalkyl, C₁-C₆-trifluoromethylalkyl, C₁-C₆perfluoroalkoxy-C₁-C₆-alkyl, C₁-C₆-alkoxy-C₁-C₆alkyl, C3-C6-cycloalkyl, heteroarycarbonyl,

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heterocyclocarbonyl, C₃-C₈-heterocycloalkyl, C₃-C₈-heterocycloalkylcarbonyl, aryl, C₅-C₆-heterocyclo, C₅-C₆-heteroaryl, C₃-C₈-cycloalkyl-C₁-C₆-alkyl, aryloxy-C₁-C₆-alkyl, heteroaryloxy-C₁-C₆-alkyl, heteroarylcxy-C₁-C₆-alkyl, heteroarylthio-C₁-C₆-alkyl, arylsulfonyl, C₁-C₆-alkylsulfonyl, C₅-C₆-heteroarylsulfonyl, carboxy-C₁-C₆-alkyl, C₁-C₄-alkoxycarbonyl-C₁-C₆-alkyl, aminocarbonyl, C₁-C₆-alkyl(R⁸N)iminocarbonyl, aryl(R⁸N)iminocarbonyl, C₅-C₆-alkyl, C₁-C₆-alkylthio-C₁-C₆-alkyl, arylthio-C₃-C₆-alkyl, C₁-C₆-alkylthio-C₃-C₆-alkenyl, C₅-C₆-heteroaryl-C₁-C₆-alkyl, halo-C₁-C₆-alkanoyl, hydroxy-heteroaryl-C₁-C₆-alkyl, halo-C₁-C₆-alkanoyl, hydroxy-

15 C₃-C₆-alkynyl, C₁-C₄-alkoxy-C₁-C₄-alkyl, C₁-C₅-alkoxycarbonyl, aryloxycarbonyl, NR⁸R⁹-(R⁸)iminomethyl, NR⁸R⁹-C₁-C₅-alkylcarbonyl, hydroxy-C₁-C₅-alkyl, R⁸R⁹-aminocarbonyl, R⁸R⁹-aminocarbonyl, R⁸R⁹-aminocarbonyl, R⁸R⁹-aminosulfonyl, R⁸R⁹-aminosulfon-C₁-C₆-alkyl, R⁸R⁹-

 C_1-C_6 -alkanoyl, thiol- C_1-C_6 -alkanoyl, C_3-C_6 -alkenyl,

aminosulfonyl, R^8R^9 -aminosulfon- C_1 - C_6 -alkyl, R^8R^9 -amino- C_1 - C_6 -alkylsulfonyl and an R^8R^9 -amino- C_1 - C_6 -alkyl group;

 R^7 is selected from the group consisting of a arylalkyl, aryl, heteroaryl, heterocyclo, C_1 - C_6 -alkyl, C_3 - C_6 -alkynyl, C_3 - C_6 -alkenyl, C_1 - C_6 -carboxyalkyl and a C_1 - C_6 -hydroxyalkyl group;

 R^8 and R^9 and R^{10} and R^{11} are independently selected from the group consisting of a hydrido, hydroxy, C₁-C₆-alkyl, C₁-C₆-alkanoyl, aroyl, aryl, ar-C₁-C₆-alkyl, heteroaryl, heteroar-C₁-C₆-alkyl, C₂-5 C_6 -alkynyl, C_2 - C_6 -alkenyl, thiol- C_1 - C_6 -alkyl, C_1 - C_6 alkylthio-C₁-C₆-alkyl, cycloalkyl, cycloalkyl-C₁-C₆alkyl, heterocycloalkyl-C₁-C₆-alkyl, C₁-C₆-alkoxy-C₁- C_6 -alkyl, aralkoxy- C_1 - C_6 -alkyl, C_1 - C_6 -alkoxy- C_1 - C_6 alkoxy-C₁-C₆-alkyl, hydroxy-C₁-C₆-alkyl, hydroxycarbonyl-C₁-C₆-alkyl, hydroxycarbonylar-C₁-C₆alkyl, aminocarbonyl-C₁-C₆-alkyl, aryloxy-C₁-C₆alkyl, heteroaryloxy-C₁-C₆-alkyl, arylthio-C₁-C₆alkyl, heteroarylthio-C₁-C₆-alkyl, the sulfoxide or sulfone of any said thio substituents, perfluoro-C1- C_6 -alkyl, trifluoromethyl- C_1 - C_6 -alkyl, halo- C_1 - C_6 -15 alkyl, alkoxycarbonylamino-C₁-C₆-alkyl and an amino-C₁-C₆-alkyl group wherein the aminoalkyl nitrogen is (i) unsubstituted or (ii) substituted with one or two radicals independently selected from the group consisting of C_1-C_6 -alkyl, ar- C_1-C_6 -alkyl, cycloalkyl 20 and C_1-C_6 -alkanoyl, or wherein \mathbb{R}^8 and \mathbb{R}^9 or \mathbb{R}^{10} and ${\tt R}^{11}$ and the carbon to which they are bonded form a carbonyl group, or wherein ${\bf R}^{\bf 8}$ and ${\bf R}^{\bf 9}$ or ${\bf R}^{\bf 10}$ and ${\bf R}^{\bf 11}$, or ${\bf R}^{\bf 8}$ and ${\bf R}^{\bf 10}$ together with the atoms to which they are bonded form a 5- to 8-membered carbocyclic ring, 25 or a 5- to 8-membered heterocyclic or heteroaryl ring containing one or two heteroatoms that are nitrogen,

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oxygen, or sulfur, with the proviso that only one of ${\bf R}^8$ and ${\bf R}^9$ or ${\bf R}^{10}$ and ${\bf R}^{11}$ is hydroxy;

 ${\bf R}^{12}$ and ${\bf R}^{12}$ are independently selected from the group consisting of a hydrido, C₁-C₆-alkyl, aryl, ar-C₁-C₆-alkyl, heteroaryl, heteroaralkyl, C₂- C_6 -alkynyl, C_2 - C_6 -alkenyl, thiol- C_1 - C_6 -alkyl, cycloalkyl, cycloalkyl-C1-C6-alkyl, heterocycloalkyl- $C_1-C_6-alkyl$, $C_1-C_6-alkoxy-C_1-C_6-alkyl$, aryloxy- $C_1-C_6-alkyl$ alkyl, amino- C_1 - C_6 -alkyl, C_1 - C_6 -alkoxy- C_1 - C_6 -alkoxy-10 C_1-C_6 -alkyl, hydroxy- C_1-C_6 -alkyl, hydroxycarbonyl- C_1 -C6-alkyl, hydroxycarbonylar-C1-C6-alkyl, aminocarbonyl-C₁-C₆-alkyl, aryloxy-C₁-C₆-alkyl, heteroaryloxy- C_1 - C_6 -alkyl, C_1 - C_6 -alkylthio- C_1 - C_6 alkyl, arylthio- C_1 - C_6 -alkyl, heteroarylthio- C_1 - C_6 alkyl, the sulfoxide or sulfone of any said thio 15 substituents, perfluoro-C₁-C₆-alkyl, trifluoromethyl- $C_1-C_6-alkyl$, halo- $C_1-C_6-alkyl$, alkoxycarbonylamino- C_1-C_6 -alkyl and an amino- C_1-C_6 -alkyl group wherein the aminoalkyl nitrogen is (i) unsubstituted or (ii) substituted with one or two radicals independently 20 selected from the group consisting of C_1 - C_6 -alkyl, ar-C₁-C₆-alkyl, cycloalkyl and C₁-C₆-alkanoyl;

 $\rm R^{13}$ is selected from the group consisting of a hydrido, benzyl, phenyl, $\rm C_1-C_6-alkyl,\ C_2-C_6-alkynyl,\ C_2-C_6-alkenyl$ and a $\rm C_1-C_6-hydroxyalkyl$ group; and

G-A-R-E-Y is a substituent that preferably has a length greater than that of a pentyl group, and more preferably has a length greater than that of a

hexyl group. The substituent G-A-R-E-Y preferably has a length that is less than that of an icosyl group, and is more preferably less than that of a stearyl group. In this substituent:

G is an aryl or heteroaryl group;
A is selected from the group consisting of

- (1) -0-;
- (2) -S-;
- $(3) NR^{17} -;$
- 10 (4) $-CO-N(R^{17})$ or $-N(R^{17})-CO-$, wherein R^{17} is hydrogen, C_1-C_4 -alkyl, or phenyl;
 - (5) -CO-O- or -O-CO-;
 - (6) —O-CO-O-;
 - (7) -HC=CH-;
- 15 (8) —NH-CO-NH-;
 - (9) —C≡C-;
 - (10) -NH-CO-O- or -O-CO-NH-;
 - (11) -N=N-;
 - (12) -NH-NH-; and
- 20 (13) $-CS-N(R^{18})-$ or $-N(R^{18})-CS-$, wherein R^{18} is hydrogen C_1-C_4 -alkyl, or phenyl; or
 - (14) A is absent and G is bonded directly to R;
- 25 R is a moiety selected from the group consisting of alkyl, alkoxyalkyl, aryl, heteroaryl, cycloalkyl, heterocycloalkyl, aralkyl, heteroaralkyl, heterocycloalkylalkyl, cycloalkylalkyl, cycloalkoxyalkyl, heterocycloalkoxyalkyl, arylthicalkyl, arylthicalkyl
- 30 aryloxyalkyl, heteroaryloxyalkyl, arylthioalkyl, heteroarylthioalkyl, cycloalkylthioalkyl, and a

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heterocycloalkylthioalkyl group wherein the aryl or heteroaryl or cycloalkyl or heterocycloalkyl substituent is (i) unsubstituted or (ii) substituted with one or two radicals selected from the group consisting of a halo, alkyl, perfluoroalkyl, perfluoroalkoxy, perfluoroalkylthio, trifluoromethylalkyl, amino, alkoxycarbonylalkyl, alkoxy, C₁-C₂-alkylene-dioxy, hydroxycarbonylalkyl, hydroxycarbonylalkylamino, nitro, hydroxy, hydroxyalkyl, alkanoylamino, and a alkoxycarbonyl group, and R is other than alkyl or alkoxyalkyl when A is -O- or -S-;

E is selected from the group consisting of

- (1) $-CO(R^{19})$ or $-(R^{19})CO$ -, wherein R^{19} is a heterocycloalkyl, or a cycloalkyl group;
 - (2) -CONH- or -HNCO-; and
 - (3) -CO-;
 - (4) $-SO_2-R^{19}$ or $-R^{19}-SO_2-$;
- 20 (5) -SO₂-;
 - (6) $-NH-SO_2- \text{ or } -SO_2-NH-;$
 - (7) -S-;
 - (8) -NH-CO-O- or -O-CO-NH-; or
 - (9) E is absent and R is bonded directly

25 to Y; and

the moiety Y is absent or is selected from the group consisting of a hydrido, alkyl, alkoxy, haloalkyl, aryl, aralkyl, cycloalkyl, heteroaryl, hydroxy, aryloxy, aralkoxy, heteroaryloxy, heteroaralkyl, perfluoroalkoxy, perfluoroalkylthio, trifluoromethylalkyl, alkenyl, heterocycloalkyl,

cycloalkyl, trifluoromethyl, alkoxycarbonyl, and a

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aminoalkyl group, wherein the aryl, heteroaryl, aralkyl or heterocycloalkyl group is (i) unsubstituted or (ii) substituted with one or two radicals independently selected from the group consisting of an alkanoyl, halo, nitro, aralkyl, aryl, alkoxy, trifluoroalkyl, trifluoroalkoxy and an amino group wherein the amino nitrogen is (i) unsubstituted or (ii) substituted with one or two groups independently selected from hydrido, alkyl, and an aralkyl group.

The substituent -G-A-R-E-Y preferably contains two to four carbocyclic or heterocyclic rings, including the aryl or heteroaryl group, G. More preferably, each of those rings is 6-membered.

15 Additional separate preferences for a compound of formula II include: (a) that A is -O- or -S-, (b) R is an aryl, heteroaryl, cycloalkyl or heterocycloalkyl group, (c) E is absent, and (d) Y is selected from the group consisting of hydrido, an alkyl, alkoxy, perfluoroalkoxy and a perfluoroalkylthio group.

A more preferred compound for use in a contemplated process has a structure that corresponds to formula III, below:

$$R^{14}O$$
— HN
 $(CH_2)_m$
 $(CH_2)_p$
 $(CH_2)_p$
 R^3
 $(CH_2)_m$
 $(CH_2)_p$
 $(CH_2)_p$

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wherein R³ is a single-ringed aryl or heteroaryl group that is 5- or 6-membered, and is itself substituted at its own 4-position when a 6-membered ring and at its own 3- or 4-position when a 5-membered ring with a substituent selected from the group consisting of a thiophenoxy, 4-chlorophenoxy, 3-chlorophenoxy, 4-methoxyphenoxy, 3benzodioxol-5-yloxy, 3,4-dimethylphenoxy, 4-fluorophenoxy, 4-fluorothiophenoxy, phenoxy, 4-trifluoro-10 methoxyphenoxy, 4-trifluoromethylphenoxy, 4-(trifluoromethylthio)phenoxy, 4-(trifluoromethylthio)thiophenoxy, 4-chloro-3-fluorophenoxy, 4isopropoxyphenoxy, 4-isopropylphenoxy, (2-methyl-1,3benzothiazol-5-yl)oxy, 4-(1H-imidazol-1-yl)phenoxy, 15 4-chloro-3-methylphenoxy, 3-methyl-phenoxy, 4ethoxyphenoxy, 3,4-difluorophenoxy, 4-chloro-3methylphenoxy, 4-fluoro-3-chlorophenoxy, 4-(1H-1,2,4triazol-1-yl)phenoxy, 3,5-difluorophenoxy, 3,4dichlorophenoxy, 4-cyclopentylphenoxy, 4-bromo-3methylphenoxy, 4-bromophenoxy, 4-methylthiophenoxy, 20 4-phenylphenoxy, 4-benzylphenoxy, 6-quinolinyloxy, 4amino-3-methylphenoxy, 3-methoxyphenoxy, 5,6,7,8tetrahydro-2-naphthalenyloxy, 3-hydroxymethylphenoxy, and a 4-benzyloxyphenoxy group;

25 R¹⁴ is hydrido, a pharmaceutically acceptable cation or C(W)R¹⁵ where W is O or S and R¹⁵ is selected from the group consisting of an C₁-C₆-alkyl, aryl, C₁-C₆-alkoxy, heteroaryl-C₁-C₆-alkyl, C₃-C₈-cycloalkyl-C₁-C₆-alkyl, aryloxy, ar-C₁-C₆-alkyl, heteroaryl and amino C₁-C₆-alkyl group wherein the aminoalkyl nitrogen is (i)

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unsubstituted or (ii) substituted with one or two substituents independently selected from the group consisting of an C_1 - C_6 -alkyl, aryl, ar- C_1 - C_6 -alkyl, C_3 - C_8 -cycloalkyl- C_1 - C_6 -alkyl, ar- C_1 - C_6 -

- alkoxycarbonyl, C₁-C₆-alkoxycarbonyl, and a C₁-C₆-alkanoyl radical, or (iii) wherein the amino C₁-C₆-alkyl nitrogen and two substituents attached thereto form a 5- to 8-membered heterocyclo or heteroaryl ring;
- m is zero, 1 or 2;
 n is zero, 1 or 2;
 p is zero, 1 or 2;
 the sum of m + n + p = 1, 2, 3 or 4;
- (a) one of X, Y and Z is selected from the group consisting of C(O), NR^6 , O, S, S(O), S(O)₂ and $NS(O)_2R^7$, and the remaining two of X, Y and Z are CR^8R^9 , and $CR^{10}R^{11}$, or
- (b) X and Z or Z and Y together constitute a moiety that is selected from the group consisting of NR⁶C(O), NR⁶S(O), NR⁶S(O)₂, NR⁶S, NR⁶O, SS, NR⁶NR⁶ and OC(O), with the remaining one of X, Y and Z being CR⁸R⁹, or
 - (c) n is zero and X, Y and Z together constitute a moiety selected from the group consisting of

wherein wavy lines are bonds to the atoms of the depicted ring;

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 R^6 and $R^{6'}$ are independently selected from the group consisting of hydrido, formyl, sulfonic-C₁-C₆alkyl, c_1-c_6 -alkoxycarbonyl- c_1-c_6 -alkyl, hydroxycarbonyl-C₁-C₆-alkyl, C₁-C₆-alkylcarbonyl-C₁-C₆-alkyl, R⁸R⁹-aminocarbonyl-C₁-C₆-alkyl, C₁-C₆-

$$\label{eq:condition} \begin{split} &\text{alkoxycarbonyl-} \textbf{C}_1-\textbf{C}_6-\text{alkylcarbonyl}, \text{ hydroxycarbonyl-} \\ &\textbf{C}_1-\textbf{C}_6-\text{alkylcarbonyl}, \text{ } \textbf{C}_1-\textbf{C}_6-\text{alkylcarbonyl-} \textbf{C}_1-\textbf{C}_6-\\ &\text{alkylcarbonyl}, \text{ } \textbf{C}_1-\textbf{C}_6-\text{alkoxycarbonylcarbonyl}, \\ &\text{hydroxycarbonylcarbonyl}, \text{ } \textbf{C}_1-\textbf{C}_6-\text{alkylcarbonylcarbonyl}, \end{split}$$

- $\begin{array}{lll} & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\$
- alkyl, C₃-C₆-cycloalkyl, heteroarycarbonyl,
 heterocyclocarbonyl, C₃-C₈-heterocycloalkyl, C₃-C₈heterocycloalkylcarbonyl, aryl, C₅-C₆-heterocyclo,
 C₅-C₆-heteroaryl, C₃-C₈-cycloalkyl-C₁-C₆-alkyl,
 aryloxy-C₁-C₆-alkyl, heteroaryloxy-C₁-C₆-alkyl,
- heteroaryl- C_1 - C_6 -alkoxy- C_1 - C_6 -alkyl, heteroarylthio- C_1 - C_6 -alkyl, arylsulfonyl, C_1 - C_6 -alkylsulfonyl, C_5 - C_6 -heteroarylsulfonyl, carboxy- C_1 - C_6 -alkyl, C_1 - C_4 -alkoxycarbonyl- C_1 - C_6 -alkyl, aminocarbonyl, C_1 - C_6 -alkyl(R^8 N)iminocarbonyl, aryl(R^8 N)iminocarbonyl, C_5 -
- C₆-heterocyclo(R^8N)iminocarbonyl, arylthio- C_1 - C_6 -alkyl, C_1 - C_6 -alkylthio- C_1 - C_6 -alkyl, arylthio- C_3 - C_6 -alkenyl, C_1 - C_4 -alkylthio- C_3 - C_6 -alkenyl, C_5 - C_6 -heteroaryl- C_1 - C_6 -alkyl, halo- C_1 - C_6 -alkanoyl, hydroxy- C_1 - C_6 -alkanoyl, thiol- C_1 - C_6 -alkanoyl, C_3 - C_6 -alkenyl,
- C₃-C₆-alkynyl, C₁-C₄-alkoxy-C₁-C₄-alkyl, C₁-C₅-alkoxycarbonyl, aryloxycarbonyl, NR⁸R⁹- $(R^8) iminomethyl, NR^8R^9-C_1-C_5-alkylcarbonyl, hydroxy-$

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 C_1-C_5 -alkyl, R^8R^9 -aminocarbonyl, R^8R^9 -aminocarbonyl- C_1-C_6 -alkylcarbonyl, hydroxyaminocarbonyl, R^8R^9 -aminosulfonyl, R^8R^9 -aminosulfon- C_1-C_6 -alkyl, R^8R^9 -amino- C_1-C_6 -alkylsulfonyl and an R^8R^9 -amino- C_1-C_6 -alkyl group;

 R^7 is selected from the group consisting of a arylalkyl, aryl, heteroaryl, heterocyclo, C_1 - C_6 -alkyl, C_3 - C_6 -alkynyl, C_3 - C_6 -alkenyl, C_1 - C_6 -carboxyalkyl and a C_1 - C_6 -hydroxyalkyl group;

 R^8 and R^9 and R^{10} and R^{11} are independently 10 selected from the group consisting of a hydrido, hydroxy, C₁-C₆-alkyl, C₁-C₆-alkanoyl, aroyl, aryl, ar-C₁-C₆-alkyl, heteroaryl, heteroar-C₁-C₆-alkyl, C₂- C_6 -alkynyl, C_2 - C_6 -alkenyl, thiol- C_1 - C_6 -alkyl, C_1 - C_6 alkylthio-C₁-C₆-alkyl, cycloalkyl, cycloalkyl-C₁-C₆-15 alkyl, heterocycloalkyl-C₁-C₆-alkyl, C₁-C₆-alkoxy-C₁- C_6 -alkyl, aralkoxy- C_1 - C_6 -alkyl, C_1 - C_6 -alkoxy- C_1 - C_6 alkoxy-C₁-C₆-alkyl, hydroxy-C₁-C₆-alkyl, hydroxycarbonyl-C₁-C₆-alkyl, hydroxycarbonylar-C₁-C₆alkyl, aminocarbonyl-C₁-C₆-alkyl, aryloxy-C₁-C₆-20 alkyl, heteroaryloxy-C₁-C₆-alkyl, arylthio-C₁-C₆alkyl, heteroarylthio-C1-C6-alkyl, the sulfoxide or sulfone of any said thio substituents, perfluoro-C1-C₆-alkyl, trifluoromethyl-C₁-C₆-alkyl, halo-C₁-C₆-25 alkyl, alkoxycarbonylamino-C₁-C₆-alkyl and an amino-C₁-C₆-alkyl group wherein the aminoalkyl nitrogen is (i) unsubstituted or (ii) substituted with one or two

radicals independently selected from the group consisting of C₁-C₆-alkyl, ar-C₁-C₆-alkyl, cycloalkyl and C₁-C₆-alkanoyl, or wherein R⁸ and R⁹ or R¹⁰ and R¹¹ and the carbon to which they are bonded form a carbonyl group, or wherein R⁸ and R⁹ or R¹⁰ and R¹¹, or R⁸ and R¹⁰ together with the atoms to which they are bonded form a 5- to 8-membered carbocyclic ring, or a 5- to 8-membered heterocyclic or heteroaryl ring containing one or two heteroatoms that are nitrogen, oxygen, or sulfur, with the proviso that only one of R⁸ and R⁹ or R¹⁰ and R¹¹ is hydroxy;

 R^{12} and R^{12} are independently selected from the group consisting of a hydrido, C1-C6-alkyl, aryl, ar-C₁-C₆-alkyl, heteroaryl, heteroaralkyl, C₂- C_6 -alkynyl, C_2 - C_6 -alkenyl, thiol- C_1 - C_6 -alkyl, 15 cycloalkyl, cycloalkyl-C1-C6-alkyl, heterocycloalkyl- $C_1-C_6-alkyl$, $C_1-C_6-alkoxy-C_1-C_6-alkyl$, aryloxy- $C_1-C_6-alkyl$ alkyl, amino-C₁-C₆-alkyl, C₁-C₆-alkoxy-C₁-C₆-alkoxy- C_1-C_6 -alkyl, hydroxy- C_1-C_6 -alkyl, hydroxycarbonyl- C_1 -C6-alkyl, hydroxycarbonylar-C1-C6-alkyl, 20 aminocarbonyl-C₁-C₆-alkyl, aryloxy-C₁-C₆-alkyl, heteroaryloxy-C₁-C₆-alkyl, C₁-C₆-alkylthio-C₁-C₆alkyl, arylthio-C₁-C₆-alkyl, heteroarylthio-C₁-C₆alkyl, the sulfoxide or sulfone of any said thio substituents, perfluoro-C₁-C₆-alkyl, trifluoromethyl-25 C₁-C₆-alkyl, halo-C₁-C₆-alkyl, alkoxycarbonylamino- C_1-C_6 -alkyl and an amino- C_1-C_6 -alkyl group wherein the aminoalkyl nitrogen is (i) unsubstituted or (ii)

substituted with one or two radicals independently selected from the group consisting of C_1 - C_6 -alkyl, ar- C_1 - C_6 -alkyl, cycloalkyl and C_1 - C_6 -alkanoyl; and

R¹³ is selected from the group consisting of a hydrido, benzyl, phenyl, C₁-C₆-alkyl, C₂-C₆-alkynyl, C₂-C₆-alkenyl and a C₁-C₆-hydroxyalkyl group. Again, the use of a compound of formula III as a pharmaceutically acceptable salt is also contemplated.

10 Preferences related to a compound of formula III that also apply to a compound of formula II include the following, which are independently preferred: (a) the sum of m + n + p = 1 or 2, and more preferably 2; (b) Z is O, S or NR⁶; (c) R⁶ is selected from the group consisting of C₃-C₆-cycloalkyl, C₁-C₆-alkyl, C₃-C₆-alkenyl, C₃-C₆-alkynyl, C₁-C₆-alkoxy-C₁-C₆-alkyl, amino-C₁-C₆-alkyl, aminosulfonyl, heteroaryl-C₁-C₆-alkyl, aryloxycarbonyl, and C₁-C₆-alkoxycarbonyl; and (d) m

20 = n = zero, p = 1, and Y is NR⁶. Another preference for a compound of both of formulas II and III is that

for a compound of both of formulas II and III is that R^{14} be hydrido, or that W of the $C(W)R^{15}$ pro-drug form be O and R^{15} be a C_1 - C_6 -alkyl, aryl, C_1 - C_6 -alkoxy, heteroaryl- C_1 - C_6 -alkyl, C_3 - C_8 -cycloalkyl- C_1 - C_6 -alkyl, or aryloxy group.

A still more preferred compound for use in a contemplated process corresponds in structure to formula IV, below:

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HO—HN
$$SO_2$$
 R^3

Here, R^3 is as defined above as to formulas I, III and more preferably as defined as to formula II (wherein the R^3 radical is the substituent G-A-R-E-Y). Most preferably, R^3 is as defined in formula III.

Z is selected group the group consisting of O, S, NR^6 , SO, SO₂, and NSO_2R^7 ,

wherein R⁶ is selected from the group consisting of hydrido, C₁-C₅-alkyl, C₁-C₅-alkanoyl, benzyl, benzoyl, C₃-C₅-alkynyl, C₃-C₅-alkenyl, C₁-C₃-alkoxy-C₁-C₄-alkyl, C₃-C₆-cycloalkyl, heteroaryl-C₁-C₆-alkyl, C₁-C₅-hydroxyalkyl, C₁-C₅-carboxyalkyl, C₁-C₅-alkylcarbonyl, and NR⁸R⁹-C₁-C₅-alkylcarbonyl or NR⁸R⁹-C₁-C₅-alkyl wherein R⁸ and R⁹ are independently hydrido, C₁-C₅-alkyl, C₁-C₅-alkoxycarbonyl or aryl-C₁-C₅-alkoxycarbonyl, or NR⁸R⁹ together form a heterocyclic ring containing 5- to 8-atoms in the ring; and

 R^7 is selected from the group consisting of an arylalkyl, aryl, heteroaryl, heterocyclo, C_1 - C_6 -alkyl, C_3 - C_6 -alkyl, C_3 - C_6 -alkelyl, C_1 - C_6 -

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carboxyalkyl and a C_1 - C_6 -hydroxyalkyl group. Most preferably, Z is O or NR^6 . Here too, the use of a compound of formula IV as a pharmaceutically acceptable salt is contemplated.

A still more preferred group of contemplated compounds for use in a contemplated process correspond in structure to formula V, below;

10

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wherein

Z is as previously defined for formula IV;

W and Q are independently oxygen (0), NR⁶ or

sulfur (S), and R⁶ is as defined in formula IV; and

q is zero or one such that when q is zero, Q is

absent and the trifluoromethyl group is bonded

directly to the depicted phenyl ring. Here again, the

use of a compound of formula IV as a pharmaceutically
acceptable salt is contemplated.

Further compounds of formula A are also particularly preferred. One group of these compounds corresponds in structure to formula B, formula VIC, and more still particularly to formula VIC-1 and formula VIC-2, and formula VIII, below. In those formulas, ring structure Q including the depicted nitrogen atom is a heterocylic ring that contains 5-or 7-members, preferably 6-members, and can contain

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zero or one nitrogen atom in addition to that depicted. The members of substituent -A-R-E-Y (or -R-E-Y or -E-Y) are as defined elsewhere in the definition of the members of the substituent -G-A-R-5 E-Y. Furthermore, substituent -A-R-E-Y (or substituent -R-E-Y or -E-Y) is bonded at the 4-position relative to that depicted nitrogen atom when Q is a 6- or 7-membered ring and at the 3- or 4-position relative to that depicted nitrogen when Q is a 5-membered ring. Still fruther, R²⁰, X, Y, Z, m, n, and p of the ring system and g are as before described.

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$$\begin{array}{c} (CH_2)_{n} - Z \\ X \\ (CH_2)_{m} (CH_2)_{p} \\ S(O)_{g} \\ B-1 \\ O \end{array}$$

$$\begin{array}{c} Z \\ B-1 \\ O \end{array}$$

$$\begin{array}{c} Z \\ B-1 \\ O \end{array}$$

$$\begin{array}{c} A \\ B \end{array}$$

$$R^{20} \xrightarrow{(CH_2)_p} (CH_2)_p$$

$$S(O)_g$$

$$VIII$$

More particularly preferred among the compounds of formula VIC, formula VIC-1, formula VIC-5 2, and formula VIII, are the compounds of formulas IX, IX-1, IX-2, X, XI, XI-1, XI-2 and XII, below, wherein Z is as before described and the members of substituent group -E-Y and -R-E-Y are as defined for the substituent group -G-A-R-E-Y.

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HONH
$$SO_2$$
 X $R-E-Y$ O XI

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The use of a compound of formulas A and I-VI, VI VIC, VIC-1, VIC-2, VIII, IX, IX-1, IX-2 and X, or a pharmaceutically acceptable salt of one of those compounds is contemplated in a before-described process. In addition, the compounds of those formulas and their pharmaceutically acceptable salts are contemplated compounds of this invention.

Particularly preferred compounds within the group defined by formula B have the structural formulas shown below:

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Several particularly preferred compounds whose structures correspond to formulas I through XII are illustrated in the Tables and examples provided hereinafter.

As was noted before, the compounds of formulas I-XII, and their pharmaceutically acceptable salts are themselves contemplated compounds of the invention.

In preferred practice, an SO₂-linked R³ 10 radical is an aryl or heteroaryl group that is a 5or 6-membered single-ring that is itself substituted with one other single-ringed aryl or heteroaryl group or, with an alkyl or alkoxy group having a chain length of 3 to about 16 carbon atoms (and more 15 preferably a length of up to about 14 carbon atoms), a phenoxy group, a thiophenoxy [C6H5-S-] group, a phenylazo [C₆H₅-N₂-] group, a N-piperidyl [C₅H₁₀N-] group, a N-piperazyl [NC4H9N-] group or a benzamido 20 [-NHC(O)C₆H₅] group. The SO₂-linked single-ringed aryl or heteroaryl R³ group here is substituted at its own 4-position when a 6-membered ring and at its own 3- or 4-position when a 5-membered ring.

The SO₂-linked aryl or heteroaryl group of

25 a R³ radical is preferably itself substituted at the
4-position when a 6-membered ring or the 3- or 4position when a 5-membered ring. A particularly
preferred substituent is a single-ringed aryl or
heteroaryl, phenoxy, thiophenoxy, phenylazo, N
30 piperidyl, N-piperazyl or benzamido group that is
unsubstituted or can itself be substituted.

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The 4- and 3-positions of rings discussed here are numbered from the sites of substituent bonding as compared to formalized ring numbering positions used in heteroaryl nomenclature, as is discussed further hereinbelow. Here, single atoms such as halogen moieties (fluoro, chloro, bromo, or iodo) or substituents that contain one to a chain length of about five atoms other than hydrogen such as phenyl, C₁-C₄ alkyl, trifluoromethyl,

trifluoromethoxy, trifluorothiomethyl or carboxyethyl groups are preferred, although longer substituents can be accommodated up to a total length of an icosyl group.

Exemplary particularly preferred

- substituted SO₂-linked R³ radicals include

 4-(phenyl)phenyl [biphenyl], 4-(4'-methoxyphenyl)
 phenyl, 4-(phenoxy)phenyl, 4-(thiophenyl)phenyl [4(phenylthio)phenyl], 4-(azophenyl)phenyl, 4-[(4'
 trifluoromethylthio)phenoxy]phenyl, 4-[(4'-

Inasmuch as a contemplated SO₂-linked aryl or heteroaryl radical of an R³ group is itself preferably substituted with a 6-membered ring, two nomenclature systems are used together herein for ease in understanding substituent positions. The first system uses position numbers for the ring

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directly bonded to the SO₂-group, whereas the second system uses ortho, meta or para for the position of one or more substituents of a 6-membered ring bonded to a SO₂-linked aryl or heteroaryl radical. Although ortho, meta and para positional nomenclature is normally not used with aliphatic ring systems, it is believed more readily understood for describing the present compounds when used in conjunction with the numerical system for the first ring bonded to the SO₂-group. When a R³ radical is other than a 6-membered ring, substituent positions are numbered from the position of linkage to the aromatic or heteroaromatic ring. Formal chemical nomenclature is used in naming particular compounds.

Thus, the 1-position of an above-discussed SO₂-linked aryl or heteroaryl group is the position at which the SO₂-group is bonded to the ring. The 4-and 3-positions of rings discussed here are numbered from the sites of substituent bonding from the SO₂-linkage as compared to formalized ring numbering positions used in heteroaryl nomenclature.

When examined along its longest chain of atoms, an R³ radical including its own substituent has a total length that is greater than a saturated chain of five carbon atoms (a pentyl group), and preferably has a length greater than that of a saturated chain of six carbon atoms (a hexyl group); i.e., a length of about a heptyl chain or longer. An R³ radical also has a length that is less than that of a saturated chain of about 20 carbon atoms [an icosyl group (icosyl was formerly spelled eicosyl)]

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and more preferably about 18 carbon atoms (a stearyl group). Most preferably, the length of R³ is about that of an 8 to about 12 carbon atom chain, even though many more atoms may be present in ring structures or substituents. This length requirement is discussed further below.

Looked at more generally, and aside from specific moieties from which it is constructed, an R³ radical (group or moiety) has a length that is

10 greater than that of a pentyl group. Such an R³ radical also has a length that is less than that of an icosyl (didecyl) group. That is to say that R³ is a radical having a minimal length longer that a saturated five carbon chain, and preferably greater

15 than a hexyl group, but is shorter than the length of a saturated twenty carbon atom chain, and preferably shorter than an eighteen carbon chain. Most preferably, R³ has a length greater than that of an octyl group and less than that of a lauryl group.

More specifically, an R³ group has a minimal length of a hexyl group only when that substituent is comprised of two rings that can be fused or simply covalently linked together by exocyclic bonding. When R³ does not contain two

25 linked or fused rings, e.g., where a R³ radical includes an alkyl or second, third or fourth ring substituent, R³ has a length that is greater than that of a hexyl group. Exemplary of such two ring R³ groups are a 2-naphthyl group or a 2-quinolinyl group

30 (each with a six carbon chain length) and 8-purinyl (with a five carbon atom chain length). Without

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wishing to be bound by theory, it is believed that the presence of multiple rings in \mathbb{R}^3 enhances selectivity of the enzyme activity inhibitor profile.

The radical chain lengths are measured

along the longest linear atom chain in the radical,
following the skeletal atoms around a ring where
necessary. Each atom in the chain, e.g. carbon,
oxygen, sulfur or nitrogen, is presumed to be carbon
for ease in calculation.

10 Such lengths can be readily determined by using published bond angles, bond lengths and atomic radii, as needed, to draw and measure a desired, usually staggered, chain, or by building models using commercially available kits whose bond angles, 15 lengths and atomic radii are in accord with accepted, published values. Radical (substituent) lengths can also be determined somewhat less exactly by assuming that all atoms have bond lengths saturated carbon, that unsaturated bonds have the same lengths as 20 saturated bonds and that bond angles for unsaturated bonds are the same as those for saturated bonds, although the above-mentioned modes of measurement are preferred. For example, a phenyl or pyridyl group has a length of a four carbon chain, as does a 25 propoxy group, whereas a biphenyl group has a length of about an eight carbon chain using such a measurement mode.

In addition, a R³ group when rotated about an axis drawn through the SO₂-bonded 1-position and the 4-position of a 6-membered ring or the SO₂-bonded position and substituent-bonded 3- or 4-position of a 5-membered ring defines a three-dimensional volume

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whose widest dimension has the width of about one furanyl ring to about two phenyl rings in a direction transverse to that axis to rotation.

Thus, a 2-naphthyl substituent or an 8
5 purinyl substituent is an appropriately sized R³
group when examined using the above rotational width criterion as well as the before-discussed criterion.

On the other hand, a 1-naphthyl group or a 7- or 9purinyl group is too wide upon rotation and is

10 excluded from being an R³ group.

As a consequence of these length and width requirements, R³ radicals such as 4-(phenyl)phenyl [biphenyl], 4-(4'-methoxyphenyl)-phenyl, 4-(phenoxy)phenyl, 4-(thiophenyl)phenyl [4-(phenylthio)phenyl], 4-(azophenyl)phenyl, 4-[(4'-

15 (phenylthio)phenyl], 4-(azophenyl)phenyl, 4-[(4'trifluoromethylthio)phenoxy]phenyl, 4-[(4'trifluoromethylthio)thiophenyl]phenyl, 4-[(4'trifluoromethyl)phenoxy]phenyl, 4-[(4'trifluoromethyl)thiophenyl]phenyl, 4-[(4'-

trifluoromethoxy)phenoxy]phenyl, 4-[(4'-

trifluoromethoxy)thiophenyl]phenyl, 4-[(4'-phenyl)N-piperidyl]phenyl, 4-[(4'-acetyl)N-piperazyl]phenyl and 4-(benzamido)phenyl are particularly preferred R³ radicals. Those substituents can themselves also be substituted in the second ring from the SO₂ group at the meta- or para-position or both with a single atom or a substituent containing a longest chain length that is preferably of up to five atoms, excluding hydrogen.

Without wishing to be bound by theory, the length of a R³ radical substituent bonded to the SO₂ group is believed to play a role in the overall activity of a contemplated inhibitor compound against MMP enzymes generally. The length of the R³ radical

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group also appears to play a role in the selective activity of an inhibitor compound against particular MMP enzymes.

In particularly preferred practice, R³ is a 5 PhR²³ group, wherein Ph is phenyl. The phenyl ring (Ph) of a PhR²³ group is substituted at its paraposition (4-position) by an R²³ group that can be another single-ringed aryl or heteroaryl group, a piperidyl group, a piperazinyl group, a phenoxy group, a thiophenoxy [C₆H₅-S-] group, a phenylazo [C₆H₅-N₂-] group or a benzamido [-NHC(O)C₆H₅] group.

In one embodiment of a particularly preferred aromatic sulfone hydroxamate inhibitor compound, an R²³ substituent is phenoxy and is itself substituted at its own para-position with a moiety 15 that is selected from the group consisting of a halogen, a C₁-C₄ alkoxy group, a C₁-C₄ alkyl group, a dimethylamino group, a carboxyl C1-C3 alkylene group, a C_1-C_4 alkoxy carbonyl C_1-C_3 alkylene group, a trifluoromethylthio group, a trifluoromethoxy group, 20 a trifluoromethyl group and a carboxamido C1-C3 alkylene group, or is substituted at the meta- and para-positions by a methylenedioxy group. It is to be understood that any R²³ substituent can be substituted with a moiety from the above list. 25 Such substitution at the para-position is preferred.

The present invention also contemplates a compound that corresponds in structure to formula VI, below, that is useful in preparing a compound of formulas I-V, as well as as an active MMP-inhibiting compound and as a pro-drug form of an inhibitor.

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wherein g is zero, 1 or 2;

 R^{20} is (a) $-0-R^{21}$, where R^{21} is selected 5 from the group consisting of a hydrido, C₁-C₆-alkyl, aryl, ar- C_1 - C_6 -alkyl group and a pharmaceutically acceptable cation, (b) $-NH-O-R^{22}$ wherein R^{22} is a selectively removable protecting group such as a 2-10 tetrahydropyranyl, benzyl, p-methoxybenzyl (MOZ), carbonyl-C₁-C₆-alkoxy, trisubstituted silyl group or o-nitrophenyl group, peptide synthesis resin and the like, wherein the trisubstituted silyl group is substituted with C₁-C₆-alkyl, aryl, or ar-C₁-C₆-alkyl or a mixture thereof, (c) $-NH-O-R^{14}$, where R^{14} is 15 hydrido, a pharmaceutically acceptable cation or $C(W)R^{25}$ where W is O (oxo) or S (thioxo) and R^{25} is selected from the group consisting of an C1-C6-alkyl, aryl, C1-C6-alkoxy, heteroaryl-C1-C6-alkyl, C3-C8-20 cycloalkyl-C₁-C₆-alkyl, aryloxy, ar-C₁-C₆-alkoxy, ar- C_1 - C_6 -alkyl, heteroaryl and amino C_1 - C_6 -alkyl group wherein the amino C_1-C_6 -alkyl nitrogen is (i) unsubstituted or (ii) substituted with one or two substituents independently selected from the group

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consisting of an C₁-C₆-alkyl, aryl, ar-C₁-C₆-alkyl,

C₃-C₈-cycloalkyl-C₁-C₆-alkyl, ar-C₁-C₆
alkoxycarbonyl, C₁-C₆-alkoxycarbonyl, and C₁-C₆
alkanoyl radical, or (iii) wherein the amino C₁-C₆
5 alkyl nitrogen and two substituents attached thereto form a 5- to 8-membered heterocyclo or heteroaryl ring, or (d) -NR²⁶R²⁷, where R²⁶ and R²⁷ are independently selected from the group consisting of a hydrido, C₁-C₆-alkyl, amino C₁-C₆-alkyl, hydroxy C₁
10 C₆-alkyl, aryl, ar-C₁-C₆-alkyl group, or R²⁶ and R²⁷ together with the depicted nitrogen atom form a 5- to 7-membered ring containing zero or one additional heteroatom that is oxygen, nitrogen or sulfur;

m is zero, 1 or 2;

n is zero, 1 or 2;

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p is zero, 1 or 2;

the sum of m + n + p = 1, 2, 3 or 4;

- (a) one of X, Y and Z is selected from the group consisting of C(O), NR^6 , O, S, S(O), $S(O)_2$ and
- NS(0)₂ \mathbb{R}^7 , and the remaining two of X, Y and Z are $\mathbb{C}\mathbb{R}^8\mathbb{R}^9$, and $\mathbb{C}\mathbb{R}^{10}\mathbb{R}^{11}$, or
 - (b) X and Z or Z and Y together constitute a moiety that is selected from the group consisting of $NR^6C(0)$, $NR^6S(0)$, $NR^6S(0)_2$, NR^6S , NR^6O , SS, NR^6NR^6 and OC(0), with the remaining one of X, Y and Z being CR^8R^9 , or
 - (c) n is zero and X, Y and Z together constitute a moiety selected from the group consisting of

$$R^{6}$$
, R^{6} , R^{12} , R^{12} , R^{12} , R^{13}

wherein wavy lines are bonds to the atoms of the 5 depicted ring;

 R^6 and R^6 ' are independently selected from the group consisting of hydrido, formyl, sulfonic- C_1 - C_6 -alkyl, C_1 - C_6 -alkoxycarbonyl- C_1 - C_6 -alkyl, C_1 - C_6 - C_6 -alkyl, C_1 - C_6 -C

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 $\label{eq:convergence} alkoxycarbonyl-C_1-C_6-alkylcarbonyl, hydroxycarbonyl-C_1-C_6-alkylcarbonyl-C_1-C_6-alkylcarbonyl, C_1-C_6-alkoxycarbonylcarbonyl, hydroxycarbonylcarbonyl, C_1-C_6-alkylcarbonylcarbonyl,$

- $\begin{array}{lll} & R^8R^9-aminocarbonylcarbonyl, & C_1-C_6-alkanoyl, & aryl-C_1-C_6-alkyl, & aroyl, & bis(C_1-C_6-alkoxy-C_1-C_6-alkyl)-C_1-C_6-alkyl, & C_1-C_6-alkyl, & C_1-C_6-alkoxy-C_1-C_6-alkyl, & C_1-C_6-alkyl, & C_1-C_6-al$
- alkyl, C₃-C₆-cycloalkyl, heteroarycarbonyl,
 heterocyclocarbonyl, C₃-C₈-heterocycloalkyl, C₃-C₈heterocycloalkylcarbonyl, aryl, C₅-C₆-heterocyclo,
 C₅-C₆-heteroaryl, C₃-C₈-cycloalkyl-C₁-C₆-alkyl,
 aryloxy-C₁-C₆-alkyl, heteroaryloxy-C₁-C₆-alkyl,
- heteroaryl- C_1 - C_6 -alkoxy- C_1 - C_6 -alkyl, heteroarylthio- C_1 - C_6 -alkyl, arylsulfonyl, C_1 - C_6 -alkylsulfonyl, C_5 - C_6 -heteroarylsulfonyl, carboxy- C_1 - C_6 -alkyl, C_1 - C_4 -alkoxycarbonyl- C_1 - C_6 -alkyl, aminocarbonyl, C_1 - C_6 -alkyl(R^8N)iminocarbonyl, aryl(R^8N)iminocarbonyl, C_5 -
- 25 C_3-C_6 -alkynyl, C_1-C_4 -alkoxy- C_1-C_4 -alkyl, C_1-C_5 -alkoxycarbonyl, aryloxycarbonyl, NR^8R^9 - $(R^8) iminomethyl, NR^8R^9-C_1-C_5-alkylcarbonyl, hydroxy-$

 C_1 - C_5 -alkyl, R^8R^9 -aminocarbonyl, R^8R^9 -aminocarbonyl- C_1 - C_6 -alkylcarbonyl, hydroxyaminocarbonyl, R^8R^9 -aminosulfonyl, R^8R^9 -aminosulfon- C_1 - C_6 -alkyl, R^8R^9 -amino- C_1 - C_6 -alkylsulfonyl and an R^8R^9 -amino- C_1 - C_6 -alkyl group;

 R^7 is selected from the group consisting of a arylalkyl, aryl, heteroaryl, heterocyclo, C_1 - C_6 -alkyl, C_3 - C_6 -alkynyl, C_3 - C_6 -alkenyl, C_1 - C_6 -carboxyalkyl and a C_1 - C_6 -hydroxyalkyl group;

 R^8 and R^9 and R^{10} and R^{11} are independently 10 selected from the group consisting of a hydrido, hydroxy, C₁-C₆-alkyl, C₁-C₆-alkanoyl, aroyl, aryl, $ar-C_1-C_6-alkyl$, heteroaryl, heteroar- $C_1-C_6-alkyl$, $C_2-alkyl$ C_6 -alkynyl, C_2 - C_6 -alkenyl, thiol- C_1 - C_6 -alkyl, C_1 - C_6 alkylthio- C_1 - C_6 -alkyl, cycloalkyl, cycloalkyl- C_1 - C_6 -15 alkyl, heterocycloalkyl-C₁-C₆-alkyl, C₁-C₆-alkoxy-C₁- C_6 -alkyl, aralkoxy- C_1 - C_6 -alkyl, C_1 - C_6 -alkoxy- C_1 - C_6 alkoxy-C₁-C₆-alkyl, hydroxy-C₁-C₆-alkyl, hydroxycarbonyl-C₁-C₆-alkyl, hydroxycarbonylar-C₁-C₆alkyl, aminocarbonyl-C₁-C₆-alkyl, aryloxy-C₁-C₆-20 alkyl, heteroaryloxy-C₁-C₆-alkyl, arylthio-C₁-C₆alkyl, heteroarylthio-C₁-C₆-alkyl, the sulfoxide or sulfone of any said thio substituents, perfluoro-C1- C_6 -alkyl, trifluoromethyl- C_1 - C_6 -alkyl, halo- C_1 - C_6 alkyl, alkoxycarbonylamino-C₁-C₆-alkyl and an amino-25 C_1-C_6 -alkyl group wherein the aminoalkyl nitrogen is (i) unsubstituted or (ii) substituted with one or two

radicals independently selected from the group consisting of C_1-C_6 -alkyl, ar- C_1-C_6 -alkyl, cycloalkyl and C_1-C_6 -alkanoyl, or wherein \mathbb{R}^8 and \mathbb{R}^9 or \mathbb{R}^{10} and ${\tt R}^{11}$ and the carbon to which they are bonded form a 5 carbonyl group, or wherein R^8 and R^9 or R^{10} and R^{11} , or ${\bf R}^{\bf 8}$ and ${\bf R}^{\bf 10}$ together with the atoms to which they are bonded form a 5- to 8-membered carbocyclic ring, or a 5- to 8-membered heterocyclic or heteroaryl ring containing one or two heteroatoms that are nitrogen, oxygen, or sulfur, with the proviso that only one of R^8 and R^9 or R^{10} and R^{11} is hydroxy:

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R¹² and R¹² are independently selected from the group consisting of a hydrido, C₁-C₆-alkyl, aryl, ar-C₁-C₆-alkyl, heteroaryl, heteroaralkyl, C₂- C_6 -alkynyl, C_2 - C_6 -alkenyl, thiol- C_1 - C_6 -alkyl, 15 cycloalkyl, cycloalkyl-C1-C6-alkyl, heterocycloalkyl- $C_1-C_6-alkyl$, $C_1-C_6-alkoxy-C_1-C_6-alkyl$, $aryloxy-C_1-C_6-alkyl$ alkyl, amino- C_1 - C_6 -alkyl, C_1 - C_6 -alkoxy- C_1 - C_6 -alkoxy-C₁-C₆-alkyl, hydroxy-C₁-C₆-alkyl, hydroxycarbonyl-C₁-C6-alkyl, hydroxycarbonylar-C1-C6-alkyl, 20 aminocarbonyl-C₁-C₆-alkyl, aryloxy-C₁-C₆-alkyl, heteroaryloxy-C₁-C₆-alkyl, C₁-C₆-alkylthio-C₁-C₆alkyl, arylthio-C₁-C₆-alkyl, heteroarylthio-C₁-C₆alkyl, the sulfoxide or sulfone of any said thio 25 substituents, perfluoro-C₁-C₆-alkyl, trifluoromethyl-C₁-C₆-alkyl, halo-C₁-C₆-alkyl, alkoxycarbonylamino- C_1-C_6 -alkyl and an amino- C_1-C_6 -alkyl group wherein the aminoalkyl nitrogen is (i) unsubstituted or (ii)

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substituted with one or two radicals independently selected from the group consisting of C_1 - C_6 -alkyl, ar- C_1 - C_6 -alkyl, cycloalkyl and C_1 - C_6 -alkanoyl;

 R^{13} is selected from the group consisting of a hydrido, benzyl, phenyl, C_1 - C_6 -alkyl, C_2 - C_6 -alkynyl, C_2 - C_6 -alkenyl and a C_1 - C_6 -hydroxyalkyl group; and

R²⁴ is R³ as defined in formulas I, III, IV or is the substituent G-A-R-E-Y of formula II

10 (formula VIA). Alternatively, R²⁴ is R³, an aryl or heteroaryl group that is substituted with a coupling substituent reactive for coupling with another moiety (formula VIB), such as a nucleophilically displaceable leaving group, D.

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$$(CH_2)_n - Z$$
 $(CH_2)_n - Z$
 $(CH_2)_n - Z$

Exemplary nucleophilically displaceable leaving groups, D, include a halo (fluoro, chloro, bromo, or iodo) nitro, azido, phenylsulfoxido, aryloxy, C₁-C₆-20 alkoxy, a C₁-C₆-alkylsulfonate or arylsulfonate group and a trisubstituted ammonium group in which the three substituents are independently aryl, ar- C₁-C₆-alkyl or C₁-C₆-alkyl. Additional coupling substituents include, without limitation, a hydroxyl

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group and an amino group that can be coupled with carbonyl-containing moieties to form esters, urethanes, carbonates, amides and ureas. Similarly, a carboxyl coupling substituent can be used to form 5 an ester, thioester or amide. Thus, a coupling substituent is useful in converting a coupling substituent-containing aryl or heteroaryl group into a substituent such as a G-A-R-E-Y substituent discussed hereinabove by the formation of a covalent bond.

A compound of formula VI can be coupled with another moiety at the R3' coupling substituent to form a compound whose newly formed R³ group is that of formulas I, III, IV or -G-A-R-E-Y. Exemplary of such couplings are the nucleophilic displacement to form ethers and thioethers, as well as the formation of ester, amide, urea, carbonate, urethane and the like linkages.

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More particularly, where a R²⁰ group is -0- R^{21} , with R^{21} being selected from the group 20 consisting of a hydrido, C_1-C_6 -alkyl, aryl, ar- C_1-C_6 alkyl group and a pharmaceutically acceptable cation, a precursor carboxylic acid or ester compound is defined that can be readily transformed into a 25 hydroxamic acid, as is illustrated in several examples hereinafter.

Where a R^{20} group is -NH-O- R^{22} , wherein R^{22} is a selectively removable protecting group such as a 2-tetrahydropyranyl, benzyl, p-methoxybenzyl (MOZ), carbonyl-C₁-C₆-alkoxy, trisubstituted silyl group, an o-nitrophenyl group, or a peptide synthesis resin and

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the like, a synthetic intermediate is typically defined. In these compounds, a trisubstituted silyl group is substituted with C₁-C₆-alkyl, aryl, ar-C₁-C₆-alkyl or a mixture thereof, such as a trimethylsilyl, dimethylisopropylsilyl, triethylsilyl, triphenylsilyl, t-butyldiphenylsilyl, diphenylmethylsilyl, a tribenzylsilyl group, and the like. Exemplary trisubstituted silyl protecting groups and their uses are discussed at several places in Greene et al., Protective Groups In Organic Synthesis, 2nd ed., John Wiley & Sons, Inc., New York (1991).

A contemplated peptide synthesis resin is solid phase support also known as a so-called

15 Merrifield's Peptide Resin that is adapted for synthesis and selective release of hydroxamic acid derivatives as is commercially available from Sigma Chemical Co., St. Louis , MO. An exemplary peptide synthesis resin so adapted and its use in the

20 synthesis of hydroxamic acid derivatives is discussed in Floyd et al., Tetrahedron Let., 37(44):8048-8048(1996).

a 2-tetrahydropyranyl (THP) protecting group is a particularly preferred selectively

removable protecting group. A contemplated THPprotected hydroxamate compound of formula VII can be prepared by reacting the carboxylic acid precursor compound of formula VII [where R²⁰ is -O-R²¹ and R²¹ is a hydrido group] in water with O-(tetrahydro-2Hpyran-2-yl)hydroxylamine in the presence of Nmethylmorpholine, N-hydroxybenzotriazole hydrate and a water-soluble carbodiimide such as 1-(3-

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dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride. The THP protecting group is readily removable in an aqueous acid solution such as an aqueous mixture of p-toluenesulfonic acid or HCl and acetonitrile or methanol. An illustrative THP-protected compound corresponds in structure to formula VIIB, below, wherein m, n, p, g, X, Z, Y, and D are as defined previously.

Where R^{20} is $-NR^{26}R^{27}$, and R^{26} and R^{27} are as defined before, an amide compound is defined that can be used as a precursor intermediate and surprisingly as a MMP inhibitor compound. R^{26} and R^{27} are both preferably hydrido.

Where a R²⁰ group is -NH-O-R¹⁴, and R¹⁴ is hydrido, or a pharmaceutically acceptable cation, an active hydroxamic acid or hydroxamate is defined.

Where a R²⁰ group is -NH-O-R¹⁴, and R¹⁴ is a C(W)R²⁵ group as defined before, a pro-drug form of the hydroxamic acid is defined that can form a hydroxamic acid or hydroxamate form of the inhibitor in situ.

A particularly preferred precursor intermediate to an intermediate compound of formula VI is an intermediate compound of formula VII, below

$$(CH_2)_n - Z$$
 $(CH_2)_m (CH_2)_p$
 $S(O)_g$

VII

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wherein m, n, p, g, X, Z, Y, D and \mathbb{R}^{20} are as defined above for formula VI.

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In regard to a compound of each of formulas VI and VII, the subscript letter "q" is used to show the oxidation state of the sulfur atom. Where q is 10 zero, the sulfur is unoxidized, and the compound depicted is typically the sulfide reaction product of a sulfur-containing synthon as is illustrated in the examples hereinafter. Where g is 1, the sulfur is oxidized to a sulfoxide, whereas when g is 2, the 15 sulfur is oxidized to a sulfone as is also illustrated hereinafter. A compound of formulas VI or VII wherein g is zero or 1 as itself typically an intermediate in the formation of a similar compound wherein g is 2 and the intermediate is a preferred 20 sulfone.

A preferred intermediate corresponds in structure to formula VIIA, below, wherein R^{20} , X, Y, Z, m, n, p and D are as defined previously.

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$$R^{20} \xrightarrow{(CH_2)_m} \xrightarrow{(CH_2)_p} SO_2$$
VIIA

In the written descriptions of molecules and groups, molecular descriptors can be combined to 5 produce words or phrases that describe structural groups or are combined to describe structural groups. Such descriptors are used in this document. Common illustrative examples include such terms as aralkyl (or arylalkyl), heteroaralkyl, heterocycloalkyl, cycloalkylalkyl, aralkoxyalkoxycarbonyl and the like. 10 A specific example of a compound encompassed with the latter descriptor aralkoxyalkoxycarbonyl is C6H5-CH2- $CH_2-O-CH_2-O-(C=O)-$ wherein C_6H_5- is phenyl. also to be noted that a structural group can have more than one descriptive word or phrase in the art, for example, heteroaryloxyalkylcarbonyl can also be termed heteroaryloxyalkanoyl. Such combinations are used herein in the description of the processes, compounds and compositions of this invention and 20 further examples are described below. The following list is not intended to be exhaustive or drawn out but provide illustrative examples of words or phrases (terms) that are used herein.

As utilized herein, the term "alkyl", alone or in combination, means a straight-chain or branched-chain alkyl radical containing 1 to about 12 carbon atoms, preferably 1 to about 10 carbon atoms, and more preferably 1 to about 6 carbon atoms.

Examples of such radicals include methyl, ethyl, n-

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propyl, isopropyl, n-butyl, isobutyl, sec-butyl, tert-butyl, pentyl, iso-amyl, hexyl, octyl and the like.

The term "alkenyl", alone or in

5 combination, means a straight-chain or branched-chain hydrocarbon radical having one or more double bonds and containing 2 to about 12 carbon atoms preferably 2 to about 10 carbon atoms, and more preferably, 2 to about 6 carbon atoms. Examples of suitable alkenyl radicals include ethenyl (vinyl), 2-propenyl, 3-propenyl, 1,4-pentadienyl, 1,4-butadienyl, 1-butenyl, 2-butenyl, 3-butenyl, decenyl and the like.

The term "alkynyl", alone or in combination, means a straight-chain hydrocarbon radical having one or more triple bonds and containing 2 to about 12 carbon atoms, preferably 2 to about 10 carbon atoms, and more preferably, 2 to about 6 carbon atoms. Examples of alkynyl radicals include ethynyl, 2-propynyl, 3-propynyl, decynyl, 1-butynyl, 2-butynyl, 3-butynyl, and the like.

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The term "carbonyl" or "oxo", alone or in combination, means a -C(=0)- group wherein the remaining two bonds (valences) can be independently substituted. The term carbonyl is also intended to encompass a hydrated carbonyl group $-C(OH)_2$ -.

The term "thiol" or "sulfhydryl", alone or in combination, means a -SH group. The term "thio" or "thia", alone or in combination, means a thiaether group; i.e., an ether group wherein the ether oxygen is replaced by a sulfur atom.

The term "amino", alone or in combination, means an amine or -NH2 group whereas the term monosubstituted amino, alone or in combination, means a substituted amine -N(H)(substituent) group wherein one hydrogen atom is replaced with a substituent, and disubstituted amine means a -N(substituent)2 wherein

two hydrogen atoms of the amino group are replaced with independently selected substituent groups.

Amines, amino groups and amides are compounds that can be designated as primary (I°), secondary (II°) or tertiary (III°) or unsubstituted, mono-substituted or N,N-disubstituted depending on the degree of substitution of the amino nitrogen. Quaternary amine (ammonium)(IV°) means a nitrogen with four substituents [-N+(substituent)] that is positively charged and accompanied by a counter ion, whereas N-oxide means one substituent is oxygen and the group is represented as [-N+(substituent),-O-]; i.e., the charges are internally compensated.

The term "cyano", alone or in combination,

means a -C-triple bond-N (-C=N) group. The term

"azido", alone or in combination, means a -N-triple

bond-N (-N=N) group. The term "hydroxyl", alone or

in combination, means a -OH group. The term "nitro",

alone or in combination, means a -NO2 group. The

term "azo", alone or in combination, means a -N=N
group wherein the bonds at the terminal positions can

be independently substituted.

The term "hydrazino", alone or in combination, means a -NH-NH- group wherein the depicted remaining two bonds (valences) can be independently substituted. The hydrogen atoms of the hydrazino group can be replaced, independently, with substituents and the nitrogen atoms can form acid addition salts or be quaternized.

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The term "sulfonyl", alone or in combination, means a -SO2- group wherein the depicted remaining two bonds (valences) can be independently substituted. The term "sulfoxido", alone or in combination, means a -SO- group wherein the remaining two bonds (valences) can be independently substituted.

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The term "sulfone", alone or in combination, means a -SO,- group wherein the depicted remaining two bonds (valences) can be independently substituted. The term "sulfenamide", alone or in combination, means a -SON= group wherein the 5 remaining three depicted bonds (valences) can be independently substituted. The term "sulfide", alone or in combination, means a -S- group wherein the remaining two bonds (valences) can be independently substituted.

The term "alkoxy", alone or in combination, means an alkyl ether radical wherein the term alkyl is as defined above. Examples of suitable alkyl ether radicals include methoxy, ethoxy, n-propoxy, isopropoxy, n-butoxy, iso-butoxy, sec-butoxy, tertbutoxy and the like.

The term "cycloalkyl", alone or in combination, means a cyclic alkyl radical that contains 3 to about 8 carbon atoms. The term "cycloalkylalkyl" means an alkyl radical as defined above that is substituted by a cycloalkyl radical containing 3 to about 8, preferably 3 to about 6, carbon atoms. Examples of such cycloalkyl radicals include cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl and the like.

A heterocyclic (heterocyclo) or heterocyclo portion of a heterocyclocarbonyl, heterocyclooxycarbonyl, heterocycloalkoxycarbonyl, or heterocycloalkyl group or the like is a saturated or partially unsaturated monocyclic, bicyclic or tricyclic heterocycle that contains one or more hetero atoms selected from nitrogen, oxygen and sulphur. Heterocyclo compounds include benzofused heterocyclic compounds such as benzo-1,4-dioxane. Such a moiety can be optionally substituted on one or more ring carbon atoms by halogen, hydroxy, hydroxycarbonyl, alkyl, alkoxy, oxo, and the like,

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and/or on a secondary nitrogen atom (i.e., -NH-) of the ring by alkyl, aralkoxycarbonyl, alkanoyl, aryl or arylalkyl or on a tertiary nitrogen atom (i.e., =N-) by oxido and that is attached via a carbon atom. The tertiary nitrogen atom with three substituents can also attached to form a N-oxide [=N(O)-] group.

The term "aryl", alone or in combination, means a 5- or 6-membered carbocyclic aromatic ring-containing moiety or a fused ring system containing two or three rings that have all carbon atoms in the ring; i.e., a carbocyclic aryl radical. Exemplary carbocyclic aryl radicals include phenyl, indenyl and naphthyl radicals.

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The term "heteroaryl", alone or in combination means a 5- or 6-membered aromatic ring-15 containing moiety or a fused ring system (radical) containing two or three rings that have carbon atoms and also one or more heteroatoms in the ring(s) such as sulfur, oxygen and nitrogen. Examples of such 20 heterocyclic or heteroaryl groups are pyrrolidinyl, piperidyl, piperazinyl, morpholinyl, thiamorpholinyl, pyrrolyl, imidazolyl (e.g., imidazol-4-yl, 1-benzyloxycarbonylimidazol-4-yl, and the like), pyrazolyl, pyridyl, pyrazinyl, pyrimidinyl, furyl, tetrahydrofuryl, thienyl, triazolyl, tetrazolyl, 25 oxazolyl, oxadiazoyl, thiazolyl, thiadiazoyl, indolyl (e.g., 2-indolyl, and the like), quinolinyl, (e.g., 2-quinolinyl, 3-quinolinyl, 1-oxido-2-quinolinyl, and the like), isoquinolinyl (e.g., 1-isoquinolinyl, 3-isoquinolinyl, and the like), tetrahydroquinolinyl (e.g., 1,2,3,4-tetrahydro-2-quinolyl, and the like), 1,2,3,4-tetrahydroisoquinolinyl (e.g., 1,2,3,4tetrahydro-1-oxo-isoquinolinyl, and the like), quinoxalinyl, β -carbolinyl, 2-benzofurancarbonyl, 35 benzothiophenyl, 1-, 2-, 4- or 5-benzimidazolyl, and

the like radicals.

When an aryl or heteroaryl radical is a substituting moiety (group, substituent, or radical), it can itself substituted, the last-named substituent is independently selected from the group consisting of a cyano, perfluoroalkyl, trifluoromethoxy, trifluoromethylthio, haloalkyl, trifluoromethylalkyl, aralkoxycarbonyl, aryloxycarbonyl, hydroxy, halo, alkyl, alkoxy, nitro, thiol, hydroxycarbonyl, aryloxy, arylthio, aralkyl, aryl, arylcarbonylamino, heteroaryloxy, 10 heteroarylthio, heteroaralkyl, cycloalkyl, heterocyclooxy, heterocyclothio, heterocycloamino, cycloalkyloxy, cycloalkylthio, heteroaralkoxy, heteroaralkylthio, aralkoxy, aralkylthio, 15 aralkylamino, heterocyclo, heteroaryl, arylazo, hydroxycarbonylalkoxy, alkoxycarbonylalkoxy, alkanoyl, arylcarbonyl, aralkanoyl, alkanoyloxy, aralkanoyloxy, hydroxyalkyl, hydroxyalkoxy, alkylthio, alkoxyalkylthio, alkoxycarbonyl, aryloxyalkoxyaryl, arylthioalkylthioaryl, 20 aryloxyalkylthioaryl, arylthioalkoxyaryl, hydroxycarbonylalkoxy, hydroxycarbonylalkylthio, alkoxycarbonylalkoxy, alkoxycarbonylalkylthio, amino, wherein the amino nitrogen is (i) unsubstituted, or (ii) substituted with one or two substituents 25 that are independently selected from the group consisting of an alkyl, aryl, heteroaryl, aralkyl, cycloalkyl, aralkoxycarbonyl, alkoxycarbonyl, arylcarbonyl, aralkanoyl, 30 heteroarylcarbonyl, heteroaralkanoyl and an alkanoyl group, or (iii) wherein the amino nitrogen and two substituents attached thereto form a 5- to 8-membered heterocyclo or heteroaryl ring containing zero to two

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additional heteroatoms that are nitrogen, oxygen or sulfur and which ring itself is (a) unsubstituted or (b) substituted with one or two groups independently selected from the group consisting of an aryl, alkyl, heteroaryl, aralkyl, heteroaralkyl, hydroxy, alkoxy, alkanoyl, cycloalkyl, heterocycloalkyl, alkoxycarbonyl, hydroxyalkyl, trifluoromethyl, benzofused heterocycloalkyl, hydroxyalkoxyalkyl, aralkoxycarbonyl, hydroxycarbonyl, aryloxycarbonyl, benzofused heterocycloalkoxy, benzofused cycloalkylcarbonyl, heterocycloalkylcarbonyl, and a cycloalkylcarbonyl group, carbonylamino

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wherein the carbonylamino nitrogen is (i) 15 unsubstituted, or (ii) is the reacted amine of an amino acid, or (iii) substituted with one or two radicals selected from the group consisting of an alkyl, hydroxyalkyl, hydroxyheteroaralkyl, 20 cycloalkyl, aralkyl, trifluoromethylalkyl, heterocycloalkyl, benzofused heterocycloalkyl, benzofused heterocycloalkyl, benzofused cycloalkyl, and an N,N-dialkylsubstituted alkylamino-alkyl group, or (iv) the carboxamido 25 nitrogen and two substituents bonded thereto together form a 5- to 8-membered heterocyclo, heteroaryl or benzofused heterocycloalkyl ring that is itself unsubstituted or substituted with one or two radicals independently selected from 30 the group consisting of an alkyl, alkoxycarbonyl, nitro, heterocycloalkyl, hydroxy, hydroxycarbonyl, aryl, aralkyl, heteroaralkyl and an amino group,

wherein the amino nitrogen is (i) unsubstituted, or (ii) substituted with one or two substituents that are independently selected from the group

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consisting of alkyl, aryl, and heteroaryl, or (iii) wherein the amino nitrogen and two substituents attached thereto form a 5- to 8-membered heterocyclo or heteroaryl ring,

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- and an aminoalkyl group
 wherein the aminoalkyl nitrogen is (i) unsubstituted,
 or (ii) substituted with one or two substituents
 independently selected from the group consisting of
 an alkyl, aryl, aralkyl, cycloalkyl,
- aralkoxycarbonyl, alkoxycarbonyl, and an alkanoyl group, or (iii) wherein the aminoalkyl nitrogen and two substituents attached thereto form a 5- to 8-membered heterocyclo or heteroaryl ring.

The term "aralkyl", alone or in

combination, means an alkyl radical as defined above in which one hydrogen atom is replaced by an aryl radical as defined above, such as benzyl, 2-phenylethyl and the like.

The term "aralkoxycarbonyl", alone or in combination, means a radical of the formula aralkyl-O-C(O)- in which the term "aralkyl" has the significance given above. An example of an aralkoxycarbonyl radical is benzyloxycarbonyl.

The term "aryloxy" means a radical of the

25 formula aryl-O- in which the term aryl has the
significance given above. The phenoxy radical is an
exemplary aryloxy radical.

The terms "heteroaralkyl" and
"heteroaryloxy" mean radicals structurally similar to
aralkyl and aryloxy that are formed from heteroaryl
radicals. Exemplary radicals include 4-picolinyl and
2-pyrimidinoxy, respectively.

The terms "alkanoyl" or "alkylcarbonyl", alone or in combination, means an acyl radical derived from an alkanecarboxylic acid, examples of

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naphthoyl.

which include formyl, acetyl, propionyl, butyryl, valeryl, 4-methylvaleryl, and the like.

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The term "cycloalkylcarbonyl" means an acyl group derived from a monocyclic or bridged cycloalkanecarboxylic acid such as cyclopropanecarbonyl, cyclohexanecarbonyl, adamantanecarbonyl, and the like, or from a benz-fused monocyclic cycloalkanecarboxylic acid that is optionally substituted by, for example, alkanoylamino, such as 1,2,3,4-tetrahydro-2-naphthoyl, 2-acetamido-1,2,3,4-tetrahydro-2-

The terms "aralkanoyl" or "aralkylcarbonyl" mean an acyl radical derived from an aryl-substituted alkanecarboxylic acid such as phenylacetyl, 3-phenylpropionyl (hydrocinnamoyl), 4-phenylbutyryl, (2-naphthyl)acetyl, 4-chlorohydrocinnamoyl, 4-aminohydrocinnamoyl, 4-methoxyhydrocinnamoyl and the like.

The terms "aroyl" or "arylcarbonyl" means an acyl radical derived from an aromatic carboxylic acid. Examples of such radicals include aromatic carboxylic acids, an optionally substituted benzoic or naphthoic acid such as benzoyl, 4-chlorobenzoyl, 4-carboxybenzoyl, 4-(benzyloxycarbonyl)benzoyl, 1-naphthoyl, 2-naphthoyl, 6-carboxy-2 naphthoyl, 6-(benzyloxycarbonyl)-2-naphthoyl, 3-benzyloxy-2-naphthoyl, 3-hydroxy-2-naphthoyl, and the like.

The term "cycloalkylalkoxycarbonyl" means an acyl group of the formula cycloalkylalkyl-O-CO-wherein cycloalkylalkyl has the significance given above. The term "aryloxyalkanoyl" means an acyl radical of the formula aryl-O-alkanoyl wherein aryl and alkanoyl have the significance given above. The term "heterocyclooxycarbonyl" means an acyl group

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having the formula heterocyclo-O-CO- wherein heterocyclo is as defined above.

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The term "heterocycloalkanoyl" is an acyl radical of the formula heterocyclo-substituted alkane carboxylic acid wherein heterocyclo has the significance given above. The term "heterocycloalkoxycarbonyl" means an acyl radical of the formula heterocyclo-substituted alkane-O-COwherein heterocyclo has the significance given above. The term "heteroaryloxycarbonyl" means an acyl radical represented by the formula heteroary1-0-COwherein heteroaryl has the significance given above.

The term "aminocarbonyl" (carboxamide) alone or in combination, means an amino-substituted carbonyl (carbamoyl) group derived from an amine reacted with a carboxylic acid wherein the amino (amido nitrogen) group is unsubstituted (-NH2) or a substituted primary or secondary amino group containing one or two substituents selected from the group consisting of hydrogen, alkyl, aryl, aralkyl, cycloalkyl, cycloalkylalkyl radicals and the like, as recited. A hydroxamate is a N-hydroxycarboxamide.

The term "aminoalkanoyl" means an acyl group derived from an amino-substituted alkanecarboxylic acid wherein the amino group can be a primary or secondary amino group containing substituents independently selected from hydrogen, alkyl, aryl, aralkyl, cycloalkyl, cycloalkylalkyl radicals and the like.

The term "halogen" means fluoride, chloride, bromide or iodide. The term "haloalkyl" means an alkyl radical having the significance as defined above wherein one or more hydrogens are replaced with a halogen. Examples of such haloalkyl 35 radicals include chloromethyl, 1-bromoethyl, fluoromethyl, difluoromethyl, trifluoromethyl, 1,1,1-trifluoroethyl and the like.

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The term "perfluoroalkyl" means an alkyl group wherein each hydrogen has been replaced by a fluorine atom. Examples of such perfluoroalkyl groups, in addition to trifluoromethyl above, are perfluorobutyl, perfluoroisopropyl, perfluorododecyl and perfluorodecyl.

The term "perfluoroalkoxy" alone or in combination, means a perfluoroalkyl ether radical wherein the term perfluoroalkyl is as defined above. Examples of such perfluoroalkoxy groups, in addition to trifluoromethoxy (F₃C-O-), are perfluorobutoxy, perfluoroisopropoxy, perfluorododecoxy and perfluorodecoxy.

The term "perfluoroalkylthio" alone or in

combination, means a perfluoroalkyl thioether radical wherein the term perfluoroalkyl is as defined above.

Examples of such perfluoroalkylthio groups, in addition to trifluoromethylthio (F₃C-S-), are perfluorobutylthio, perfluoroisopropylthio,

perfluorododecylthio and perfluorodecylthio.

The term "aromatic ring" in combinations such as substituted-aromatic ring sulfone or substituted-aromatic ring sulfoxide means aryl or heteroaryl as defined before.

The term "pharmaceutically acceptable" is used adjectivally herein to mean that the modified noun is appropriate for use in a pharmaceutical product. Pharmaceutically acceptable cations include metallic ions and organic ions. More preferred metallic ions include, but are not limited to appropriate alkali metal (Group Ia) salts, alkaline earth metal (Group IIa) salts and other physiological acceptable metal ions. Exemplary ions include aluminum, calcium, lithium, magnesium, potassium, sodium and zinc in their usual valences. Preferred

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organic ions include protonated tertiary amines and quaternary ammonium cations, including in part, trimethylamine, diethylamine, N,N'dibenzylethylenediamine, chloroprocaine, choline, diethanolamine, ethylenediamine, meglumine (Nmethylglucamine) and procaine. Exemplary pharmaceutically acceptable acids include without limitation hydrochloric acid, hydrobromic acid, phosphoric acid, sulfuric acid, methanesulfonic acid, acetic acid, formic acid, tartaric acid, maleic acid, 10 malic acid, citric acid, isocitric acid, succinic acid, lactic acid, gluconic acid, glucuronic acid, pyruvic acid oxalacetic acid, fumaric acid, propionic acid, aspartic acid, glutamic acid, benzoic acid, and the like. 15

"M" utilized in the reaction schemes that follow represents a leaving group such as halogen, phosphate ester or sulfate ester.

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Preparation of Useful Compounds

Schemes A through C and Schemes 1 through 19 hereinbelow illustrate chemical processes and transformations that can be useful for the preparation of compounds useful in this invention; i.e., compounds of formulas I, II, III, IV and V and similar cyclic inhibitors. In addition, the preparation of compounds of formula VI and formula VII is illustrated. Compounds of formula VI and formula VII can be used as intermediates in the preparation of the compounds of formulas I, II, III, IV and V or pro-drugs or MMP inhibitors.

In Schemes A through C, the symbol J

independently represents R²⁰ or other synthetically

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useful groups such as amides, acid chlorides, mixed anhydrides and the like. The n is 0, 1 or 2 and is preferred to be 1 or 2 in Scheme C. The n of these schemes corresponds to q in formulas VI and VII., and is zero, 1 or 2. The symbol m is 1 or 2. The symbol r is independently 1, 2 or 3. The symbol P represents a protecting group that can also be a member of the group R⁶. In Scheme A, for simplicity and clarity of illustration positional isomers are 10 illustrated with a bond through the ring in standard fashion. Later Schemes typically only show one positional isomer but positional isomers are represented by these structures and reactions in a manner consistent with Formula I, II, III, IV, V, VI, VII above. Similarly, the symbol B represents O, S, 15 SO, SO₂ and NR⁶. The symbols C and C' independently are electrophilic groups or groups capable of participating in a condensation reaction. Here to it should be noted that the six-membered ring is shown for illustrative purposes but the procedures and/or 20 reagents are applicable to and represent combinations the permit the preparation of 5- to 8-membered rings.

The structures in Schemes 1 through 19 are also shown with compounds that represent the other compounds of this invention. The aromatic ring in Scheme C is aryl and heteroaryl. The moieties of -A-R-E-Y are as defined before. Reactions illustrated involving a spiroheterocyclic nitrogen atom may not be applicable to those compounds with sulfur or oxygen.

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Scheme A

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Scheme A shows in step 1 the reduction of a heteraryl compound to a carboxyl derivative. Generally, the first product is a hydrogen-containing amine heterocycle when the starting material is aromatic or an R⁶-containing heterocycle when a partially unsaturated heterocycle is the starting material.

Compound 2 can be treated in several ways depending on the needs of the chemist. In Step 2, the nitrogen can be protected by preparing, for 10 example, a carbobenzoxy (Z) or tert-butoxycarbonyl derivative. Such acylations can be carried out by methods well known in the art, especially the art of amino acid and peptide synthesis. The process of acylation with activated carboxyl group- or activated 15 sulfonyl group-containing reagents to prepare contemplated compounds is carried out in the same manner. Examples of such acylating groups are carbonyl azides, halides, anhydrides, mixed anhydrides, carbodiimide derivatives or other less 20 traditional activated ester groups such as the hydroxybenzotriazole derivative. These acylations can be run in the presence of base including mild bases such as triethylamine or N-ethylmorpholine if desired. The preparation of some activated ester 25 reagents and their use to prepare other compounds useful in this invention is discussed below. should be recalled that the groups constituting P and serving as a selectively removable protecting group can also be included as part of the group R6. 30

Step 4 of Scheme A shows the alkylation or acylation of Compound 2 to produce compound 5. The

process of acylation and alkylation are as discussed herein. In Step 5, the group J can be changed if desired. An example of such a change is exchange of an ester for a THP-protected hydroxamate conversion of a THP-protected hydroxamate inot a hydroxamate or conversion of an acid into a protected hydroxamate or the like.

Steps 3, 7 and 8 show the preparation of sulfur-containing derivatives of the contemplated compounds or intermediates to those compounds. 10 starting material for the above steps (e.g., compounds 2, 5 and 6) can be treated with a base to deprotonate the carbon alpha to the carbonyl function. This anion can be reacted with a sulfur electrophile to produce a sulfone, sulfoxide or 15 Such electrophiles can be of the form of, sulfide. for example, $R^{24}S-SR^{24}$, $R^{24}SO_2C_1$, $R^{24}SC_1$, $R^{24}SOC_1$, $R^{24}S(0)-SR_{24}$ and the like where R^{24} is as defined before or is an aryl or heteroaryl sulfur-containing material containing a coupling substituent, R3', that 20 can be used to prepare one of the R²⁴-containing groups. Preparation of the anion requires a base and a strong base may be required such as one of the metal amides, hydrides or alkyls discussed herein. 25 The solvents are nonprotic, and dipolar aprotic solvents are preferred along with an inert atmosphere. Subsequent schemes usually utilize R3 for the R^{24} group for ease of illustration.

It should be noted that these processes 30 produce sulfides (thio ethers), sulfoxides or sulfones depending on starting material. In

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addition, the sulfides can be oxidized to sulfoxides or sulfones, and the sulfoxides can be oxidized to their corresponding sulfone derivatives. The choice of position in the synthetic sequence to change the oxidation state of sulfur as well as the decision to change oxidation state is under the control of the chemist skilled in the art. Methods of oxidizing sulfur are discussed hereinbelow.

Scheme A, Steps 6, 9, 10 and 12 10 independently illustrate the interconversion of groups within J. Examples of such interconversions include exchange of an ester for hydroxamic acid or hydroxamic acid derivative, conversion of a carboxylic acid into an activated carbonyl derivative or into a hydroxamic acid or hydroxamic acid 15 derivative(pro-drug or protected derivative), or removal of a protecting group from a hydroxamate derivative. The preparation of activated carbonyl compounds their reaction with nucleophiles such as hydroxamic acid, protected hydroxamates or hydroxamic 20 acid pro-drugs is discussed below as is the conversion of protected hydroxamic acid derivatives into hydroxamic acids. The preparation of, for example, hydroxybenzotriazole/carbodiimide, derived products is discussed herein. The preparation or 25 hydrolysis of esters, amides, amide derivatives, acid chlorides, acid anhydrides, mixed anhydrides and the like are synthetic methods very well known in the art, andare not discussed in detail herein. Step 6 illustrates the conversion of compound 4 into 30 compound 9, without first being converted into compound 7.

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Scheme B

Scheme B illustrates an alternate method of preparing contemplated compounds. The reagent shown above the arrow in Step 1 is a reagent with two

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active groups in addition to the heteroatoms (B) noted before. Here again, the particular reagent illustrated was selected to permit a clear illustration of the reaction, but it is also intended to represent reagents that permit the preparation of the heteroatom position, and 5-, 7- and 8-membered ring size compounds. These reagents are readily selected by those skilled in the art.

C and C' in this Step 1 reagent are independently an electophile or a group convertible 10 into an electrophile. Such groups include halides, sulfonic acid esters, epoxides, thioepoxides, hydroxyl groups, and the like. This reagent is reacted with a nucleophilic anion of a sulfur containing carbonyl compound such as compound 1. - 15 anion is formed by deprotonation of compound 1 and examples of bases suitable for such a deprotonation are discussed below. Treatment with the above electrophilic reagent is carried out under alkylating conditions well known in the art and discussed 20 herein. The product of this reaction can be either Compound 2 or Compound 3; i.e., the reaction can be carried out as a pot or two step process as required.

Step 3 illustrates the interconversion of J
groups if desired as discussed above for Scheme A.
Step 4 uses reagent where C, for example, represents
a nucleophile as discussed above and C' represents an
electrophile or a nucleophile such as hydroxyl, thiol
or R⁶-amino. It is noted that C' can be,
independently, a nucleophile or an electrophile when
m is 2; i.e., the C' groups are not required to be
the same when m is 2. When m is 2, treatment with a
second mole of base provides the skilled chemist an

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alternative preparation of Compound 5. When C' is hydroxyl, thiol, or R⁶-amino and m is 2, the person skilled in the art can condense Compound 4 with, for example, an aldehyde or ketone, under reductive conditions or with subsequent reduction to form a contemplated compound. As above, the compound where m is 2 can be made in one step (one pot process) or two steps, thus permitting the chemist the choice of having the reagent(s) be the same (one pot) or different (two step).

Scheme B also illustrates the interconversions of the groups within J, the oxidation state of the sulfur and groups on nitrogen; i.e., R⁶ groups, to provide the contemplated compounds. These methods and processes are discussed above for the reactions of Scheme A.

Scheme C

Scheme C illustrates the nucleophilic displacement of a group D as defined herein. This reaction is carried out in a similar manner to the displacement reactions discussed herein. The choice of oxidation state of the sulfur is made by the person skilled in the art, but sulfoxide or sulfone groups are preferred, and the sulfone is most preferred. The displacement can be carried out either before or after the methylene next to the carbonyl group is reacted to form a spiro heterocyclic group.

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Steps 1, 2 and 3 also illustrate that although the nucleophilic displacement can be carried out with one nucleophile (Nu), the product of this reaction can be modified by methods well known in the art and as shown herein to provide the group -A-R-E-Y as defined hereinbefore.

A non-limiting illustration of such a

20 process is provided when D is fluoride. The fluoride
leaving group can be directly displaced with the
anion of 4-trifluoromethylphenol, 4trifluoromethoxyphenol, 4-trifluoromethylthiophenol
and the like to provide a contemplated compound.

25 This is a one pot process from Compound 4. Other
compounds included in -A-R-E-Y can be prepared by
displacing the fluoride leaving group with ammonia to
provide an amine, which can then be acylated by
methods discussed wherein with, for example, 4
30 trifluoromethylbenzoyl chloride, to form another
contemplated product compound.

The R⁶ function can be changed and/or further modified in compounds or at steps in the Schemes as desired or required by the person skilled in the art to prepare the contemplated compounds. Interconversion of dual purpose functional groups 5 such as short or long term protecting groups into other R⁶ groups has been mentioned. Many other routine and/or useful conversions, including the preparation of synthetic intermediates, are very well known in the art. A few non-limiting examples of 10 such conversions or reactions include: reductions; nucleophilic displacement/substitution reactions; exchange or preparation of carboxylic or sulfonic acids, amides, esters, acid halides, mixed anhydrides and the like; electrophilic displacement/substitution reactions; oxidations; ring/chain conversions, ring opening reactions, condensation reactions, including those involving sulfonyl or carbonyl groups and/or carbon-hydrogen bonds influenced by either or both of those groups. The selection of preparative methods 20 or conversion methods of the contemplated compounds and the order of the reaction(s) is made by the skilled person. It is expected that should a particular sequence or method prove to be undesirable that an alternative will be selected and used. 25 Included is the choice of preparing/adding the groups in a single step using a convergent inhibitor strategy or preparing the final R⁶ group following a stepwise strategy.

Thus, in general, the choices of starting material and reaction conditions can vary as is well known to those skilled in the art. Usually, no

single set of conditions is limiting because variations can be applied as required. Conditions are also selected as desired to suit a specific purpose such as small scale preparations or large scale preparations. In either case, the use of less safe or less environmentally sound materials or reagents is usually be minimized. Examples of such materials are diazomethane, diethyl ether, heavy metal salts, dimethyl sulfide, chloroform, benzene and the like.

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These reactions can be carried out under a dry inert atmosphere such a nitrogen or argon if desired. Selected reactions known to those skilled in the art, can be carried out under a dry atmosphere such as dry air whereas other synthetic steps, for 15 example, aqueous acid or base ester or amide hydrolysis, can be carried out under laboratory air. In addition, some processes of these syntheses can be carried out in a pressure apparatus at pressures above, equal to or below atmospheric pressure. The 20 use of such an apparatus aids in the control of gaseous reagents such as hydrogen, ammonia, trimethylamine, methylamine, oxygen and the like, and can also help prevent the leakage of air or humidity into a reaction in progress. This discussion is not 25 intended to be exhaustive as it is readily noted that additional or alternative methods, conditions, reactions or systems can be identified and used by a chemist of ordinary skill.

30 The illustrated reactions are usually carried out at a temperature of between -25°C to solvent reflux under an inert atmosphere such as nitrogen or argon. The solvent or solvent mixture

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can vary widely depending upon reagents and other conditions and can include polar or dipolar aprotic solvents as listed or mixtures of these solvents. Reactions can be carried out at lower temperatures such as dry ice/acetone or liquid nitrogen temperature if desired to carry out such reactions as metalations or anion formations using strong bases.

In some cases, amines such as triethylamine, pyridine or other non-reactive bases can serve as reagents and/or solvents and/or co-10 solvents. In some instances, in these reactions and other reactions in these Schemes, protecting groups can be used to maintain or retain groups in other parts of a molecule(s) at locations that is(are) not desired reactive centers. Examples of such groups 15 that the skilled person can maintain or retain include, amines, other hydroxyls, thiols, acids and the like. Such protecting groups can include acyl groups, arylalkyl groups, carbamoyl groups, ethers, alkoxyalkyl ethers, cycloalkyloxy ethers, arylalkyl 20 groups, silyl groups including trisubstituted silyl groups, ester groups and the like. Examples of such protecting groups include acetyl, trifluoroacetyl, tetrahydropyran (THP), benzyl, tert-butoxy carbonyl 25 (BOC or TBOC), benzyloxycarbonyl (Z or CBZ), tertbutyldimethylsilyl (TBDMS) or methoxyethoxymethylene (MEM) groups. The preparation of such protected compounds as well as their removal is well known in the art. The protecting groups can also be used as 30 substituents in the contemplated compounds whose utility is as a drug rather than as a synthetic intermediate.

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Many reactions or processes involve bases that can act as reactants, reagents, deprotonating agents, acid scavengers, salt forming reagents, solvents, co-solvents and the like. Bases that can be used include, for example, metal hydroxides such 5 as sodium, potassium, lithium, cesium or magnesium hydroxide, oxides such as those of sodium, potassium, lithium, calcium or magnesium, metal carbonates such as those of sodium, potassium, lithium, cesium, calcium or magnesium, metal bicarbonates such as 10 sodium bicarbonate or potassium bicarbonate, primary (I°), secondary (II°) or tertiary (III°) organic amines such as alkyl amines, arylalkyl amines, alkylarylalkyl amines, heterocyclic amines or heteroaryl amines, ammonium hydroxides or quaternary 15 ammonium hydroxides. As non-limiting examples, such amines can include triethylamine, trimethylamine, diisopropylamine, methyldiisopropylamine, diazabicyclononane, tribenzylamine, dimethylbenzylamine, morpholine, N-methylmorpholine, 20 N,N'-dimethylpiperazine, N-ethylpiperidine, 1,1,5,5tetramethylpiperidine, dimethylaminopyridine, pyridine, quinoline, tetramethylethylenediamine, and the like. Non-limiting examples of ammonium hydroxides, usually made from amines and water, can 25 include ammonium hydroxide, triethylammonium hydroxide, trimethylammonium hydroxide, methyldiiospropylammonium hydroxide, tribenzylammonium hydroxide, dimethylbenzylammonium 30 hydroxide, morpholinium hydroxide, Nmethylmorpholinium hydroxide, N,N'dimethylpiperazinium hydroxide, N-ethylpiperidinium

hydroxide, and the like. As non-limiting examples,

quaternary ammonium hydroxides can include tetraethylammonium hydroxide, tetramethylammonium hydroxide, dimethyldiiospropyl-ammonium hydroxide, benzylmethyldiisopropylammonium hydroxide, methyldiazabicyclononylammonium hydroxide, methyltribenzylammonium hydroxide, N,N-dimethylmorpholiniumhydroxide, N,N,N',N'-tetramethylpiperazinium hydroxide, and N-ethyl-N'-hexylpiperidinium hydroxide and the like.

10 Metal hydrides, amides or alcoholates such as calcium hydride, sodium hydride, potassium hydride, lithium hydride, aluminum hydride, diisobutylaluminum hydride (DIBAL) sodium methoxide, potassium tert-butoxide, calcium ethoxide, magnesium ethoxide, sodium amide, potassium diisopropyl amide 15 and the like can also be suitable reagents. Organometallic deprotonating agents such as alkyl or aryl lithium reagents such as methyl lithium, phenyl lithium, tert-butyl lithium, lithium acetylide or butyl lithium, Grignard reagents such as 20 methylmagnesium bromide or methymagnesium chloride, organocadmium reagents such as dimethylcadmium and the like can also serve as bases for causing salt formation or catalyzing the reaction. Quaternary 25 ammonium hydroxides or mixed salts are also useful for aiding phase transfer couplings or serving as phase transfer reagents. Pharmaceutically acceptable bases can be reacted with acids to form contemplated pharmaceutically acceptable salts. It should also be 30 noted that optically active bases can be used to make optically active salts which can be used for optical resolutions.

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Generally, reaction media can comprise a single solvent, mixed solvents of the same or different classes or serve as a reagent in a single or mixed solvent system. The solvents can be protic, 5 non-protic or dipolar aprotic. Non-limiting examples of protic solvents include water, methanol (MeOH), denatured or pure 95% or absolute ethanol, isopropanol and the like. Typical non-protic solvents include acetone, tetrahydrofuran (THF), 10 dioxane, diethyl ether, tert-butylmethyl ether (TBME), aromatics such as xylene, toluene, or benzene, ethyl acetate, methyl acetate, butyl acetate, trichloroethane, methylene chloride, ethylenedichloride (EDC), hexane, heptane, isooctane, 15 cyclohexane and the like. Dipolar aprotic solvents include compounds such as dimethylformamide (DMF), dimethylacetamide (DMAc), acetonitrile, DMSO, hexamethylphosphorus triamide (HMPA), nitromethane, tetramethylurea, N-methylpyrrolidone and the like. 20 Non-limiting examples of reagents that can be used as solvents or as part of a mixed solvent system include organic or inorganic mono- or multi-protic acids or bases such as hydrochloric acid, phosphoric acid, sulfuric acid, acetic acid, formic acid, citric acid, succinic acid, triethylamine, morpholine, N-25 methylmorpholine, piperidine, pyrazine, piperazine, pyridine, potassium hydroxide, sodium hydroxide, alcohols or amines for making esters or amides or thiols for making contemplated products and the like.

The preparation of compounds contemplated herein can require the oxidation of nitrogen or sulfur to N-oxide derivatives or sulfoxides or sulfones. Reagents for this process can include, in

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a non-limiting example, peroxymonosulfate (OXONE®), hydrogen peroxide, meta-chloroperbenzoic acid, perbenzoic acid, perbenzoic acid, perbenzoic acid, peracetic acid, perlactic acid, tert-butyl peroxide, tert-butyl hypochlorite, sodium hydrochlorite, hypochlorous acid, sodium meta-periodate, periodic acid and the like with the weaker agents being most useful for the preparation of sulfones and sulfoxides. Protic, non-protic, dipolar aprotic solvents, either pure or mixed, can be chosen, for example, methanol/water.

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The oxidation can be carried out at temperature of about -78° to about 50° degrees Centigrade, and normally selected from a range -10°C to about 40°C. Sulfoxides are best prepared using one equivalent of oxidizing agent. It can be desirable in the case of more active oxidizing agents, but not required, that the reactions be carried out under an inert gas atmosphere with or without degassed solvents. It should be noted that the oxidation of sulfides to sulfones can be carried out in one step or two steps via the sulfoxide as desired by the chemist.

Reduction is a well known process in the art with a useful method being hydrogenation. In

25 such cases (catalytic reduction), there can be a metal catalyst such as Rh, Pd, Pt, Ni or the like with or without an additional support such as carbon, barium carbonate and the like. Solvents can be protic or non-protic pure solvents or mixed solvents

30 as required. The reductions can be carried out at atmospheric pressure to a pressure of multiple atmospheres with atmospheric pressure to about 40 pounds per square inch (psi) preferred or very high

pressures in special hydrogenation equipment well known in the art.

Reductive alkylation of amines or active methylene compounds is also a useful method of

5 preparing compounds. Such alkylations can be carried out under reductive hydrogenation conditions as presented above using, for example, aldehydes or ketones. Hydride transfer reagents such as sodium cyanoborohydride, aluminum hydride, lithium

10 aluminumhydride, borane, sodium borohydride, diisobutylaluminum hydride and the like are also useful as reagents for reductive alkylation. Acyl groups can be reduced in a similar manner to produce substituted amines.

Alternative methods of alkylating carbon or nitrogen are direct alkylation. Such an alkylation, as is well known in the art, can be carried by treatment of an activated carbon containing at least one hydrogen with base to form the corresponding anion, adding an electrophilic reagent and permitting the SN2 reaction to proceed. An amine to be alkylated is treated similarly except that deprotonation may not be required. Electrophiles include halogen derivatives, sulfonate esters, epoxides and the like.

Bases and solvents for alkylation reactions are those discussed above. Preferred are bases that are hindered such that competition with the electrophile is minimized. Additional preferred bases are metal hydrides, amide anions or organometallic bases such as n-butyl lithium. The solvents, solvent mixtures or solvent/reagent mixtures discussed are satisfactory but non-protic or

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dipolar aprotic solvents such as acetone, acetonitrile, DMF and the like are examples of preferred classes.

Acids are used in many reactions during 5 various syntheses. For example, removal of the THP protecting group to produce the hydroxamic acid. acid can be a mono-, di- or tri-protic organic or inorganic acid. Examples of acids include hydrochloric acid, phosphoric acid, sulfuric acid, 10 acetic acid, formic acid, citric acid, succinic acid, hydrobromic acid, hydrofluoric acid, carbonic acid, phosphorus acid, p-toluene sulfonic acid, trifluoromethane sulfonic acid, trifluoroacetic acid, difluoroacetic acid, benzoic acid, methane sulfonic acid, benzene sulfonic acid, 2,6-dimethylbenzene 15 sulfonic acid, trichloroacetic acid, nitrobenzoic acid, dinitrobenzoic acid, trinitrobenzoic acid, and the like. They can also be Lewis acids such as aluminum chloride, borontrifluoride, antimony pentafluoride and the like. Acids in a protic can 20 also be used to hydrolyze esters, amides and the like as well as catalyze exchange reactions.

Conversion of a carboxylic acid protected as an ester or amide into a hydroxamic acid or hydroxamic acid derivative such as an O-arylalkylether or O-cycloalkoxyalkylether group is useful. In the case where hydroxylamine is used, treatment of an ester or amide with one or more equivalents of hydroxylamine hydrochloride at room temperature or above in a solvent or solvents, usually protic or partially protic, such as those listed above can provide a hydroxamic acid directly. This exchange process can be further catalyzed by the

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addition of additional acid. Alternatively, a base such as a salt of an alcohol used as a solvent, for example, sodium methoxide in methanol, can be used to form hydroxylamine from hydroxylamine hydrochloride in situ which can exchange with an ester or amide. As mentioned above, exchange can be carried out with a protected hydroxyl amine such as tetrahydropyranylhydroxyamine (THPONH2), benzylhydroxylamine (BnONH2), and the like in which 10 case compounds such as shown in Schemes A, B and C that are tetrahydropyranyl (THP) or benzyl (Bn) hydroxamic acid derivatives are the products. Removal of the protecting groups when desired, for example, following further transformations in another part of the molecule or following storage, is 15 accomplished by standard methods well known in the art such as acid hydrolysis of the THP group as discussed above or reductive removal of the benzyl group with hydrogen and a metal catalyst such as palladium, platinum, palladium on carbon or nickel. 20

In the case where R²⁰ is hydroxyl; i.e.,
where the intermediate is a carboxylic acid, standard
coupling reactions can be used. For example, the
acid can be converted into an acid chloride, mixed

25 anhydride or activated ester such as
hydroxybenzotriazole and treated with hydroxylamine
or a protected hydroxylamine in the presence of a
non-competitive base to the nitrogen acylated
compound. This is the same product as discussed

30 above. Couplings of this nature are well known in
the art and especially the art related to peptide and
amino acid chemistry.

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An amide of this invention, whether used as a drug or as a protecting group, is prepared by treatment of an acid halide, anhydride, mixed anhydride or active ester with a primary amine, secondary amine or ammonia, or their equivalent. These standard coupling reactions are well known in the art and are discussed elsewhere herein. An alternative method of preparation of amides is by the exchange of, for example, an alkoxycarbonyl (ester) or aminecarbonyl (amide) group for an amine or different amine as required. Ester exchange processes are especially useful when less hindered amines, including ammonia, are used to make the corresponding amides of this invention.

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15 Further, amides can be prepared from hydroxamic acids or protected hydroxamic acid compounds by catalytic reductions or in vivo or in vitro enzymatic processes. For example, catalytic reduction of O-benzylhydroxamic acid compounds is 20 known to produce varying ratios of amide and hydroxamic acid depending upon the catalyst used as well as other reaction conditions such as solvent, temperature, hydrogen gas pressure and the like.

Compounds contemplated herein can possess one or more asymmetric carbon atoms and are thus capable of existing in the form of optical isomers, enantiomers, diastereoisomers, as well as in the form of racemic or nonracemic mixtures. A compound can also exist in other isomeric forms such as ortho, meta and para isomers, cis and trans isomers, syn and anti isomers, E and Z isomers, tautomeric isomers, alpha and beta isomers, axial and equatorial isomers and isomers due to hindered rotation. An isomer can

-119-

exist in equilibrium with another isomer in a mammal or a test system. Such a compound can also exist as an isomeric equilibrium system with a solvent or water, for example, as a hydrated ketone or aldehyde, as is well known in the art. All isomers are included as compounds of this invention.

The chemical reactions described above are generally disclosed in terms of their broadest application to the preparation of the compounds of this invention. Occasionally, the reactions may not 10 be applicable as described to each compound included within the disclosed scope. The compounds for which this occurs will be readily recognized by those skilled in the art. In all such cases, either the reactions can be successfully performed by 15 conventional modifications known to those skilled in the art, e.g., by appropriate protection of interfering groups, by changing to alternative conventional reagents, by routine modification of reaction conditions, and the like, or other reactions 20 disclosed herein or otherwise conventional, are applicable to the preparation of the corresponding compounds that are contemplated.

5

Scheme 1

$$\begin{array}{c} \text{MeO} \\ 1 \\ \end{array} \begin{array}{c} \text{HS} \\ \\ \end{array} \begin{array}{c} \text{HS} \\ \\ \end{array} \begin{array}{c} \text{CH}_2\text{O}/\text{KHCO}_3 \\ \end{array} \begin{array}{c} \text{OHOH} \\ \\ \end{array} \begin{array}{c} \text{OHOH} \\ \end{array} \begin{array}{c} \text{RCOR}/\text{H}^+ \\ \end{array} \begin{array}{c} \text{MeO} \\ \\ \end{array} \begin{array}{c} \text{RR} \\ \end{array} \\ \end{array} \begin{array}{c} \text{OHOH} \\ \end{array} \begin{array}{c} \text{RR} \\ \text{OHOH} \\ \end{array} \begin{array}{c} \text{OHOH} \\ \end{array} \begin{array}{c} \text{RR} \\ \text{OHOH} \\ \end{array} \begin{array}{c} \text{OHOH} \\ \end{array} \begin{array}{c} \text{RR} \\ \text{OHOH} \\ \end{array} \begin{array}{c} \text{OHOH} \\ \end{array} \begin{array}{c} \text{OHOH} \\ \end{array} \begin{array}{c} \text{RR} \\ \text{OHOH} \\ \end{array} \begin{array}{c} \text{OHOH$$

$$\begin{array}{c} \text{MeO}_2\text{C} \\ \text{Br} \end{array}$$

$$\begin{array}{c} \text{MeO}_2\text{C} \\ \text{SH} \end{array}$$

$$\begin{array}{c} \text{MeO}_2\text{C} \\ \text{SH} \end{array}$$

$$\begin{array}{c} \text{MeO}_2\text{C} \\ \text{OH} \end{array}$$

$$\begin{array}{c} \text{OH} \\ \text{OH} \end{array}$$

$$\begin{array}{c} \text{OH} \\ \text{OH} \end{array}$$

$$\begin{array}{c} \text{OH} \\ \text{OH} \end{array}$$

$$\begin{array}{c} \text{MeO}_2\text{C} \\ \text{OH} \end{array}$$

$$\begin{array}{c} \text{OH} \\ \text{OH} \end{array}$$

$$\begin{array}{c} \text{MeO}_2\text{C} \\ \text{OH} \end{array}$$

$$\begin{array}{c} \text{OH} \\ \text{OH} \end{array}$$

$$\begin{array}$$

5

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$$\begin{array}{c|c}
 & 60^{\circ}C \\
\hline
DMSO
\end{array}$$

$$\begin{array}{c|c}
CO_2Et & CO_2Et \\
\hline
N & BOC_2O \\
\hline
3 & BOC \\
\hline
4 & CO_2Et \\
\hline
DA & EtO_2C \\
\hline
S & BOC \\
\hline
BOC \\
5 & BOC
\end{array}$$

In a similar manner, the following analogs can be made.

5

Table 1 through Table 165, below, show several contemplated aromatic sulfone hydroxamic acid inhibitor compounds or structural formulas that illustrate substituent groups. Each group of

compounds is illustrated by a generic formula, or formulae, followed by a series of preferred moieties or groups that constitute various substituents that can be attached at the position clearly shown in the generic structure. The substituent symbols, e.g., R1 and R2 and R3, are as shown in each Table, and are typically not those used before. One or two bonds (wavy lines) are shown with those substituents to indicate the respective positions of attachment in the illustrated compound. This system is well known in the chemical communication arts and is widely used in scientific papers and presentations. For example in Table 2, R1 and R2 together with the atoms to which they are bonded is the variable group with the structural entities that can substitute for R1 and R2 together shown in the balance of that table.

10

15

Table 1

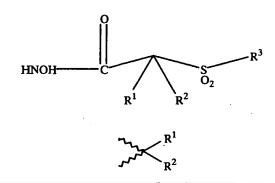


Table 2

HO-HN
$$SO_2$$
 R^3

Table 3

Table 4

Table 5

$$\begin{array}{c} CH_3 \\ O \\ N \\ O \\ N \\ O \\ R^3 \end{array}$$

Table 8

HO
$$R^3$$
 CH_3
 CH_3
 C
 R^3

Table 10

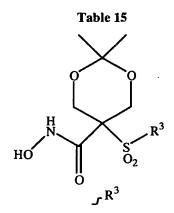
$$HO \longrightarrow S R^3$$

Table 11

HO
$$R^3$$

Table 12

HO
$$R^3$$



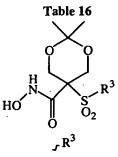


Table 17

Table 18

$$H_3C_{M_{N_0}}$$
 H_1
 H_2
 H_3
 H_3
 H_4
 H_4
 H_5
 H_4
 H_5
 H_5
 H_5
 H_5
 H_5
 H_5
 H_6
 H_7
 H_7

 $_{\mathcal{F}}R^3$

Table 20

Table 22

$$H_3C_{M_{M_n}}$$
 $H_3C_{M_{M_n}}$
 $H_3C_{M_n}$
 $H_3C_{M_$

Table 23
$$H_3C_{I_{M_n}}, \stackrel{H}{\underset{O_2}{\bigvee}} O$$

$$R^3$$

Table 24

$$HO \xrightarrow{H} S \xrightarrow{R^3} R^3$$

 r^{R^3}

Table 25

Table 26

$$\begin{array}{c|c}
H & H \\
H & O \\
O & O_2
\end{array}$$

Table 27

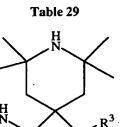
Table 28

$$HO \xrightarrow{H} O_2$$

$$G_2$$

$$G_2$$

$$G_2$$



√R³

$$\begin{cases} 1 & \text{obs} \\ 1 & \text{obs} \\ 2 & \text{obs} \\ 3 & \text{obs} \\ 4 & \text{obs} \\ 4$$

Table 30

Table 31

Table 32

HO
$$R^3$$

Table 36

HO
$$R^3$$

Table 37

$$\begin{array}{c|c} & NH_2 \\ N & NH \\ NH & NH \\$$

$$\begin{array}{c|c}
 & NH_2 \\
 & NH \\
 &$$

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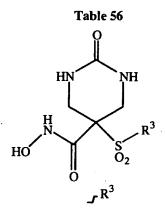
Table 41

Table 44

Table 45

Table 46

Table 49



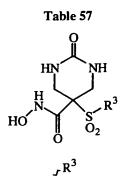


Table 58

$$\begin{array}{c|c} H & O \\ N & O \\ O & O_2 \end{array}$$

Table 59

$$HO = \begin{pmatrix} & & & \\ & & &$$

HO
$$R^3$$

Table 61

Table 62

HO
$$R^3$$

Table 63

HO
$$R^3$$
 R^3

$$2 \qquad \sum_{N=1}^{N} \sum_{j=1}^{N} \sum_{i=1}^{N} \sum_{j=1}^{N} \sum_{j=1}^{N}$$

9
$$S \times N$$

$$4 \qquad \begin{array}{c} & & \\ & \\ & \\ \\ & \\ \\ & \\ \end{array}$$

$$11 \longrightarrow S \longrightarrow O$$

Table 64

$$\begin{array}{c|c} H & & \\ O & & \\ N & & \\ O & & \\ O_2 & & \\ & &$$

Table 65

$$HO \xrightarrow{H} S \xrightarrow{N - CH_2} R^3$$

Table 66

Table 67

Table 69

HO
$$R^3$$
 CH_3 CH_3 R^3

Table 70

HO
$$R^3$$
 R^3

HO
$$R^3$$
 R^3

Table 72

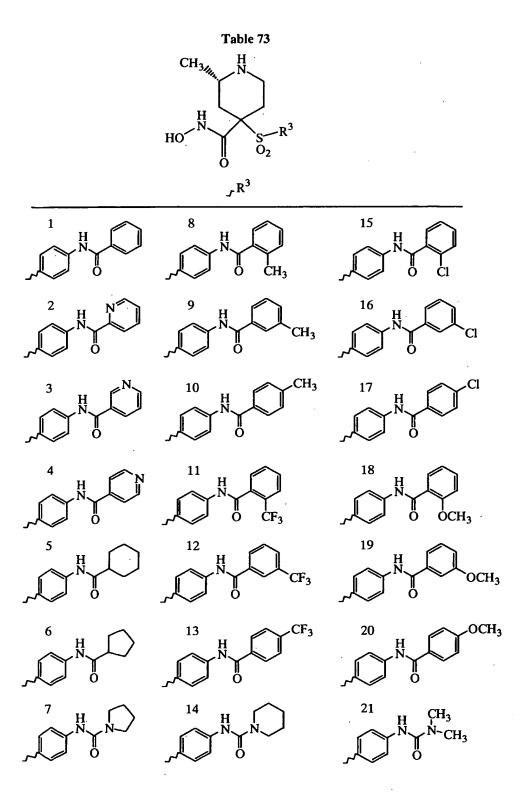


Table 74

$$R^3$$

 $_{\mathcal{I}}R^3$

Table 77

Table 78

$$\begin{array}{c|c} CH_{3_{M_{N_{N_{N}}}}} & H \\ HO & & \\ &$$

Table 79

√R³

Table 80

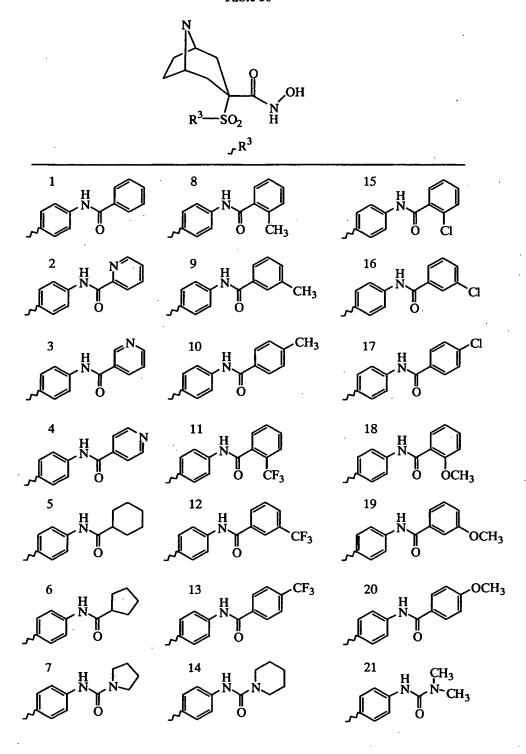


Table 81

$$R^3$$
 SO_2 H OH

1 0 CH₃ 9 Ph 16 S CH₃

2 0 CH₃ 10 17 S CH₃

3 0 CH₃ 11 N 18 S CH₃

4 0 CF₃ 12 N 19 S Ph

5 0 CF₃ 13 N 20 S Ph

7 0 Ph

14 N 21 S N

8 0 Ph

15 S N

22 S N

Table 82

Table 83

$$R^3$$
 O_2S R^3 OH

Table 84

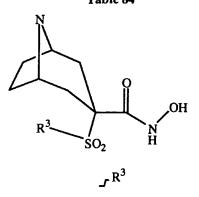


Table 85

$$R^3$$
— SO_2 H N

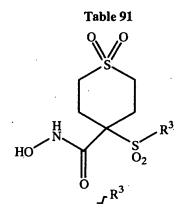
Table 86

$$\begin{array}{c|c}
 & O & O \\
 & S & \\
 & R^3
\end{array}$$

Table 87

Table 88

HO
$$R^3$$



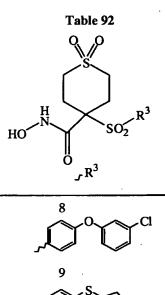


Table 93

$$\begin{array}{c|c}
H & S \\
R^3 & R^3
\end{array}$$

Table 94

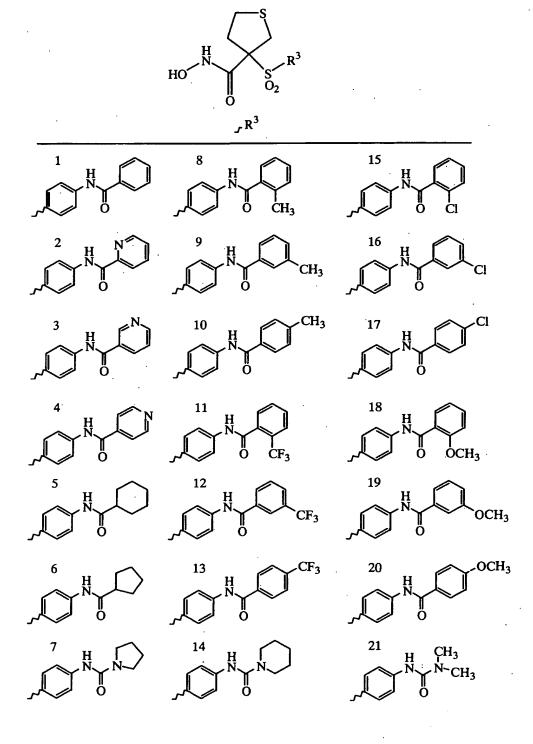
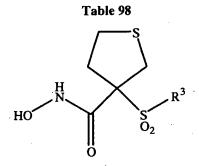


Table 95

$$\begin{array}{c|c}
H & & \\
N & & \\
O & & \\
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 $_{-}R^{3}$



$$_{r}R^{3}$$

Table 99

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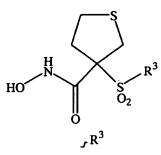


Table 100

HO
$$R^3$$

Table 101

Table 102

$$R^3$$
HO
 R^3
 Ph

Table 104

Table 105

 $\mathcal{L}^{\mathbb{R}^3}$

$$\frac{1}{3}$$
 $\frac{1}{N}$ $\frac{1}{N}$ $\frac{1}{N}$ $\frac{1}{N}$ $\frac{1}{N}$

$$4 \qquad \begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \\ \end{array}$$

Table 106

HO
$$R^3$$

Table 107

Table 108

Table 109

Table 111

Table 112

HO
$$R^3$$

Table 113

Table 114

Table 115

$$\begin{array}{c} CH_3 \\ \downarrow \\ N \\ \downarrow \\ N \\ C \\ S \\ S \\ R^3 \end{array}$$

Table 120

$$HO = \begin{pmatrix} 0 & & & \\ & &$$

Table 121

Table 122

HO
$$R^3$$

Table 124

HO
$$R^3$$

Table 125

HO
$$R^3$$

Table 127

HO
$$R^3$$

Table 128

HO
$$R^3$$

Table 129

Table 130

Table 131

Table 132

Table 133

Table 134

Table 135

Table 137

Table 139

CH₂CH₂—SCF₃

Table 151

Table 153

Table 155

Table 156

Table 157

Table 158

Table 159

Table 160

Table 161

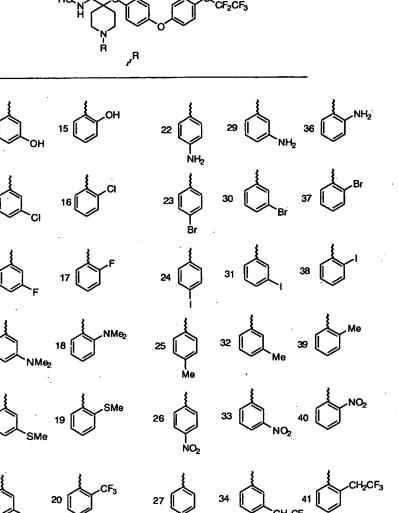


Table 163

Table 163

Table 164

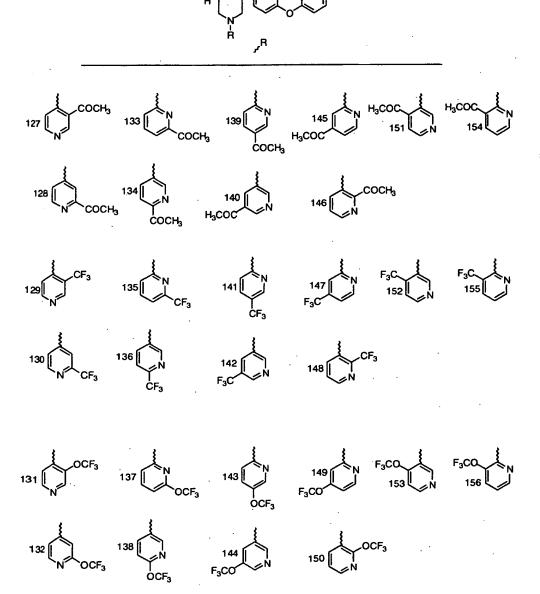


Table 165

157
$$\downarrow$$
 161 \downarrow CO₂CH₃ 165 \downarrow CONH₂ 169 \downarrow 158 \downarrow CO₂H 166 \downarrow COCH₃ 170 \downarrow CONH₂ 171 \downarrow CN 160 \downarrow CO₂H 164 \downarrow COCH₃ 168 \downarrow COCH₃ 172 \downarrow CN

5 Treatment Method

10

A contemplated inhibitor compound is used for treating a host mammal such as a mouse, rat, rabbit, dog, horse, primate such as a monkey, chimpanzee or human that has a condition associated with pathological matrix metalloprotease activity.

Also contemplated is use of a contemplated metalloprotease inhibitor compound in the treatment of a disease state that can be affected by the activity of metalloproteases $TNF-\alpha$ convertase.

Exemplary of such disease states are the acute phase responses of shock and sepsis, coagulation responses, hemorrhage and cardiovascular effects, fever and inflammation, anorexia and cachexia.

In treating a disease condition associated with pathological matrix metalloproteinase activity, 10 a contemplated MMP inhibitor compound can be used in the form of an amine salt derived from an inorganic or organic acid. Exemplary salts include but are not limited to the following: acetate, adipate, alginate, citrate, aspartate, benzoate, benzenesulfonate, 15 bisulfate, butyrate, camphorate, camphorsulfonate, digluconate, cyclopentanepropionate, dodecylsulfate, ethanesulfonate, glucoheptanoate, glycerophosphate, hemisulfate, heptanoate, hexanoate, fumarate, hydrochloride, hydrobromide, hydroiodide, 2-hydroxy-20 ethanesulfonate, lactate, maleate, methanesulfonate, nicotinate, 2-naphthalenesulfonate, oxalate, palmoate, pectinate, persulfate, 3-phenylpropionate, picrate, pivalate, propionate, succinate, tartrate,

Also, a basic nitrogen-containing group can be quaternized with such agents as lower alkyl halides, such as methyl, ethyl, propyl, and butyl chloride, bromides, and iodides; dialkyl sulfates

30 like dimethyl, diethyl, dibuytl, and diamyl sulfates, long chain halides such as decyl, lauryl, myristyl and stearyl chlorides, bromides and iodides, aralkyl halides like benzyl and phenethyl bromides, and others to provide enhanced water-solubility. Water

35 or oil-soluble or dispersible products are thereby

thiocyanate, tosylate, mesylate and undecanoate.

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obtained as desired. The salts are formed by combining the basic compounds with the desired acid.

Other compounds useful in this invention that are acids can also form salts. Examples include salts with alkali metals or alkaline earth metals, such as sodium, potassium, calcium or magnesium or with organic bases or basic quaternary ammonium salts.

In some cases, the salts can also be used 10 as an aid in the isolation, purification or resolution of the compounds of this invention.

Total daily dose administered to a host mammal in single or divided doses can be in amounts, for example, for 0.001 to 30 mg/kg body weight daily and more usually 0.01 to 10 mg. Dosage unit compositions can contain such amounts or submultiples thereof to make up the daily dose. A suitable dose can be administered, in multiple sub-doses per day. Multiple doses per day can also increase the total daily dose, should this be desired by the person prescribing the drug.

The dosage regimen for treating a disease condition with a compound and/or composition of this invention is selected in accordance with a variety of factors, including the type, age, weight, sex, diet and medical condition of the patient, the severity of the disease, the route of administration, pharmacological considerations such as the activity, efficacy, pharmacokinetic and toxicology profiles of the particular compound employed, whether a drug delivery system is utilized and whether the compound is administered as part of a drug combination. Thus, the dosage regimen actually employed can vary widely and therefore can deviate from the preferred dosage regimen set forth above.

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A compound of the present invention can be formulated as a pharmaceutical composition. Such a

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composition can then be administered orally, parenterally, by inhalation spray, rectally, or topically in dosage unit formulations containing conventional nontoxic pharmaceutically acceptable carriers, adjuvants, and vehicles as desired. Topical administration can also involve the use of transdermal administration such as transdermal patches or iontophoresis devices. The term parenteral as used herein includes subcutaneous injections, intravenous, intramuscular, intrasternal 10 injection, or infusion techniques. Formulation of drugs is discussed in, for example, Hoover, John E., Remington's Pharmaceutical Sciences, Mack Publishing Co., Easton, Pennsylvania; 1975 and Liberman, H.A. and Lachman, L., Eds., Pharmaceutical Dosage Forms, 15 Marcel Decker, New York, N.Y., 1980.

Injectable preparations, for example, sterile injectable aqueous or oleaginous suspensions can be formulated according to the known art using suitable dispersing or wetting agents and suspending 20 agents. The sterile injectable preparation can also be a sterile injectable solution or suspension in a nontoxic parenterally acceptable diluent or solvent, for example, as a solution in 1,3-butanediol. Among the acceptable vehicles and solvents that can be 25 employed are water, Ringer's solution, and isotonic sodium chloride solution. In addition, sterile, fixed oils are conventionally employed as a solvent or suspending medium. For this purpose any bland fixed oil can be employed including synthetic mono-30 or diglycerides. In addition, fatty acids such as oleic acid find use in the preparation of injectables. Dimethyl acetamide, surfactants including ionic and non-ionic detergents, polyethylene glycols can be used. Mixtures of 35 solvents and wetting agents such as those discussed above are also useful.

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Suppositories for rectal administration of the drug can be prepared by mixing the drug with a suitable nonirritating excipient such as cocoa butter, synthetic mono- di- or triglycerides, fatty acids and polyethylene glycols that are sold at ordinary temperatures but liquid at the rectal temperature and will therefore melt in the rectum and release the drug.

5

35

Solid dosage forms for oral administration can include capsules, tablets, pills, powders, and 10 granules. In such solid dosage forms, the compounds of this invention are ordinarily combined with one or more adjuvants appropriate to the indicated route of administration. If administered per os, a contemplated aromatic sulfone hydroximate inhibitor 15 compound can be admixed with lactose, sucrose, starch powder, cellulose esters of alkanoic acids, cellulose alkyl esters, talc, stearic acid, magnesium stearate, magnesium oxide, sodium and calcium salts of phosphoric and sulfuric acids, gelatin, acacia gum, 20 sodium alginate, polyvinylpyrrolidone, and/or polyvinyl alcohol, and then tableted or encapsulated for convenient administration. Such capsules or tablets can contain a controlled-release formulation as can be provided in a dispersion of active compound 25 in hydroxypropylmethyl cellulose. In the case of capsules, tablets, and pills, the dosage forms can also comprise buffering agents such as sodium citrate, magnesium or calcium carbonate or bicarbonate. Tablets and pills can additionally be 30 prepared with enteric coatings.

For therapeutic purposes, formulations for parenteral administration can be in the form of aqueous or non-aqueous isotonic sterile injection solutions or suspensions. These solutions and suspensions can be prepared from soerile powders or granules having one or more of the carriers or

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diluents mentioned for use in the formulations for oral administration. A contemplated aromatic sulfone hydroximate inhibitor compound can be dissolved in water, polyethylene glycol, propylene glycol, ethanol, corn oil, cottonseed oil, peanut oil, sesame oil, benzyl alcohol, sodium chloride, and/or various buffers. Other adjuvants and modes of administration are well and widely known in the pharmaceutical art.

Liquid dosage forms for oral administration

10 can include pharmaceutically acceptable emulsions,
solutions, suspensions, syrups, and elixirs
containing inert diluents commonly used in the art,
such as water. Such compositions can also comprise
adjuvants, such as wetting agents, emulsifying and

15 suspending agents, and sweetening, flavoring, and
perfuming agents.

The amount of active ingredient that can be combined with the carrier materials to produce a single dosage form varies depending upon the mammalian host treated and the particular mode of administration.

Best Mode For Carrying Out The Invention

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Without further elaboration, it is believed
that one skilled in the art can, using the preceding
description, utilize the present invention to its
fullest extent. The following preferred specific
embodiments are, therefore, to be construed as merely
illustrative, and not limiting of the remainder of
the disclosure in any way whatsoever.

Abbreviations are often used for reagents and solvents in the specific examples that follow. Those abbreviations and their meanings are as follows:

35 BOC = t-butoxycarbonyl
DEAD = diethyl azodicarboxylate

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DMF = dimethylformamide DMPU = 1,3-dimethyl-3,4,5,6-tetrahydro-2(1H)-pyrimidinone EtOAc = ethyl acetate EDC = 1-ethyl-3-[3-(dimethylamino)-5 propyl]carbodiimide hydrochloride Et,O = diethyl ether HOBT = 1-hydroxybenzotriazole MeOH = methanol MeCl, = methylene chloride 10 MsCl = methanesulfonyl chloride NMM = N-methyl morpholine THF = tetrahydrofruan TsCl = toluenesulfonyl chloride THP-O-hydroxylamine = O-tetrahydropyran-15 hydroxylamine and O-tetrahydro-2Hpyran-2-yl-hydroxylamine

The preparation of compounds useful in the synthesis of compounds of the invention are provided 20 herein below in Preparative Examples I through XI.

Preparative Example I: Preparation of 1,1dimethylethyl ester 4-[(hydroxyamino)carbonyl]-4-[(phenoxyphenyl)-sulfonyl]-1-25 piperidinecarboxylic acid

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Part A: A solution of 4-(phenoxy)benzenethiol (2.03 g, 10.0 mmol) in DMSO (DMSO; 20 mL) was heated to sixty-five degrees Celsius for 5 hours. The solution remained at ambient temperature for 18 hours. The solution was extracted with ethyl acetate and the combined organic layers were washed with H₂O and saturated NaCl and dried over magnesium sulfate. Concentration in vacuo provided the disulfide as a yellow oil (2.3 g, quantitative yield).

isonipecotate (15.7 g, 0.1 mol) in THF (100 mL) was added a solution of di-tert-butyl dicarbonate (21.8 g, 0.1 mol) in THF (5 mL) drop-wise over 20 minutes. The solution was stirred overnight (about eighteen hours) at ambient temperature and concentrated in vacuo to yield a light oil. The oil was filtered through silica gel (7:3 ethyl acetate/hexanes) and concentrated in vacuo to give the BOC-piperidine compound (26.2 g, quantitative yield) as a clear, colorless oil.

Part C: To a solution of diisopropylamine (2.8 mL, 20 mmoL) in THF (30 mL), cooled to minus seventy-eight degrees Celsius, was added n-butyl lithium (12.5 mL, 20 mmol) drop-wise. After 15 minutes, the BOC-piperidine compound of part B (2.6 g, 10 mmol) in THF (10 mL) was added drop-wise. After 1.5 hours the solution was cooled to minus sixty degrees Celsius and the disulfide of part A (2.0 g, 10 mmol) in THF (7 mL). The solution was stirred at ambient temperature for 2 hours. The solution was diluted with H₂O and extracted with ethyl acetate. The organic layer was washed with H₂O and saturated NaCl and dried over magnesium sulfate.

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Chromatography (on silica, ethyl acetate/hexane) provided the sulfide as an oil (1.8 g, 40%).

part D: To a solution of the sulfide of part C (1.8 g, 3.95 mmol) in dichloromethane (75 mL) cooled to zero degrees Celsius, was added m-chloroperbenzoic acid (1.7 g, 7.9 mmol). The solution was stirred for 1.5 hours followed by dilution with H₂O and extraction with dichloromethane. The organic layer was washed with 10 percent Na₂SO₄, H₂O, and saturated NaCl and dried over magnesium sulfate. Chromatography (on silica, ethyl acetate/hexane) provided the sulfone as a solid (1.15 g, 59%).

Part E: To a solution of the sulfone of
part D (800 mg, 1.63 mmol) in THF (9 mL) and ethanol
(9 mL) was added NaOH (654 mg, 16.3 mmol) in H₂O (3
mL). The solution was heated at sixty-five degrees
Celsius for 18 hours. The solution was concentrated
in vacuo and the residue was dissolved in H₂O.

20 Following acidification with 2N HCl to pH 4, the solution was extracted with ethyl acetate and the organic layer was washed with saturated NaCl and dried over magnesium sulfate. Concentration in vacuo provided the acid as a white foam (790 mg,

25 quantitative yield). Analytical calculated for
C23H27NO7S: C, 59.86; H, 5.90; N, 3.04; S, 6.95. Found:
C, 59.49; H, 6.37; N, 2.81; S, 6.59.

Part F: To a solution of the acid of part G (730 mg, 1.58 mmol) in DMF (9 mL) was added HOBT (256 mg, 1.90 mmol) followed by EDC (424 mg, 2.21 mmol), 4-methylmorpholine (0.521 mL, 4.7 mmol) and 50 percent aqueous hydroxylamine (1.04 mL, 15.8 mmol). The solution was stirred for 20 hours and additional

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N-hydroxybenzotriazole*H₂O (256 mg), EDC (424 mg) and 50 percent aqueous hydroxylamine (1.04 mL) were added. After an additional 24 hours of stirring the solution was diluted with H₂O and extracted with ethyl acetate and the organic layer was washed with saturated NaCl and dried over magnesium sulfate. Reverse phase chromatography (on silica, acetonitrile/H₂O) provided the title compound as a white solid (460 mg, 61%). HPLC purity: >99%. Analytical calculated for C₂₃H₂₈N₂O₇S: C, 57.97; H, 5.92; N, 5.88; S, 6.73. Found: C, 57.95; H, 6.02; N, 5.81; S, 6.85.

Preparative Example II: Preparation of N-hydroxy-4[[4-(phenylthio)phenyl]sulfonyl]-1(2-propynyl)-4-piperidinecarboxamide,
monohydrochloride

20 Part A: To a solution of ethyl isonipecotate
(15.7 g, 0.1 mol) in THF (100 mL) was added a
solution of di-tert-butyl dicarbonate (21.8 g, 0.1
mol) in THF (5 mL) drop-wise over 20 minutes. The
solution was stirred overnight (about eighteen hours)
25 at ambient temperature and concentrated in vacuo to
yield a light oil. The oil was filtered through
silica gel (thyl acetate/hexanes) and concentrated

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in vacuo to give the BOC-piperidine compound as a clear, colorless oil (26.2 g, quantitative yield).

Part B: A solution of 4-fluorothiophenol (50.29 g, 390 mmol) in DMSO (500 mL) was heated to 65 degrees Celsius for 6 hours. The reaction was quenched into wet ice and the resulting solid was collected by vacuum filtration to provide the disulfide as a white solid (34.4 g, 68.9%).

Part C: To a solution of the BOC-piperdine 10 compound of part A (16 g, 62 mmol) in THF (300 mL) cooled to minus 50 degrees Celsius was added lithium diisopropylamide (41.33 mL, 74 mmol) and the solution was stirred for 1.5 hours at zero degrees Celsius. To this solution was added the disulfide of part B (15.77 g, 62 mmol), and the resulting solution was 15 stirred at ambient temperature for 20 hours. The reaction was quenched with the addition of H2O and the solution was concentrated in vacuo. The aqueous residue was extracted with ethyl acetate and the organic layer was washed with 0.5N KOH, $\rm H_2O$, and 20 saturated NaCl. Chromatography (on silica, hexane/ethyl acetate) provided the sulfide as an oil (18.0 g, 75%).

Part D: To a solution of the sulfide of

25 part C (16.5 g, 43 mmol) in dichloromethane (500 mL)

cooled to zero degrees Celsius was added 3
chloroperbenzoic acid (18.0 g, 86 mmol) and the

solution was stirred for 20 hours. The solution was

diluted with H₂O and extracted with dichloromethane.

30 The organic layer was washed with 10 percent Na₂SO₃,

H₂O, and saturated NaCl and dried over magnesium

sulfate. Chromatography (on silica, ethyl

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acetate/hexane) provided the sulfone as a solid (10.7 g, 60%).

Part E: Into a solution of the sulfone of part D (10 g, 24.0 mmol) in ethyl acetate (250 mL)

was bubbled HCl gas for 10 minutes followed by stirring at ambient temperature for 4 hours.

Concentration in vacuo provided the amine hydrochloride salt as a white solid (7.27 g, 86%).

Part F: To a solution of the amine

10 hydrochloride salt of part E (5.98 g, 17.0 mmol) in

DMF (120 mL) was added potassium carbonate (4.7 g,

34.0 mmol) followed by propargyl bromide (2.02 g,

17.0 mmol) and the solution was stirred for 4 hours

at ambient temperature. The solution was partitioned

15 between ethyl acetate and H₂O, and the organic layer

was washed with H₂O and saturated NaCl and dried over

magnesium sulfate. Chromatography (on silica, ethyl

acetate/hexane) provided the propargyl amine as a

yellow oil (5.2 g, 86%).

20 Part G: To a solution of the propargyl amine of part F in DMF (15 mL) was added thiophenol (0.80 mL, 7.78 mmol) and CsCO₃ (2.79 g, 8.56 mmol) and the solution was heated to 70 degrees Celsius for 6 hours. The solution was partitioned between ethyl ether and H₂O. The organic layer was washed with H₂O and saturated NaCl, and dried over magnesium sulfate. Chromatography (on silica, ethyl acetate/hexane) provided the S-phenoxyphenyl compound as an oil (1.95 g, 56%).

30 Part H: To a solution of the S-phenoxyphenyl of part G (1.81 g, 4.06 mmol) in ethanol (21 mL) and H_2O (3.5 mL) was added KOH (1.37 g, 24.5 mmol) and the solution was heated to 105

degrees Celsius for 4.5 hours. The solution was acidified to a pH value of 1 with concentrated HCl solution and then concentrated to provide the acid as a yellow residue that was used without additional purification (1.82 g).

Part I: To a solution of the acid of part H (1.82 g, 4.06 mmol) in acetonitrile (20 mL) was added O-tetrahydro-2H-pyran-2-yl-hydroxylamine (723 mg, 6.17 mmol) and triethylamine (0.67 mL, 4.86 mmol). To this stirring solution was added EDC (1.18 g, 6.17 mmol) and the solution was stirred for 18 hours. The solution was partitioned between H₂O and ethyl acetate. The organic layer was washed with H₂O, saturated NaHCO₃ and saturated NaCl and dried over magnesium sulfate. Chromatography (on silica, ethyl acetate/hexane) provided the protected hydroxamate as a white solid (1.32 g, 63%).

Part J: To a solution of the protected hydroxamate of part I (9.65 g, 18.7 mmol) in methanol (148 mL) cooled to zero degrees Celsius was added acetyl chloride (4.0 mL, 56.2 mmol), and the solution was stirred for 45 minutes at ambient temperature. Concentration in vacuo followed by trituration with ethyl ether provided the title compound as a white solid (8.10 g, 94%). MS(CI) MH⁺ calculated for C₂₁H₂₂N₂O₄S₂: 431, found 431.

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Preparative Example III: Preparation of N-hydroxy-4[(4-phenoxyphenyl)sulfonyl]-1-(2propynyl)-4-piperidinecarboxamide,
monohydrochloride

Part A: A solution of 4-(phenoxy)benzenethiol (2.03 g, 10.0 mmol) in DMSO (20 mL) was heated to 65 degrees Celsius for 5 hours. The solution remained at ambient temperature for 18 hours. The solution was extracted with ethyl acetate and the combined organic layers were washed with $\rm H_2O$ and saturated NaCl, and dried over magnesium sulfate.

10 Concentration in vacuo provided the disulfide as a yellow oil (2.3 g, quantitative yield).

Part B: To a solution of ethyl isonipecotate (15.7 g, 0.1 mol) in THF (100 mL) was added a solution of di-tert-butyl dicarbonate (21.8 g, 0.1 mol) in THF (5 mL) dropwise over 20 minutes. The solution was stirred overnight at ambient temperature and concentrated in vacuo to yield a light oil. The oil was filtered through silica gel (ethyl acetate/hexane) and concentrated in vacuo to give the BOC-piperidine compound as a clear, colorless oil (26.2 g, quantitative yield).

Part C: To a solution of disopropylamine (2.8 mL, 20 mmoL) in THF (30 mL), cooled to minus seventy-eight degrees Celsius, was added n-butyl lithium (12.5 mL, 20 mmol) dropwise. After 15 minutes, the BOC-piperidine compound of Part B (2.6 g, 10 mmol) in THF (10 mL) was added dropwise. After 1.5 hours, the solution was cooled to minus sixty

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degrees Celsius and the disulfide of Part A (2.0 g, 10 mmol) in THF (7 mL) was added. The solution was stirred at ambient temperature for 2 hours. The solution was diluted with $\rm H_2O$ and extracted with ethyl acetate. The organic layer was washed with $\rm H_2O$ and saturated NaCl and dried over magnesium sulfate. Chromatography (on silica, ethyl acetate/hexane) provided the sulfide as an oil (1.8 g, 40%).

Part D: To a solution of the sulfide of

10 Part C (1.8 g, 3.95 mmol) in dichloromethane (75 mL)

cooled to zero degrees Celsius, was added m
chloroperbenzoic acid (1.7 g, 7.9 mmol). The

solution was stirred for 1.5 hours followed by

dilution with H₂O and extraction with dichloromethane.

15 The organic layer was washed with 10 percent Na₂SO₄,

H₂O, and saturated NaCl and dried over magnesium

sulfate. Chromatography (on silica, ethyl

acetate/hexane) provided the sulfone as a solid (1.15

g, 59%).

Part E: Into a solution of the sulfone of Part D (3.56 g, 7.0 mmol) in ethyl acetate (100 mL) cooled to zero degrees Celsius was bubbled HCl gas for 5 minutes. Concentration in vacuo followed by trituration with ethyl ether provided the amine hydrochloride salt as a white solid (3.5 g, quantitative yield). MS(CI) MH⁺ calculated for C₂₀H₂₃NO₅S: 390, found 390.

Part F: To a solution of the amine hydrochloride salt of part E (2.6 g, 6 mmol) and K_2CO_3 (1.66 g, 12 mmol) in DMF (50 mL) was added propargyl bromide (892 mg, 6 mmol) and the solution was stirred at ambient temperature for 4 hours. The solution was diluted with H_2O and extracted with ethyl acetate.

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The combined organic layers were washed with saturated NaCl and dried over magnesium sulfate. Chromatography (on silica, ethyl acetate/hexane) provided the propargyl amine as a white solid (2.15 q, 82%).

Part G: To a solution of the propargyl amine of part F (2.15 g, 5 mmol) in THF (30 mL) and ethanol (30 mL) was added NaOH (2.0 g, 50 mmol) and the solution was heated at 65 degrees Celsius for 48 hours. The solution was concentrated in vacuo and the aqueous residue was acidified to a pH value of 5. Vacuum filtration of the resulting precipitate provided the acid as a white solid (2.04 g, quantitative yield).

- Part H: To a solution of the acid of part 15 G (559 mg, 1.4 mmol) in dichloromethane (5 mL) was added triethylamine (0.585 mL, 4.2 mmol) and 50 percent aqueous hydroxylamine (0.925 mL, 14.0 mmol) followed by bromotris(pyrrolidino)phosphonium
- hexafluourphosphate (PyBroP; 718 mg, 1.54 mmol). 20 The solution was stirred at ambient temperature for 4 hours. The solution was diluted with $\mathrm{H}_2\mathrm{O}$ and extracted with dichloromethane. The organic layer was washed with saturated NaCl and dried over
- magnesium sulfate. Reverse phase chromatography (on 25 silica, acetonitrile/H2O) provided the hydroxamate as a white solid (140 mg, 25%). Analytical calculation for $C_{21}H_{22}N_2O_5S$: C, 60.85; H, 5.37; N, 6.76; S, 7.74. Found: C, 60.47; H, 5.35; N, 6.61; S, 7.46.
- Part I: To a solution of the hydroxamate 30 of part H (121 mg, 0.292 mmol) in methanol (2 mL) cooled to zero degrees Celsius was added acetyl chloride (0.228 mL, 0.321 mmol) in methanol (1 mL).

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After stirring at ambient temperature for 30 minutes the solution was concentrated under a stream of N₂.

Trituration with ethyl ether provided the title compound as a white solid (107 mg, 81%). Analytical calculation for C₂₁H₂₂N₂O₅S*HCl*0.3H₂O: C, 55.27; H, 5.21; N, 6.14. Found: C, 54.90; H, 5.37; N, 6.07.

Preparative Example IV: Preparation of 4-[(4-fluorophenyl)sulfonyl]tetrahydro-N[(tetrahydro-2H-pyran-2-yl)oxy]-2Hpyran-4-carboxamide

15 Part A: In dry equipment under nitrogen, sodium metal (8.97 g, 0.39 mol) was added to methanol (1000 mL) at two degrees Celsius. The reaction was stirred at ambient temperature for forty five minutes at which time the sodium had dissolved. The solution 20 was chilled to five degrees Celsius and p-fluorothiophenol (41.55 mL, 0.39 mmol) was added, followed by methyl 2-chloroacetate (34.2 mL, 0.39 mol). The reaction was stirred at ambient temperature for four hours, filtered, and concentrated in vacuo to give the sulfide as a clear colorless oil (75.85 g, 97%).

Part B: To a solution of the sulfide from part A (75.85 g, 0.38 mol) in methanol (1000 mL) were added water (100 mL) and Oxone (720 g, 1.17 mol) at 20 degrees Celsius. An exotherm to 67 degrees

Celsius was noted. After two hours, the reaction was filtered and the cake was washed well with methanol. The filtrate was concentrated in vacuo. The residue was taken up in ethyl acetate and washed with brine, dried over MgSO₄, filtered, and concentrated in vacuo to give the sulfone as a crystalline solid (82.74 g, 94%).

Part C: To a solution of the sulfone from part B (28.5 q, 0.123 mol) in N, N-dimethylacetamide (200 mL) were added potassium carbonate (37.3 g, 0.27 10 mol), bis-(2-bromoethyl)ether (19.3 mL, 0.147 mol), 4-dimethylaminopyridine (0.75 g, 6 mmol), and tetrabutylammonium bromide (1.98 g, 6 mmol). The reaction was stirred overnight (about 18 hours) at ambient temperature. The reaction was slowly poured 15 into 1N HCl (300 mL), the resultant solid filtered and the cake washed well with hexanes. The solid was recrystallized from ethyl acetate/hexanes to give the pyran compound as a beige solid (28.74 g, 77%). MS 20 (ES+) MH+ calculated for $C_{13}H_{15}O_5S_1F_1$; 303, found 303.

Part D: In dry equipment under nitrogen, the pyran compound from part C (8.0 g, 26.5 mmol) was dissolved in dry tetrahydrofuran (250 mL) and a solution of potassium trimethylsilonate (10.2 g, 79.5 mmol) in dry tetrahydrofuran (15 mL) was added at ambient temperature. After ninety minutes, water (100 mL) was added and the solution concentrated in vacuo. The residue was taken up in water and extracted with ethyl acetate to remove unreacted starting material. The aqueous solution was treated with 6N HCl until pH=1. The slurry was extracted with ethyl acetate and the combined extracts washed with

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water, dried over Na_2SO_4 , filtered, and concentrated in vacuo. The residue was heated in diethyl ether, the solid filtered and dried to give the carboxylic acid as a crystalline solid (5.78 g, 76%). HRMS (ES-) M-H calculated for $C_{12}H_{13}O_5$ S_1F_1 : 287.04, found 287.04.

Part E: In dry equipment under nitrogen, the carboxylic acid from part D (9.1g, 31.6 mmol) was dissolved in dry N, N-dimethylformamide (70 mL) and the remaining reagents were added to the solution in 10 the following order: N-hydroxybenzotriazole hydrate (5.1 g, 37.9 mmol), N-methylmorpholine (10.4 mL, 94.8 mmol), O-tetrahydro-2H-pyran-2-yl-hydroxylamine (11.5 g, 98 mmol), and 1-(3-dimethylaminopropyl)-3ethylcarbodiimide hydrochloride (8.48 g, 44.2 mmol). After three hours at ambient temperature, the reaction was concentrated in vacuo. The residue was taken up in ethyl acetate, washed with water, 5% KHSO₄, saturated NaHCO₃, brine, dried over Na₂SO₄, filtered, and concentrated in vacuo. Chromatography 20 (on silica, ethyl acetate/hexanes) provided the title compound as a crystalline solid (9.7 g, 80%). HRMS

25 Preparative Example V: Preparation of tetrahydro-N-hydroxy-4-[[4-[4-trifluoromethoxy)-phenoxy)phenyl]sulfonyl]-2H-pyran-4-carboxamide

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(ES+) MH+ calculated for $C_{17}H_{22}NO_6$ S_1F_1 : 388.12, found

Part A: To a solution of the title compound of Preparative Example IV (3.1 g, 8 mmol) in N,N-dimethylacetamide (20 mL) were added cesium carbonate (8.8 g, 27 mmol) and p-(trifluoromethoxy)phenol (2.1 mL, 16 mmol). The slurry was stirred at 95 degrees Celsius for nineteen hours. The reaction was concentrated in vacuo. The residue was taken up in ethyl acetate, washed with brine, dried over Na₂SO₄, filtered, and concentrated in vacuo. Chromatography (on silica, ethyl acetate/hexanes) provided the substituted THP-protected hydroxamate as a white foam (4.2 g, 96%). HRMS (ES+) MH+ calculated for C₂₄H₂₆N₁O₈ S₁F₃: 546.14, found 546.14.

15 Part B: To a slurry of the THP-protected hydroxamate from part A (4.0 g, 7.3 mmol) in dioxane (20 mL) were added a 4N HCl dioxane solution (20 mL) and methanol (20 mL). After fifteen minutes at ambient temperature, the reaction was diluted with 20 ethyl acetate and washed with water, dried over Na₂SO₄, filtered, and concentrated in vacuo. The product was recrystallized (acetone/hexanes) to give the title compound as a white solid (2.2 g, 65%). HRMS (ES+) M+ NH₄ calculated for C₁₉H₁₈N₁O₇S₁F₃: 479.11, found 479.11.

Preparative Example VI: Preparation of 1
cyclopropyl-N-hydroxy-4-[[4-(2-phenoxyethoxy)phenyl]sulfonyl]-4-piperidine

carboxamide, monohydrochloride

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Part A: To a solution of the product of Preparative Example II, part E, (14.36 g, 40 mmol) in methanol (50 mL) was added acetic acid (24.5 g, 400 mmol), a portion (about 2 g) of 4-Ångstrom molecular sieves, (1-ethoxycyclopropyl)-oxytrimethyl silane (25.8 mL, 148 mmol) and sodium cyanoborohydride (7.05 g, 112 mmol). The solution was heated at reflux for 8 hours. The precipitated solids were removed by filtration and the filtrate was concentrated in 10 The residue was diluted with H_2O (400 mL) and extracted with ethyl acetate. The organic layer was washed with saturated NaCl and dried over MgSO4, filtered and concentrated in vacuo. The solid was filtered, washed with H2O/diethyl ether to give the 15 desired cyclopropyl amine {ethyl 4-[(4-fluorophenylsulfonyl)]-1-cyclopropyl-4-piperidinecarboxylate} as a white solid (11.83 g, 81.5%). MS MH⁺ calculated for $C_{17}H_{22}NO_4SF$: 356, found: 356.

Part B: A solution of the cyclopropyl amine of Part A (2.0 g, 5.6 mmol), ethylene glycol phenyl ether (2.8 mL, 23 mmol), and cesium carbonate (7.3 g, 23 mmol) in DMAC (10 mL) was heat at 125-135 degrees Celsius for 18 hours under an atmosphere of nitrogen. The mixture was concentrated in vacuo, diluted with 25 water, and extracted with ethyl acetate. combined ethyl acetate layers were washed with water and brine, dried over magnesium sulfate, concentrated

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in vacuo, dissolved in diethyl ether, precipitated as the hydrochloride salt, and dried at 40 degrees Celsius in a vacuum oven. The solid was dissolved into a mixture of water, acetonitrile, and ethanol and then the pH was adjusted to 12 with 1N NaOH solution. The mixture was concentrated in vacuo to remove ethanol and acetonitrile. The solid was isolated by filtration, washed with water, and dried at 50 degrees Celsius in a vacuum oven to afford the ether as a white solid (1.8 g, 68%): MS+ calcd. for C25H31NO6S 474, found 474. Anal. calcd. for C25H31NO6S: C, 63.40; H, 6.60; N, 2.96; S, 6.77. Found: C, 63.35; H, 6.59; N, 2.99; S, 6.61.

Part C: A mixture of the ether of part B 15 (1.8 g, 3.7 mmol) and a 50% NaOH aqueous solution (3.0 g, 37 mmol) in THF (32 mL), EtOH (32 mL), and H_2O (16 mL) was heated at 60 degrees Celsius under a nitrogen atmosphere for 24 hours. The material was concentrated in vacuo and triturated with diethyl ether to give a solid. The tan solid was dissolved 20 into a mixture of water, ethanol, and THF, precipitated by adjusting the pH to 3 with concentrated hydrochloric acid, concentrated in vacuo, triturated with water, and dried at 50 degrees 25 Celsius in a vacuum oven to give a crude white solid acid (2.3 g).

A mixture of the crude white solid acid

(2.3 g), N-hydroxybenzotriazole (1.9 g, 14 mmol), 4methylmorpholine (1.6 mL, 14 mmol), O-tetrahydro-2H
30 pyran-2-yl-hydroxylamine (1.1 g, 9.4 mmol), and 1-(3dimethylaminopropyl)-3-ethylcarbodiimide
hydrochloride (2.7 g, 14 mmol) in DMF (90 mL) was
stirred at ambient temperature under a nitrogen

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atmosphere for 2 days. The mixture was concentrated in vacuo, diluted with water, and extracted with ethyl acetate. The organic layer was washed with 1N NaOH solution, water, and brine, dried over magnesium sulfate, concentrated in vacuo, and purification by flash chromatography (20:80 to 40:60 ethyl acetate/toluene) to afford the protected hydroxamate as a white solid: (0.43 g, 21%): MS MH+ calcd. for C28H36N2O7S 545, found 545. Anal. calcd. for C28H36N2O7S: C, 61.74; H, 6.66; N, 5.14; S, 5.89. Found: C, 61.72; H, 6.75; N, 5.06; S, 5.91.

Additional compound was isolated by acidifying the aqueous layer to pH of 3, collecting the solid by filtration, and drying to give a white solid (0.80 g).

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Part D: To an ambient temperature solution of acetyl chloride (0.31 mL, 4.4 mmol) in methanol (11 mL) under a nitrogen atmosphere was added the protected hydroxamate of part C (0.80 g, 1.5 mmol). After stirring for 2.5 hours, the precipitate was collected by filtration, washed with diethyl ether,

and dried at 45 degrees Celsius in a vacuum oven to afford the title compound as a white solid (0.58 g, 79%): MS MH+ calcd. for C₂₃H₂₈N₂O₆S 461, found 461.

25 Anal. calcd. for C₂₃H₂₈N₂O₆S 1.5HCl: C, 53.62; H, 5.77; N, 5.44; S, 6.22. Found: C, 53.47; H, 5.79; N, 5.41; S, 6.16.

Preparative Example VII: Preparation of N-hydroxy-1
(2-methoxyethyl)-4-[[4-[4-(trifluoromethoxy)phenoxy]phenyl]sulfonyl}-4piperidinecarboxamide, monohydrochloride

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Part A: To a solution of the product of Preparative Example II, Part D (30 g, 161 mmol) in dichloromethane (50 mL) cooled to zero degrees 5 Celsius was added trifluroacetic acid (25 mL) and the solution was stirred at ambient temperature for 1 hour. Concentration in vacuo provided the amine trifluoroacetate salt as a light yellow gel. To the solution of the trifluoroacetate salt and K₂CO₃ (3.6 10 q, 26 mmol) in N, N-dimethylformamide (50 mL) cooled to zero degrees Celsius was added 2-bromoethyl methyl ether (19 mL, 201 mmol), and solution was stirred at ambient temperature for 36 hours. Then, N,Ndimethylformamide was evaporated under high vacuum 15 and the residue was diluted with ethyl acetate. organic layer was washed with water and dried over MgSO4. Concentration in vacuo provided the methoxyethyl amine as a light yellow gel (26.03 g, 20 86.8%).

Part B: To a solution of methoxyethyl amine (6.0 g, 16.0 mmol) of Part A and powdered K_2CO_3 (4.44 g, 32 mmol) in N,N-dimethylformamide (30 mL) was added 4-(trifluoromethoxy)phenol (5.72 g, 32 mmol) at ambient temperature and the solution was heated to ninety degrees Celsius for 25 hours. The solution was concentrated under high vacuum and the residue was

dissolved in ethyl acetate. The organic layer was washed with 1N NaOH, H₂O and dried over MgSO₄. Chromatography on silica eluting with ethyl acetate/hexane provided trifluoromethoxy phenoxyphenyl sulfone as a light yellow gel (7.81 g, 91.5%).

Part C: To a solution of trifluoromethoxy phenoxyphenyl sulfone of Part B (7.81 g, 14.7 mmol) in ethanol (14 mL) and tetrahydrofuran (14 mL) was added NaOH (5.88 g, 147 mmol) in H_2O (28 mL) from an addition funnel at ambient temperature. The solution was then heated to sixty degrees Celsius for 18 hours. The solution was concentrated in vacuo and diluted with water. The aqueous layer was extracted with ether and acidified to pH = 2. Vacuum filtration of white precipitation provided the acid as a white solid (5.64 g, 73.3%).

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To a solution of the acid of Part C Part D: (5.64 g, 10.8 mmol), N-methyl morpholine (4.8 mL, 43.1 mmol), 1-hydroxybenzotriazole (4.38 g, 32.4 20 mmol) and O-tetrahydropyranyl hydroxyl amine (2.5 g, 21.6 mmol) in N,N-dimethylformamide (50 mL) was added 1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide hydrochloride (6.2 g, 32.4 mmol), and the solution was stirred at ambient temperature for 24 hours. 25 solution was concentrated under high vacuum and the residue was dissolved in ethyl acetate. The organic layer was washed with saturated aqueous NaHCO3, H2O and dried over MgSO4. Concentration in vacuo and chromatography on silica eluting with ethyl 30 acetate/hexane provided the tetrahydropyranylprotected hydroxamate as a white foam (6.65 g, quantitative yield).

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Part E: To a solution of 4N HCl in dioxane (28 mL, 110 mmol) was added a solution of the tetrahydropyranyl-protected hydroxamate of Part D (6.65 g, 11.03 mmol) in methanol (3 mL) and dioxane 5 (9 mL) and was stirred at ambient temperature for 3 hours. Concentration in vacuo and trituration with diethyl ether provided the title compound as a white solid (4.79 g, 78.2%). Analytical calculation for $C_{22}H_{25}N_2O_7SF_3$.HCl.0.5H₂O: C, 46.85; H, 4.83; N, 4.97; S, 5.69. Found: C, 46.73; H, 4.57; N, 4.82; S, 5.77.

Preparative Example VIII: Preparation of N-hydroxy-(trifluoromethyl)phenoxy]-phenyl] sulfonyl]-4-piperidinecarboxamide, dihydrochloride

Part A: To a suspension of 4-bromopiperidine 20 hydrobromide (107.0 g, 0.436 mol) in tetrahydrofuran (1 L) was slowly added triethylamine (122 mL, 0.872 mol) followed by di-tert-butyl dicarbonate (100 g, 0.458 mol), which was added in several portions. The resulting mixture was stirred at ambient temperature 25

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for 22 hours then filtered and concentrated in vacuo. The solids were washed with hexanes and then collected by filtration to give the Boc-piperidine compound as an amber oil (124 g, >100 %).

Part B: To a solution of 4-fluorophenol (50.0 g, 0.390 mol) in acetone (400 mL), degassed with N_2 , was added Cs_2CO_3 (159 g, 0.488 mol). After degassing the resulting mixture with N_2 for 5 minutes, the Bocpiperidine compound of Part A (85.9 g, 0.325 mol) was added. The resulting mixture was stirred at ambient temperature for 18 hours and then filtered through a pad of Celite®, washing with acetone. The filtrate was concentrated in vacuo to provide the sulfide as a tan residue (98.5 g, 97%).

Part C: To a solution of the sulfide of Part B 15 (8.00 g, 25.7 mmol) in dichloromethane (90 mL) and methanol (15 mL) was added monoperoxyphthalic acid magnesium salt hexahydrate (19.1 g, 38.6 mmol) in two portions. The resulting mixture was stirred at 20 ambient temperature for 1.5 hours and then filtered. The filtrate was washed with saturated NaHCO3 and then with saturated NaCl. The combined aqueous layers were extracted with dichloromethane (100 mL). combined organic layers were dried over Na₂SO₄ and then concentrated in vacuo. The resulting solids 25 were washed with hexanes then dissolved in dichloromethane and filtered through a pad of Celite®, washing with dichloromethane. The filtrate was concentrated in vacuo and recrystallization from ethyl acetate provided the sulfone as a white 30 crystalline solid (4.45 g, 50%).

Part D: To a solution of sulfone of Part C (7.00 g, 20.4 mmol) in N,N-dimethylformamide (40 mL)

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was added Cs₂CO₃ (19.9 g, 61.2 mmol) and α,α,α-trifluoro-p-cresol (3.97 g, 24.5 mmol). The resulting mixture was heated at eighty degrees Celsius for 16 hours. After cooling to ambient temperature the reaction mixture was concentrated in vacuo. The resulting residue was treated with H₂O and the solids were collected by filtration. The solids were then washed with hexanes then methanol to provide the biaryl ether as a tan solid (8.60 g, 87%).

Part E: To a solution of the biaryl ether of Part D (8.59 q, 17.7 mmol) in tetrahydrofuran (100 mL), cooled to zero degrees Celsius, was slowly added lithium bis(trimethylsilyl)amide (22.0 mL, 1.0M in tetrahydrofuran, 22.0 mmol), at such a rate that the 15 temperature of the reaction never exceeded one degree Celsius. The resulting mixture was stirred at zero degrees Celsius for 1 hour then a solution of methyl chloroformate (2.05 mL, 26.6 mmol) in tetrahydrofuran (5.0 mL) was slowly added, at such a rate that the 20 temperature of the reaction mixture never exceeded four degrees Celsius. After the addition was complete, the mixture was slowly permitted to warm to ambient temperature. Saturated NH4Cl (50 mL) was added and the tetrahydrofuran was removed in vacuo. 25 Water (50 mL) was added to the residue which was then extracted with ethyl acetate. The combined organic layers were washed with saturated NaCl and dried over Na₂SO₄. Recrystallization from methanol provided the methyl ester as a pale yellow crystalline solid (7.66 30 g, 80%).

Part F: To a solution of the methyl ester of Part E (7.66 g, 14.1 mmol) in dioxane (30 mL) and

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methanol (10 mL) was added a solution of 4N HCl in dioxane (10 mL, 40 mmol). After stirring at ambient temperature for 2 hours additional 4N HCl in dioxane (10 mL, 40 mmol) was added. After stirring at ambient temperature for 2.5 hours, the reaction mixture was concentrated in vacuo to provide the amine as an off-white solid (6.80 g, >100%).

Part G: To a suspension of the amine of Part F
(3.00 g, 6.25 mmol) in acetonitrile (20 mL) was added

10 K₂CO₃ (3.46 g, 25.0 mmol), 4-(2-chloroethyl)morpholine
hydrochloride (1.22 g, 6.56 mmol) and a catalytic
amount of NaI. The resulting mixture was heated at
reflux for 22 hours. After cooling to ambient
temperature, the reaction mixture was filtered

15 through a pad of Celite®, washing with ethyl acetate.
The filtrate was concentrated in vacuo to provide the
morpholinyl ethyl amine as a tan solid (3.45 g,
>100%).

Part H: To a solution of the morpholinyl ethyl
amine of Part G (3.45 g, 6.25 mmol) in
tetrahydrofuran (60 mL) was added potassium
trimethylsilanolate (1.60 g, 12.50 mmol). After
stirring at ambient temperature for 25 hours, H₂O was
added. The reaction mixture was then neutralized (pH
7) with 1N HCl. The tetrahydrofuran was removed in
vacuo and the resulting precipitate was collected by
filtration and washed with diethyl ether to provide
the amino acid as an off-white solid (2.87 g, 85%).

Part I: To a suspension of the amino acid of

Part H (2.87 g, 5.29 mmol) in dichloromethane (25 mL)

was added N-methylmorpholine (1.74 mL, 15.9 mmol), O
(tetrahydropuranyl) hydroxylamine (0.682 g, 5.82

mmol) and PyBroP® (2.96 g, 6.35 mmol). After

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stirring at ambient temperature for 19 hours additional N-methylmorpholine (0.872 mL, 7.94 mmol), O-(tetrahydropuranyl) hydroxylamine (0.310 g, 2.65 mmol) and PyBroP® (1.48 g, 3.17 mmol) were added.

5 The resulting mixture was stirred at ambient temperature for 3 hours and then concentrated in vacuo. The residue was partitioned between ethyl acetate and H₂O. The organic layers were washed with saturated NaCl and dried over Na₂SO₄. Chromatography (on silica, methanol/chloroform) provided the protected hydroxamate as an off-white solid (2.62 g, 77%).

Part J: To a solution of the protected hydroxamate of Part I (2.62 g, 4.08 mmol) in dioxane (9 mL) and methanol (3 mL) was added a solution of 4N HCl in dioxane (10 mL, 40.0 mmol). The resulting mixture was stirred at ambient temperature for 2 hours and then diethyl ether (20 mL) was added. The resulting solids were collected by filtration to give the title compound as an off-white solid (2.31 g, 90%). MS MH+ calculated for C₂₅H₃₁O₆N₃SF₃: 558, found 558.

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Part A: To a solution of the product of Preparative Example VI, Part A, (6.97 g, 19.6 mmol) in DMF (500 mL) was added K₂CO₃ (3.42 g, 18.0 mmol) and 4-(triflouromethoxy)phenol (3.7 g, 24.8 mmol). The solution was stirred at ninety degrees Celsius for 40 hours. The solution was diluted with H₂O (600 mL) and extracted with ethyl acetate. The organic layer was washed with water, saturated NaCl and dried over MgSO₄, filtered and concentrated in vacuo to afford the desired diaryl ether as an oil (8.5 g, quantitative). HRMS MH⁺ calculated for C₂₄H₂₆NSO₆F₃: 514.1511. Found 514.1524.

To a solution of diaryl ether from Part Part B: A (8.4 g, 16.4 mmol) in ethanol (50 mL) and 15 tetrahydrofuran (50 mL) was added a solution of NaOH (6.54 q, 164 mmol) in water (20 mL) and the solution was heated at sixty degrees Celsius for 18 hours. The solution was concentrated in vacuo to remove most Of organic solvents and the aqueous residue was 20 acidified to pH = 4.0. The resulting precipitate was filtered to give the desired filtered to give the hydrochloride salt as a white solid (5.01 g, 63%). HRMS MH calculated for C₂₂H₂₂NSO₆F₃: 486.1198, found 25 486.1200.

Part C: To a solution of the hydrochloride salt Of Part B (5.0 g, 10.3 mmol) in DMF (80 mL) were

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added 1-hydroxybenzotriazole (1.65 g, 12.3 mmol), Nmethyl morpholine (3.4 mL, 30.9 mmol) and Otetrahydropyranyl hydroxylamine hydrochloride (1.8 g, 15.4 mmol) followed by 1-3-(dimethylamino)propyl]-3-5 ethylcarbodiimide hydrochloride (1.60 g, 12.3 mmol). The solution was stirred at ambient temperature for 42 hours. The solution was diluted with H₂O (400 mL) and extracted with ethyl acetate. The organic layer was washed with saturated NaCl and dried over MgSO4, filtered and concentrated in vacuo. Chromatography on silica gel, eluting with 30% ethyl acetate/hexane provided the desired tetrahydropyranyl-protected hydroxamate as a white solid (5.41 g, 89%).

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Part D: To a solution of tetrahydropyranylprotected hydroxamate of Part C (5.4 g, 9.2 mmol) in 15 dioxane (80 mL) and methanol (20 mL) was added 4 N HCl/dioxane (50 mL). The reaction was stirred at ambient temperature for 2.5 hours, the solution was concentrated in vacuo. Trituration with diethyl ether afforded the title compound as a white solid 20 (4.02 g, 81%). HRMS MH calculated for C₂₂H₂₃N₂SO₆F₃: 501.1307, found 501.1324.

Preparative Example X: Preparation of 1-cyclopropyl-N-hydroxy-4-[[4-[4-(trifluoromethyl) 25 phenoxy phenyl | sulfonyl | -4piperidinecarboxamide, monohydrochloride

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Part A: To a solution of the product of Preparative Example VI, Part A, (5.96 g, 15.0 mmol) in DMF (100 mL) was added K_2CO_3 (12.34 g, 38.0 mmol) 5 and α, α, α -trifluoromethyl phenol (3.65 g, 22.5 mmol). The solution was stirred ninety degrees Celsius for 28 hours. The solution was diluted with H₂O (400 mL) and extracted with ethyl acetate. The organic layer was washed with water, saturated NaCl and dried over MqSO₄, filtered and concentrated in vacuo to afford desired aryl ether as an oil (7.54 g, quantitative)

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Part B: To a solution of aryl ether from Part A (7.54 g, 15.0 mmol) in ethanol (40 mL) and tetrahydrofuran (40 mL) was added a solution of NaOH (6.06 g, 151.0 mmol) in water (20 mL) and the solution was heated at sixty degrees Celsius for 18 hours. The solution was concentrated in vacuo and the aqueous residue was acidified to pH = 2.0. resulting precipitate was filtered to give the desired hydrochloride salt as a white solid (7.98 g, quantitative). MS MH+ calculated for C22H22NSO5F3: 470, found 470.

Part C: To a solution of the hydrochloride salt of Part B (7.60 g, 15.0 mmol) in DMF (100 mL) were 25 added 1-hydroxybenzotriazole (2.44 g, 18.0 mmol), Nmethyl morpholine (3.4 mL, 30.9 mmol) and O-

tetrahydropyranyl hydroxyl amine hydrochloride (2.63 g, 22.5 mmol) followed by 1-3-(dimethylamino)propyl]3-ethylcarbodiimide hydrochloride (4.02 g, 21.0 mmol). The solution was stirred at ambient

5 temperature for 96 hours. The solution was diluted with H₂O (400 mL) and extracted with ethyl acetate. The organic layer was washed with saturated NaCl and dried over MgSO₄, filtered and concentrated in vacuo. Chromatography on silica eluting with 30% ethyl

10 acetate/hexane provided the desired tetrahydropyranyl-protected hydroxamate as a white solid (5.93g, 69%).

Part D: To a solution of tetrahydropyranyl-protected hydroxamate of Part C (3.8 g, 6.7 mmol) in dioxane (100 mL) was added 4 N HCl/dioxane (30 mL). The reaction was stirred at ambient temperature for 2 hours, then the solution was concentrated *in vacuo*. Trituration with diethyl ether afforded the title compound as a white solid (3.33 g, 96%). MS MH $^+$ calculated for $C_{22}H_{23}N_2SO_5F_3$: 485 , found 485.

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Preparative Example XI: Preparation of Resin II:

Step 1: Attachment of Compound of

Preparative Example IV to Resin I

with of resin I [Floyd et al., Tetrahedron Lett.
1996, 37, 8045-8048] (8.08 g, 9.7 mmol) and 1-methyl2-pyrrolidinone (50 mL). A magnetic stirring bar was
added, and the resin slurry slowly stirred. A

separate solution of the compound of Part D,
Preparative Example IV (5.58 g,19.4 mmol) in 1methyl-2-pyrrolidinone (35 mL) was added to the
slurry followed by addition of benzotriazole-1-yl-

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oxy-tris-pyrrolidino-phosphonium hexafluorophosphate (10.1 g, 19.4 mmol) in one portion. Once the hexafluorophosphate salt had dissolved, 4methylmorpholine (4.26 mL, 39 mmol) was added dropwise. The reaction slurry was stirred at room temperature for 24 hours, then the resin was collected in a sintered-disc funnel and washed with N, N-dimethylformamide, methanol, methylene chloride and diethyl ether (3x30 mL each solvent). The resin was dried in vacuo to yield 10.99 g polymer-bound 10 hydroxymate as a tan polymeric solid. Theoretical loading on polymer was 0.91 mmol/g. FTIR microscopy showed bands at 1693 and 3326 cm⁻¹ indicative of the hydroxamate carbonyl and nitrogen-hydrogen stretches, respectively.

Step 2: Preparation of Resin III: Reaction of Resin II With Nucleophiles Resin II (50 mg, 0.046 mmol) was weighed into an 8 mL glass vial, and a 0.5 M solution of a 20 nucleophile in 1-methyl-2-pyrrolidinone (1 mL) was added to the vessel. In the case of phenol and thiophenol nucleophiles, cesium carbonate (148 mg, 0.46 mmol) was added, and in the case of substituted piperazine nucleophiles, potassium carbonate (64 mg, 0.46 mmol) was added. The vial was capped and heated 25 to 70 to 155 degrees Celsius for 24-48 hours, then cooled to room temperature. The resin was drained and washed with 1-methyl-2-pyrrolidinone, 1-methyl-2pyrrolidinone/water (1:1), water, 10% acetic acid/water, methanol, and methylene chloride (3x3 mL 30 each solvent).

Large Scale Preparation of Resin IIIa:

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Resin II (5 g, 0.91 mmol) was weighed into an oven-dried three-necked round bottom flask fitted with a temperature probe, an overhead stirring paddle, and a nitrogen inlet. Anhydrous 1-methyl-2-pyrrolidinone (35 mL) was added to the flask followed by ethyl isonipecotate (7.0 mL, 45.5 mmol). The resin slurry was stirred slowly with the overhead stirrer, and the mixture was heated to 80 degrees Celsius with a heating mantle for 65 hours. The flask was thereafter cooled to room temperature.

The resin was collected in a sintered-disk glass funnel and washed with N,N-dimethylformamide, methanol and methylene chloride (3X30 mL each solvent). The resin was dried in vacuo to provide 5.86 g of resin IIIa as off-white resin beads. The theoretical loading of the polymer was 0.81 mmol/g. TFA cleavage performed on 50 mg of resin IIIa as described in step 3 yielded 10.4 mg of off-white solid spectroscopically indistinguishable from a known sample.

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Step 3: Cleavage of Hydroxamic Acids From The Polymer-Support

trifluoroacetic acid/ water mixture (19:1, 1 mL) for 1 hour at room temperature. During that time, the resin became a deep red color. The resin was then drained and washed with trifluoroacetic acid/water (19:1) and methylene chloride (2x1 mL each solvent),

Resin III was treated with a

30 collecting the combined filtrates in a tared vial.

The volatiles were removed in vacuo, then a
toluene/methylene chloride mixture (2 mL each) was
added to the residue. The mixture was again

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concentrated *in vacuo*. The product was characterized by electrospray mass spectroscopy.

Step 4: Hydrolysis of Polymer-Bound
Ester: Preparation of Resin IVa

Resin IIIa (5.8 g, 4.5 mmol) was weighed into a three-necked round bottomed flask fitted with an overhead stirring paddle. 1,4-Dioxane was added to the flask, and the resin slurry was stirred for 15 minutes. Then, a 4 M solution of KOH (5 mL, 20 mmol) 10 was added, and the mixture was stirred for 44 hours. The resin was thereafter collected in a sintered-disk glass funnel and washed with dioxane/water (9:1), water, 10% acetic acid/water, methanol and methylene chloride (3X30 mL each solvent). The resin was dried in vacuo to yield 5.64 g of resin IVa as off-white polymer beads. FTIR microscopy showed bands at 1732 and 1704 cm^{-1} and a broad band from $2500-3500 \text{ cm}^{-1}$. The theoretical loading of the polymer-bound acid was 20 0.84 mmol/g.

Example 1: Preparation of 1-(2-methoxyethyl)
4-[[4-[4-(trifluoromethoxy)
phenoxy]phenyl]sulfonyl]
4-piperidinecarboxamide

Part A: To a solution of the product of Preparative Example II, part D, (30 g, 161 mmol) in dichloromethane (50 mL) cooled to zero degrees Celsius was added trifluroacetic acid (25 mL) and the solution was stirred at ambient temperature for 1 Concentration in vacuo provided the amine trifluoroacetate salt as a light yellow gel. To the solution of the trifluoroacetate salt and K2CO3 (3.6 g, 26 mmol) in N,N-dimethylformamide (50 mL) cooled 10 to zero degrees Celsius was added 2-bromoethyl methyl ether (19 mL, 201 mmol) and solution was stirred at ambient temperature for 36 hours. Then N, Ndimethylformamide was evaporated under high vacuum 15 and the residue was diluted with ethyl acetate. organic layer was washed with water and dried over MgSO₄. Concentration in vacuo provided the methoxyethyl amine as a light yellow gel (26.03 g, 86.8%).

20 Part B: To a solution of the methoxyethyl amine (6.0 g, 16.0 mmol) of part A and powdered K₂CO₃ (4.44 g, 32 mmol) in N,N-dimethylformamide (30 mL) was added 4-(trifluoromethoxy)phenol (5.72 g, 32 mmol) at ambient temperature and the solution was 25 heated to ninety degrees Celsius for 25 hours. The solution was concentrated under high vacuum and the residue was dissolved in ethyl acetate. The organic

layer was washed with 1N NaOH, H_2O and dried over MgSO₄. Chromatography on silica eluting with ethyl acetate/hexane provided trifluoromethoxy phenoxyphenyl sulfone as a light yellow gel (7.81 g, 91.5%).

Part C: To a solution of trifluoromethoxy phenoxyphenyl sulfone of part B (7.81 g, 14.7 mmol) in ethanol (14 mL) and tetrahydrofuran (14 mL) was added NaOH (5.88 g, 147 mmol) in H₂O (28 mL) from an addition funnel at ambient temperature. The solution was then heated to sixty degrees Celsius for 18 hours. The solution was concentrated in vacuo and diluted with water. The aqueous layer was extracted with ether and acidified to pH = 2. Vacuum filtration of the white precipitation provided the carboxylic acid as a white solid (5.64 g, 73.3%).

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Part D: To a suspension of the carboxylic acid of part C (200 mg, 0.397 mmol) in methylene chloride (4 mL) was added oxalyl chloride (101 mg, 20 0.80 mmol). After 15 minutes at ambient temperature the volatiles were removed under vacuum. The solid residue was resuspended in methylene chloride (4 mL) and gaseous ammonia was bubbled through the suspension. Triethylamine (81 mg, 0.80 mmol) was 25 added and the stream of ammonia gas through the reaction was continued for 1 minute. Concentration afforded a solid which was chromatographed (reverse phase C₁₈ silica eluting with a gradient of 30% acetonitrile/water to 100% acetonitrile) to afford 30 the desired primary amide as a colorless powder (6 mg, 3 mg). MS MH^+ calculated for $C_{22}H_{25}N_2$ F_3O_6S : 503, found 503. HRMS M+ calculated for $C_{22}H_{25}N_2$ F_3O_6S : 503.1464, found 503.1472.

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Example 2: Preparation of 4-[(4-phenylthiophenyl)
sulfonyl]-1-(2-propynyl)4-piperidinecarboxamide

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$$H_2N$$
 SO_2
 S
 S

A mixture of the acid from Preparative Example II, part H, (1.29 g, 2.85 mMol), Nhydroxybenzotriazole (1.15 g, 8.54 mMol), 4-10 methylmorpholine (0.94 mL, 14 mMol), concentrated NH_4OH (3 mL), and 1-(3-dimethylaminopropyl)-3ethylcarbodiimide hydrochloride (1.64 g, 8.54 mMol) in DMF (25 mL) was stirred at ambient temperature for The mixture was concentrated in vacuo, 15 20 hours. diluted with water, and extracted with ethyl acetate. The organic layer was washed with saturated NaHCO3, water, and brine, dried over magnesium sulfate, and concentrated in vacuo. Chromatography (on silica, MeOH/CHCl3) afford the title amide as a white solid 20 (0.143 q, 12%). Analytical calculation for $C_{21}H_{22}N_2O_3S_2$: C, 60.84; H, 5.35; N, 6.76; S, 15.47. Found: C, 60.74; H, 5.31; N, 6.74; S, 15.43.

25 Examples 3-58

The following compounds were prepared by parallel synthesis (resin based synthesis, automated synthesis) using parallel synthesis from Resin IVa as

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PCT/US00/06719

described previously in Preparative Example XI the following compounds were prepared:

Example	Amine	R	MS (M+H)
3	3,5-Dimethylpiperidine	! —√	508
4	N-Methylpropargylamine	}- <u>(</u>	464
5	N-Methylallylamine	} -<	466
6	1-(1-phenylethyl)- piperazine	TTA TTA	585
7	1-(2-phenylethyl)- piperazine	}	585
8	1-(2-chlorophenyl)- piperazine	} -√-√-	591
9	<pre>1-(4-methoxyphenyl)-2- methylpiperazine</pre>	F-\(\)\(\)	585
10	1-(5-Chloro-2- methylphenyl)piperazin e	├	605
11	1-(2-methoxyphenyl)- piperazine	>-~	587
12	1-Acetylpiperazine	\ \-\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	523

13	1-(2,4- Dimethylphenyl)- piperazine	\-\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	585
14	N-(2-hydroxyethyl)- piperazine	Ş−N N- TFA	525
15	l-(Ethoxy- carbonylmethyl)- piperazine	},	567
16	1-(2-Fluorophenyl)- piperazine		575
17	1-(2-Furoy1)- piperazine	⊱- ∵	575
18	1-(Cyclopentyl)- piperazine		549
19	1-(2-Propyl)- piperazine	\$-~\N	523
20.	N-(2-(1-Piperazino)- acetyl)pyrrolidine	}, ™	592
21	1-(3-Dimethyl- aminopropyl)- piperazine	Ş-N_N-TFA N-	566
22	1-(2-Methoxyethyl)- piperazine	\$\\	539
23	1-(2-Dimethyl- aminoethyl)- piperazine	Ş−N TFA TFA N	552
24	1-(2-Ethoxyphenyl)- piperazine	⊱-∵- -	601
25	1-(4-Fluorphenyl)- piperazine	├ ~\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	575
26	1-(2-Pyridyl)- piperazine	\$\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	558
27	2-(1-piperazinyl)- pyrimidine	>(\)	559
28	4-Piperazino- acetophenone	\-_\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	- 599
29	1-(4-Nitrophenyl)- piperazine	├-\\ ;	602
30	1-(3,5-Dichloropyrid- 4-yl)piperazine	>- \	626
31	4-(2-Methoxyphenyl)- piperidine	⊱ -\	586

32	N-[2-Nitro-4- (trifluoromethyl)- phenyl]piperazine	\$-\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	670
33	<pre>1-[3-(Trifluormethyl)- pyrid-2-yl]- piperazine</pre>	⊱	626
34	cis-3,5-Dimethyl- morpholine	}- √	510
35	N-Propylcyclopropane- methylamine	>-	508
36	1-(2,4-Difluorphenyl)- piperazine	}- -,	593
37	1-(4-Pyridyl)- piperazine	\$\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	558
38	<pre>1-(4-Trifluoromethyl- phenyl)-piperazine</pre>	\$-n\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	625
39	1-Allylpiperazine	{\\\\\\	521
40	1-(2-Pyrazinyl)- piperazine	\$\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	559
41	<pre>1-[3-Chloro-5- (trifluoromethyl)pyrid -2-yl)]piperazine</pre>	Ş−N N− CI N− CF,	660
42	<pre>1-(2-(4-Morpholino)- ethyl)piperazine</pre>	⊱"—"—"—"	594
43	3-Chlorophenyl- piperazine	⊱. ○○	591
44	4-(Hydroxymethyl)- piperidine		510
45	Diisobutylamine	⊱ .(524
46	cis-2,6-Dimethyl- piperazine	>-N NH TFA	509
47	3-Methylpiperidine	⊱ -√	494

48	N,N-Diallylamine	}- :	492
49	<pre>1-[4-(Trifluormethyl)- 2-pyrimidyl]- piperazine</pre>	\ }- ~\\\\\	627
50	1-[4-(Trifluormethyl)- 2-pyridyl]- piperazine	}\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	626
51	N,N,N'-Trimethyl- ethylenediamine	`	497
52	(4-Ethylaminomethyl)- pyridine	},	53,1
53	Methyl- cyclopropylamine	} -₩H	466
54	3,5-Dimethyl- piperidine	}- -\	508
55	3,5-Dimethyl- piperidine	>- -√	508
56	Isobutylamine	\$-NH	468
57	Propylamine	\$-NH	454
58	N-Methyl- isobutylamine	>- <	482

Examples 59-78:

Step 5: Preparation of Resin V

Into a fritted reaction vessel was weighed

resin IVa (100 mg, 0.083 mmol), and the vessel was
capped under nitrogen and cooled to zero degrees
Celsius. A 1.0 M solution of 2-chloro-4,6-dimethoxy1,3,5-triazine in methylene chloride (0.4 mL, 0.4
mmol) was added followed by a 1.0 M solution of Nmethylmorpholine in methylene chloride (0.6 mL, 0.6
mmol). The solutions were stirred for 4 hours at
zero degrees Celsius and warmed to ambient

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temperature. A 0.7 M solution of the appropriate amine to be reacted in methylene chloride (0.4 mL, 0.28 mmol) was added and the reaction mixture stirred for 24 hours. The reaction mixture was stirred for 5 24 hours, then the resin was drained and washed with 1-methyl-2-pyrrolidinone and methylene chloride (4X3 mL each solvent). The reaction was repeated using the same amounts of reagents described above. The reaction was stirred for 4 hours at zero degrees Celsius after the activating step and ambient temperature for 24 hours following amine solution addition. After 24 hours, the resin was drained and washed with 1-methyl-2-pyrrolidinone, 1:1 1-methyl-2pyrrolidinone/water, water, 1:9 acetic acid/water, methanol and methylene chloride (3X3 mL each solvent).

The following hydroxamic acids were synthesized using the indicated polymer-bound acid and the indicated amine in Step 5 followed by release 20 from the polymer using Step 3, before:

Example	Amine	R	MS (M+H)
59	Aniline		488
60	N-Methylaniline		502

61	4-(Trifluoromethyl)- aniline	55 os. 55	6
62	4-Aminopyridine	HN TFA 48	9
63	2-(Trifluoromethoxy)- aniline	,cs, 57	'2
64	2-Chloroaniline	52	:2
65	2-Fluoroaniline	50	16
66	o-Anisole	51	. 8
67	2-(Methylamino)- pyridine	TFA 50	13
68	3-(Trifluoromethoxy)- aniline	°−°. 57	'2
69	3-(Trifluoromethyl)- aniline	55 cr, 55	6
70	3-Chloroaniline	52	2
71	3-Fluoroaniline	50 ~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~)6
72	m-Anisole	51	. 8
73	4-(Trifluoromethoxy)- aniline	9 c _F ,	12
74	4-Aminopyrmidine	19 TFA 49	10
75	4-Fluoroaniline	50	16
76	p-Anisole	51 ~	.8

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Examples 79-88

Step 12: Further Synthesis of Resin III.

5 Into a 8 mL glass vial was placed resin II (200 mg, 0.18 mmol) and cesium carbonate (0.98g mg, 3 mmol) (no cesium carbonate used with piperidine and pyrrolidine nucleophiles). One mL of a 1.8 M solution of the amine nucleophile to be reacted in 1methyl-2-pyrrolidinone (1.8 mmol) was added and the 10 vial was capped and heated to 100 degrees Celsius for 30 hours. Then the vessel was cooled to room temperature, and the resin was drained and washed with 1-methyl-2-pyrrolidinone, 1:1 1-methyl-2-15 pyrrolidinone/water, water, 1:9 acetic acid/water, methanol and methylene chloride (3X3 mL each solvent).

The following hydroxamic acids were

20 synthesized from Resin III using Step 11 with the indicated amines, followed by release from the polymer using the reaction conditions in Step 3.

Example	Amine	R	MS (M+H)
79	1-(2-Methoxyphenyl)- piperidine	├- ○ - -○	475
80	4-(4-Methoxybenzoyl)- piperidine	} ○ᢤ	503
81	Pyrrolidine	}- ⊘	355
82	1-(4-Methoxyphenyl)-2- piperazine	\$,_____\	490
83	1-(2-Fluorophenyl)- piperazine	·	464
84	l-(2,4- Diemthylphenyl)- piperazine	}\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	474
85	1-(2-Methoxyphenyl- piperazine		476
86	<pre>1-(4-Trifluoromethyl- phenyl)piperazine</pre>	}	514
87	1-(2,4- Difluorophenyl)- piperazine	} -√√	482
88	1-(2-Chlorphenyl)- piperazine		480

Example 89: Preparation of N-hydroxy-4 [[4-(4-trifluoromethoxyphenoxy)phenyl]

sulfonyl]-1-(9-fluorenylmethoxy-carbonyl)-4-piperidinecarboxamide

To a solution of 4-[[4-(4-trifluoromethoxyphenoxy)phenyl]sulfonyl]-1-[(1,1diemthylethoxy)carbonyl]piperidinecarboxylic acid (6.25g, 11.5 mmol) prepared using techniques discussed elsewhere herein was added 50% trifluoroacetic acid solution in dichloromethane (100 mL) and stirred 1 hour at room temperature. The solvent was evaporated to afford 9.91 g of an oil. The oil was dissolved in acetonitrile (50 mL) and water (50 mL). To the 10 solution was added sodium carbonate to a pH~9-10 followed by a solution of N-(9-fluorenylmethoxycarbonyloxy) succinimide (3.88 q, 11.5 mmol) in acetone (25 mL). The pH value of the solution was adjusted to 9-10 with sodium carbonate. The reaction mixture was stirred 16 hours. To the reaction 15 mixture was added 2M aqueous hydrochloric acid to a pH value of about 3. The solution was extracted with dichloromethane (3x100 mL). The combined organics were dried over magnesium sulfate, filtered and the solvent evaporated to afford N-hydroxy-4 [[4-(4-20 trifluoromethoxyphenoxy)phenyl] sulfonyl]-1-(9fluorenylmethoxycarbonyl)-4-piperidinecarboxamide (8.15 g) as a yellow oil. MS (ES) m/z 668 (M+H)⁺.

25 Example 90: Preparation of N-hydroxy-4 [[4-(4-trifluoromethylphenoxy)phenyl]sulfonyl]
-1-(9-fluorenylmethoxycarbonyl)-4piperidinecarboxamide

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$$HO \xrightarrow{O_2} CF_3$$

Using the method of Example 89, N-hydroxy4-[[4-(4-trifluoromethyl-phenoxy)phenyl] sulfonyl]-15 (9-fluorenyl-methoxycarbonyl)-4-piperidinecarboxamide
was prepared from 4-[[4-(4-trifluoromethylphenoxy)phenyl]-sulfonyl]-1-[(1,1-dimethylethoxy)carbonyl]piperidinecarboxylic acid, which itself was prepared
using techniques discussed elsewhere herein. MS (ES)
10 m/z 652 (M+H)⁺.

Example 91 Preparation of N-hydroxy-4-[[4-(4-trifluoromethoxyphenoxy)phenyl]
sulfonyl]-1-(phenylcarbonyl)-4piperidinecarboxamide

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Step 1: Preparation of Resin MT-I. To a solution of N-hydroxy-4-[[4-(4-trifluoromethoxy-phenoxy)phenyl]sulfonyl]-1-(9-fluorenylmethoxy-carbonyl)-4-piperidinecarboxamide of Example 89 (11.5 mmol) in dimethylformamide (75 mL) were added resin I (Floyd et al., Tetrahedron Lett. 1996, 37, 8045-8048) (7.0 g, 7.67 mmol), pyBOP (8.0 g) and N-methylmorpholine (5.05 mL), and the mixture was

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stirred with an overhead stirrer 4 days. The resin was filtered and washed with dimethylformamide (3x50 mL), methanol (3x50 mL), dichloromethane (3x50 mL) and ether (3x50 mL). The resin was dried in vacuo to provide resin MT-I.

Resin MT-I was swelled with dimethylformamide (2x100 mL) and drained. To swollen resin MT-1, was added a 20% solution of piperidine in dimethylformamide (100 mL). After 1 hour, the resin was drained and retreated with 20% piperidine in dimethylformamide (100 mL). After 15 minutes the resin was filtered and washed with dimethylformamide (3x100 mL), methanol (3x100 mL), dichloromethane (3x100 mL) and ether (3x100 mL). The resin was dried in vacuo to afford resin MT-II (7.23 g).

Step 3: Preparation of N-hydroxy-4-[[4-(4trifluoromethoxyphenoxy)phenyl]sulfonyl]-1-(phenylcarbonyl)-4-piperidinecarboxamide from Resin MT-II. To a suspension of resin MT-II (250 mg) in 20 dichloromethane (2 mL) was added diisopropylethylamine (165 μ L) and benzoyl chloride (110 μ L) and agitated 3 hours. The resin was filtered and washed with dichloromethane (2x10 mL) and methanol (2x10 mL). To the resin was added a solution of 95% 25 trifluoroacetic acid in water and agitated for 1 hour. The resin was drained and washed with methanol (1x 2 mL) and dichloromethane (1x2 mL). The filtrate was evaporated. The residue was purified by RPHPLC to afford N-hydroxy-4-[[4-(4-trifluoromethoxyphenoxy)phenyl]sulfonyl]-1-(phenylcarbonyl)-4piperidinecarboxamide (9.8 mg) as a solid. MS (ES) m/z 565 $(M+H)^+$.

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Example 92: Preparation of N-hydroxy-4-[[4-(4-trifluoromethylphenoxy)phenyl]
sulfonyl]-1-(phenylcarbonyl)-4piperidinecarboxamide

HOHN S CF₃

N-hydroxy-4-[[4-(4-trifluoromethyl
phenoxy)phenyl] sulfonyl]-1-(phenylcarbonyl)-4piperidinecarboxamide was prepared by the method of
Example 91 from N-hydroxy-4-[[4-(4trifluoromethylphenoxy)phenyl]sulfonyl]-1-(9fluorenylmethoxycarbonyl)-4-piperidinecarboxamide

(the product of Example 90). MS (ES) m/z 549 (M+H)⁺.

Example 93: Preparation of N-(2-tetrahydropyranoxy)
-4-[[4-(4-trifluoromethoxyphenoxy)phenyl]sulfonyl]-4-piperidinecarboxamide

 $\begin{array}{c|c} O & O_2 \\ \hline \\ O & S \\ \hline \\ O & \\ O$

Step 1: Boc deprotection of ethyl 4-[[4-(4-trifluoromethoxyphenoxy)phenyl]sulfonyl]-1-[(1,1-dimethylethoxy)carbonyl]piperidinecarboxylate. To a solution of ethyl 4-[[4-(4-trifluoromethoxy-

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phenoxy)phenyl]sulfonyl]-1-[(1,1-dimethylethoxy)-carbonyl]piperidinecarboxylate (12.58 g, 19.1 mmol; see Example 89) in dichloromethane (50 mL) was added trifluoroacetic acid (50 mL) and the mixture was stirred at room temperature for 1 hour. The reaction mixture was evaporated to afford a pale yellow oil.

Step 2: Cbz protection of step 1. The material from step 1 was dissolved in dichloromethane (200 mL). To this solution was added diisopropylethylamine (33.3 mL) and benzyl chloroformate (5.5 mL) and the mixture was stirred at room temperature for 1 hour. To the reaction mixture was added 2M aqueous hydrochloric acid to a pH value of about 1 and extracted with dichloromethane (2x100 mL). The combined organics were washed with 2M aqueous HCl (1x100 mL) and brine (1x100 mL), dried over magnesium sulfate, filtered and the solvent evaporated to afford a pale yellow oil.

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Step 3: Hydrolysis of the product of step The material prepared in step 2 was dissolved in 20 tetrahydrofuran (100 mL) and ethanol (50 mL). To this solution was added 1M aqueous sodium hydroxide (50 mL) and 50% aqueous sodium hydroxide (10 mL) and stirred 16 hours. To the solution was added 50% aqueous sodium hydroxide (2 mL) and stirred and 25 additional 24 hours. The tetrahydrofuran and ethanol were evaporated. The pH value of the solution was adjusted to pH about 1 with concentrated hydrochloric acid. The reaction mixture was extracted with ethyl acetate (2x100 mL), washed with brine (1x100 mL), 30 dried over magnesium sulfate, filtered and the solvent evaporated to afford a pale yellow oil.

Step 4: Cbz deprotection of step 3. The material prepared in step 3 was dissolved in ethanol (100 mL). This solution was added to 10% palladium on carbon (1.0 g). The solution was placed under 45 psi hydrogen. Additional catalyst was added at 6 hours (1.75 g) and 20 hours (1.0 g 4% Pd/C). After 48 hours the reaction mixture was filtered through a plug of Celite. The filtrate was evaporated and triturated with ether to afford N-(2-10 tetrahydropyranoxy)-4[[4-(4-trifluoromethoxy-phenoxy)phenyl]sulfonyl]-4-piperidinecarboxamide (4.47 g) as a white solid. MS (ES) m/z 545 (M+H)⁺.

Example 94: Preparation of N-(2-tetrahydropyranoxy)-4[[4-(4-trifluoromethylphenoxy)phenyl]sulfonyl]-4-piperidinecarboxamide

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N-(2-tetrahydropyranoxy)-4[[4-(4-trifluoromethylphenoxy)phenyl]sulfonyl]-4-piperidinecarboxamide was prepared by the method of Example 93 starting from ethyl 4-[[4-(4-trifluoromethylphenoxy)phenyl]sulfonyl]-1-[(1,1-dimethylethoxy)carbonyl]piperidinecarboxylate (see Example 90). MS (ES) m/z 529 (M+H)⁺.

Example 95: Preparation of N-hydroxy-4-[[4-(4-trifluoromethylphenoxy)phenyl]
sulfonyl]-1-(2-fluorophenyl-

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carbonyl)-4-piperidinecarboxamide

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To a solution of N-(2-tetrahydropyranoxy)-4 [[4-(4-trifluoromethylphenoxy)phenyl]sulfonyl]-4-piperidinecarboxamide, the product of Example 94, (50 mg) dissolved in dichloromethane (2.5 mL) was added PS-NMM (135 mg, Argonaut) and 2-fluorobenzoyl chloride (12.1 μL) and stirred for 2 hours. To the reaction mixture was added PS-trisamine (50 mg, Argonaut) and the mixture was stirred 1 hour. The reaction mixture was filtered and washed with dichloromethane (2x2 mL) and methanol (1x2 mL). The combined organics were evaporated to afford N-hydroxy-4[[4-(4-trifluoromethylphenoxy)phenyl]-sulfonyl]-1-(2-fluorophenylcarbonyl)-4-piperidinecarboxamide (53.5 mg) as a white solid. MS (ES) m/z 583 (M+H)⁺.

20

Examples 96-124

The following hydroxamic acids were prepared by the method of Example 95 using the appropriate acylating agent.

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96			
96			m/z
	3-fluorophenyl	3-fluorobenzoyl	583 (M+H)
		chloride	
97	4-fluorophenyl	4-fluorobenzoyl	583 (M+H)
		chloride	
98	2-trifluoro-	2-trifluoromethyl-	633 (M+H)
	methylphenyl	benzoyl chloride	
99	3-trifluoro-	3-trifluoro-	633 (M+H)
	methylphenyl	methylbenzoyl	
		chloride	
100	phenylmethyl	phenylacetyl chloride	579 (M+H)
101	2-tolyl	2-toluoyl chloride	579 (M+H)
102	4-tolyl	4-toluoyl chloride	579 (M+H)
103	4-methoxy-	methyl 4-	623 (M+H)
	carbonylphenyl	chlorocarbonyl	•
		benzoate	
104	4-methoxyphenyl	4-anisoyl chloride	595 (M+H)
105	2-thienyl	2-thiophenecarbonyl	571 (M+H)
		chloride	
106	2-furyl	2-furoyl chloride	555 (M+H)
107	3-pyridyl	nicotinoyl chloride	566 (M+H)
108	4-pyridyl	isonicotinoyl	566 (M+H)
		chloride	
109	c-propyl	cyclopropanecarbonyl	529 (M+H)
		chloride	•
110	trichloromethyl	trichloroacetic	622 (M+H)
•	·	anhydride	
111	trifluoromethyl	trifluoroacetic	574 (M+H)
		anhydride	
112	pentafluorophenyl	pentafluorobenzoyl	655 (M+H)
		chloride	
113	4-nitrophenyl	4-nitrobenzoyl	610 (M+H)

		chloride	
114	4-trifluoro-	4-trifluoromethyl-	633 (M+H) ⁺
	methylphenyl	benzoyl chloride	
115	4-trifluoro-	4-trifluoromethoxy-	649 (M+H)*
•	methoxyphenyl	benzoyl chloride	
116	4-methoxy-	4-methoxyphenyl-	609 (M+H) ⁺
	phenylmethyl	acetyl chloride	
117	3-methoxyphenyl	3-anisoyl chloride	595 (M+H)*
118	2-methoxyphenyl	2-anisoyl chloride	595 (M+H) ⁺
119	3,5-	3,5-dimethoxybenzoyl	625 (M+H) ⁺
	dimethoxyphenyl	chloride	٠
120	3,4-	3,4-dimethoxybenzoyl	625 (M+H) ⁺
•	dimethoxyphenyl	chloride	
121	2,5-	2,5-difluorobenzoyl	601 (M+H) ⁺
	difluorophenyl	chloride	
122	methoxy-	methyl malonyl	561 (M+H)+
	carbonylmethyl	chloride	
123	4-dimethyl-	4-dimethylamino-	608 (M+H)*
	aminophenyl	benzoyl chloride	
124	1,1-dimethylethyl	pivaloyl chloride	545 (M+H) ⁺

Examples 125-138

The following hydroxamic acids were

5 prepared by the method of Example of 95 using the appropriate isocyanate as the acylating agent.

Example	RNCO	Isocyanate	MS (ES)
			m/z
125	/=_NCO	Phenyl	
		isocyanate	580 (M+H)
126	5 / NCO	4-Fluorophenyl	
	F—NCO	isocyanate	598 (M+H)
127		4-Methoxybenzyl	
	O-/ NCO	isocyanate	624 (M+H)
128	∕ NCO	Ethyl isocyanate	532 (M+H)
129	F₃Cੑ		
	.)=\	3-Trifluoromethyl	648 (M+H)
	NCO	phenyl isocyanate	
130	O	3-Isocyanate	
	HO NCO	propionic acid	576 (M+H
131	<u></u>	3-Pyridyl	
	N——NCO	isocyanate	581 (M+H
132		4-Chlorophenyl	
	CI—NCO	isocyanate	614 (M+H
133	Ę		
	NCO	3-Fluorophenyl	598 (M+H
		isocyanate	
134			
	NCO	4-Acetylphenyl	622 (M+H
	` <u></u>	isocyanate	
135	/=<	2-Fluorophenyl	598 (M+H
	√NCO	isocyanate	550 (Hrn
136	\ <u>\</u>	4-(Methylthio)	
	's-\NCO	phenyl isocyanate	626 (M+H
137		Benzyl	
	NCO	isocyanate	594 (M+H
138	NC,		
	NCO	3-Cyanophenyl	605 (M+H
		isocyanate	

Examples 140-143

The following hydroxamic acids were prepared by the method of Example 95 using the appropriate acylating agent (electophile) and starting from N-(2-tetrahydropyranoxy)-4[[4-(4-trifluoromethylphenoxy)phenyl]sulfonyl]-4-piperidinecarboxamide, the product of Example 94.

Example	R	Electrophile	MS (ES) m/z
140	OCF3	4-trifluoro- methoxybenzoyl chloride	633 (M+H) ⁺
141	TH CF3	4-trifluoromethyl- phenyl isocyantate	632 (M+H) ⁺
142	SH CF3	4-trifluoro- methylphenyl thioisocyanate	648 (M+H) ⁺
143	0, 0 \\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	4-trifluoromethyl- benzenesulfonyl chloride	653 (M+H) ⁺

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Example 144: Preparation of N-hydroxy-4[[4-(4-trifluoromethylphenoxy)phenyl]
sulfonyl]-1-(4-aminophenylcarbonyl)-4piperidinecarboxamide

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A solution of N-hydroxy-4[[4-(4trifluoromethylphenoxy)phenyl]sulfonyl]-1-(4nitrophenylcarbonyl)-4-piperidinecarboxamide, the

5 product of Example 113, (56.0 mg) dissolved in acetic
acid (2.5 mL) was added to 4% palladium on carbon (20
mg) and placed under 43 psi hydrogen gas for 2.5 h.
The reaction mixture was filtered through a pad of
celite. The solvent was evaporated to afford N10 hydroxy-4-[[4-(4-trifluoromethylphenoxy)phenyl]
sulfonyl]-1-(4-aminophenylcarbonyl)-4piperidinecarboxamide (50.2 mg) as a pale yellow
solid. MS (ES) m/z 580 (M+H)⁺.

15 Example 145: Preparation of N-hydroxy-4[[4-(4-trifluoromethylphenoxy)phenyl]sulfonyl]-1-(4-carboxyphenylcarbonyl)4-piperidinecarboxamide

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To a solution of the product of Example 103 (57 mg) dissolved in tetrahydrofuran (1 mL) and ethanol (1 mL) was added 1M aqueous sodium hydroxide

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solution (1 mL) plus 50% aqueous sodium hydroxide (50 µL) and the reaction mixture was stirred 2 hours. The pH value of the reaction mixture was adjusted to 1 with 6M hydrochloric acid. The solution was extracted with ethyl acetate. The organics were dried over sodium sulfate, filtered and the solvent evaporated. The residue was purified by RPHPLC to afford the acid N-hydroxy-4[[4-(4-trifluoromethyl-phenoxy)phenyl]sulfonyl]-1-(4-carboxyphenylcarbonyl)-4-piperidinecarboxamide (12.8 mg). MS (ES) m/z 631 (M+NH₄)⁺.

Example 146: Preparation of N-hydroxy-4-[[4-(4-methoxyphenoxy)phenyl]sulfonyl]-4-thianecarboxamide

Step 1: Hydrolysis of methyl 4-[[4-(420 methoxyphenoxy)phenyl]sulfonyl]-4-thianecarboxylate.
To a solution of methyl 4-[[4-(4-methoxyphenoxy)phenyl]sulfonyl]-4-thianecarboxylate (10.0 g, 31
mmol) dissolved in tetrahydrofuran (150 mL) was added
potassium trimethylsilanolate (12.1 g) and stirred 2
25 hours. Water was added to the reaction mixture and
extracted with ethyl acetate (2x100 mL). The pH
value of the aqueous layer was adjusted to 2 with 2M
hydrochloric acid and extracted with ethyl acetate
(2x100 mL). The latter organics were washed with
30 brine, dried over magnesium sulfate, filtered and the

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solvent evaporated to afford a pale yellow solid (8.20 g).

Step 2: Loading on resin. The compound obtained in step 1 (4.0 g, 13.1 mmol) was dissolved in 1-methyl-2-pyrrolidinone (15 mL) and added to a suspension of resin I (6.0 g, 6.6 mmol; Preparative Example XI) in 1-methyl-2-pyrrolidinone (40 mL). To this solution were added pyBOP (6.85 g) and N-methylmorpholine (2.9 mL), and the mixture was stirred with overhead stirring 16 hours. The resin was filtered and washed with dimethylformamide (3x50 mL), methanol (3x50 mL), dichloromethane (3x50 mL) and ether (3x50 mL). The resin was dried in vacuo to provide resin MT-III (6.79 g).

15 Step 3: Aryl fluoride displacement of resin MT-III. A suspension of resin MT-III (200 mg, 0.17 mmol), 1-methyl-2-pyrrolidinone (2 mL), cesium carbonate (560 mg) and 4-methoxyphenyl (306 mg) were stirred at 105 °C for 16 hours. The reaction mixture 20 was cooled and the resin filtered. The resin was washed with dimethylformamide (3x5 mL), methanol (3x5 mL), 10% aqueous acetic acid (3x5 mL), methanol (3x5 mL) and dichloromethane (3x5 mL). To the resin was added 95% aqueous trifluoroacetic acid and the reaction mixture was agitated for 1 hour. The resin 25 was drained and washed with dichloromethane (2x1 mL). The solvent was evaporated. The residue was purified by RPHPLC to provide N-hydroxy-4-[[4-(4-methoxyphenoxy)phenyl]sulfonyl]-4-thianecarboxamide (17.9) 30 mg) as a pale yellow oil.

Examples 147-151

The following hydroxamic acids were prepared by the method of Example 146 using the appropriate alcohol.

Example	R	Alcohol	MS (ES) m/z
147	4-trifluoro-	4-trifluoro-	495 (M+NH ₄) ⁺
	methoxyphenyl	methoxyphenol	•
148	4-isopropyl- phenyl	4-isopropylphenol	453 (M+NH ₄) ⁺
149	3-pyridyl	3-hydroxypyridine	395 (M+H) ⁺
150	3,4-dimethoxy- phenyl	3,4-dimethoxyphenol	471 (M+NH ₄) ⁺
151	4-pyridyl	4-hydroxypyridine	395 (M+H)+

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Examples 152-155

The following hydroxamic acids were prepared by the method of Example 146 using the appropriate amine.

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Example	R	Amine	MS (ES)
			m/z
152	4-(4-fluoro- benzoyl) piperidyl	4-(4-fluorobenzoyl)- piperidine	507 (M+H) ⁺
153	4-(2-methoxy- phenyl) piperidyl	4-(2-methoxyphenyl)- piperidine	491 (M+H) ⁺
154	x, N	N-cyclopropyl- methyl-N-methyl-4- piperidine	496 (M+H) ⁺
155	O pyrrolidinyl	carboxamide pyrrolidine	371 (M+H) ⁺

Example 156: Preparation of N-hydroxy-4-[[4-(4-methoxyphenoxy)phenyl]sulfonyl]-4-thianecarboxamide-1,1-dioxide

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Step 1: Oxidation of Resin MT-III. A

10 suspension of resin MT-III (2.0 g, 1.72 mmol), mchloroperbenzoic acid (4.37 g) and dichloromethane
(25 mL) was stirred at room temperature for 20 hours.
The resin was filtered and washed with
dichloromethane (3x25 mL), dimethylformamide (3x25

15 mL), methanol (3x25 mL), lM aqueous sodium
bicarbonate (2x25 mL), methanol (3x25 mL),
dichloromethane (3x25 mL) and ether (3x25 mL). The
resin was dried in vacuo to afford resin MT-IV
(2.16 g).

Step 2: Aryl fluoride displacement of resin MT-IV. N-hydroxy-4-[[4-(4-methoxyphenoxy)-phenyl]sulfonyl]-4-thianecarboxamide 1,1-dioxide was prepared by the method of Example 146 using resin MT-IV in the place of resin MT-III. ES (MS) m/z 473 (M+NH₄)⁺.

Examples 156-160

The following hydroxamic acids were

10 prepared by the method of Example 156 using the appropriate alcohol.

Example	R	Alcohol	MS (ES) m/z
157	4-trifluoro-	4-trifluoro-	527 (M+NH ₄)
	methoxyphenyl	methoxyphenol	
158	4-isopropylphenyl	4-isopropylphenol	485 (M+NH ₄)
159	3-pyridyl	3-hydroxypyridine	427 (M+H) ⁺
160	4-pyridyl	4-hydroxypyridine	427 (M+H) ⁺

Example 161

The following hydroxamic acids were prepared by the method of Example 156 using the appropriate amine.

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Example	R	Amine	MS (ES) m/z
161	4-(4-fluorobenzoyl)	4-(4-fluoro-	539 (M+H) ⁺
	piperidyl	benzoyl)-	
		piperidine	•

Example 162: Preparation of N-hydroxy-4-[[4-[4-[4-[(3,5-dimethylpiperidyl)carbonyl]-piperidyl]phenyl]sulfonyl]-4-thianecarboxamide

10 Step 1: Aryl fluoride displacement of Resin MT-III. To a suspension of resin MT-III (4.06 g, 3.4 mmol) in 1-methyl-2-pyrrolidinone (40 mL) was added ethyl isonipecotate (5.25 mL), and the mixture was heated to 100 °C for 16 hours. The cooled

15 reaction mixture was filtered and the resin was washed with methanol (3x25 mL), dichloromethane (1x10 mL) and ether (3x25 mL). The resin was dried in vacuo to afford resin MT-V (4.21 g).

Step 2: Hydrolysis of resin MT-V. To a

20 suspension of resin MT-V (4.13 g) in tetrahydrofuran

(20 mL) was added 4M aqueous potassium hydroxide (10 mL) and stirred at room temperature for 5 days. The

resin was filtered and washed with methanol (3x25 mL), dichloromethane (3x25 mL) and ether (3x25 mL). The resin was dried in vacuo to afford resin MT-VI.

Step 3: Conversion to amide. suspension of resin MT-VI (268 mg) in 1-methyl-2pyrrolidinone (2 mL) were added 3,5-dimethylpiperidine (299 μL), pyBOP (587 mg) and diisopropylethyl amine (393 µL), and mixture was stirred 40 hours. The resin was filtered and washed with dimethylformamide (3x2 mL), methanol (3x2 mL), 10 10% aqueous acetic acid (3x2 mL), methanol (3x2 mL), dichloromethane (3x2 mL) and glacial acetic acid (1x2 mL). The resin was treated with 95% aqueous trifluoroacetic acid (2 mL) and agitated 1 hour. 15 resin was washed with dichloromethane (2 mL) and methanol (2 mL). The filtrate was evaporated. residue was purified by RPHPLC to afford N-hydroxy-4-[[4-[4-[(3,5-dimethylpiperidyl)carbonyl]piperidyl] phenyl]sulfonyl]- 4-thianecarboxamide (7.5 mg) MS 20 $(ES) m/z 524 (M+H)^{+}$.

Example 163: Preparation of N-hydroxy-4-[[4-[4-[4-[(3,5-dimethylpiperidyl)carbonyl]-piperidyl]phenyl]sulfonyl]-4-thianecarboxamide

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N-hydroxy-4-[[4-[4-[(3,5-dimethyl-piperidyl)carbonyl]piperidyl]phenyl]sulfonyl]-4-thianecarboxamide was prepared by the method of using cis-2,6-dimethylmorpholine as the amine. MS (ES) m/z 526 (M+H)⁺.

Example 164: N-hydroxy-4[[[4-[4-(4-fluorophenyl)methoxy]piperidyl]phenyl]sulfonyl]-1tetrahydropyrancarboxamide

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Step 1: Preparation of amine 4-(4fluorophenyl) methoxy piperidine. Ninety-five percent dry sodium hydride is weighted in a 25 mL vial. Boc-(4-hydroxy)-piperidine (1g, 4.97 mmol) in 10 mL of dimethyl formamide is added and the reaction mixture is stirred at room temperature for 15 minutes 4fluoro benzyl bromide (1.4g, 7.5 mmol) is added and the reaction mixture is stirred at room temperature for 16hours, then guenched with water and diluted with ethyl acetate. The organic layer was washed with brine, then dried over MgSO4, and concentrated in The crude product was purified by flash column chromatography on silica gel eluting with ethyl acetate:hexane 1:10. The Boc-protected amine is dissolved in 3 mL of dichloromethane and 3 mL of trifluoroacetic acid and the reaction mixture is stirred at room temperature for 16 hours and the solvent is evaporated to give 1.8 g of 4-(4fluorophenyl)-methoxy piperidine. MS: M+H=210.1319.

Step 2: Preparation of N-hydroxy-4 [[[4-[4-(4-fluorophenyl)methoxy] piperidyl] phenyl]sulfonyl]-1-tetrahydropyrancarboxamide. To a solution of N-tetrahydropyranoxy-4-fluorophenyl-

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sulfonyl-1-tetrahydropyrancarboxamide (100 mg, 0.26 mmol) in 1.5 mL of DMA are added the amine from step 1 (0.52 mmol, 2 eq.) and cesium carbonate (420 mg, 1.29 mmol). The reaction mixture is stirred at 100 °C 5 for 48 hours. The reaction is treated with water and filtered through Celite eluting with dichloromethane. The solvent was evaporated and the residue is dissolved in 2 mL of 4M HCl in dioxane. The mixture is stirred at room temperature for 1 hour and 1 mL of 10 methanol is added. After stirring 15 minutes at room temperature, the solvent is evaporated and the residue was purified by RPHPLC eluting with 10% to 90% acetonitrile/water to give N-hydroxy-4-[[[4-[4-(4-fluorophenyl)methoxy]piperidyl]phenyl]sulfonyl]-1tetrahydropyrancarboxamide. MS: M+H= 493.1792. 15

Examples 165-181

The following hydroxamic acids were synthesized by the procedure of Example 164:

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Example	Halide starting	R	HI RES MS
	material		M+H
165	benzyl bromide	بر ک	475.1913
166	ethyl iodide	<u></u> ኢ^	413.1764

167	4-fluoro benzyl bromide	,1; F	493.1792
168	iodopropane	≒ ~	427.1918
169	3,5-dimethyl benzyl bromide	7,	144.1391
170	4-chloro benzyl bromide	h. C	509.1515
171	3-methyl benzyl bromide	77	489.2059
172	4-methyl benzyl bromide	15	489.2074
173	3-trifluoro- methoxy benzyl bromide	TY OCE	559.1738
174	2-trifluoro- methyl benzyl bromide	الراب CF3 الراب الرا	543.1780
175	4-trifluoro- methoxy benzyl bromide	h Oce	559.1730
176	3,4-dichloro- benzyl bromide	***************************************	543.1155
177	3-trifluoro- methyl benzyl bromide	7.	543.1779
178	3,5-dimethoxy- benzyl bromide	7.	535.2120
179	3,4-difluoro- benzyl bromide	:\	511.1705
180	4-cyano- benzyl bromide	-TC CN	500.1835

5

181 2-phenyl benzyl bromide

551.2196

Example 182: N-hydroxy-4-[[[4-[3-(4-fluorophenyl)-methoxy]piperidyl] phenyl]sulfonyl]-1-tetrahydropyrancarboxamide

N-hydroxy-4[[[4-[3-(4-fluorophenyl)-

10 methoxy]piperidyl]phenyl]sulfonyl]-1-tetrahydropyrancarboxamide is prepared by the method of Example
164 starting from Boc-(3-hydroxy)-piperidine in step
1.

15 Examples 183-184

The following hydroxamic acids were synthesized using a procedure similar to that of Example 182:

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Example	Halide starting material	R	HI RES MS
183	4-fluroro benzyl bromide	نام المارية ا	M+H=475.1913
184	benzyl bromide	77	M+H=551.2196

5 Example 185: N-hydroxy-4[[[4-(4-phenoxy)piperidyl]phenyl]sulfonyl]-1tetrahydropyrancarboxamide

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N-hydroxy-4[[[4-(4-phenoxy)piperidyl] phenyl]sulfonyl]-1-tetrahydropyrancarboxamide is prepared by the method of Example 164 starting from 4-phenoxypiperidine in step 2.

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Examples 186-187

The following hydroxamic acids were synthesized using a procedure similar to that of Example 185:

Example	Amine starting material	R	HI RES MS
186		Н	M+H=461.1749
187	HNOO	3,5-dimethyl	M+H=489.2065

Example 188: Preparation of N-hydroxy-4[[[4-[(3-trifluoromethyl)phenylcarbamoxy]-piperidyl]phenyl]sulfonyl]-1-tetrahydropyrancarboxamide

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Step 1: A solution of N-tetrahydro-10 pyranoxy-4-fluorophenylsulfonyl-1-tetrahydropyrancarboxamide (1 g, 2.58 mmol), 4-hydroxypiperidine (392 mg, 3.87 mmol) and cesium carbonate (2.52g, 7.74 mmol) in 20 mL of NMP is stirred at 100 °C for 48 hours. The reaction mixture is treated with 15 water and neutralized to pH 4 with 5% aqueous HCl. The aqueous layer is extracted twice with ethyl acetate and the combined organic layer is dried using magnesium sulfate and concentrated in vacuo. The 20 crude product was purified by flash column chromatography on silica gel eluting with ethyl acetate:hexane 1:10 to give N-tetrahydropyranoxy-4-[[(4-hydroxypiperidyl) phenyl] sulfonyl]-1tetrahydropyrancarboxamide. MS: M+Na= 491.2.

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Step 2: To a solution of alcohol N-

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tetrahydro-pyranoxy-4[[(4-hydroxypiperidyl)phenyl]sulfonyl]-1-tetrahydropyrancarboxamide (50 mg, 0.107 mmol) in 2 mL of dichloromethane is added alpha, alpha, alpha-trifluoro-M-tolyl isocyanate (21 mg, 0.112 mmol). The reaction mixture is stirred for 16 hours at room temperature and 21 mg of alpha, alpha, alpha-trifluoro-m-tolyl isocyanate is added. The mixture is stirred 48 hours at room temperature and treated with water. The solvent is 10 evaporated and the residue is dissolved in 2 mL of 4M HCl in dioxane. The mixture is stirred at room temperature for 1 hour and 1 mL of methanol is added. After stirring 15 minutes at room temperature the solvent is evaporated and the residue was purified by RPHPLC eluting with 10% to 90% acetonitrile/water to give N-hydroxy-4-[[[4-[(3-trifluoromethyl)phenylcarbamoxy]piperidyl]phenyl]sulfonyl]-1-

20-

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Examples 189-191

The following hydroxamic acids were synthesized using a procedure similar to that of Example 188:

tetrahydropyrancarboxamide. MS: M+Na= 594.1.

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Example	Isocyanate starting material	R	MS
189	alpha,alpha,alpha- trifluoro-M-tolyl isocyanate	-}{_CF ₃	M+Na=594.1
190	4-ethoxyphenyl isocyanate	-><_>	M+Na=570.2
191	4-fluorophenyl isocyanate	- }	M+H=522.1742

Example 192: Preparation of N-hydroxy-4[[4-(4-trifluoromethoxyphenoxy)-phenyl]-sulfonyl]-1-[[(2-trifluoromethoxy)-phenyl]-sulfonyl-4-piperidinecarboxamide

N-hydroxy-4[[4-(4-trifluoromethoxyphenoxy)
10 phenyl]sulfonyl]-1-[[(2-trifluoromethoxy)phenyl]
sulfonyl-4-piperidinecarboxamide can be prepared

using the method of Example 93 starting from 2
trifluoromethoxybenzene sulfonyl chloride.

15 Examples 193-197

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The following hydroxamic acids were synthesized using a procedure similar to that of Example 192:

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Example	Sulfonyl chloride	R	MS
,	starting material		
193	2-trifluoro-	,CF ₃	M+NH4=
	methoxybenzene	٩	702.1003
	sulfonyl chloride	-}	
194	benzene	, /	M+NH4=
	sulfonyl chloride	-}{_}	618.1216
195	alpha-toluenesulfonyl		M+NH4=
	chloride	/ <u>~</u> (_)	632.1337
196	3-trifluoro-	,CF3	M+NH4=
	methylbenzene	√	686.1027
	sulfonyl chloride	-3/	
197	3-trifluoromethane	5	M-H= 591.1
	sulfonyl chloride	_{	

Example: 198

5

 $\label{lem:n-hydroxy-4} N-hydroxy-4 \mbox{$[$ 4-(4-trifluoromethoxyphenoxy)-phenyl]sulfonyl]-1-(N-methylthiourea)-4-}$

10 piperidinecarboxamide was prepared by the method of

5

Example 192 starting with methyl isothiocyanate. M+H= 534.0977.

Examples 199-202

The following hydroxamic acids were synthesized using the procedure of Example 198:

Example	Sulfonyl chloride	R	MS
	starting material		M+H
199	2-morpholinoethyl isothiocyanate	22~N	633.1643
200	2-piperidinoethyl isothiocyanate	, _{t,} N	653.1694
201	pyridine-3- isothiocyanate	-}{ = N	.597.1094
202	4-dimethylaminophenyl isothiocyanate	-\$-{__N	639.1526

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Example 203: Preparation of 1,1-dimethylethyl-3,6-dihydro-4-[2-(trifluoromethyl)phenyl]-1(2H)-pyridinecarboxylate

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Part A: An oven-dried 1.0 liter flask fitted with a thermometer and nitrogen inlet was charged with 55 mL of a 2 M solution of lithium diisopropoylamide in tetrahydrofuran and 50 mL of 10 tetrahydrofuran. The flask was immersed in a dry ice/acetone bath. When the temperature of the solution was less than -70 degrees, a solution of Nt-butoxycarbonylpiperidinone (20.0 g, 0.1 mole) in 100 mL tetrahydrofuran was added dropwise, 15 maintaining the temperature less than -65 degrees. After complete addition, the flask was stirred with cooling for 20 minutes. Then a solution of Ntrifluoromethanesulfonimide (38.2 g, 0.107 mole) was added drop-wise maintaining the temperature less than 20 -65 degrees. After complete addition, the dry ice/acetone bath was swapped with an ice/water bath. The reaction was stirred overnight (about eighteen hours), slowly warming to room temperature. After 16 hours, the solvent was removed in vacuo, and the 25 residue was purified by column chromatography on neutral alumina, yielding 26.53 g of product as a yellow oil. Electrospray mass spectroscopy showed m/z 332 (M+H).

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Part B: A three-necked 15 mL round-bottom flask was charged with the product from Part A (6 g, 18.1 mmol), o-trifluorobenzeneboronic acid (4.94 g, 26 mmol), lithium chloride (2.34 g, 55 mmol), 2 M sodium 5 carbonate (26 mL, 52 mmol) and ethylene glycol dimethyl ether (60 mL). Nitrogen was bubbled through the solution for 10 minutes, then palladium tetrakistriphenylphosphine (1.06 g, 0.92 mmol) was added. The mixture was heated to reflux for 1.5 hours, then cooled to room temperature. The solvent 10 was removed in vacuo, then the residue was partitioned between 100 mL of methylene chloride and 100 mL of 2 M sodium carbonate with 3 mL concentrated ammonium hydroxide. The aqueous layer was extracted with an additional 100 mL methylene chloride, then 15 the combined organic layers were dried over magnesium sulfate and concentrated to give 8.42 g of crude product as a dark brown oil. Purification via flash column chromatography (10% ethyl acetate3/hexanes) yielded 2.76 g of pure product as a yellow oil. 20 Electrospray mass spectroscopy showed m/z 328 (M+H).

Example 204: Preparation of 1,2,3,6-tetrahydro-4[2-trifluoromethyl)phenyl]pyridine

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The title compound of Example 203 (300 mg, 0.92 mmol) was dissolved in methylene chloride (5 mL) in a 15 mL round-bottom flask, and 5 mL of

-381-

trifluoroacetic acid was added dropwise. After 15 minutes, the solvent was removed in vacuo, and the residue partitioned between 20 mL of ethyl acetate and 20 mL of 2 M sodium carbonate. The organic layer was washed with additional 2 M sodium carbonate, dried over magnesium carbonate and concentrated in vacuo to yield 195 mg of pure product as a colorless oil. Electrospray mass spectroscopy showed m/z 228 (M+H).

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Example 205: Preparation of 4-[2-(trifluoromethyl) phenyl]piperidine

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Part A: A solution of the title compound of Example 203 (2.3 g, 7 mmol) in 20 mL ethanol was added to a hydrogenation flask containing 1 g of 4% palladium on carbon (0.38 mmol). The mixture was placed under 100 PSI hydrogen and heated to 50 degrees Celsius for 5 hours. Then the mixture was cooled to room temperature and filtered through Celite. The filtrate was concentrated in vacuo to give 2.27 g of pure product as a colorless oil. Electrospray mass spectroscopy showed m/z 330 (M+H).

part B: The product from Part A above (2.24 g, 6.8 mmol) was dissolved in 100 mL methylene chloride, and 100 mL of trifluoroacetic acid was added dropwise. After 15 minutes, the solvent was removed in vacuo, and the residue partitioned between

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100 mL of ethyl acetate and 100 mL of 2 M sodium carbonate. The organic layer was washed with additional 2 M sodium carbonate, dried over magnesium carbonate and concentrated in vacuo to yield 1.12 g of pure product as a colorless oil. Electrospray mass spectroscopy showed m/z 230 (M+H).

Example 206: General Description for Preparation of
Hydroxamic Acids via Aryl
Fluoride Displacement with Amines.

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Part A: A 2 dram vial was charged with aryl fluoro compound of Preparative Example IV (170 mg, 0.44 mmol), 1 ml of 2-methylpyrrolidinone, cesium carbonate (360 mg, 1.1 mmol) and 0.66 mmol of an 15 amine. A small magnetic stirring bar was added, then the vial was capped and placed in a Pierce Reactitherm at 115 degrees Celsius. The reaction progress was followed by analytical HPLC. When the reaction was greater than 90% complete, the vial was cooled to 20 room temperature. The reaction mixture was diluted with 5 mL of water, then 1.2 mL of 5% hydrogen chloride/water was added dropwise. Then, the entire mixture was poured onto a column of Celite. column was washed exhaustively with ethyl acetate 25 (30-40 mL) and the filtrate was collected and concentrated to give the crude products.

Part B: The product from above was dissolved in 2 mL 1,4-dioxane and 2 mL of methanol in a 4 dram vial with a small magnetic stirring bar. A solution of 4 N hydrogen chloride in 1,4-dioxane was carefully added to the reaction, and the mixture was stirred for 2 hours. Then the solvent was removed in

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vacuo and the residue purified by preparative
reversed-phase HPLC.

Examples 207-214

The following hydroxamic acids were prepared using the method described above in Example 106 with the indicated amine as the starting material.

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			m/z from
Example	amine	R	electrospray
	•		mass spectroscopy
		F ₃ C	spectroscopy
	Decades a E		
	Product of	>	
207	Example 205	> \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	513.3 (M+H)
		F₃Cੑ	
	Product of		
208		}_n'	511.2 (M+H)
208	Example 204		311.2 (MTD)
,			
209	piperidine	}-n'	369.2 (M+H)
203	biberiaine	` \	303.2 ()
		, /	
210	tetrahydro-	}-v >>	367.2 (M+H)
	piperidine		
	piperiume	_	
		~	
	•	, /\ \\	•
211	4-(2-keto-	⟨-N	501 (M+H)
	benzimid-		
	azolinyl)-		
		_	•
	piperidine		

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212	hexamethyl- eneimine	}-n-	383.2 (M+H)
213	1-methylhomo- piperazine	} —NN	398.2 (M+H)
214	1,3,3- trimethyl-6- azabicyclo- [3.2.1]octane	}-n	437.3 (M+H)

Examples 215-223

Using the procedures outlined in Examples
5 203, 204, 206 and other methods outlined above, the
following analogs are made from the indicated boronic
acid:

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Example	Boronic acid	R
215	B(OH) ₂ OCF ₃	CF ₃
216	B(OH) ₂	

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Example 224: Preparation of Tetrahydro-N-hydroxy-4[[4-(pentaflourooxy)phenyl]sulfonyl]2H-thiopyran-4-carboxamide

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Part A: To a solution of the product of
Preparative Example IV (2.5 g, 6 mmol) in
dimethylformamide (50 mL) was added 4pentafluroethyloxy phenol (2.0 g, 6 mmol) followed by
cesium carbonate (5 g, 12 mmol). The reaction was
heated at eighty degrees Celsius for twelve hours.
Stripping the dimethylformamide in vacuo afforded a
brown solid (5.5 g). The product was dissolvent in
ethylacetate (150ml) and extracted with water, brine
and dried over sodium sulfate. The ¹H NMR, MS, and
HPLC was consistent with desired compound.

Part B: To the product of part A, crude THP
20 protected hydroxamate was disolved in acetonitrile/

water (40 ml) was slowly added 10% aq HCl (10 ml).

After stirring two hours, the acetonitrile was

stripped. The resultant precipitate was collected,

giving the title compound as a white solid (2.1 g).

25 The ¹H NMR, MS, and HPLC was consistent with desired

compound. This solid was recrystallized from

ethylacetate and hexanes (1.8g). The ¹H NMR, MS, and

HPLC was consistent with desired compound. MS (CI)

M+H calculated for C₂₃H₂₇BrNO₆S: 511, found 511.

Example 225: Preparation of Tetrahydro-4-[[4(pentaflourooxy)phenyl]sulfonyl]-2Hthiopyran-4-carboxamide

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Part A. The product of Preparative Example V

(2.5 g) was dissolved in methanol (60 mL). To this

solution ammonium formate (3 g) was added, followed
by Pd on charcoal 20% catalyst. The mixture was
heated to reflux for 24 hour. After complete
reaction the mixture was cooled filtered through a
plug of Celite and the solvent removed under reduced

pressure to give pure amide (1.7g). The ¹H NMR, MS,
and HPLC was consistent with desired compound. MS

(CI) M+H calculated for C₂₃H₂₇BrNO₆S: 445, found 445.

Example 226: Preparation of 4-(4-pyridyloxy) thiophenol hydrochloride:

Part A: Phenol (1500 g, 15.9 mol) and 425 chloropyridine hydrochloride (800 g, 7.1mol) were
combined in a melt at 150°C under a nitrogen
atmosphere. After fifteen hours, the reaction was

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dissolve in 3N sodium hydroxide solution (5400 mL) and extracted with methylene chloride (4X). The organic extracts were combined, washed with 1N sodium hydroxide solution, water and brine, dried over Na₂SO₄, filtered, and concentrated *in vacuo*. The isolated oil was dissolved in hexanes (1000 mL) and cooled to -60°C. The precipitate was collected and dried *in vacuo* to yield 452 g (38%) of the 4-phenoxypyridine as a white solid.

10 Part B: A solution of the 4-phenylpyridine from part A (400 g, 2.3 mol) in 1,2-dichloroethane (1250 mL) was cooled in an ice bath under a nitrogen atmosphere and treated with chlorosulfonic acid (400 mL, 6.0 mol). The reaction temperature was held below 12°C during the addition. The reaction was then heated to 45°C for 15 hours. The standard work-up procedure afforded 270 grams (40%) of the desired 4-[(pyrid-4-yl)oxy]benzenesulfonic acid.

Part C: A slurry of the sulfonic acid part B

20 (420 g, 1.5 mol) in acetonitrile (2500 mL) and DMF

(40 mL) was warmed to 75°C under a nitrogen atmosphere
and treated with thionyl chloride (243 mL, 3.3 mol)
added dropwise over 3 hours. After stirring for onehalf hour, the standard work-up procedure afforded

25 483 grams (100%) of the desired 4-[(pyrid-4yl)oxy]benzenesulfonyl chloride hydrochloride.

Part D: A solution of triphenylphosphine (65.6 g, 250.28 mmol) in dry methylene chloride (240 mL) was cooled to zero degrees C in an ice-water bath, then treated with dimethylformamide (3.4 mL, 3.2 g, 43.40 mmol). The reaction mixture was then treated with the sulfonyl chloride from part C (25.5 g, 83.43

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mmol), added as a solid over one-half hour. After
two hours in the ice bath, the reaction was treated
with 1 N aqueous hydrochloric acid solution (150 mL)
and stirred vigorously for one hour. The layers were
separated and the aqueous layer was extracted with
methylene chloride (1X). The aqueous layer was
concentrated in vacuo to yield 17.9 grams (90%) of
the 4-(4-pyridyloxy)thiophenol hydrochloride as a tan
solid, m/z = 204 (M + H).

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Example 227: Preparation of

Part A: A solution of 4-(4-pyridyloxy)-15 thiophenol (2.0 g, 8.34 mmol) and tertbutylbromoacetate (1.2 mL, 1.6 g, 8.34 mmol) in dry methanol (30 mL) was cooled to zero degrees C and treated with triethylamine (2.4 mL, 1.8 g, 17.52 mmol). The addition was done at a rate which held 20 the reaction temperature below 10°C. The ice bath was removed and after two hours at ambient temperature, the reaction was concentrated in vacuo. The residue was partitioned between ethyl acetate and saturated sodium bicarbonate, the layers were separated and the 25 aqueous layer was extracted with ethyl acetate (2X). The organic extracts were combined, washed with water and brine, dried over Na₂SO₄, filtered, and concentrated in vacuo to yield 2.3 grams of the tert10

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butyl ester of the sulfide acid suitable for the next step.

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Part B: To a solution of the tert-butyl ester of the sulfide acid from part A (2.3 g, 7.25 mmol) in dry anisole (85 mL, 8.1 g, 74.67 mmol) was added trifluoroacetic acid (25.5 mL, 37.7 g, 330.6 mmol). After one-half hour at ambient temperature, the reaction was concentrated in vacuo to 3.7 g of the TFA salt of the sulfide acid suitable for the next step.

Part C: To a solution of the TFA salt of the acid obtained from part B (2.7 g, 7.19 mmol) in dimethylformamide (10 mL) was added Nhydroxybenzotriazole hydrate (1.5 g, 10.79 mmol), Nmethylmorpholine (4.7 mL, 4.4 g, 43.16 mmol), O-(tetrahydro-2H-pyran-2-yl)hydroxylamine (2.5 g, 21.58 mmol), and 1-(3-dimethylaminopropyl)-3ethylcarbodiimide hydrochloride (1.8 g, 9.35 mmol). After sixteen hours at ambient temperature, the reaction was concentrated in vacuo. The residue was 20 partitioned between ethyl acetate and saturated sodium bicarbonate, the layers were separated and the aqueous layer was extracted with ethyl acetate (3X). The organic extracts were combined, washed with brine, dried over Na₂SO₄, filtered, and concentrated 25 in vacuo. Chromatography (on silica, methanol-ethyl acetate/hexanes) afforded 2.1 g (81%) of the THP sulfide hydroxamate as a dry, white foam, m/z = 361(M + H).

Part D: To a solution of the THP sulfide 30 hydroxamate from part C (2.1 g, 5.83 mmol) in methanol/water (13 mL/2 mL) was added tetrabutylammonium Oxone (5.8 g, 61.29 mmol). After

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2 days at ambient temperature, the reaction was concentrated in vacuo. The residue was partitioned between ethyl acetate and saturated sodium bicarbonate, the layers were separated and the aqueous layer was extracted with ethyl acetate (6X). The organic extracts were combined, washed with water and brine, dried over Na₂SO₄, filtered, and concentrated in vacuo. Chromatography (on silica, methanol-ethyl acetate/hexanes) afforded 0.9 g (40%) of the THP sulfone hydroxamate as a dry, white foam, m/z = 393 (M + H).

Part E: To a slurry of the THP sulfone hydroxamate from part D (0.9 g, 2.29 mmol) in methanol (0.6 mL) was added 4N HCl dioxane solution (6 mL). After one hour at ambient temperature, the reaction mixture was slowly poured into diethyl ether (200 mL). Filtration afforded 0.6 grams (78%) of the title compound as a white solid, m/z = 309 (M + H).

20 Example 228: Preparation of

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Part A: A solution of 4-(4-pyridyloxy)
thiophenol (18.0 g, 75.08 mmol) and tertbutylbromoacetate (10.5 mL, 13.9 g, 71.33 mmol) in
dry methanol (250 mL) was cooled to 0°C and treated
with triethylamine (22.0 mL, 16.0 g, 157.68 mmol).
The addition was done at a rate which held the

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reaction temperature below 1°C. The ice bath was removed and after one-half hour at ambient temperature, the reaction was concentrated in vacuo. The residue was partitioned between ethyl acetate and saturated sodium bicarbonate, the layers were separated and the aqueous layer was extracted with ethyl acetate (2X). The organic extracts were combined, washed with water and brine, dried over Na₂SO₄, filtered, and concentrated in vacuo to yield 21.7 grams of the tert-butyl ester of the sulfide acid suitable for the next step.

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Part B: To a solution of the tert-butyl ester of the sulfide acid from part A (221.7 g, 68.37mmol) in dry anisole (76.5 mL, 76.1 g, 704.12 mmol) was added trifluoroacetic acid (240 mL, 355 g, 3,117 mmol). After one hour at ambient temperature, the reaction was concentrated in vacuo to yield 34.7 g of the TFA salt of the sulfide acid suitable for the next step.

Part C: To a solution of the TFA salt of the 20 sulfide acid from part B (34.7 g, 68.37 mmol) in dry methanol (100 mL) was added thionyl chloride (7.5 mL, 12.2 g, 102.5 mmol). After twelve hours at ambient temperature, the reaction was concentrated in vacuo. The residue was partitioned between ethyl acetate and 25 saturated sodium bicarbonate, the layers were separated and the aqueous layer was extracted with ethyl acetate (3X). The organic extracts were combined, washed with water and brine, dried over Na₂SO₄, filtered, and concentrated in vacuo to yield 30 18.7 grams of the methyl ester of the sulfide acid suitable for the next step.

Part D: To a solution of the methyl ester of the sulfide acid obtained from part C (18.7 g, 67.92 mmol) in methylene chloride (325 mL) was added tetrabutylammonium Oxone (193 g, 543.4 mmol). After 2 days at ambient temperature, the reaction was concentrated in vacuo. The residue was partitioned between ethyl acetate and saturated sodium bicarbonate, the layers were separated and the aqueous layer was extracted with ethyl acetate (9X). The organic extracts were combined, washed with water 10 and brine, dried over Na₂SO₄, filtered, and concentrated in vacuo. Chromatography (on silica, methanol-ethyl acetate/hexanes) afforded 7.3 g (35%) of the methyl ester of the sulfone acid as a dry, white foam, m/z = 308 (M + H). 15

Part E: To a solution of the methyl ester of the sulfone acid obtained from part D (2.7 g, 8.79 mmol) in dry dimethylformamide (20 mL) was added 18crown-6 ether (0.5 g, 1.90 mmol) and potassium carbonate (4.9 g, 35.14 mmol). The reaction slurry 20 was treated with bis-(2-bromoethyl)ether (1.1 mL, 2.0 g, 8.79 mmol) and then heated to 60°C. After fifteen hours at 60°C, the reaction was concentrated in vacuo. The residue was partitioned between ethyl acetate and water, the layers were separated and the aqueous 25 layer was extracted with ethyl acetate (3X). The organic extracts were combined, washed with brine (3%), dried over Na₂SO₄, filtered, and concentrated in vacuo. Chromatography (on silica, NH3-methanol-ethyl acetate/hexanes) afforded 1.6 g (48%) of the THP 30 sulfone methyl ester as a tan solid, m/z = 378 (M + H).

Part F: To a solution of the THP sulfone methyl ester from part E (1.6 g, 4.24 mmol) in dry tetrahydrofuran (20 mL) was added potassium trimethylsilanoate (1.6 g, 12.72 mmol). After five hours at ambient temperature, the reaction was concentrated in vacuo to yield the potassium salt of the THP sulfone acid as a tan solid suitable for use in the next step.

- Part G: To a slurry of the potassium salt of 10 the THP sulfone acid obtained from part F (1.7 q, 4.24 mmol) in dimethylformamide (20 mL) was added Nhydroxybenzotriazole hydrate (1.1 q, 8.48 mmol) and 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (1.6 g, 8.48 mmol). After heating the 15 reaction mixture at 40°C for one-half hour, Nmethylmorpholine (1.4 mL, 1.3 g, 12.72 mmol) and O-(tetrahydro-2H-pyran-2-yl)hydroxylamine (1.0 g, 8.48 mmol) were added. After heating at 45°C for 15 hours, the reaction was concentrated in vacuo. The residue 20 was partitioned between ethyl acetate and 10% potassium carbonate, the layers were separated and the aqueous layer was extracted with ethyl acetate (13X). The organic extracts were combined, washed with water and brine (3X), dried over Na₂SO₄, 25 filtered, and concentrated in vacuo. Chromatography (on silica, (2M ammonia in methanol-ethyl acetate)/hexanes) afforded 0.7 g (35%) of the THPprotected THP sulfone hydroxamate as a dry, white
- 30 Part H: To a slurry of the THP-protected THP sulfone hydroxamate from part G (0.7 g, 1.43 mmol) in methanol (0.4 mL) was added 4N HCl dioxane solution

foam, m/z = 463 (M + H).

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(4 mL). After thirty minutes at ambient temperature, the reaction mixture was slowly poured into diethyl ether (200 mL) and stirred for fifteen minutes. Filtration afforded 0.5 grams (83%) of the title compound as the HCl salt, $m/z = 379 \, (M + H)$.

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Example 229: Preparation of N-hydroxy-1-(4-methyl-phenyl)-4-[[4-[4-(trifluoromethoxy)-phenoxy]phenyl]sulfonyl]-4-piperidine-carboxamide monohydrochloride

HO N S O CF₃

HCI

Part A: To a suspension of ethyl 4-(4fluorophenylsulfonyl]-4-piperidinecarboxylate, 15 hydrochloride Preparative Example II (2.56 g, 7.28 mmol) in H_2O (50 mL) was added 1.25N NaOH (pH = 9.0). The agueous layer was extracted with diethyl ether (2 x 75 mL). The combined organic layers were washed with saturated NaCl and dried over Na₂SO₄. 20 Concentration in vacuo provided the free amine as an off-white solid (1.72 g). To a solution of the free amine (1.70 g, 5.39 mmol) in toluene (25 mL) was added Cs_2CO_3 (2.34 g, 7.19 mmol) and a solution of 4bromotoluene (0.877 g, 5.13 mmol) in toluene (5 mL). 25 This was followed by the addition of tris(dibenzyldeneacetone)dipallidium (0) (0.047 g,

0.0513 mmol) and BINAP (0.096 g, 0.154 mmol). The resulting mixture was then heated to one hundred degress Celsius for 17 hours. After cooling to ambient temperature, the reaction mixture was filtered through a pad of Celite®, washing with ethyl acetate and the filtrate was concentrated in vacuo. Chromatography (on silica, ethyl acetate/hexane) provided the aniline as a yellow oil (1.59 g, 76%).

part B: To a solution of the aniline of

part A (1.56 g, 3.85 mmol) in N,N-dimethylformamide

(8.0 mL) was added K₂CO₃ (1.06 g, 7.70 mmol) and 4
(trifluoromethoxy)phenol (0.823 g, 4.62 mmol). The

resulting mixture was heated to ninety degrees

Celsius for 19 hours. The reaction was cooled to

ambient temperature and concentrated in vacuo. The

residue was partitioned between H₂O and diethyl ether.

The organic layer was washed with saturated NaCl and

dried over Na₂SO₄. Concentration in vacuo provided

the biaryl ether as a brown oil (2.42 g, >100 %).

Part C: To a solution of the biaryl ether of part B (2.42 g, 3.85 mmol) in tetrahydrofuran (10 mL) and H_2O (10 mL) was added NaOH (1.54 g, 38.50 mmol) in H_2O (5.0 mL). The mixture was heated to sixty degrees Celsius for 6 hours then cooled to ambient temperature. The mixture was then acidified (pH = 7) with 1N HCl. The solids were collected by vacuum filtration, then suspended in acetonitrile and concentrated in vacuo to give the acid as a tan solid (1.95 g, 95%).

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Part D: To a suspension of the acid of part C (1.95 g, 3.64 mmol) in N,N-dimethylformamide (15 mL) was added 1-hydroxybenzotriazole (0.596 g, 4.37 mmol), N-methylmorpholine (1.19 mL, 10.92 mmol), O-

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(tetrahydropuranyl) hydroxylamine (1.28 g, 10.92
mmol) and 1-3-[(dimethylamino)propyl]-3ethylcarbodiimide hydrochloride (0.977 g, 5.10 mmol).
The resulting mixture was stirred at ambient

5 temperature for 16 hours then concentrated in vacuo.
The residue was partitioned between H2O and ethyl
acetate. The combined organic layers were washed
with H2O, saturated NaHCO3, saturated NaCl and dried
over Na2SO4. Chromatography (on silica,
methanol/ethyl acetate) provided the protected
hydroxamate as a pale-yellow foam (1.90 g, 83%).

Part E: To the protected hydroxamate of part D (1.89 g, 3.00 mmol) was added 4N HCl in dioxane (7.50 mL, 30.0 mmol) and methanol (1.22 mL, 30.0 mmol). The resulting mixture was stirred at ambient temperature for 2 hours, then diethyl ether (5 mL) was added and the precipitate was collected by filtration to provide the title compound as a fine white solid (1.56 g, 89%). MS MH calculated for C₂₆H₂₅O₆N₂S₁F₃: 551, found 551.

Example 230: Preparation of N-hydroxy-1-(2-hydroxyethyl)-4-[4-(4-trifluoro-methoxyphenoxy)phenyl]sulfonyl]-4-piperidinecarboxamide, hydrochloride

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part A: Ethyl 4-(4-fluorophenylsulfonyl]-4piperidinecarboxylate, hydrochloride (3.95 g, 11.3

mmol) Preparative Example II, powdered potassium carbonate (3.45 g, 25 mmol), and N,N-dimethylformamide (11.3 mL) were combined. 2-(2-Bromoethoxy)tetrahydro-2H-pyran (1.85 mL, 12 mmol) was added and the mixture was stirred for 48 hours at ambient temperature. The reaction was diluted with water (100 mL) and extracted with ethyl acetate (100 mL, then 50 mL). The combined organic layers were dried over magnesium sulfate, concentrated, and chromatographed to afford the desired tetrahydropyranyl ether as an oil (4.44 g, 88%)

Part B: The tetrahydropyranyl ether from Part A was stirred at 110 degrees Celsius for 20 hours in the presence of powdered potassium carbonate (2.07 g, 15 mmol), 4-(trifluoromethoxy)phenol (2.67 mL, 15 mmol), and N,N-dimethyformamide (5 mL). The mixture was diluted with saturated sodium bicarbonate (50 mL) and was extracted with ethyl acetate (150, then 50 mL). The combined organic layers were dried over magnesium sulfate, concentrated, and chromatographed to afford the desired aryl ether as an oil (5.72 g, quantitative).

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Part C: The aryl ether from Part C (1.28 g, 2.1 mmol) was refluxed in the presence of potassium

25 hydroxide (954 mg, 16.8 mmol), ethanol (9 mL), and water (3 mL). After 2 hours, the reaction vessel was cooled to zero degrees Celsius. Concentrated hydrochloric acid was added drop-wise to adjust the pH to 4.0. The acidified reaction was concentrated, azeotroped with acetonitrile, and dried in vacuo, affording the crude carboxylic acid, which was used directly in Part D.

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Part D: The carboxylic acid from Part C was converted to O-tetrahydropyranyl hydroxamate using O-tetrahydropyranyl hydroxylamine (351 mg, 3 mmol), N-methylmorpholine (0.5 mL), N-hydroxybenzotriazole (405 mg, 3 mmol), and 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (573 mg, 3 mmol) in N,N-dimethylformamide (9 mL). The tetrahydropyranyl hydroxamate (855 mg, 60 %) was obtained as an oil.

Part E: The tetrahydropyranyl hydroxamate (855 mg, 1.26 mmol) was dissolved in absolute methanol (10 mL). Acetyl chloride (0.78 mL, 11 mmol) was added over 2-3 minutes. After 4 hours both tetrahydropyranyl groups had been cleaved. The reaction was concentrated, azeotroped with chloroform/acetonitrile, and dried in vacuo affording the title compound as a white foam (676 mg, 98%). MS (EI) MH+ calculated for C21H23F3N2O7S: 505, found 505.

20 Example 231: Preparation of N-hydroxy-4-[[4-[4-[(trifluoromethyl)thio]phenoxy]phenyl]sulfonyl]-4-piperidinecarboxamide, monohydrochloride

Part A: To a solution of the compound of example N-tert-butoxycarbonyl-ethyl 4-(4-fluorophenylsulfonyl)-4-piperidinecarboxylate,

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hydrochloride of Preparative Example II (1.50 g, 3.61 mmol) in N,N-dimethylformamide (10 mL) was added cesium carbonate (2.94 g, 9.03 mmol) and (4trifluoromethylthio) phenol (1.05 g, 5.41 mmol) and 5 the solution was heated to 100 degrees Celsius for 24 The solution was partitioned between ethyl acetate and water and the organic layer was washed with water and dried over sodium sulfate. Filtration through silica gel (ethyl acetate) provided the phenoxyphenol compound as an oil (2.35 g, quantitative yield). MS(CI) MH⁺ calculated for $C_{26}H_{30}NO_7S_2F_3$: 590, found 590.

Part B: To a solution of phenoxyphenol compound of part A (2.35 g, <3.61 mmol) in tetrahydrofuran (10 15 mL) and ethanol (10 mL) was added sodium hydroxide (1.44 g, 36.1 mmol) in water (5 mL). The solution was heated to sixty degrees Celsius for 20 hours. The solution was concentrated under a stream of nitrogen to remove the solvents and the residue was 20 dissolved in water and acidified to pH = 1 with 10% hydrochloric acid. The solution was extracted with ethyl acetate and washed with saturated sodium chloride and dried over magnesium sulfate. Concentration in vacuo provided the carboxylic acid 25 as an oil (2.0 g, quantitative yield).

Part C: To a solution of the carboxylic acid of part B (2.0 g, <3.61 mmol) in N, N-dimethylformamide (10 mL) was added 1-hydroxybenzotriazole hydrate (586 mg, 4.33 mmol), 4-methylmorpholine (1.19 mL, 10.8 mmol) and O-tetrahydropyranyl hydroxylamine (634 mg, 5.41 mmol) and the solution was stirred for 30 minutes. The 1-[3-(dimethylamino)propyl]-3ethylcarbodiimide hydrochloride (969 mg, 5.05 mmol)

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was added and the solution was stirred for seven days. The solution was partitioned between ethyl acetate and water and the organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Chromatography (on silica, ethyl acetate/hexane) provided the protected hydroxamate as a clear, colorless oil (1.07 g, 45 % yield). MS(CI) MNa⁺ calculated for C₂₉H₃₅N₂O₈S₂F₃: 683, found 683.

Part D: To a solution of the protected 10 hydroxamate of part C (1.05 g, 1.60 mmol) in 1,4dioxane (10 mL) was added 4M hydrochloric acid in dioxane (10 mL) and the solution was stirred for 1.5 hours. The solution was diluted with ethyl ether and the resulting white precipitate was collected by 15 vacuum filtration to provide the title compound as a white solid (330 mg, 40 % yield). MS(CI) MH⁺ calculated for $C_{19}H_{19}N_2O_5S_2F_3$: 477, found 477. HRMS calculated for $C_{19}H_{19}N_2O_5S_2F_3$: 477.0766, found 477.0766. Analytical calculation for $C_{19}H_{19}N_2O_5S_2$ HCl: C, 44.49; 20 H, 3.93; N, 5.46; Cl, 6.91. Found: C, 44.51; H, 3.90; N, 5.38; Cl, 6.95.

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Part A: To a solution of ethyl N-cyclopropyl-4-(4-fluorophenylsulfonyl]-4-

piperidinecarboxylate (Preparative Example VI, Part A) (2.0 g, 5.11 mmol) in dimethylacetamide (10 mL) was added methyl isonipectotate (1.03 mL, 7.66 mmol) and cesium carbonate (4.16 g, 12.78 mmol) and was heated to one hundred ten degrees Celsius for 18

10 hours. The solution was cooled to ambient temperature and partitioned between ethyl acetate and water. The organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Concentration in vacuo provided the

phenylamine as an oil (1.81 g, 74 %). MS(CI) MH^{+} calculated for $C_{24}H_{34}N_{2}O_{6}S$: 479, found 479.

Part B: To a solution of the phenylamine of part A (1.79 g, 3.74 mmol) in tetrahydrofuran (20 mL) was added potassium trimethylsilanoate (960 mg, 7.49 mmol) and the resulting solution was stirred for 18 hours at ambient temperature. The solution was concentrated in vacuo and the residue was dissolved into water. The solution was acidified with 3N hydrochloric acid to pH = 3. The resulting

25 precipitate was collected and washed with ethyl ether to provide the acid as a light yellow solid (1.09 g,

63 %). MS(CI) MH $^{+}$ calculated for $C_{23}H_{32}N_{2}O_{6}S$: 465, found 465.

Part C: To a solution of the acid of part B (500 mg, 1.08 mmol) in dichloromethane (10 mL) was 5 added 1-hydroxybenzotriazole hydrate (160 mg, 1.19 mmol), triethylamine (0.15 mL, 1.19 mmol) and Nbenzylmethylamine (0.33 mL, 2.38 mmol). After thirty minutes the 1-[3-(dimethylamino)propyl]-3ethylcarbodiimide hydrochloride was added and the 10 solution was stirred for 20 hours at ambient temperature. The solution was diluted with ethyl acetate and washed with water and saturated sodium chloride and dried over sodium sulfate. Chromatography (on silica, ethyl acetate) provided 15 the amide as a white solid (480 mg, 78 %). MS(CI) MH⁺ calculated for $C_{31}H_{41}N_3O_5S$: 568, found 568.

Part D: To a solution of the amide of part C (400 mg, 0.71 mmol) in ethanol (5 mL) and tetrahydrofuran (5 mL) was added sodium hydroxide (282 mg, 7.1 mmol) in water (3 mL). The solution was heated to sixty degrees Celsius for 24 hours. The solution was concentrated under a stream of nitrogen and the residue was diluted with water and acidified with 3N hydrochloric acid to pH=2. The solution was concentrated to provide the acid as a crude white solid which is used in the next step without further purification. MS(CI) MH⁺ calculated for C₂₉H₃₇N₅O₅S: 540, found 540.

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Part E: To a solution of the crude acid of
part D (<0.71 mmol) in N,N-dimethylformamide (10 mL)
was added 1-hydroxybenzotriazole hydrate (115 mg,
0.85 mmol), 4-methylmorpholine (0.39 mL) and 0tetrahydropyranyl hydroxylamine (124 mg, 1.06 mmol).

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After thirty minutes 1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide hydrochloride (190 mg, 0.99 mmol) was added and the solution was stirred for 18 hours. The solution was partitioned between ethyl acetate and water and the organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Chromatography (on silica, ethyl acetate) provided the protected hydroxamate as an oil (184 mg, 41 %). MS(CI) MH⁺ calculated for C₃₄H₄₆N₄O₆S: 639, found 639.

Part F: To a solution of the protected hydroxamate of part E (180 mg, 0.28 mmol) in dioxane (10 mL) was added 4M hydrochloric acid in dioxane (10 mL) and the solution was stirred for one hour. Trituration (ethyl ether) and vacuum filtration provided the title compound as a white solid (96.5 mg, 58 %). MS(CI) MH⁺ calculated for C₂₉H₃₈N₄O₅S: 555, found 555. HRMS calc. 555.2641, found 555.2644.

20 Example 233: Preparation of 4-[[4-[4-[(3,5-dimethyl1-piperidinyl)carbonyl]-1-piperidinyl]phenyl]sulfonyl]-N-hydroxy-1-(2methoxyethyl)-4-piperidinecarboxamide,
monohydrochloride

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Part A: To a solution of isonipecotic acid (5.8 g, 44.9 mmol) in water (200 mL) was added sodium carbonate (4.62 g, 44.9 mmol) followed by the drop-wise addition of di-tert-butyl-dicarbonate (10.1 g, 46.3 mmol) in dioxane (40 mL). After four hours the solvent was concentrated in vacuo and the solution was extracted with ethyl ether. The aqueous layer was acidified with 3N hydrochloric acid to pH=2. The solution was extracted with ethyl ether and the organic layer was washed with saturated aqueous sodium chloride and dried over magnesium sulfate. Concentration in vacuo provided N-Boc-isonipecotic acid as a white solid (9.34 g, 90 %).

Part B: To a solution of the N-Boc-15 isonipecotic acid of part A (1.0 g, 4.37 mmol) in dichloromethane (10 mL) was added 1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide hydrochloride (853 mg, 4.45 mmol), 1hydroxybenzotriazole hydrate (620 mg, 4.59 mmol) 3,5-20 dimethylpiperdine (0.67 mL, 5.03 mmol) and diisopropylethylamine (1.67 mL, 9.61 mmol) and was stirred for 21 hours. The solution was concentrated in vacuo. The residue was diluted with ethyl acetate and washed with 1M hydrochloric acid, saturated 25 sodium bicarbonate and saturated aqueous sodium chloride and dried over sodium sulfate. Concentration in vacuo provided the amide as a clear colorless oil (1.21 g, 89 %).

Part C: To a solution of the amide of

part B (1.20 g, 3.84 mmol) in dichloromethane (5 mL)

was added trifluoroacetic acid (5 mL) and the

solution was stirred for 1 hour. Concentration in

vacuo provided an oil which was added directly to a

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solution of the compound of Preparative Example VII, Part A (956 mg, 2.56 mmol) in dimethylacetamide (10 mL). Cesium carbonate (2.92 g, 8.96 mmol) was added and the solution was heated to one hundred degrees 5 Celsius for 18 hours. The solution was partitioned between ethyl acetate and water and the organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Concentration in vacuo provided the phenylamine as an oil (1.53 g, 68 %). MS(CI) MH⁺ calculated for C₃₀H₄₇N₃O₆S: 578, found 578.

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Part D: To a solution of the phenylamine of part C (1.5 g, 2.6 mmol) in ethanol (9 mL) and tetrahydrofuran (9 mL) was added sodium hydroxide (1.02 g, 26 mmol) in water (5 mL) and the solution was heated to sixty degrees Celsius for 20 hours. The solution was concentrated and the residue was diluted with water and acidified to pH = 3 with 3N hydrochloric acid. Vacuum filtration provided the acid as a beige solid (500 mg, 33 %). MS(CI) MH+ calculated for C28H43N3O6S: 550, found 550.

Part E: To a solution of the acid of part D (492 mg, 0.84 mmol) in N,N-dimethylformamide (10 mL) was added 1-hydroxybenzotriazole hydrate (136 mg, 1.01 mmol), 4-methylmorpholine (0.46 mL, 4.20 mmol), and O-tetrahydropyranyl hydroxylamine (147 mg, 1.26 mmol). After one hour 1-[3-(dimethylamino)propyl]-3ethylcarbodiimide hydrochloride (225 mg, 1.18 mmol) was added and the solution was stirred for 72 hours at ambient temperature. The solution was partitioned between ethyl acetate and water. The organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Concentration in

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vacuo provided the protected hydroxamate as an oil (524 mg, 96 %). MS(CI) MH^+ calculated for $C_{33}H_{51}N_4O_7S$: 649, found 649.

Part F: To a solution of the protected

hydroxamate of part E (514 mg, 0.79 mmol) in 1,4dioxane (10 mL) was added 4M hydrochloric acid in
dioxane (10 mL) and the solution was stirred for 1.5
hours. The solution was concentrated in vacuo and
trituration (ethyl ether) provided the title compound

as a white solid (360 mg, 76 %). MS(CI) MH⁺ calculated
for C₂₈H₄₄N₄O₆S: 565, found 565. HRMS calculated for
C₂₈H₄₄N₄O₆S: 565.3060, found 565.3070. Analytical
calculation for C₂₈H₄₄N₄O₆S 2HCl:2H₂O: C, 49.92; H,
7.48; N, 8.32; S, 4.76; Cl, 10.52. Found: C, 49.41;

H, 7.55; N, 7.85; S, 4.53; Cl, 10.78.

Example 234: Preparation of 4-[[4-[4-[(3,5-dimethyl1-piperidinyl)carbonyl]-1-piperidinyl]phenyl]sulfonyl]-N-hydroxy-1-(2methoxyethyl)-4-piperidinecarboxamide

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Part A: A solution of the hydroxamate of Example 233, part F (50 mg, 0.08 mmol) in water (2 mL) was neutralized with saturated sodium bicarbonate. The aqueous solution was extracted with ethyl acetate. Concentration in vacuo provided the

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hydroxamate free base as an orange solid (35 mg, 75%).

Example 235: Preparation of 1-[4-[[4[(hydroxyamino)5 carbonyl]-1-(2-methoxyethyl)-4piperidinyl]sulfonyl]phenyl]-N-methylN-[2-(2-pyridinyl)ethyl]-4-piperidinecarboxamide, dihydrochloride

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Part A: To a solution of the N-Bocisonipecotic acid of Example 233, part A (1.0 g, 4.37 mmol) in dichloromethane (10 mL) was added 1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide hydrochloride (853 mg, 4.45 mmol), 1-hydroxybenzotriazole hydrate (620 mg, 4.59 mmol), 2-(2methylaminoethyl)pyridine (0.69 mL, 5.03 mmol) and diisopropylethylamine (1.67 mL, 9.61 mmol) and was stirred for 21 hours. The solution was concentrated in vacuo. The residue was diluted with ethyl acetate and washed with 1M hydrochloric acid, saturated sodium bicarbonate and saturated sodium chloride and dried over sodium sulfate. Concentration in vacuo provided the amide as a clear colorless oil (1.03 g, 68 %). MS(CI) MH^+ calculated for $C_{19}H_{29}N_3O_3$: 348, found 348.

Part B: To a solution of the amide of part A (1.0 g, 2.88 mmol) in dichloromethane (5 mL) was added trifluoroacetic acid (5 mL) and the solution was stirred for 1 hour. Concentration in vacuo provided an oil which was added directly to a solution of the compound of Preparative Example VII, Part A (716 mg, 1.92 mmol) in dimethylacetamide (10 mL). Cesium carbonate (2.20 g, 6.72 mmol) was added and the solution was heated to one hundred degrees Celsius for 18 hours. The solution was partitioned 10 between ethyl acetate and water and the organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Concentration in vacuo provided the phenylamine as a yellow oil (1.20 q, quantitative yield). MS(CI) MH+ calculated for 15 $C_{31}H_{44}N_4O_6S$: 601, found 601.

Part C: To a solution of the phenylamine of part B (1.20 g, 2.00 mmol) in ethanol (8 mL) and tetrahydrofuran (8 mL) was added sodium hydroxide (800 mg, 20 mmol) in water (5 mL) and the solution was heated to sixty degrees Celsius for 20 hours. The solution was concentrated and the residue was diluted with water and acidified to pH = 1 with 3N hydrochloric acid. Concentration in vacuo provided the crude acid as an oil. MS(CI) MH $^+$ calculated for $C_{29}H_{40}N_4O_6S$: 573, found 573.

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Part D: To a solution of the acid of part C (<2.0 mmol) in N,N-dimethylformamide (10 mL) was added 1-hydroxybenzotriazole hydrate (324 mg, 2.04 mmol), 4-methylmorpholine (1.1 mL, 10.0 mmol), and O-t trahydropyranyl hydroxylamine (351 mg, 3.00 mmol). After one hour 1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide hydrochloride (536 mg, 2.80 mmol)

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was added and the solution was stirred for 18 hours at ambient temperature. The solution was partitioned between ethyl acetate and water. The organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Reverse phase chromatography (on silica, acetonitrile/water) provided the protected hydroxamate as an oil (170 mg, 13 % yield over two steps). MS(CI) MH⁺ calculated for C₃₄H₄₉N₅O₇S: 672, found 672.

10 Part E: To a solution of the protected hydroxamate of part D (160 mg, 0.24 mmol) in dioxane (7 mL) was added 4M hydrochloric acid in dioxane (10 mL) and the solution was stirred for 30 minutes. The resulting solid was collected by vacuum filtration.

15 Washing with ethyl ether provided the title compound as a white solid (90 mg, 57 %). MS(CI) MH⁺ calculated for C₂₉H₃₇N₅O₆S: 588, found 588. HRMS calculated for C₂₉H₃₇N₅O₆S: 558.2856, found 588.2857.

20 Example 236: Preparation of N-hydroxy-1-(2methoxyethyl)-4-[[4-[4-[(phenylamino)carbonyl]-1-piperidinyl]phenyl]sulfonyl]-4-piperidinecarboxamide
monohydrochloride)

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Part A: To a solution of the N-Bocisonipecotic acid of Example 233, part A (1.0 g, 4.37 mmol) in dichloromethane (4 mL) was added 2-chloro-4,6-dimethoxy-1,3,5-triazine (752 mg, 4.28 mmol). The solution was cooled to zero degrees Celsius and 4-methylmorpholine (0.47 mL, 4.28 mmol) was added. After two hours aniline (0.39 mL, 4.28 mmol) was added and the solution was stirred for 20 hours at ambient temperature. The solution was concentrated 10 in vacuo. The residue was diluted with ethyl acetate and washed with 1M hydrochloric acid, saturated sodium bicarbonate and saturated sodium chloride and dried over sodium sulfate. Concentration in vacuo provided the amide as a pink solid (1.48 g, quantitative yield). 15

Part B: To a solution of the amide of part A (1.48 g, 4.28 mmol) in dichloromethane (5 mL) was added trifluoroacetic (5 mL) and the solution was stirred for 1 hour. Concentration in vacuo provided an oil which was added directly to a solution of the 20 compound of Preparative Example VII, Part A (1.06 mg, 2.85 mmol) in dimethylacetamide (10 mL). Cesium carbonate (3.25 g, 9.97 mmol) was added and the solution was heated to one hundred ten degrees 25 Celsius for 18 hours. The solution was partitioned between ethyl acetate and water and the organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Concentration in vacuo provided the phenylamine as a yellow oil (1.74 g, quantitative yield). MS(CI) MH+ calculated for 30 $C_{29}H_{39}N_3O_6S: 558$, found 558.

Part C: To a solution of the phenylamine of part B (1.74 g, 2.85 mmol) in ethanol (10 mL) and

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tetrahydrofuran (10 mL) was added sodium hydroxide (1.14 g, 28.5 mmol) in water (7 mL) and the solution was heated to sixty degrees Celsius for 20 hours. The solution was concentrated and the residue was diluted with water and acidified to pH = 1 with 3N hydrochloric acid producing a solid. Vacuum filtration provided the acid as a beige solid (1.62 g, quantitative yield). MS(CI) MH $^+$ calculated for $C_{27}H_{35}N_3O_6S$: 530, found 530.

Part D: To a solution of the acid of part 10 C (1.60 q, 2.83 mmol) in N,N-dimethylformamide (10 mL) was added 1-hydroxybenzotriazole hydrate (458 mg, 3.40 mmol), 4-methylmorpholine (1.56 mL, 14.2 mmol), and O-tetrahydropyranyl hydroxylamine (497 mg, 4.24 mmol). After one hour, 1-[3-(dimethylamino)propyl]-15 3-ethylcarbodiimide hydrochloride (759 mg, 3.96 mmol) was added and the solution was stirred for 18 hours at ambient temperature. The solution was partitioned between ethyl acetate and water. The organic layer was washed with water and saturated sodium chloride 20 and dried over sodium sulfate. Chromatography (on silica, ethyl acetate/methanol) provided the protected hydroxamate as a yellow oil (790 mg, 44 %). MS(CI) MH^{+} calculated for $C_{32}H_{44}N_{4}O_{7}S$: 629, found 629.

Part E: To a solution of the protected hydroxamate of part D (780 mg, 1.24 mmol) in dioxane (10 mL) was added 4M hydrochloric acid in dioxane (10 mL) and the solution was stirred for two hours. The resulting solid was collected by vacuum filtration. Washing with ethyl ether provided the title compound as a white solid (580 mg, 80 %). MS(CI) MH $^+$ calculated for $C_{27}H_{36}N_4O_6S$: 545, found 545. HRMS calculated for $C_{27}H_{36}N_4O_6S$: 545.2434, found 545.2429.

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Example 237: Preparation of N-hydroxy-1-(2methoxyethyl)-4-[[4-[4-[[(3-phenylpropyl)amino]carbonyl]-1-piperidinyl]phenyl]sulfonyl]-4-piperidinecarboxamide, monohydrochloride

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Part A: To a solution of the N-Boc-10 isonipecotic acid of Example 233, part A (1.0 g, 4.37 mmol) in dichloromethane (10 mL) was added 1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide hydrochloride (853 mg, 4.45 mmol), 1hydroxybenzotriazole hydrate (620 mg, 4.59 mmol), 3phenyl-1-propylamine (0.72 mL, 5.03 mmol) and diisopropylethylamine (1.67 mL, 9.61 mmol) and was stirred for 18 hours. The solution was concentrated in vacuo. The residue was diluted with ethyl acetate and washed with 1M hydrochloric acid, saturated 20 sodium bicarbonate and saturated sodium chloride and dried over sodium sulfate. Concentration in vacuo provided the amide as a yellow oil (1.4 g, 93 %).

Part B: To a solution of the amide of part

5 A (1.4 g, 4.05 mmol) in dioxane (10 mL) was added 4M
hydrochloric acid in dioxane (10 mL) and the solution
was stirred for 1 hour. The resulting solid was
collected by vacuum filtration and washed with ethyl

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ether. The solid was added to a solution of the compound of Preparative Example VII, Part A (1.01 mg, 2.70 mmol) in dimethylacetamide (10 mL). Cesium carbonate (3.07 g, 9.45 mmol) was added and the 5 solution was heated to one hundred degrees Celsius for 18 hours. The solution was partitioned between ethyl acetate and water and the organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Concentration in vacuo provided the phenylamine as an orange oil (1.71 g, quantitative yield). MS(CI) MH calculated for $C_{32}H_{45}N_3O_6S$: 600, found 600.

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Part C: To a solution of the phenylamine of part B (1.70 g, 2.70 mmol) in ethanol (10 mL) and tetrahydrofuran (10 mL) was added sodium hydroxide 15 (1.08 g, 27.0 mmol) in water (5 mL) and the solution was heated to sixty degrees Celsius for 20 hours. The solution was concentrated and the residue was diluted with water and acidified to pH = 1 with 3N hydrochloric acid producing a solid. Vacuum 20 filtration provided the acid as a white solid (1.15 q, 75%). MS(CI) MH⁺ calculated for $C_{30}H_{41}N_3O_6S$: 572, found 572.

Part D: To a solution of the acid of part C (1.02 g, 1.68 mmol) in N, N-dimethylformamide (10 25 mL) was added 1-hydroxybenzotriazole hydrate (272 mg, 2.02 mmol), 4-methylmorpholine (0.92 mL, 8.4 mmol), and O-tetrahydropyranyl hydroxylamine (295 mg, 2.52 rnmol). After one hour 1-[3-(dimethylamino)propyl]-3ethylcarbodiimide hydrochloride (451 mg, 2.35 mmol) 30 was added and the solution was stirred for 18 hours at ambient temperature. The solution was partitioned between ethyl acetate and water. The organic layer

-415-

was washed with water and saturated sodium chloride and dried over sodium sulfate. Chromatography (on silica, ethyl acetate/methanol) provided the protected hydroxamate as an oil (490 mg, 41 %).

5 MS(CI) MH^+ calculated for $C_{35}H_{50}N_4O_7S$: 671, found 671.

Part E: To a solution of the protected hydroxamate of part D (480 mg, 0.72 mmol) in dioxane (10 mL) was added 4M hydrochloric acid in dioxane (10 mL) and the solution was stirred for one hour. The resulting solid was collected by vacuum filtration. Washing with ethyl ether provided the title compound as a white solid (400 mg, 90 %). MS(CI) MH⁺ calculated for C₃₀H₄₂N₄O₆S: 587, found 587. Analytical calculation for C₃₀H₄₂N₄O₆S 2HCl: 2H₂O: C, 51.79; H, 6.95; N, 8.05; S, 4.61; Cl, 10.19. Found: C,51.34; H, 6.72; N, 7.82;

Example 238: Preparation of rel-4-[[4-[4-[(3R,5R)-3,5-dimethyl-1-piperidinyl]carbonyl]-1-piperidinyl]phenyl]sulfonyl]-N-hydroxy-4-piperidinecarboxamide,
monohydrochloride

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S, 4.59; Cl, 10.92.

Part A: To a solution of the N-Bocisonipecotic acid of Example 233, Part A (1.0 g, 4.37 mmol) in dichloromethane (10 mL) was added 1-[3-

(dimethylamino)propyl]-3-ethylcarbodiimide
hydrochloride (853 mg, 4.45 mmol), 1hydroxybenzotriazole hydrate (620 mg, 4.59 mmol) 3,5dimethylpiperdine (0.67 mL, 5.03 mmol) and
diisopropylethylamine (1.67 mL, 9.61 mmol) and was
stirred for 21 hours. The solution was concentrated
in vacuo. The residue was diluted with ethyl acetate
and washed with 1M hydrochloric acid, saturated
sodium bicarbonate and saturated sodium chloride and
dried over sodium sulfate. Concentration in vacuo
provided the amide as a clear colorless oil (1.4 g,
quantitative yield).

To a solution of the amide of Part B: part A (1.4 g, 4.49 mmol) in dioxane (10 mL) was added 4M hydrochloric acid in dioxane (10 mL) and the 15 solution was stirred for 1 hour. Concentration in vacuo provided a solid that was added directly to a solution of the compound of Preparative Example II, Part D, (1.24 mg, 2.99 mmol) in dimethylacetamide (10 mL). Cesium carbonate (3.42 g, 10.5 mmol) was added 20 and the solution was heated to one hundred degrees Celsius for 20 hours. The solution was partitioned between ethyl acetate and water and the organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Concentration in 25 vacuo provided the phenylamine as a yellow solid (1.90 g, quantitative yield). MS(CI) MH⁺ calculated for C₃₂H₄₉N₃O₇S: 620, found 620.

Part C: To a solution of the phenylamine

30 of part B (1.9 g, 3.0 mmol) in ethanol (10 mL) and
tetrahydrofuran (10 mL) was added sodium hydroxide
(1.2 g, 30 mmol) in water (5 mL) and the solution was
heated to sixty degrees Celsius for 20 hours. The

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solution was concentrated and the residue was diluted with water and acidified to pH=1 with 3N hydrochloric acid. The solution was extracted with ethyl acetate and washed with 1M hydrochloric acid and saturated sodium chloride and dried over magnesium sulfate. Concentration in vacuo provided the acid as a yellow oil (1.9 g, quantitative yield). MS(CI) MH^+ calculated for $C_{30}H_{45}N_{3}O_{7}S$: 592, found 592.

Part D: To a solution of the acid of part C (1.87 g, 3.00 mmol) in N, N-dimethylformamide (10 10 mL) was added 1-hydroxybenzotriazole hydrate (486 mg, 3.6 mmol), 4-methylmorpholine (1.65 mL, 15 mmol), and O-tetrahydropyranyl hydroxylamine (526 mg, 4.5 mmol). After one hour 1-[3-(dimethylamino)propyl]-3ethylcarbodiimide hydrochloride (805 mg, 4.2 mmol) 15 was added and the solution was stirred for 18 hours at ambient temperature. The solution was partitioned between ethyl acetate and water. The organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Chromatography (on 20 silica, ethyl acetate/hexane) provided the protected hydroxamate as an oil (1.63 g, 79 %).

hydroxamate of part D (1.61 g, 2.33 mmol) in dioxane

(10 mL) was added 4M hydrochloric acid in dioxane (10 mL) and the solution was stirred for 45 minutes. The solution was concentrated in vacuo and trituration (ethyl ether) a white solid. Reverse phase chromatography (on silica, acetonitrile/

water(hydrochloric acid)) produced fractions A, B, C and D. Concentration in vacuo of fraction A provided the title compound as a white solid (59 mg). MS(CI)

MH* calculated for C₂₅H₃₈N₄O₅S: 507, found 507.

Part E: To a solution of the protected

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Example 239: Preparation of rel-1,1-dimethylethyl 4
[[4-[4-[[(3R,5R)-3,5-dimethyl-1piperidinyl]carbonyl]-1-piperidinyl]phenyl]sulfonyl]-4-[(hydroxyamino)carbonyl]-1-piperidinecarboxylate

10 Part A: From the reverse phase chromatography of Example 238, Part E, fraction C was concentrated in vacuo to provide the title compound as a white solid (49 mg). MS(CI) MH⁺ calculated for C₃₀H₄₆N₄O₇S: 607, found 607.

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Example 240: Preparation of rel-4-[[4-[4-[(3R,5S)-3,5-dimethyl-1-piperidinyl]carbonyl]-1-piperidinyl]phenyl]sulfonyl]-N-hydroxy-4-piperidinecarboxamide,
monohydrochloride

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Part A: From the reverse phase chromatography of Example 238, Part E, fraction B was concentrated *in vacuo* to provide the title compound as a white solid (198 mg). MS(CI) MH⁺ calculated for C₂₅H₃₈N₄O₅S: 507, found 507.

Example 241: Preparation of rel-1,1-dimethylethyl 4
[[4-[4-[[(3R,5S)-3,5-dimethyl-1piperidinyl]carbonyl]-1-piperidinyl]phenyl]sulfonyl]-4-[(hydroxyamino)carbonyl]-1-piperidinecarboxylate

Part A: From the reverse phase chromatography of Example 238, Part E, fraction D was concentrated *in vacuo* to provide the title compound as a white solid (242 mg). MS(CI) MH⁺ calculated for C₃₀H₄₆N₄O₇S: 607, found 607.

Example 242: Preparation of 4-[[4-[4-[(2,3-dihydro-1H-inden-2-yl)amino]carbonyl]-1piperidinyl]phenyl]sulfonyl]-N-hydroxy1-(2-methoxyethyl)-4-piperidinecarboxamide, monohydrochloride

-420-

Part A: To a solution of the N-Bocisonipecotic acid of Example 233, Part A (1.0 g, 4.37 mmol) in dichloromethane (10 mL) was added 1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide hydrochloride (853 mg, 4.45 mmol), 1hydroxybenzotriazole hydrate (620 mg, 4.59 mmol) 2aminoindane hydrochloride (853 mg, 5.03 mmol) and diisopropylethylamine (1.67 mL, 9.61 mmol) and was 10 stirred for 21 hours. The solution was concentrated in vacuo. The residue was diluted with ethyl acetate and washed with 1M hydrochloric acid, saturated sodium bicarbonate and saturated sodium chloride and dried over sodium sulfate. Concentration in vacuo 15 provided the amide as a white solid (1.35 g, 90 %).

Part B: To a solution of the amide of part A (1.35 g, 3.92 mmol) in 1,4-dioxane (10 mL) was added 4M hydrochloric acid in dioxane (10 mL) and the solution was stirred for 1 hour. Concentration in vacuo provided a solid which was added directly to a solution of the title compound of Preparative Example VII, Part A, (976 mg, 2.61 mmol) in dimethylacetamide (10 mL). Cesium carbonate (2.97 g, 9.14 mmol) was added and the solution was heated to one hundred degrees Celsius for 18 hours. The solution was partitioned between ethyl acetate and water and the organic layer was washed with water and saturated

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sodium chloride and dried over sodium sulfate. Concentration *in vacuo* provided the phenylamine as an orange oil (1.65 g, quantitative yield). MS(CI) MH^{+} calculated for $C_{32}H_{43}N_{3}O_{6}S$: 598, found 598.

Part C: To a solution of the phenylamine of part B (1.60 g, 2.61 mmol) in ethanol (10 mL) and tetrahydrofuran (10 mL) was added sodium hydroxide (1.04 g, 26 mmol) in water (5 mL) and the solution was heated to sixty degrees Celsius for 18 hours. The solution was concentrated and the residue was diluted with water and acidified to pH = 3 with 3N hydrochloric acid. Vacuum filtration provided the acid as a beige solid (1.06 g, 71 %). MS(CI) MH † calculated for C30H39N3O6S: 570, found 570.

Part D: To a solution of the acid of part 15 E (1.0 q, 1.65 mmol) in N,N-dimethylformamide (10 mL) was added 1-hydroxybenzotriazole hydrate (267 mg, 1.98 mmol), 4-methylmorpholine (0.91 mL, 8.25 mmol), and O-tetrahydropyranyl hydroxylamine (289 mg, 2.48 20 mmol). After one hour 1-[3-(dimethylamino)propyl]-3ethylcarbodiimide hydrochloride (443 mg, 2.31 mmol) was added and the solution was stirred for 18 hours at ambient temperature. The solution was partitioned between ethyl acetate and water. The organic layer was washed with water and saturated sodium chloride 25 and dried over sodium sulfate. Chromatography (on silica, ethyl acetate, methanol) provided the protected hydroxamate as an oil (575 mg, 52 %). MS(CI) MH^+ calculated for $C_{35}H_{48}N_4O_7S$: 669, found 669.

Part E: To a solution of the protected hydroxamate of part D (565 mg, 0.85 mmol) in dioxane (10 mL) was added 4M hydrochloric acid in dioxane (10 mL) and the solution was stirred for 1.5 hours. The

solution was concentrated in vacuo and trituration (ethyl ether) provided the title compound as a white solid (450 mg, 86 %). MS(CI) MH⁺ calculated for C₃₀H₄₀N₄O₆S: 585, found 585. HRMS calculated for C₃₀H₄₀N₄O₆S: 585.2747, found 585.2776. Analytical calculation for C₃₀H₄₀N₄O₆S 2HCl :2H₂O: C, 51.94; H, 6.68; N, 8.08; S, 4.62; Cl, 10.22. Found: C, 51.66; H, 6.25; N, 7.80; S, 4.73; Cl, 10.33.

10 Example 243: Preparation of 1-cyclopropyl-N-hydroxy4-[[4-[4-[(phenylamino)carbonyl]-1piperidinyl]phenyl]sulfonyl]-4piperidinecarboxamide, monohydrochloride

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Part A: To a solution of the product of Example 232, Part B (562 mg, 1.12 mmol) in dichloromethane (3 mL) was added 2-chloro-4,6
20 dimethoxy-1,3,5-triazine (164 mg, 0.93 mmol) and 4-methylmorpholine (0.21 mL, 1.87 mmol). The solution was stirred for 45 minutes and aniline (0.085 mL, 0.93 mmol) was added. The solution was stirred for 72 hours. The solution was partitioned between ethyl acetate and water and the organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Concentration in vacuo provided

the amide as an oil (434 mg, 86%). MS(CI) MH^+ calculated for $C_{29}H_{37}N_3O_5S$: 540, found 540.

Part B: To a solution of the amide of part A (425 mg, 0.79 mmol) in ethanol (5 mL) and tetrahydrofuran (5 mL) was added sodium hydroxide (315 mg, 7.89 mmol) in water (2 mL) and the solution was heated to sixty degrees Celsius for 18 hours. The solution was concentrated and the residue was diluted with water and acidified to pH = 1 with 3N hydrochloric acid producing a solid. Vacuum filtration provided the acid as a beige solid (261 mg, 60%). MS(CI) MH⁺ calculated for C₂₇H₃₃N₃O₅S: 512, found 512.

Part C: To a solution of the acid of part 15 B (245 mg, 0.45 mmol) in N, N-dimethylformamide (10 mL) was added 1-hydroxybenzotriazole hydrate (73 mg, 0.54 mmol), 4-methylmorpholine (0.25 mL, 2.25 mmol), and O-tetrahydropyranyl hydroxylamine (79 mg, 0.68 mmol). After one hour 1-[3-(dimethylamino)propyl]-3ethylcarbodiimide hydrochloride (121 mg, 0.63 mmol) 20 was added and the solution was stirred for 18 hours at ambient temperature. The solution was partitioned between ethyl acetate and water. The organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Chromatography (on 25 silica, ethyl acetate) provided the protected hydroxamate as a yellow oil (242 mg, 88 %). MS(CI) MH⁺ calculated for C32H42N4O6S: 611, found 611.

Part D: To a solution of the protected

30 hydroxamate of part C (235 mg, 0.38 mmol) in dioxane

(5 mL) was added 4M hydrochloric acid in dioxane (10 mL) and the solution was stirred for two hours. The resulting solid was collected by vacuum filtration.

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Washing with ethyl ether provided the title compound as a white solid (114 mg, 53 %). MS(CI) MH^+ calculated for $\mathrm{C}_{27}\mathrm{H}_{34}\mathrm{N}_4\mathrm{O}_5\mathrm{S}$: 527, found 527. HRMS calculated for $\mathrm{C}_{27}\mathrm{H}_{34}\mathrm{N}_4\mathrm{O}_5\mathrm{S}$: 527.2328, found 527.2339.

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Example 244: Preparation of 1-[4-[[4-[(hydroxyamino)-carbonyl]-1-(2-methoxyethyl)-4-piperidinyl]-N-methyl-N-phenyl-4-piperidinecarboxamide,
monohydrate

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Part A: To a solution of the N-Bocisonipecotic acid of Example 233, Part A (500 mg, 15 2.18 mmol) in dichloromethane (2 mL) was added 2chloro-4,6-dimethoxy-1,3,5-triazine (319 mg, 1.82 mmol). The solution was cooled to zero degrees Celsius and 4-methylmorpholine (0.20 mL, 1.82 mmol) 20 was added. After two hours, N-methylaniline (0.20 mL, 1.82 mmol) was added and the solution was stirred for 20 hours at ambient temperature. The solution was concentrated in vacuo. The residue was diluted with ethyl acetate and washed with 1M hydrochloric acid, saturated sodium bicarbonate and saturated 25 sodium chloride and dried over sodium sulfate. Concentration in vacuo provided the amide as a pink solid (445 mg, 77%).

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Part B: To a solution of the amide of part A (440 g,1.38 mmol) in 1,4-dioxane (10 mL) was added 4M hydrochloric acid in dioxane (10 mL) and the solution was stirred for 1 hour. Concentration in vacuo provided an oil which was added directly to a solution of the compound of Preparative Example VII, Part A (344 mg, 0.92 mmol) in dimethylacetamide (10 mL). Cesium carbonate (1.05 g, 3.22 mmol) was added and the solution was heated to one hundred ten degrees Celsius for 18 hours. The solution was 10 partitioned between ethyl acetate and water and the organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Concentration in vacuo provided the phenylamine as a yellow oil (440 mg, 84%). . 15

Part C: To a solution of the phenylamine of part B (440 mg, 0.77 mmol) in ethanol (7 mL) and tetrahydrofuran (7 mL) was added sodium hydroxide (308 mg, 7.7 mmol) in water (3 mL) and the solution was heated to sixty degrees Celsius for 20 hours. The solution was concentrated and the residue was diluted with water and acidified to pH = 1 with 3N hydrochloric acid producing a solid. Vacuum filtration provided the acid as a yellow solid and carried on to the next step without additional purification.

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Part D: To a solution of the acid of part C (<0.77 mmol) in N,N-dimethylformamide (10 mL) was added 1-hydroxybenzotriazole hydrate (125 mg, 0.92 mmol), 4-methylmorpholine (0.43 mL, 3.85 mmol), and O-tetrahydropyranyl hydroxylamine (135 mg, 1.16 mmol). After one hour, 1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide hydrochloride (207 mg, 1.08 mmol)

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was added and the solution was stirred for 24 hours at ambient temperature. The solution was partitioned between ethyl acetate and water. The organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Chromatography (on silica, ethyl acetate/methanol) provided the protected hydroxamate as a yellow oil (150 mg, 30%).

MS(CI) MH⁺ calculated for C₃₃H₄₆N₄O₇S: 643, found 643.

Part E: To a solution of the protected

10 hydroxamate of part D (150 mg, 0.23 mmol) in dioxane

(2 mL) was added 4M hydrochloric acid in dioxane (3

mL) and the solution was stirred for two hours. The

resulting solid was collected by vacuum filtration.

Washing with ethyl ether provided the title compound

15 as a yellow solid (75 mg, 55 %). MS(CI) MH⁺ calculated

for C₂₈H₃₈N₄O₆S: 559, found 559. HRMS calculated for

C₂₈H₃₈N₄O₆S: 559.2590, found 559.2613.

Example 245: Preparation of 1-acetyl-N-hydroxy-4
[[4-[4-[(phenylamino)carbonyl]-1piperidinyl]phenyl]sulfonyl]-4piperidinecarboxamide

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Part A: To a solution of the N-Boc-amide of Preparative Example III, Part B, (6.9 g, 11.4 mmol) in 1,4-dioxane (10 mL) was added 4M hydrochloric acid

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in dioxane (10 mL) and the solution was stirred for 1 hour. Concentration in vacuo provided an oil which was added directly to a solution of the product of Preparative Example II, Part D (3.15 g, 7.6 mmol) in dimethylacetamide (30 mL). Cesium carbonate (8.65 g, 26.6 mmol) was added and the solution was heated to one hundred ten degrees Celsius for 18 hours. The solution was partitioned between ethyl acetate and water. The organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Concentration in vacuo provided the phenylamine as a tan solid (3.92 g, 86%).

Part B: To a solution of the phenylamine of part A (3.90 g, 6.51 mmol) in methanol (20 mL) was added 4M hydrochloric acid in dioxane (10 mL) and the solution was stirred for 3 hours. Concentration in vacuo followed by trituration (ethyl ether) provided the amine hydrochloride salt as a yellow solid (3.25 g, 93%).

Part C: To a solution of the amine hydrochloride salt of part B (500 mg, 0.93 mmol) in dichloromethane (5 mL) was added triethylamine (0.40 mL, 2.79 mmol) followed by acetyl chloride (0.07 mL, 1.02 mmol). The solution was stirred for 3 hours.

The solution was diluted with ethyl acetate and washed with 1M hydrochloric acid, saturated sodium bicarbonate and saturated sodium chloride and dried over sodium sulfate. Concentration in vacuo provided the acylated compound as an oil (390 mg, 77 %).

30 MS(CI) MH⁺ calculated for $C_{28}H_{35}N_3O_6S$: 542, found 542.

Part D: To a solution of the acylated compound of part C (390 mg, 0.72 mmol) in ethanol (5 mL) and tetrahydrofuran (5 mL) was added sodium

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hydroxide (58 mg, 1.44 mmol) in water (1 mL) and th solution was heated to sixty degrees Celsius for 3 hours. The solution was concentrated and the residue was diluted with water and acidified to pH = 1 with 3N hydrochloric acid. The solution was extracted with ethyl acetate and washed with water and saturated sodium chloride and dried over magnesium sulfate. Concentration in vacuo provided the acid as a white solid (137 mg, 37 %). MS(CI) MH⁺ calculated for C₂₆H₃₁N₃O₆S: 514, found 514.

Part E: To a solution of the acid of part D (137 mg, 0.27 mmol) in N,N-dimethylformamide (DMF) (10 mL) was added 1-hydroxybenzotriazole hydrate (44 mg, 0.32 mmol), 4-methylmorpholine (0.10 mL, 1.08 mmol), and O-tetrahydropyranyl hydroxylamine (47 mg, 15 0.41 mmol). After one hour 1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide hydrochloride (72 mg, 0.38 mmol) was added and the solution was stirred for 24 hours at ambient temperature. The solution was partitioned between ethyl acetate and water. 20 organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Chromatography (on silica, ethyl acetate/methanol) provided the protected hydroxamate as a white solid 25 (140 mg, 85%). MS(CI) MH^{+} calculated for $C_{31}H_{40}N_{4}O_{7}S$: 613, found 613.

Part F: To a solution of the protected hydroxamate of part E (130 mg, 0.21 mmol) in dioxane (2 mL) was added 4M hydrochloric acid in dioxane (3 mL) and the solution was stirred for two hours. The resulting solid was collected by vacuum filtration. Washing with ethyl ether provided the title compound

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as a yellow solid (51 mg, 48 %). MS(CI) MH^+ calculated for $\mathrm{C}_{26}\mathrm{H}_{32}\mathrm{N}_4\mathrm{O}_6\mathrm{S}$: 528, found 528.

Example 246: Preparation of4-[[4-[4-[(2,3-dihydro1H-indol-1-yl)carbonyl]-1-piperidinyl]phenyl]sulfonyl]-N-hydroxy-1-(2methoxyethyl)-4-piperidinecarboxamide,
monohydrate

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Part A: To a solution of the N-Bocisonipecotic acid of Preparative Example I, Part B
(750 mg, 3.27 mmol) in dichloromethane (3 mL) was
added 2-chloro-4,6-dimethoxy-1,3,5-triazine (564 mg,
3.21 mmol). The solution was cooled to zero degrees
Celsius and 4-methylmorpholine (0.35 mL, 3.21 mmol)
was added. After two hours, indoline (0.36 mL, 3.21
mmol) was added and the solution was stirred for 22
hours at ambient temperature. The solution was
concentrated in vacuo. The residue was diluted with
ethyl acetate and washed with 1M hydrochloric acid,
saturated sodium bicarbonate and saturated sodium
chloride and dried over sodium sulfate.

25 Concentration in vacuo provided the amide as a pink solid (940 mg, 89 %).

Part B: To a solution of the amide of part A (935 g, 2.83 mmol) in 1,4-dioxane (10 mL) was added

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4M hydrochloric acid in dioxane (10 mL) and the solution was stirred for 1 hour. Concentration in vacuo provided an oil which was added directly to a solution of the compound of Preparative Example VII, Part A, (705 mg, 1.89 mmol) in dimethylacetamide (10 mL). Cesium carbonate (2.15 g, 6.61 mmol) was added and the solution was heated to one hundred ten degrees Celsius for 18 hours. The solution was partitioned between ethyl acetate and water and the organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Concentration in vacuo provided the phenylamine as an orange oil (893 mg, 81 %). MS(CI) MH+ calculated for C₃₁H₄₁N₃O₆S: 584, found 584.

of part C: To a solution of the phenylamine of part B (885 g, 1.52 mmol) in ethanol (10 mL) and tetrahydrofuran (10 mL) was added sodium hydroxide (607 mg, 15.2 mmol) in water (5 mL) and the solution was heated to sixty degrees Celsius for 20 hours.

The solution was concentrated and the residue was diluted with water and acidified to pH = 1 with 3N hydrochloric acid producing a solid. Vacuum filtration provided the acid as a beige solid (475 g, 53 %). MS(CI) MH⁺ calculated for C₂₉H₃₇N₃O₆S: 556, found 556.

Part D: To a solution of the acid of part C (465 g, 0.79 mmol) in N,N-dimethylformamide (10 mL) was added 1-hydroxybenzotriazole hydrate (128 mg, 0.95 mmol), 4-methylmorpholine (0.43 mL, 3.95 mmol), and O-tetrahydropyranyl hydroxylamine (139 mg, 1.18 mmol). After one hour, 1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide hydrochloride (212 mg, 1.10 mmol) was added and the solution was stirred for 18 hours

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at ambient temperature. The solution was partitioned between ethyl acetate and water. The organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Chromatography (on silica, ethyl acetate/methanol) provided the protected hydroxamate as a yellow oil (305 mg, 60 %). MS(CI) MH⁺ calculated for C₃₄H₄₆N₄O₇S: 655, found 655.

Part E: To a solution of the protected hydroxamate of part D (300 mg, 0.46 mmol) in dioxane (5 mL) was added 4M hydrochloric acid in dioxane (5 mL) and the solution was stirred for two hours. The resulting solid was collected by vacuum filtration. Washing with ethyl ether provided the title compound as a white solid (260 mg, 94 %). MS(CI) MH⁺ calculated for C₂₉H₃₄N₄O₆S: 571, found 571.

Example 247: Preparation of N-hydroxy-1-(2-methoxyethyl)-4-[[4-[4-[(phenylmethyl)amino]carbonyl]-1-piperidinyl]phenyl]-sulfonyl]-4-piperidinecarboxamide,monohydrochloride

Part A: To a solution of the N-Boc-isonipecotic acid of Preparative Example I, Part B, (750 mg, 3.27 mmol) in dichloromethane (10 mL) was added 1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide

hydrochloride (640 mg, 3.34 mmol), 1hydroxybenzotriazole hydrate (463 mg, 3.43 mmol) and diisopropylethylamine (1.25 mL, 7.19 mmol). After thirty minutes, benzylamine (0.41 mL, 3.76 mmol) was 5 added and the solution was stirred for 22 hours at ambient temperature. The solution was concentrated in vacuo. The residue was diluted with ethyl acetate and washed with 1M hydrochloric acid, saturated sodium bicarbonate and saturated sodium chloride and dried over sodium sulfate. Concentration in vacuo provided the amide as an oil (320 mg, 31 %).

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Part B: To a solution of the amide of part A (320 g, 1.0 mmol) in 1,4-dioxane (10 mL) was added 4M hydrochloric acid in dioxane (10 mL) and the solution was stirred for 1 hour. Concentration in 15 vacuo provided an oil which was added directly to a solution of the product of Preparative Example II, Part D, (288 mg, 0.77 mmol) in dimethylacetamide (10 mL). Cesium carbonate (878 g, 2.7 mmol) was added and the solution was heated to one hundred ten 20 degrees Celsius for 18 hours. The solution was partitioned between ethyl acetate and water and the organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Concentration in vacuo provided the phenylamine as an 25

Part C: To a solution of the phenylamine of part B (367 g, 0.64 mmol) in ethanol (5 mL) and tetrahydrofuran (5 mL) was added sodium hydroxide (257 mg, 6.4 mmol) in water (2 mL) and the solution was heated to sixty degrees Celsius for 20 hours. The solution was concentrated and the residue was

orange oil (367 mg, 83 %). MS(CI) MH calculated for

 $C_{30}H_{41}N_3O_6S$: 572, found 572.

diluted with water and acidified to pH = 1 with 3N hydrochloric acid producing a solid. Vacuum filtration provided the acid as a beige solid (415 g, quantitative yield). MS(CI) MH $^+$ calculated for $C_{28}H_{37}N_{3}O_{6}S$: 544, found 544.

Part D: To a solution of the acid of part C (415 g, <0.64 mmol) in N, N-dimethylformamide (10 mL) was added 1-hydroxybenzotriazole hydrate (104 mg, 0.77 mmol), 4-methylmorpholine (0.35 mL, 3.20 mmol), and O-tetrahydropyranyl hydroxylamine (112 mg, 0.96 10 mmol). After one hour, 1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide hydrochloride (172 mg, 0.90 mmol) was added and the solution was stirred for 18 hours at ambient temperature. The solution was partitioned 15 between ethyl acetate and water. The organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Chromatography (on silica, ethyl acetate/methanol) provided the protected hydroxamate as a yellow oil (9 mg, 2 %). MS(CI) MH^{+} calculated for $C_{33}H_{46}N_{4}O_{7}S$: 643, found 643. 20

Part E: To a solution of the protected hydroxamate of part D (9 mg, 0.014 mmol) in dioxane (1 mL) was added 4M hydrochloric acid in dioxane (1 mL) and the solution was stirred for two hours. The resulting solid was collected by vacuum filtration. Washing with ethyl ether provided the title compound as a white solid (2.5 mg, 30 %). MS(CI) MH $^+$ calculated for $C_{28}H_{34}N_4O_6S$: 559, found 559.

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Example 248: Preparation of N-hydroxy-1-(2-methoxy-ethyl)-4-[[4-[4-[[4-(trifluoromethoxy)-phenyl]amino]carbonyl]-1-piperidinyl]-phenyl]sulfonyl]-4-piperidine-carboxamide, monohydrochloride

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HOHN HCI OCF3

Part A: To a solution of the N-Boc-isonipecotic acid of Preparative Example I, Part B, (750 mg, 3.27 10 mmol) in dichloromethane (3 mL) was added 2-chloro-4,6-dimethoxy-1,3,5-triazine (564 mg, 3.21 mmol). The solution was cooled to zero degrees Celsius and 4-methylmorpholine (0.35 mL, 3.21 mmol) was added. After two hours, 4-(trifluoromethoxy)aniline (0.43 15 mL, 3.21 mmol) was added and the solution was stirred for 22 hours at ambient temperature. The solution was concentrated in vacuo. The residue was diluted with ethyl acetate and washed with 1M hydrochloric acid, saturated sodium bicarbonate and saturated 20 sodium chloride and dried over sodium sulfate. Concentration in vacuo provided the amide as a pink solid (1.16 g, 93 %).

Part B: To a solution of the amide of part

25 A (1.16 g, 2.99 mmol) in 1,4-dioxane (10 mL) was
added 4M hydrochloric acid in dioxane (10 mL) and the
solution was stirred for 1 hour. Concentration in

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vacuo provided an oil which was added directly to a solution of the product of Preparative Example VII, Part A (743 mg, 1.99 mmol) in dimethylacetamide (10 mL). Cesium carbonate (2.26 g, 6.90 mmol) was added and the solution was heated to one hundred ten degrees Celsius for 18 hours. The solution was partitioned between ethyl acetate and water and the organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate.

10 Concentration in vacuo provided the phenylamine as an orange oil (1.38 g, quantitative yield). MS(CI) MH⁺ calculated for C₃₀H₃₈N₃O₇SF₃: 642, found 642.

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Part C: To a solution of the phenylamine of part B (1.38 g, 2.00 mmol) in ethanol (10 mL) and tetrahydrofuran (10 mL) was added sodium hydroxide (800 mg, 20 mmol) in water (5 mL), and the solution was heated to sixty degrees Celsius for 20 hours. The solution was concentrated and the residue was diluted with water and acidified to pH = 1 with 3N hydrochloric acid producing a solid. Vacuum filtration provided the acid as a beige solid (536 g, 41 %). MS(CI) MH⁺ calculated for $C_{28}H_{34}N_{3}O_{7}SF_{3}$: 614, found 614.

Part D: To a solution of the acid of part

C (536 g, 0.83 mmol) in N,N-dimethylformamide (10 mL)

was added 1-hydroxybenzotriazole hydrate (134 mg,
0.99 mmol), 4-methylmorpholine (0.46 mL, 4.15 mmol),

and O-tetrahydropyranyl hydroxylamine (145 mg, 1.24

mmol). After one hour 1-[3-(dimethylamino)propyl]-3
ethylcarbodiimide hydrochloride (223 mg, 1.16 mmol)

was added and the solution was stirred for 18 hours

at ambient temperature. The solution was partitioned
between ethyl acetate and water. The organic layer

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was washed with water and saturated sodium chloride and dried over sodium sulfate. Chromatography (on silica, ethyl acetate/methanol) provided the protected hydroxamate as a yellow oil (287 mg, 48 %). MS(CI) MH⁺ calculated for C₃₃H₄₃N₄O₈SF₃: 713, found 713.

Part E: To a solution of the protected hydroxamate of part D (280 mg, 0.39 mmol) in dioxane (10 mL) was added 4M hydrochloric acid in dioxane (10 mL) and the solution was stirred for two hours. The resulting solid was collected by vacuum filtration. Washing with ethyl ether provided the title compound as a white solid (228 mg, 88 %). MS(CI) MH $^+$ calculated for $C_{28}H_{35}N_4O_7SF_3$: 629, found 629.

15 Example 249: Preparation of N-hydroxy-1-(2methoxyethyl)-4-[[4-[4-[[[3-(trifluoromethoxy)phenyl]amino]carbonyl]-1piperidinyl]phenyl]sulfonyl]-4piperidinecarboxamide, monohydrochloride

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Part A: To a solution of the N-Bocisonipecotic acid of Preparative Example I, Part B, (750 mg, 3.27 mmol) in dichloromethane (3 mL) was ,added 2-chloro-4,6-dimethoxy-1,3,5-triazine (564 mg, 3.21 mmol). The solution was cooled to zero degrees Celsius and 4-methylmorpholine (0.35 mL, 3.21 mmol)

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was added. After two hours 3-(trifluoromethoxy)aniline (0.43 mL, 3.21 mmol) was added and the
solution was stirred for 22 hours at ambient
temperature. The solution was concentrated in vacuo.
The residue was diluted with ethyl acetate and washed
with 1M hydrochloric acid, saturated sodium
bicarbonate and saturated sodium chloride and dried
over sodium sulfate. Concentration in vacuo provided
the amide as a pink solid (1.20 g, 97 %).

Part B: To a solution of the amide of part 10 A (1.20 g, 3.10 mmol) in 1,4-dioxane (10 mL) was added 4M hydrochloric acid in dioxane (10 mL) and the solution was stirred for 1 hour. Concentration in vacuo provided an oil which was added directly to a solution of the product of Preparative Example VII, 15 Part A, (770 mg, 2.06 mmol) in dimethylacetamide (10 mL). Cesium carbonate (2.34 g, 7.21 mmol) was added and the solution was heated to one hundred ten degrees Celsius for 18 hours. The solution was partitioned between ethyl acetate and water and the 20 organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Concentration in vacuo provided the phenylamine as an orange oil (1.72 g, quantitative yield). MS(CI) MH⁺ calculated for C₃₀H₃₈N₃O₇SF₃: 642, found 642. 25

part C: To a solution of the phenylamine of part B (1.72 g, <2.06 mmol) in ethanol (10 mL) and tetrahydrofuran (10 mL) was added sodium hydroxide (824 mg, 20.6 mmol) in water (5 mL) and the solution was heated to sixty degrees Celsius for 18 hours. The solution was concentrated and the residue was diluted with water and acidified to pH = 1 with 3N hydrochloric acid. Concentration in vacuo provided

the acid as a crude brown oil which was used in the next step without additional purification.

Part D: To a solution of the acid of part C (<2.06 mmol) in N,N-dimethylformamide (10 mL) was added 1-hydroxybenzotriazole hydrate (334 mg, 2.47 mmol), 4-methylmorpholine (1.13 mL, 10.3 mmol), and O-tetrahydropyranyl hydroxylamine (361 mg, 3.09 mmol). After one hour, 1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide hydrochloride (553 mg, 2.88 mmol) was added and the solution was stirred for 18 hours 10 at ambient temperature. The solution was partitioned between ethyl acetate and water. The organic layer was washed with water and saturated sodium chloride and dried over sodium sulfate. Chromatography (on silica, ethyl acetate/methanol) provided the 15 protected hydroxamate as a yellow oil (64 mg, 4 % for 2 steps). MS(CI) MH⁺ calculated for C₃₃H₄₃N₄O₈SF₃: 713, found 713.

Part E: To a solution of the protected

10 hydroxamate of part D (63 mg, 0.089 mmol) in dioxane

13 (5 mL) was added 4M hydrochloric acid in dioxane (5 mL) and the solution was stirred for two hours. The resulting solid was collected by vacuum filtration.

13 Washing with ethyl ether provided the title compound

14 as a white solid (48 mg, 81 %). MS(CI) MH+ calculated for C₂₈H₃₅N₄O₇SF₃: 629, found 629.

Example 250: Preparation of 1-(2-ethoxyethyl)-Nhydroxy-4-[[4-[4-(trifluoromethoxy)phenoxy]phenyl]sulfonyl]-4-piperidinecarboxamide monohydrochloride

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Part A: To a solution of the product of Preparative Example II, Part D, (1.0 g, 2.4 mmol) in dichloromethane (10 mL) was added trifluoroacetic acid (10 mL) and the solution was stirred at ambient temperature for 1 hour. Concentration in vacuo provided the amine trifluoroacetate salt as a light To the solution of the amine yellow gel. trifluoroacetate salt and potassium carbonate (0.99 10 g, 7.2 mmol) in N,N-dimethylformamide (5 mL) was added 2-bromoethyl ethyl ether (0.33 mL, 2.87 mmol) and the solution was stirred at ambient temperature for 36 hours. Then N, N-dimethylformamide was evaporated under high vacuum and the residue was 15 diluted with ethyl acetate. The organic layer was washed with water and dried over magnesium sulfate. Concentration in vacuo provided the ethoxyl ethyl amine as a light yellow gel (0.68 g, 65.4%).

Part B: To a solution of ethoxyl ethyl amine (0.68 g, 1.56 mmol) of part A and powdered potassium carbonate (0.43 g, 3.1 mmol) in N,N-dimethylformamide (5 mL) was added 4-(trifluoromethoxy)phenol (0.4 mL, 3.08 mmol) at ambient temperature and the solution was heated to ninety degrees Celsius for 25 hours. The solution was concentrated under high vacuum and

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the residue was dissolved in ethyl acetate. The organic layer was washed with 1N sodium hydroxide, water and dried over magnesium sulfate.

Chromatography on silica eluting with ethyl acetate/hexane provided the desired trifluoromethoxy phenoxyphenyl sulfone as a light yellow gel (1.0 g, quantitative).

Part C: To a solution of trifluoromethoxy phenoxyphenyl sulfone of Part B (1.0 g, 1.72 mmol) in ethanol (2 mL) and tetrahydrofuran (2 mL) was added sodium hydroxide (0.688 g, 17.2 mmol) in water (4 mL) at ambient temperature. The solution was then heated to sixty degrees Celsius for 18 hours. The solution was concentrated in vacuo and diluted with water. The aqueous layer was extracted with ether and acidified to pH=2. Vacuum filtration of the white precipitate provided the acid as a white solid (0.94 g, quantitative yield).

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Part D: To a solution of the acid of part C (0.94 q, 1.86 mmol), N-methyl morpholine (0.61 mL, 20 5.55 mmol), 1-hydroxybenzotriazole (0.76 g, 5.59 mmol) and O-tetrahydropyranyl hydroxyl amine (0.33 g, 2.7 mmol) in N,N-dimethylformamide (40 mL) was added 1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide 25 hydrochloride (1.06 g, 5.59 mmol) and the solution was stirred at ambient temperature for 24 hours. solution was concentrated under high vacuum and the residue was dissolved in ethyl acetate. The organic layer was washed with saturated aqueous Sodium bicarbonate, water and dried over magnesium sulfate. 30 Concentration in vacuo and chromatography on silica eluting with ethyl acetate/hexane provided the

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tetrahydropyranyl amide as a white foam (0.74 g, 66.1%).

Part E: To a solution of 4N hydrochloric acid (3 mL, 12 mmol)) in dioxane was added a solution of the tetrahydropyranyl amide of part D (0.74 g, 1.2 mmol) in methanol (0.4 ml) and dioxane (1.2 mL) and was stirred at ambient temperature for 3 hours. Filtration of precipitation gave the title compound as white solid (0.217g, 32.9%). Analytical calculation for C₂₂H₂₅N₂O₇SF₃.HCl.0.5H₂O: C, 46.85; H, 4.83; N, 4.97; S, 5.69. Found: C, 46.73; H, 4.57; N, 4.82; S, 5.77.

Example 251: Preparation of N-hydroxy-1-(2-methoxyethyl)-4-[[4-[4-(trifluoromethoxy)phenoxy]phenyl]sulfonyl]-4-piperidinecarboxamide monomethanesulfonate (salt)

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Part A: To the ethanol solution of the product of Preparative Example VII, Part D, (0.3 g, 0.5 mmol) was added methane sulfonic acid (0.042 mL, 0.65 mmol). After two hours stirring at room temperature the solution was cooled to zero degree Celsius. Filtration of the precipitate gave the title compound as a white crystalline solid (0.105 g, 35%).

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Analytical calculation for $C_{22}H_{25}N_2O_7SF_3$. CH_4O_3S . H_2O : C, 43.67; H, 4.94; N, 4.43. Found: C, 43.96; H, 4.62; N, 4.47.

5 Example 252: Preparation of N-hydroxy-1-(2-methoxyethyl)-4-[[4-[4-(trifluoro-methoxy)phenoxy]phenyl]sulfonyl]4-piperidinecarboxamide

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Part A: The title compound of Preparative
Example VII (15 g, 27 mmol) was partitioned between
ethyl acetate and saturated sodium bicarbonate
solution. The aqueous layer was extracted with ethyl
15 acetate. The combined organic layers were washed
with saturated sodium bicarbonate solution, water,
brine and dried over magnesium sulfate.
Concentration in vacuo and recrystallization from hot
toluene gave the title compound as white crystals
20 (13.14 g, 93.9%). Analytical calculation for
C₂₂H₂₅N₂O₇SF₃: C, 50.96; H, 4.86; N, 5.40; S, 6.18.
Found: C, 51.33; H, 5.11; N, 5.29; S, 6.50.

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Example 253: Preparation of N-hydroxy-1-(2-methoxyethyl)-4-[[4-[4-(trifluoro-methoxy)phenoxy]phenyl]sulfonyl]-4-piperidinecarboxamide mono(4-methyl-benzenesulfonate) (salt)

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Part A: To the ethanol solution of Preparative

Example VII (8 g, 13.32 mmol) was added ptoluenesulfonic acid (2.9 g, 15.24 mmol) and the
solution was stirred at ambient temperature for 6
hours. Evaporation of the solvent and
recrystallization from hot ethanol gave the title

compound as white crystals (6.58 g, 71.8%).
Analytical calculation for C₂₂H₂₅N₂O₇SF₃.C₇H₈SO₃: C,
50.43; H, 4.82; N, 4.06; S, 9.28. Found: C, 50.36;
H, 4.95; N, 4.00; S, 9.47.

20 Example 254: Preparation of N-hydroxy-1-(2-methoxy-ethyl)-4-[[4-[4-(trifluoromethoxy)-phenoxy]phenyl]sulfonyl]-4-piperidine-carboxamide sulfate (2:1) (salt)

Part A: To a solution of Preparative Example VII (0.35 g, 0.58 mmol) in ethanol (1.5 mL) was added sulfuric acid (17 ?L, 0.32 mmol) and the solution was stirred at ambient temperature for 6 hours.

Evaporation of solvent and recrystallization from hot acetonitrile gave the title compound as a white powder (180 mg, 54.6%). Analytical calculation for C₂₂H₂₅N₂O₇SF₃.0.7H₂SO₄: C, 45.00; H, 4.53; N, 4.77; S, 9.28. Found: C, 44.77; H, 4.97; N, 4.41; S, 9.19.

Example 255: Preparation of N-hydroxy-1-(2-methoxy-ethyl)-4-[[4-[4-(trifluoromethoxy)-phenoxy]phenyl]sulfonyl]-4-piperidine-carboxamide phosphate (1:1) (salt)

Part A: To the ethyl acetate solution (4 mL) of Example 252 (0.5 g, 0.9 mmol) was added concentrated phosphoric acid (85%, 0.1248 g, 1.08 mmol) and solution was stirred at ambient temperature for 2 hours. Evaporation of the solvent and

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recrystallization from hot ethanol gave the title compound as a white powder (0.4917 g, 82.7%). Analytical calculation for $C_{22}H_{25}N_2O_7SF_3.H_3PO_4.H_2O$: C, 41.64; H, 4.77; N, 4.42. Found: C, 41.14; H, 4.64; N, 4.25.

Example 256: Preparation of N-hydroxy-1-(2-methoxy-ethyl)-4-[[4-[4-(trifluoromethoxy)-phenoxy]phenyl]sulfonyl]-4-piperidine-carboxamide monoacetate (salt)

Part A: To a solution of Example 252 (0.5 g, 0.9 mmol) in ethyl acetate (5 mL) was added concentrated acetic acid (63.7 mg, 1.08 mmol) and solution was stirred at ambient temperature for 2 hours.

Evaporation of the solvent and recrystallization from hot ethyl acetate gave the title compound as a white crystalline solid (0.4635 g, 83.0%). Analytical calculation for C₂₂H₂₅N₂O₇SF₃.0.7C₂H₄O₂: C, 50.14; H, 5.00; N, 5.00; S, 5.72. Found: C, 50.47; H, 5.09; N, 5.00; S, 6.13.

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Example 257: Preparation of N-hydroxy-1-(2-methoxy-ethyl)-4-[[4-[4-(trifluoromethoxy)-phenoxy]phenyl]sulfonyl]-4-piperidine-

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carboxamide 2-hydroxy-1,2,3-propanetricarboxylate (3:1) (salt)

Part A: To a solution of Example 252 (0.3 g, 0.578 mmol) in ethyl acetate (5 mL) was added citric acid (41 mg, 0.21 mmol) and the solution was stirred at ambient temperature for 2 hours. Evaporation of the solvent and recrystallization from hot ethanol gave the title compound as a white crystalline solid (0.181 g, 53.7%). Analytical calculation for C₂₂H₂₅N₂O₇SF₃.(1/3)C₆H₉O₇. 0.9H₂O: C, 48.34; H, 4.99; N, 4.70; S, 5.38. Found: C, 48.42; H, 4.99; N, 4.70; S, 5.38.

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Example 258: Preparation of N-hydroxy-1-(2-methoxy-ethyl)-4-[[4-[4-(trifluoromethoxy)-phenoxy]phenyl]sulfonyl]-4-piperidine-carboxamide monobenzenesulfonate (salt)

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Part A: To a solution of Preparative Example
VII, Part D (0.4 g, 0.66 mmol) in ethanol (2.5 mL)
was added benzene sulfonic acid (0.11 g, 0.69 mmol)
and the solution was stirred at ambient temperature
for 3 hours. Evaporation of the solvent and
recrystallization from hot ethanol at minus 20
degrees Celsius gave the title compound as white
crystals (0.28 g, 64.3%). Analytical calculation for
C22H25N2O7SF3.C6H6SO3.0.2H2O: C, 49.44; H, 4.65; N, 4.12;
S, 9.43. Found: C, 49.18; H, 4.67; N, 4.08; S, 9.75.

Example 259: Preparation of N-hydroxy-1-(2-methoxy-ethyl)-4-[[4-[4-(trifluoromethoxy)-phenoxy]phenyl]sulfonyl]-4-piperidine-carboxamide (2R,3R)-2,3-dihydroxy-butanedioate (2:1) (salt)

Part A: To a solution of Example 252 (0.3 g, 0.578 mmol) in ethyl acetate (5 mL) was added tartaric acid (48 mg, 0.3 mmol) and solution was stirred at ambient temperature for 2 hours.

Evaporation of the solvent and recrystallization from hot ethanol at zero degrees Celsius gave the title compound as a white solid (0.2 g, 58.3%). Analytical calculation for C₂₂H₂₅N₂O₇SF₃.0.5C₄H₆O₆. 1.25H₂O: C,

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46.79; H, 4.99; N, 4.55; S, 5.20. Found: C, 47.17; H, 5.20; N, 4.07; S, 5.03

Example 260: Preparation of N-hydroxy-1-(2-methoxy
ethyl)-4-[[4-[4-(trifluoromethoxy)phenoxy]phenyl]sulfonyl]-4-piperidinecarboxamide phosphate (3:1) (salt)

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Part A: To a solution of Example 252 (0.5 g, 0.9 mmol) in ethyl acetate (5 mL) was added phosphoric acid (37 mg, 0.32 mmol) and solution was stirred at ambient temperature for 2 hours. Evaporation of the solvent and recrystallization from hot ethanol at zero degrees Celsius gave the title compound as a white solid (0.312 g, 59%). Analytical calculation for C₂₂H₂₅N₂O₇SF₃.0.33H₃PO₄. 0.5H₂O: C, 47.18; H, 4.86; N, 5.00. Found: C, 47.15; H, 4.73; N, 4.90.

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Example 261: Preparation of N-hydroxy-1-[2-(1H-imidazol-1-yl)ethyl]-4-[[4-[4-(trifluoromethoxy)phenoxy]phenyl] sulfonyl]-4-piperidinecarboxamide, dihydrochloride

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Part A: The aryl ether from Example 230, Part B (3.12 g, 5.2 mmol) was dissolved in absolute methanol (50 mL). Acetyl chloride (2.1 mL, 30 mmol) was added over 1 minute. The reaction was stirred for 4 hours, concentrated, azeotroped with chloroform/acetonitrile, and dried in vacuo, affording the desired hydroxyethyl compound as a white solid (2.75 g, 96%). The desired hydroxyethyl product was characterized by NMR spectroscopy.

Part B: To the dichloromethane solution of the hydroxyethyl compound of Part A (1 g, 1.9 mmol) was added thionyl chloride (3.8 mmol) and reaction solution was stirred at ambient temperature for 12 15 hours. Concentration in vacuo provided the chloride as a light yellow gel. To the solution of the chloride and potassium carbonate (0.54 g, 3.8 mmol) in N,N-dimethylformamide (5 mL) was added imidazole (0.4 g, 5.7 mmol) and solution was stirred at ambient 20 temperature for 12 hours. Then N, N-dimethylformamide was evaporated under high vacuum and the residue was diluted with ethyl acetate. The organic layer was washed with water and dried over magnesium sulfate. Concentration in vacuo and chromatography on silica eluting with ethyl acetate/hexane provided the

imidazole ethyl ester as a light yellow gel (0.82 g,
75.2%).

Part C: To a solution of imidazole ethyl ester of part A (0.82 g, 1.44 mmol) in ethanol (3 mL) and 5 tetrahydrofuran (3 mL) was added sodium hydroxide (0.57 g, 14.4 mmol) in water (6 mL) at ambient temperature. The solution was then heated to sixty degrees Celsius for 18 hours. The solution was concentrated in vacuo and the residue was dissolved in acetonitrile. Concentrated hydrochloric acid was 10 used to acidify the residue to pH = 1 and concentration in vacuo gave the carboxylic acid as the product. To a solution of the carboxylic acid, N-methyl morpholine (0.62 mL, 5.7 mmol), 1hydroxybenzotriazole (0.59 g, 4.3 mmol) and 0-15 tetrahydropyranyl hydroxyl amine (0.34 g, 2.9 mmol) in N,N-dimethylformamide (30 mL) was added 1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide hydrochloride (0.83 g, 5.7 mmol) and the solution was stirred at ambient temperature for 24 hours. The 20 solution was concentrated under high vacuum and the residue was dissolved in ethyl acetate. The organic layer was washed with saturated aqueous Sodium bicarbonate, water and dried over magnesium sulfate. Concentration in vacuo and chromatography on silica eluting with ethyl acetate/hexane provided the tetrahydropyranyl amide as a white foam (0.27 g, 29.7%).

Part D: To a solution of 4N hydrochloric acid in dioxane (2 mL, 8 mmol)) was added a solution of the tetrahydropyranyl amide of part B (0.27 g, 0.45 mmol) in methanol (1 ml) and 1,4-dioxane (3 mL) and was stirred at ambient temperature for 3 hours.

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Evaporation of solvent and trituration with ethyl ether gave the title compound as a white solid (0.179 g, 67%). Analytical calculation for $C_{24}H_{25}N_4O_6SF_3$. 2HCl.1.25H₂O: C, 44.35; H, 4.57; N, 8.62. Found: C, 44.57; H, 4.36; N, 7.95.

Example 262: Preparation of N-hydroxy-1-(2-methoxyethyl)-4-[[4-[4-(1H-1,2,4-triazol-1-yl)phenoxy]phenyl]sulfonyl]-4-piperidinecarboxamide trihydrochloride

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Part A: To a solution of the product of Preparative Example II , Part D, (1.5 g, 3.6 mmol) 15 and powdered potassium carbonate (0.99 g, 7.2 mmol) in N,N-dimethylformamide (10 mL) was added 4-(1,2,4triazole-1-yl)phenol (0.87 g, 5.4 mmol) at ambient temperature and the solution was heated to ninety 20 degrees Celsius for 32 hours. Solution was concentrated under high vacuum and the residue was dissolved in ethyl acetate. The organic layer was washed with 1N sodium hydroxide, water and dried over magnesium sulfate. Chromatography on silica eluting with ethyl acetate/hexane provided the N-Boc diaryl 25 ether as a light yellow gel (0.907 g, 44.5%).

Part B: To a solution of N-Boc diaryl ether of part A (0.907 g, 1.6 mmol) in dichloromethane (3 mL)

was added trifluoroacetic acid (3 mL) and the solution was stirred at ambient temperature for 1 hour. Concentration in vacuo provided the amine trifluoroacetate salt as a light yellow gel. To the solution of the amine trifluoroacetate salt and potassium carbonate (0.44 g, 3.2 mmol) in N,Ndimethylformamide (5 mL) was added 2-bromoethyl methyl ether (0.36 mL, 3.8 mmol) and solution was stirred at ambient temperature for 36 hours. N, N-dimethylformamide was evaporated under high vacuum and the residue was diluted with ethyl acetate. The organic layer was washed with water and dried over magnesium sulfate. Concentration in vacuo provided the methoxyl ethyl amine as a light yellow 15 gel (0.82 g, 91%).

Part C: To a solution of the methoxyl ethyl amine of part B (0.80 g, 1.4 mmol) in ethanol (3 mL) and tetrahydrofuran (3 mL) was added sodium hydroxide (0.56 g, 14 mmol) in water (6 mL) at ambient temperature. The solution was then heated to sixty 20 degrees Celsius for 18 hours. The solution was concentrated in vacuo and the residue was dissolved in acetonitrile. Concentrated hydrochloric acid was used to acidify the residue until the pH = 1 and 25 concentration in vacuo gave the carboxylic acid as product. To a solution of the carboxylic acid, Nmethyl morpholine (0.92 mL, 8.4 mmol), 1hydroxybenzotriazole (0.57 g, 4.3 mmol) and 0tetrahydropyranyl hydroxyl amine (0.34 g, 2.9 mmol) 30 in N,N-dimethylformamide (30 mL) was added 1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide hydrochloride (0.80 g, 4.2 mmol) and the solution was stirred at ambient temperature for 24 hours.

solution was concentrated under high vacuum and the residue was dissolved in ethyl acetate. The organic layer was washed with saturated aqueous sodium bicarbonate, water and dried over magnesium sulfate. Concentration in vacuo and chromatography on silica eluting with ethyl acetate/hexane provided the tetrahydropyranyl amide as a white foam (0.39 g, 47.6%).

Part D: To a solution of 4N hydrochloric acid in dioxane (1.6 mL, 6.4 mmol)) was added a solution of the tetrahydropyranyl amide of part C (0.39 g, 0.66 mmol) in methanol (2 ml) and dioxane (6 mL) and was stirred at ambient temperature for 3 hours.

Evaporation of the solvent and trituration with ethyl ether gave the title compound as a white solid (0.34 g, 83%). ESI MS calculated for C₂₃H₂₇N₅O₆S: 501, found 501.

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Part A: To a methanol solution of the product of Example 253 (1.0 g, 1.4 mmol) and 20% palladium on carbon (1.5 g) was added ammonium

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formate (2.4 g, 38 mmol) and reaction solution was heated to reflux for 72 hours. The reaction solution was filtered through Celite and the filtrate was concentrated in vacuo. The residue was dissolved in ethyl acetate and washed with saturated aqueous Sodium bicarbonate, water and dried over magnesium sulfate. Concentration in vacuo and chromatography on a C-18 reverse phase column eluting with acetonitrile/water with hydrochloric acid provided the title compound as a white powder (181 mg, 23.2%). Analytical calculation for C₂₂H₂₅N₂O₆SF₃.HCl: C, 49.03; H, 4.86; N, 5.20. Found: C, 48.80; H, 4.93; N, 5.29.

Example 264: Preparation of N-hydroxy-1-[3-(4-morpholinyl)propyl]-4-[[4-[4-(trifluoromethoxy)phenoxy]phenyl]
sulfonyl]-4-piperidinecarboxamide
dihydrochloride

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Part A: To a solution of the product of Preparative Example II, Part D, (15 g, 36 mmol) and powdered potassium carbonate (10 g, 72 mmol) in N,N-dimethylformamide (200 mL) was added 4-(trifluoromethoxy)phenol (19.3 mL, 72 mmol) at ambient temperature and the solution was heated to ninety degrees Celsius for 25 hours. The solution

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was concentrated under high vacuum and residue was dissolved in ethyl acetate. The organic layer was washed with 1N sodium hydroxide, water and dried over magnesium sulfate. Chromatography on silica eluting with ethyl acetate/hexane provided trifluoromethoxy phenoxyphenyl sulfone as a light yellow gel (20 g, quantitative).

Part B: To a solution of trifluoromethoxyl phenoxyphenyl sulfone (1.0 g, 1.75 mmol) of part A in dichloromethane (1 mL) was added trifluoroacetic acid 10 (1 mL) and the solution was stirred at ambient temperature for 1 hour. Concentration in vacuo provided the amine trifluoroacetate salt as a light yellow gel. To the solution of the amine trifluoroacetate salt and potassium carbonate (0.48 q, 3.5 mmol) in N,N-dimethylformamide (10 mL) was added morpholino propyl chloride (0.68 g, 3.5 mmol) and solution was stirred at 40 degree Celsius for 36 hours. The N, N-dimethylformamide was evaporated under high vacuum and the residue was diluted with 20 ethyl acetate. The organic layer was washed with water and dried over magnesium sulfate. Concentration in vacuo provided the morpholino propyl amine as a light yellow gel (1 g, quantitative 25 yield).

Part C: To a solution of morpholino propyl amine of part B (1 g, 1.6 mmol) in ethanol (3 mL) and tetrahydrofuran (3 mL) was added sodium hydroxide (0.67 g, 16 mmol) in water (6 mL) at ambient temperature. The solution was then heated to sixty degrees Celsius for 18 hours. The solution was concentrated in vacuo and the residue was dissolved in acetonitrile. Concentrated hydrochloric acid was

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used to acidify the residue to pH = 1 and concentration in vacuo gave the carboxylic acid as the product. To a solution of the carboxylic acid, N-methyl morpholine (0.18 mL, 4.8 mmol), 1hydroxybenzotriazole (0.45 g, 3.2 mmol) and 0tetrahydropyranyl hydroxyl amine (0.3 g, 2.5 mmol) in N, N-dimethylformamide (30 mL) was added 1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide hydrochloride (0.64 q, 3.2 mmol) and the solution was stirred at ambient temperature for 24 hours. 10 solution was concentrated under high vacuum and the residue was dissolved in ethyl acetate. The organic layer was washed with saturated aqueous Sodium bicarbonate, water and dried over magnesium sulfate. Concentration in vacuo and chromatography on silica eluting with ethyl acetate/hexane provided the tetrahydropyranyl amide as a white foam (0.56 g, 50%).

Part D: To a solution of 4N hydrogen chloride in dioxane (2 mL, 8 mmol)) was added a solution of the tetrahydropyranyl amide of part C (0.56 g, 0.83 mmol) in methanol (3 ml) and dioxane (8 mL) and was stirred at ambient temperature for 3 hours. Evaporation of solvent and tritration with ethyl ether gave the title compound as a white solid (0.476 g, 86.5%). Analytical calculation for C₂₆H₃₂N₃O₇SF₃.2HCl: C, 47.28; H, 5.19; N, 6.36; S, 4.85. Found: C, 46.86; H, 5.35; N, 6.29; S, 5.09.

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Example 265: Preparation of N-hydroxy-1-(1Himidazol-2-ylmethyl)-4-[[4-[4(trifluoromethyl)phenoxy] phenyl]sulfonyl]-4-piperidinecarboxamide
dihydrochloride

HOHN CF₃

Part A: To a suspension of the hydrochloride salt from Preparative Example VIII, Part F, (0.988 g, 10 21.6 mmol) and 2-imidazolecarboxaldehyde (315 mg, 3.28 mmol) in methanol (5 mL) at room temperature was added borane-pyridine complex (0.41 mL, 3.28 mmol). After 18 hours the reaction was concentrated under a stream of nitrogen. Saturated aqueous sodium 15 bicarbonate was then added and the mixture was extracted with ethyl acetate (3X). The combined organic extracts were washed with water and brine and dried over sodium sulfate. Concentration gave a residue which was purified on silica gel eluting with 20 ammonia-saturated methanol/methylene chloride (3/97) to afford the desired 4(5)-imidazole derivative (1.04 g, 89.7 %) as a yellow solid. MS MH calculated for $C_{25}H_{26}N_3O_5SF_3$: 538, found 538.

25 Part B: A solution of sodium hydroxide (766 mg, 19.2 mmol) in water (5 mL) was added to a solution of the 4(5)-imidazole derivative of Part A (1.03 g, 1.92 mmol) in tetrahydrofuran (5 mL) and ethanol (5 mL) and the resulting solution was stirred

at ambient temperature for 66 hours. The solution was concentrated in vacuo to afford a residue which was treated with 2 N aqueous hydrochloric acid (14.4 mL, 28.8 mmol). Concentration afforded the desired carboxylic acid as a yellow foam which was used directly without purification.

Part C: To a solution of the carboxylic acid of Part B in dimethylformamide (15 mL) was added sequentially N-methylmorpholine (1.16 g, 11.5 mmol), N-hydroxybenzotriazole (311 mg, 2.30 mmol), 1-(3-10 dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (478 mg, 2.50 mmol), and Otetrahydropyranyl hydroxylamine (303 mg, 2.6 mmol). After 16 hours at ambient temperature the reaction was warmed to 51 degrees Celsius for 2 hours and then 15 concentrated in vacuo. Water was added and the mixture was extracted sequentially with ethyl acetate and with methylene chloride. The combined organic extracts were washed with brine and dried over sodium sulfate. Concentration gave a residue which was 20 chromatographed on silica gel eluting with ammoniasaturated methanol/methylene chloride (7/93) to afford the desired tetrahydropyranyl-protected hydroxamate (0.50 q, 43%) as an off-white foam. MS MH⁺ calculated for $C_{28}H_{31}F_3N_4O_6S$: 609, found 609. 25

Part D: To a solution of tetrahydropyranylprotected hydroxamate of part C (500 mg, 0.82 mmol)
in methanol (1mL) and 1,4-dioxane (5 mL) was added 4
N hydrogen chloride/dioxane (2.5 mL). After stirring
at ambient temperature for 1 hours, the solution was
concentrated in vacuo. Trituration with diethyl
ether provided the title compound as a white solid
(490 mg, quantitative yield). HRMS MH+ calculated

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for $C_{23}H_{23}N_4SO_5F_3$: 525. Found: 525. MS MH^+ calculated for $C_{23}H_{23}F_3N_4O_5S$: 525, found 525.

Example 266: Preparation of 1-cyclopropyl-N-hydroxy
4-[[4-[4-(trifluoromethoxy)phenoxy]
phenyl]sulfonyl]-4-piperidinecarboxamide

To a solution of the product of Preparative Example IX (2.08 g, 4.0 mmol) in warm water (200 mL) was added sodium bicarbonate to pH = 8 and the solution was stirred for 1 hour. The resulting white solid was isolated by filtration, washed with water and dried at 40°C for 48 hours to afford the title compound as a white solid (1.82 g, 94%). Analytical calculation for C₂₂H₂₃N₂SF₃O₅:H₂O, 52.50; H, 5.01; N, 5.57; S, 6.38. Found: C, 52.24; H, 4.65; N, 5.46; S, 6.75.

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Example 267: Preparation of 1-cyclopropyl-N-hydroxy4-[[4-[4-(trifluoromethoxy)phenoxy]phenyl]sulfonyl]-4-piperidinecarboxamide
mono(4-methylbenzenesulfonate) (salt)

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To a solution of the product of Example 266 (550 mg, 1.10 mmol) in ethanol (5 mL) was added ptoluenesulfonic acid (240 mg, 1.26 mmol) and the
reaction was then stirred for 3.5 hour. The
resulting white solid was isolated by filtration,
washed with ethanol and dried at 40°C for 48 hours to
afford the title compound as a white solid (633 mg,
Recrystallized from methanol/water afforded
the title compound as analytically pure material.
Analytical Calculation for C₂₉H₃₁N₂S₂F₃O₉: 51.78; H,
4.64; N, 4.16. Found: C, 51.44; H, 4.32; N, 4.18.

15 Example 268: Preparation of 1-cyclopropyl-N-hydroxy4-[[4-[4-(trifluoromethoxy)phenoxy]phenyl]sulfonyl]-4-piperidinecarboxamide monomethanesulfonate (salt)

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To a solution of the product of Example 266 (550 mg, 1.13 mmol) in ethanol (5 mL) was added

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methane sulfonic acid (82 μL) and the r action was then stirred for 3.5 hours. Concentration *in vacuo* afforded the title compound as a solid (640 mg, 97%). Recrystallization from methanol afforded analytically pure title compound. Analytical Calculation for C₂₃H₂₇N₂S₂F₃O₉: 46.30; H, 4.56; N, 4.70, S, 10.75. Found: C, 46.10; H, 4.71; N, 4.65; S, 10.99.

Example 269: Preparation of 1-cyclopropyl-N-hydroxy
4-[[4-[4-(trifluoromethylphenoxy]phenyl]sulfonyl]-4-piperidinecarboxamide

To a solution of the product of Preparative Example X (2.15 g, 4.0 mmol) in warm water (200 mL) was added sodium bicarbonate to pH = 8. The solution was stirred for 1 hour. The resulting white solid was isolated by filtration, washed with water and dried at 40 degrees Celsius for 48 hours to afford the titled compound as a white solid (1.96 g, 98%). Analytical Calculation for C₂₂H₂₃N₂SF₃O₅:2H₂O: C, 51.26; H, 5.24; N, 5.44; S, 6.21. Found: C, 50.58; H, 4.72; N, 5.33; S, 6.04.

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Example 270: Preparation of 1-cyclopropyl-N-hydroxy
4-[[4-[4-(trifluoromethylphenoxy]phenyl]sulfonyl]-4-piperidinecarboxamide

mono(4-methylbenzenesulfonate)(salt)

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To a solution of the product of Example 269 (550 mg, 1.13 mmol) in ethanol (5 mL) was added p10 toluenesulfonic acid (248 mg, 1.26 mmol) and the
solution was stirred for 3.5 hours. The resulting
white solid was isolated by filtration, washed with
ethanol and dried at 40°C for 48 hours to afford the
title compound as a white solid (705 mg, 95%).
15 Recrystallized from methanol afforded analytically
pure material. Analytical Calculation for
C29H31N2S2F3O8: C, 53.04; H, 4.76; N, 4.27; S, 9.77
Found: C, 52.94; H, 4.46; N, 4.30; S, 9.99.

20 Example 271: Preparation of 1-cyclopropyl-N-hydroxy4-[[4-[4-(trifluoromethylphenoxy]phenyl]sulfonyl]-4-piperidinecarboxamide monomethanesulfonate (salt)

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To a solution of the product of Example 269 (550 mg, 1.13 mmol) in ethanol (5 mL) was added

5 methane sulfonic acid (79 μL) and the reaction was stirred for 3.5 hours. Concentration in vacuo gave the title compound as a solid (569 mg, 87%).

Analytical Calculation for C₂₃H₂₇N₂S₂F₃O₈: C, 47.58; H, 4.69; N, 4.82. Found: C, 47.15; H, 4.18; N, 4.74.

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Example 272: Preparation of 1-acetyl-N-hydroxy-4[[4-[4-(trifluoromethoxy)phenoxy]phenyl]sulfonyl]-4-piperidinecarboxamide

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Part A: To a solution of the product of Preparative Example II, Part D (33.2 g, 80.0 mmol) in dimethylformamide (150 mL) was added cesium carbonate (65.2 g, 200 mmol) and 4-(trifluromethoxy)phenol (21.4 g, 120 mmol). The solution was mechanically stirred at sixty degrees Celsius for 24 hours. The solution was then diluted with water (1 L) and extracted with ethyl acetate. The organic layer was

washed with water, saturated aqueous sodium chloride and dried over magnesium sulfate, then filtered and concentrated *in vacuo*. Chromatography on silica gel eluting with 20% ethyl acetate/hexane provided the desired diaryl sulfide as a white solid (45.0 g, quantitative yield).

Part B: To a solution of the diaryl sulfide from part A (24 g, 42.8 mmol) in ethanol (80 mL) and tetrahydrofuran (80 mL) was added a solution of NaOH (14.8 g, 370 mmol) in water (100 mL) and the solution was heated at sixty degrees Celsius for 18 hours. The solution was concentrated in vacuo and the aqueous residue was acidified to pH = 5.0 and extracted with ethyl acetate. The organic extract was washed with saturated aqueous sodium chloride and dried over magnesium sulfate, then filtered and concentrated in vacuo to give the desired carboxylic acid as a white foam (23.0 g, quantitative yield)

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Part C: To a solution of carboxylic acid of part B (22.8 g, 43.0 mmol) in ethyl acetate (400 mL) cooled to zero degrees Celsius was bubbled gaseous Hydrogen chloride for 20 minutes. The reaction was stirred at this temperature for 2.5 hours. The solution was then concentrated in vacuo to afford the desired hydrochloride salt as a white foam (21.0 g, quantitative yield).

Part D: To a solution of the hydrochloride salt of part C (17.0 g, 35.0 mmol) in acetone (125 mL) and water (125 mL) was added triethyl amine (24 mL, 175 mmol). The reaction was cooled to zero degrees Celsius and acetyl chloride (3.73 mL, 53.0 mmol) was added. The solution was then stirred at ambient temperature for 18 hours. Concentration in vacuo gave a residue which was acidified with aqueous hydrochloric acid to pH 1.0 and then extracted with ethyl acetate. The organic layer was washed with water, saturated aqueous sodium chloride and dried

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over magnesium sulfate, then filtered and concentrated *in vacuo* to give the desired methanesulfonamide as a white solid (17.0 g, quantitative yield).

Part E: To a solution of the methanesulfonamide 5 of part D (14.4 g, 29.6 mmol) in dimethylformamide (250 mL) was added 1-hydroxybenzotriazole (4.8 g, 35.5 mmol), N-methyl morpholine (12.3 mL, 88.8 mmol) and O-tetrahydropyranyl hydroxyl amine (5.2 g, 44.4 mmol) followed by 1-3-(dimethylamino) propyl]-3-ethyl 10 carbodiimide hydrochloride (7.99 q, 41.4 mmol). The solution was stirred at ambient temperature for 18 The solution was diluted with water (500 mL) and extracted with ethyl acetate. The organic layer was washed with saturated aqueous sodium chloride and 15 dried over magnesium sulfate, then filtered and concentrated in vacuo. Chromatography on a C18 reverse phase column eluting with acetonitrile/water provided the desired tetrahydropyranyl-protected hydroxamate as a white solid (12.0 g, 71%). 20

Part F: To a solution of tetrahydropyranyl-protected hydroxamate of part E (12.0 g, 20.5 mmol) in dioxane (250 mL) and methanol (50 mL) was added 4 N hydrogen chloride/dioxane (51 mL). After stirring at ambient temperature for 3.5 hours the solution was concentrated in vacuo. Trituration with diethyl ether and filtration provided the title compound as a white solid (8.84 g, 85%). HRMS MH calculated for C,H,N,SO,F: 503502.1021. Found 502.0979.

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Example 273: Preparation of N-hydroxy-1-(methyl sulfonyl)-4-[[4-[4-sulfonyl]- (trifluoromethoxy)phenoxy]phenyl]-4-piperidinecarboxamide

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Part A: To a solution of the product of Preparative Example II, Part D, (33.2 g, 80.0 mmol) in dimethylformamide (150 mL) was added cesium carbonate (65.2 gm, 200.0 mmol) and 4-(trifluromethoxy)phenol (21.4 g, 120 mmol). solution was mechanically stirred at sixty degrees Celsius for 24 hours. The solution was then diluted with water (1 L) and extracted with ethyl acetate. 10 The organic layer was washed with water, saturated aqueous sodium chloride and dried over magnesium sulfate, then filtered and concentrated in vacuo. Chromatography on silica gel eluting with 20% ethyl acetate/hexane provided the desired diaryl sulfide as a white solid (45.0 gm, quantitative yield). 1.5

Part B: To a solution of the diaryl sulfide from part A (21 g, 37.0 mmol) in ethanol (80 mL) and tetrahydrofuran (80 mL) was added a solution of NaOH (14.8 g, 370 mmol) in water (75 mL) and the solution was heated at sixty degrees Celsius for 18 hours. The solution was concentrated in vacuo and the aqueous residue was acidified to pH = 5.0, and then extracted with ethyl acetate. The organic extract was washed with saturated aqueous sodium chloride and dried over magnesium sulfate, then filtered and concentrated in vacuo to give the desired carboxylic acid as a white foam (19.3 g, 97%)

Part C: To a solution of carboxylic acid of part B (19.3 g, 37.0 mmol) in ethyl acetate (400 mL) cooled to zero degrees Celsius was bubbled gaseous hydrogen chloride for 30 minutes. The reaction was stirred at this temperature for 2.5 hours. The

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solution was then concentrated *in vacuo* to afford the desired hydrochloride salt as a white foam (15.8 g, 93%).

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Part D: To a solution of the hydrochloride salt of part C (15.8 g, 33.0 mmol) in acetone (100 mL) 5 and water (100 mL) was added triethyl amine (23 mL, 164 mmol). The reaction was cooled to zero degrees Celsius and methanesulfonyl chloride (5.1 mL, 66.0 mmol) was added. The solution was stirred at ambient temperature for 18 hours. The reaction was 10 concentrated in vacuo and acidified with aqueous hydrochloric acid to pH 1.0. The aqueous residue was extracted ethyl acetate. The organic extract was washed with water, saturated sodium chloride and dried over magnesium sulfate, then filtered and 15 concentrated in vacuo to give the desired carboxylic acid methanesulfonamide as a white solid (17.6 gm, quantitative yield).

Part E: To a solution of the methanesulfonamide of part D (18 g, 35.0 mmol) in dimethylformamide (150 20 mL) was added 1-hydroxybenzotriazole (5.66 gm, 42.0 mmol), N-methyl morpholine (14.0 mL, 105.0 mmol) and O-tetrahydropyranyl hydroxyl amine (6.1 g, 52 mmol) followed by 1-3-(dimethylamino) propyl]-3-ethyl carbodiimide hydrochloride (9.4 gm, 49.0 mmol). 25 solution was stirred at ambient temperature for 18 The solution was diluted with water (500 mL) and extracted with ethyl acetate. The organic extract was washed with saturated aqueous sodium chloride and dried over magnesium sulfate, then 30 filtered and concentrated in vacuo. Chromatography on a C18 reverse phase column eluting with acetonitrile/water provided desired tetrahydropyranyl-protected hydroxamate as a white solid (8.17 g, 41%). 35

Part F: To a solution of tetrahydropyranylprotected hydroxamate of part E (8.17 g, 13.0 mmol) WO 00/69821 PCT/US00/06719

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in dioxane (100 mL) and methanol (100 mL) was added 4 N hydrogen chloride/dioxane (50 mL). After stirring at ambient temperature for 3.5 hours the solution was concentrated in vacuo. Trituration with diethyl ether provided the title compound as a white solid (6.83 g, 92%). MS MH calculated for $C_{20}H_{21}NS_2O_8F$: 539. Found 539.

The following compounds were prepared by

10 parallel synthesis (resin based synthesis, automated synthesis) procedures utilizing reactions such as acylation and nucleophilic displacement:

Example 274:

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Example 275:

10 Example 276:

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Examples: 277-315

Example	R ₁ R ₂ NH	Amine	MS (ES)
			m/z
277	\sim NH ₂	Ethyl amine	592 (M+H)
278	/=	3-(Aminomethyl)	
	N-NH ₂	pyridine	655 (M+H)
279	HN-		
	(N)	Imidazole	615 (M+H)

280	H_2N OH	3-Amino-1-propanol	622	(M+H)
281	HN			
	NH ₂	Histamine	658	(M+H)
282	~\s\	2-Thiophene		
	H_2N	methyl amine	660	(M+H)
283				
	O NH	Morpholine	634	(M+H)
284	/==N	2-(Aminomethyl)		•
	NH ₂	pyridine	655	(M+H)
285		4-(Aminomethyl)		
	NH ₂	pyridine	655	(M+H)
286	H ₂ N OH	Ethanolamine	608	(M+H)
287	H N			
	/ N.	N,N,N-Trimethyl	649	(M+H)
200		ethylenediamine		•
288	HNN—	1-Methylpiperazine	647	(M+H)
289	H_2N	N,N-Dimethyl		
	. <i>I</i>	ethylenediamine	635	(M+H)
290	HN_NH	Piperazine	633	(M+H)
291				
	HNS	Thiomorpholine	650	(M+H)
292	√ N	N-Propylcyclopropne		
	Ĥ, Ž	methylamine	660	(M+H)
293	H ₂ N	(Aminomethyl)		
		cyclopropane	618	(M+H)
294	\Z	Dimethylamine	592	(M+H)
295	<u> </u>			
-4-	. п	Diethylamine	620	(M+H)
296	NH	Piperidine	632	(M+H)
297		·	•	
	N N N N N N N N N N N N N N N N N N N	(R)-(-)-2-	648	(M+H)
	••	Pyrrolidine		

		methanol		
298	NH			
	\smile	Pyrrolidine	618	(M+H)
299	HN N O OH	1-(2-(2-		
		Hydroxyethoxy)	721	(M+H)
		ethyl)piperazine		
300	HNCO ₂ NH ₂	Isonipecotamide	675	(M+H)
301	H ₂ N ✓ OH	2-(2-Aminoethoxy)		
	0 1	ethanol	652	(M+H)
302	N N	3,3'-Iminobis(N,N-dimethylpropylamine)	734	(M+H)
303	0-	Bis(2-Methoxy		
	-0-	ethyl)amine	680	(M+H)
304		4-Hydroxy		
	HN OH	piperidine	648	(M+H)
305	HN N	N-(Carboethoxy methylpiperazine	719	(M+H)
306	ON-NH	1-(2- Morpholinoethyl) piperazine	746	(M+H)
307		1-(2-Methoxyethyl)		
	ONNH	piperazine	691	(M+H)
308	_N	1-(2-		
	/ * N NH	Dimethylaminoethyl)	704	(M+H)
		piperazine		
309	H ₂ N	2-Methoxyethylamine	622	(M+H)
310	F NH ₂	2,2,2-Trifluoroethyl amine	646	(M+H)
311	NH	1,2,4-Triazole		(M+H)
312	N NH ₂	Methoxyamine		(M+H)
	0-141.12	Hechovyamine	J 3 4	(m'm)

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Examples: 316-332

HO N S O
$$CF_3$$
 F_2B_1N

Example	R ₁ R ₂ NH	Amine	MS (ES) m/z
316		3-(Aminomethyl)	
	NH ₂	pyridine	593 (M+H)
317	HN	Imidazole	553 (M+H)
318	HN	Piperidine	570 (M+H)
319	O_NH	Morpholine	572 (M+H)
320	/=N	2-(Aminomethyl)	
	NH ₂	pyridine	593 (M+H)
321	H ₂ N OH	Ethanolamine	546 (M+H)
322	F	2,2,2-Trifluoro	
	F NH ₂	ethylamine	584 (M+H)
323	H .	·	
	\N\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	N,N,N-Trimethyl	587 (M+H)
	,	ethylenediamine	

324	HN_N-	1-Methylpiperazine	585	(M+H)
325		4-(Aminomethyl)		
•	NH ₂	pyridine	593	(H+H)
326	NH			
	\	Pyrrolidine	556	(M+H)
327	UN	Bis(2-Methoxy		
	-0-	ethyl)amine	618	(M+H)
328	HN NH			
	'''\'''	Piperazine	571	(M+H)
329				
	N NH	4-(Ethylamino	621	(M+H)
		methyl)pyridine		
330		1-(2-Methoxy		
	ON_NH	ethyl)pyridine	629	(M+H)
331	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	N-		
	, H	Propylcyclopropane	598	(M+H)
		methylamine		•
332	H_2N O	2-Methoxyethylamine	560	(M+H)

Examples: 333-347

Example	R ₁ R ₂ NH	Amine	MS (ES) m/z
333	NH ₂	3-(Aminomethyl) pyridine	635 (M+H)

334	HN	Piperidine	612 (M+H)
335	O_NH	Morpholine	614 (M+H)
336	NH ₂	2-(Aminomethyl) pyridine	635 (M+H)
337	H ₂ N OH	Ethanolamine	588(M+H)
338	_H	N,N,N-Trimethyl ethylenediamine	629 (M+H)
339	HN_N—	1-Methylpiperazine	627 (M+H)
340	NH ₂	4-(Aminomethyl) pyridine	636 (M+H)
341	NH	Pyrrolidine	598 (M+H)
342	HN-0-	Bis(2-Methoxy ethyl)amine	660 (M+H)
343	ниин	Piperazine	613 (M+H)
344	NH	4-(Ethylamino methyl)pyridine	663 (М+Н)
345	ONNH	1-(2-Methoxy ethyl)pyridine	671 (M+H)
346	~ ₽	N-Propylcyclopropane methylamine	640 (M+H)

Examples 348-942:

The following compounds were prepared in a manner similar to that used in the preceding examples. In the tables that follow, a generic structure is shown above the table with substituent groups being illustrated in the table along with available mass spectral data.

Example	R	К	MS (ES) m/z
348	OCF ₃	G	·
349	o C OCF₃ HCI	/ =	499.1131
350	O OCF3		583
351	o CF₃	Z= CH ₃	580.1366
352	O COCF3		538.1282

353	O OCF3	Ĵ _N Coch₃	610
354	2HCI	Z	
355	O OCF3	NH CI	648
356	2HCI CF3	Z	
357	OCF3	Å N Och₃	610
358	2HCI NO ₂	<u> </u>	
359	o OCF3	J. C.	648
360	2HCi	<u> </u>	
361	OCF ₃	١	616
362	o CPOCF ₃	Ĵ _Ŋ Ĵ _G	614
363	OCF ₃	Ĵ _Ŋ Ţ ^F	616
364	O-OCF3	NH CF3	648
365	o CF ₃	Å _I Ç	614
366	OCF ₃	° NH CF₃	648

367	HCI CF3		
368	2HCI OCH ₃	\forall	
369	2HCI 2HCI	∀	
370	HCI: 0	\forall	
371	O O O O O O O O O O O O O O O O O O O	\forall	
372	O OCF3		539.1201
373	N HGI	\forall	
374	o CF₃	NO ₂	567.1120
375	o ← CF ₃	N CF₃	590.1174
376	NO HCI	∀	
378	HCI	\forall	474.1567
379	OCF₃ HCI		555
380	o CF₃	NH₂	537.1412
381	o-CF₃		523
382	N S N S	. 6	

383	NO HCI	\forall	
384	o CF₃	z civ	547.1279
385	NOCH₃)₂ HCI	∀	
386	O CF3		555.1516
387	O OCF ₃	Z CF ₃	607.1061
388	HC A CP	∀	
389	Û₅.	HCI	516
390	OCF ₃	Z Z	539
391	O CF3		538.1272
392	HCI OCF3		538.1252
393		∀	
394	HCI CF3	∀	
395	OCF3		522.1351
396	TFA OCF3	\forall	582.2245

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397	HŅ	<u> </u>	532.2280
		V	
	TFA 6		·
398			
		\triangle	
	N CH3	,	
			}
	2HCl		
399	CF ₃		528
	0	N S	
	HCI		
400	N CH₃	L	
		V	
	HCI -	•	515 2244
401			515.3344
		V	
	HOAc U		
402	N^		582.2266
	CF ₃	\triangle	
}		•	
	HOAc 🗸		
403	CF ₃	_\s\	
	0	N	
404	HCI	0	550
404	CF ₃	Ů,,,	330
			1
405	HCI	9	550
405	CF ₃	Ĭ.	330
	1101		·
106	HCI		555
406	CF ₃	<u></u>	
		ĺ	
407	HCI 0. A		
101		√ 0	
	OCF ₃		
	HCI		
408		<u> </u>	600.2162
300	N	\triangle	
	CF ₃	V	
	. TFA 🗸 🗸		<u> </u>
409	N\		548
		∇	
	HCI ³²	·	<u> </u>

410	OCF ₃		
411	OCF ₃		
412	OCF ₃	NH CH₃	502
413	OCF ₃	~ <u>``</u> `	529
414	N CF3	\forall	600.2141
415	N S → OCF ₃	4	600
416	O _{OCF3}	<mark>Д</mark> н	489
417	N S S S S S S S S S S S S S S S S S S S	\forall	530
418	HOAc OCF3	\forall	598.2200
419	HOAC CI	∀	548.2013
420	NO.		569.2259
421	100 o o o o o o o o o o o o o o o o o o		570.2186
422	NO OCF3		635.2185
423	NO OCF3	~~~	636.2104

424	NO _O		586.2059
425	NO OCF3		562.1957
426	NO CI		585.1968
427	, Co	~	586.1936
428	N 0 ← CF3		637.2137
429	NO OCF3	Z = = = = = = = = = = = = = = = = = = =	638.2072
430	CF ₃		637.2146
431	°CF3	z z	638.2075
432		i,	602.1731
433	N 0 CF₃	i,	654.1921
434	CF ₃	Ļ	654.1932
435	NOCH ₃		636
436	N → COCH₃	i c	596

437	N → OCH₃	-н	502
438	TFA OCH ₃		579
439	OCF ₃	7	411.1211
440	N OC₄H₃ HCI	\forall	480
441	TFA O OCH ₃	\forall	542
442	N OC₄H₃ HCI	Î.Ļ	540
443	NOC ₄ H ₉	–н	440
444	N OC₄H₃ HCI		518
445	NOAC CI	∕ОСН₃	566
446	NOCCF3 HOAC F	OCH ₃	618
447	N CF3	∕ОСН₃	616
4 48		∕OCH3	550.2387
4 4 9		∕ OCH₃	616
4 49	Ö.,	∕ ОСН₃	616

	0, 6		
450		N .	
İ	HCI OCF₃	, N	
451	9	1 9	580.1370
	OCF ₃	N OCH₃	
4-0-	HCI		
452	OCF ₃	Å,	
453	°Q	-	
AFA	HCI OCF ₃	,	614
454	HCI SOCF3		614
455	N 2HCI	\rightarrow	456
456	1	· I	585
	N A	N N	
ĺ			
	HCI O		
457	O ₂	1	463
·			
	HCI	w.	
458	110/	<u></u>	549
	HCI O		
459	HCI O	1 -	532
	HCI O		
460	_	-Н	574
400		-n	3/4
	HCI CCE.		
461	N CI		564
	$ \bigvee_{s} \bigvee_{s} $	\triangleleft	
	HCI C		
462	Ν̈́	<u></u>	616
,	V s ∕ s ∕ s		
	HCI CCF ₃		

463	HCI SCF3	6	598
464	N OCH₃ OCH₃ 2HCI	\rightarrow	514

Example	R	MS (ES) m/z
465	TFA O N	505.1746
466	он он	551 (+Na)
467	N OCF3	
468		463.1704
469	N O CN	486
470	ال	503

Example	R	MS (ES) m/z
		537
471		33,
472	CI .	533.2348
	هُ ح	
472		499.2304
473		499.2504
474	<u> </u>	504
	O CN	
475	l N	
	N N	
476	2-	532.2522
	TFA HN	
477	Ň	
	ĭ []	
	<u> </u>	

Example	R	MS (ES) m/z
478		
		-
·		
479	N Br	539.0842
480	NO OCF3	545.1595
481	NO ₂ NO ₂ CF ₃	574.1483
482		503.2238
483		515.2234
484		417
485		475.1910
486		383
487	TFA TFA	460
488	N N N TFA	438

Example	R	MS (ES) m/z
489	N	452
	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	
	TFA .	
490	N I	474
	Ň	
	TFA U	·
491	1/2	476
	F	
492		383
493	1/7	472
	N	·
	TFA	
494	N. 1	472
	TFA N	
495		383
496	N N	383
497	CH ₃	517
	N N N N N N N N N N N N N N N N N N N	
	Ö TFA	
498	OCH ₃	
499		503
	~ ~ ~ ~ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	

Example	R	MS (ES) m/z
500		521
	, ,	571
501		571
	OCF ₃	
502	NO CI	571
503	N CF3	571
504		489.2059
505	N O O F	507.1987
506	N O OCF3	557
507	N O CF ₃	557
508	N O CI	557
509		503.2226
510	N F	521.2122

Example	R	MS (ES) m/z
511	OCF ₃	571.2056
512	CF ₃	571.2054
513	N CI	571.1464
514	HCI N	379.0964
515		504.1831
516		532.2105
517	i , i , i , i , i , i , i , i , i , i ,	470.1935
518		576.2355
519	i, C	596.2033
520		518.1945
521	i i i i i i i i i i i i i i i i i i i	538.1372

Example	R	MS (ES) m/z
522		519
523		560
524	N O	399
525		413
526	N O F	493
527	N OCF3	581
528	N O CF3	343.1742
529	n o	399.1597
530		483
531	N O F	501
532	N CF3	551

Example	R	MS (ES) m/z
_		407
533	N\	407
534	✓ 	515
334	N)	
	N CF ₃	
535	Ņ	460
	TEA	
536	TFA	460
536	Ŋ	400
	N N	
	TFA .	
537	N O	464
	\sim \downarrow	
538		460
539	Ň,	412
	Ĭ,	
540	N N	495.4984
	Ĩ Į Į J	
	\checkmark	
541	Ŋ	479.1416
	Ų ↓	
542	I	572.2800
342		
l		
		•
Ì		
<u> </u>	<u> </u>	

Example	R	MS (ES) m/z
543		539.2017
544		489.2049
·		
545	Ī.	. 497
	NO OF	
546	NO ₂	506
547		479
548	NO ₂	524
549	NO ₂	542
550	N O NO ₂	520
551	N O NO ₂	520

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Example	R	MS (ES) m/z
552	NO ₂	506
553	N NH ₂	476
554		547.2525
555		561.2692
556		561.2679
557	NO NC	576.2184
558	NO F	511.1755
559	N CN	500.1830
560	N CN	500.1888

Example	R	MS (ES) m/z
561	N F	577.1650
	F OCH₃	
562		413.1750
563	N 00	427.1903
564	ОН	385.1457
565	N O F ₂ C CF ₃	637.2067
566	NO N	532.2448
567	N CF3	529.1631
568	N O CF ₃	529.1603
569	NO ₂ CF ₃	574.1478
570	N CI CF3	597.0849
571	NO ₂ CF ₃	574.1473

Example	R	MS (ES) m/z
572	N CF ₃	513.1228
573	N O CI	509.1536
574	N O O	509.1529
575	NO OF	493.1803
576		493.1838
577	N O O N	476.1847
578	N O N N TFA	476.1865
579	N O Br	553.1057
580	N O N TFA	476.1879

Example	R	MS (ES) m/z
581		489.2076
	l	
582	N~	507.2016
<u> </u>		
583	, F	
	CF ₃	
584	N N	
	Î	
585	l N [∼]	415.1559
	N N N	
586	N N	401.1399
	N O H	
587	N N	443
588	Ň	477
589	CI	515
707		
•	N N N N N N N N N N N N N N N N N N N	

Evample	R	MS (ES) m/z
Example	R	L
590		438
591		452
592		466
593		472
594	N N OCH3	502
595	N N N OCF3	556
596	N HCI	457
597	N ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	
598		415.1911
599	N HCI	471
600		575.2777

Example	R	MS (ES) m/z
601	'n	. 575
602	Ņ	589.2947
603	N N	589.2914
604	N N	601.2936
605	N .	587.2808
		551 0005
606	N .	551.2225
607	N N	587.2048
		·
	0_CF ₃	610 0000
608	N N	619.2098
	O_CF2CF2H	607 1070
609	N C	687.1978
	OCF ₂ CF ₂ CF ₃	

Example	R	MS (ES) m/z
610	N	857.2070
	0CF2CF2CF2CF2CF2CF2CF2	
		710 0004
611		719.2024
612	O_CF2CF2CHF2	401 1746
612	/ .	401.1746
	, the second sec	
613		581.2323
·		
614	, \	511.1900
615	Z CI	495.1368
616	N OCH3	521.1980
_	O OCH₃	
617	×-	529.0962
	CI	
618	N OC ₂ H ₅	505.2031
619		475.1898
620		529.1604
<u> </u>	CF ₃	

Example	R	MS (ES) m/z
621	N O OC ₂ H ₅	456.1761
622	N TFA	398.1751
623	N TFA OH	414.1690
624		434.1651
629	N S S S S S S S S S S S S S S S S S S S	510
634	\(\sigma_s\sigma_\)	483.1992
635		425
636		507.1910
637	NO.	489.2064
638		511.1910
639		521.1962
640		505.2006
641	N N N C I	513.1277

Example	R	MS (ES) m/z
642		517.2410
643		519.2190
644		505
645	N O	428.1821
646	N NH	428
647		503
648		506.1830
649		524
650	$N \longrightarrow SO_2$	524.1531
651	O CH3	490.1912
<u> </u>	Isomer 2 (minor)	

Example	R	MS (ES) m/z
652		487
653		487
654		491
655		503
656		473
658		
659		·
665	o CI	510.1353
666	CI	541.1815

Example	R	MS (ES) m/z
667	n d	475
668	o CI Isomer 1	510.1366
669	Isomer 2	510.1358
670		
671	$N \longrightarrow N \longrightarrow S$	524
672		535
673	F ₃ CO CN	594
674	$N \longrightarrow N \longrightarrow S_{0_2}$	524
675	N S CF ₃	578
676	N S CF3	578
677	N S CF3	578
678	N S OCH3	540

Example	R	MS (ES) m/z
679	N S O2 OCF3	594
680	N S NO ₂	555
681	N S S F	528
682	N S F	528
683	N S OCH3	570
684	N N N N N N N N N N	514
685	$N \longrightarrow S \longrightarrow S$	516
686	N N N NH TFA O2	384.1593
688		527.1658 (M+NH4)
690	N S CN	535
691	N S N N N N N N N N N N N N N N N N N N	568

Example	R	MS (ES) m/z
692	N	423.1946
693	i de la companya della companya dell	441.2080
694	OCH3	506.1857
695	N OCF3	530.1565
696	N S OCH3	540
697	N S CF ₃	592.1401
698	N S OCH3	554.1659
699	N OCF ₃	608.1355
706	Isomer 1 (major)	490.1929
707	N S	491
708	OCF ₃	

Example	R	MS (ES) m/z
714	OCF3	560.1568
720		459.1987
721	OCH ₃	508.2019
722	оснз	480.1700
723	N O O O	441.2053
724	N F	509
725	OCF ₃	557
726	N OCF3	557
727	N CF3	541
728	N F	491

Example	R	MS (ES) m/z
729	N CF3	541
730		501
731	N F	509
732		501
733		501
734	N CH₃	517
735	N OCH ₃	521
736	N F	505
737		501
738	N CF ₃	559

Example	R	MS (ES) m/z
740	ОСН ₃	
_		
741	Isomer 1 OCH ₃	
, ***		
	0 0	
752	Isomer	572
/52	N N	372
,		
755		467
756	Ö	453
730	·	455
		150
757	N .	453
	 0	
758	Ņ	451
759	N .	451
	\sim \downarrow \downarrow	
760	"	488
	N	
İ	CI	
761	ä \si	AE1
761	N)	451

Example	R	MS (ES) m/z
781		444
		·
782		444
784	CH ₃	
,	N CH ₃	
786		499
787		499
788		515
789		529
790	N N N N N N N N N N N N N N N N N N N	516
791		517

Example	R	MS (ES) m/z
793	N N	
	N S CH ₃	
794		
/34		
	TFA	
796	N N	517
797		
798	N N	
799	N	
	Î /	
802		
807	N N	
	N SIL	
	HCI N—CH₃	
811	N .	
	OCH ₃	
	осн _з	

Example	R	MS (ES) m/z
815	TFA N—S	
816		
822	HCI O'CH3	
823	HCI CH ₃	
825	F F F F F F F F F F F F F F F F F F F	
826	N CH ₃	
827	N OH	
828		
829	OCH ₃	

Example	R	MS (ES) m/z
830	OC4H9	
	N S OC4ng	
831	N	
832		
632	N O CHO	
	N S N CH₃	
	CH ₃ CH ₃	
833	N OCH₃	
024	OCH ₃	
834		
	, S	
	Ĩ CH₃	
	5113	
835	N	
836	. он СН3	
	N-CH ₀	
838		
	N .	
	"	
841		
	N N	
		i

Example	R	MS (ES) m/z
842	TFA OH	·
844		
845	O N N N S N N S O 2S	
846	N CI	
847	N OC ₄ H ₉	
848	N(C ₂ H ₅) ₂	
850		
851		
852	N(CH ₃) ₂	
853	0 0 N NCH ₃) ₂	

Example	R	MS (ES) m/z
854		
,		į
856	N S CH ₃	·
857	N C	
858	N CH ₃	
859	NOCH ₃	
860	N C 3H7	
861	N CI	
862		
863	OCH 3 OC 3H7	

Example	R	MS (ES) m/z
864		
·		
867	Ö ÇH₃	
	N N	
		•
	0	
868	A N	
869	N O	
	CH ₃	
872	CH ₃	
	O NH ₂	
873	N O	
877	CF ₃	
878	O_N CH ₃	
	N N	
	2HCI 0	

Example	R	MS (ES) m/z
881	R CH ₃	
	N N N N N N N N N N N N N N N N N N N	
882	Ü ,C₂H₅	
002	N NH	
·		,
883	\ <u></u>	
	NH NH	
884	\triangle	
	N	
	 	
885		
	NO CONTRACTOR OF THE PROPERTY	
886	N(CH ₃) ₂	
887	CH3	
	CH ₃	
888		
888		
	CH ₃	
	Ö	

Example	. R	MS (ES) m/z
889	N OCH3	
890	N ОН	
891	ОН	
892	ОН	
893	OH No.	
894	ОН	
895	N(CH ₃) ₂	
899	N(CH ₃) ₂	
901	N CH₃ CH₃	·

Example	R	MS (ES) m/z
902	N N N N N N N N N N N N N N N N N N N	
	CH ₃ N(CH ₃)₂	
905	CH ₃ CH ₃	
	CH ₃	•
906	N CH ₃	
909	» s—	
910	N OH	
911	NH OH	
912	H ₃ C CI	

Example	R	MS (ES) m/z
913	CH ₃	
914		
915	N CH ₃	
916	N S S	
920	OCF ₃	
921		·
922	C ₃ H ₇	
924		
926	N CH ₃	

Example	R	MS (ES) m/z
931	N N N	
932		
939	och ₃	

5

Example	R	K	MS
940		HCI	
941	N O CF3	нсі	

5

10

Example	R	K	MS
942	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	нсі	

Example 943: In Vitro Metalloprotease Inhibition

The compounds prepared in the manner described in the Examples above were assayed for activity by an *in vitro* assay. Following the procedures of Knight et al., FEBS Lett. 296(3):263 (1992). Briefly, 4-aminophenylmercuric acetate (APMA) or trypsin-activated MMPs were incubated with various concentrations of the inhibitor compound at room temperature for 5 minutes.

More specifically, recombinant human MMP-13, MMP-1, MMP-2 and MMP-9 enzymes were prepared in laboratories of the assignee following usual 15 laboratory procedures. MMP-13 from a full length CDNA clone was expressed as a proenzyme using a baculovirus as discussed in V.A. Luckow, Insect Cell Expression Technology, pages 183-218, in Protein Engineering: Principles and Practice, J.L.Cleland et 20 al eds., Wiley-Liss, Inc., (1996). See, also, Luckow et al., J. Virol., 67:4566-4579 (1993); O'Reilly et al., Baculovirus Expression Vectors: A Laboratory Manual, W.H. Freeman and Company, New York, (1992); and King et al., The Baculovirus Expression System: A 25 Laboratory Guide, Chapman & Hall, London (1992) for Further details on use of baculovirus expression The expressed enzyme was purified first systems.

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over a heparin agarose column and then over a chelating zinc chloride column. The proenzyme was activated by APMA for use in the assay.

MMP-1 expressed in transfected HT-1080 cells was provided by Dr. Harold Welgus of Washington 5 University, St. Louis, MO. The enzyme was also activated using APMA and was then purified over a hydroxamic acid column. Dr. Welgus also provided transfected HT-1080 cells that expressed MMP-9. Transfected cells that expressed MMP-2 were provided 10 by Dr. Gregory Goldberg, also of Washington University. Studies carried out using MMP-2 in the presence of 0.02% 2-mercaptoethanol are shown in the table below with an asterisk. Studies with MMP-7 were carried out at pH 7.5 in the presence of 0.02% 15 2-mercaptoethanol using conditions otherwise similar to those used for the other enzymes. The enzyme was obtaind from a hMMP-7-expressing E. coli clone that was a gift of Dr. Steven Shapiro of Washington University, St.Louis, MO. Further specifics for 20 preparation and use of these enzymes can be found in the scientific literature describing these enzymes. See, for example, Enzyme Nomenclature, Academic Press, San Diego, Ca (1992) and the citations therein, and Frije et al., J. Biol. Chem., 26(24): 25 16766-16773 (1994). The enzyme substrate is a methoxycoumarin-containing polypeptide having the following sequence:

MCA-ProLeuGlyLeuDpaAlaArgNH2, wherein MCA

is methoxycoumarin and Dpa is 3-(2,4-dinitrophenyl)L-2,3-diaminopropionyl alanine. This substrate is
commercially available from Baychem as product
M-1895.

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The buffer used for assays contained 100 mM Tris-HCl, 100 mM NaCl, 10 mM CaCl₂ and 0.05 percent polyethyleneglycol (23) lauryl ether at a pH value of 7.5. Assays were carried out at room temperature, and dimethyl sulfoxide (DMSO) at a final concentration of 1 percent was used to dissolve inhibitor compound.

The assayed inhibitor compound in DMSO/buffer solution was compared to an equal amount of DMSO/buffer with no inhibitor as control using Microfluor White Plates (Dynatech). The inhibitor or control solution was maintained in the plate for 10 minutes and the substrate was added to provide a final concentration of 4 μ M.

In the absence of inhibitor activity, a 15 fluorogenic peptide was cleaved at the gly-leu peptide bond, separating the highly fluorogenic peptide from a 2,4-dinitrophenyl quencher, resulting in an increase of fluorescence intensity (excitation at 328 nm/emission at 415 nm). Inhibition was 20 measured as a reduction in fluorescent intensity as a function of inhibitor concentration, using a Perkin Elmer L550 plate reader. The IC50 values were calculated from those values. The results are set forth in the Inhibition Table A, below, reported in 25 terms of IC_{50} to three significant figures, where appropriate.

		1 2	
Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
1	22.7	8.5	>10,000
2	5,500	6,000	>10,000
8	15.6	2,900	>10,000
9	15.6	2,900	>10,000
10	18.1	>10,000	>10,000
11	18.0	4,500	>10,000
12	50.0	2,500	>10,000
13	12.2	5,600	>10,000
14	40.0	6,000	>10,000
15	37.0	2,700	>10,000
16	6.70	1,400	>10,000
17	31.6	3,500	>10,000
18	45.0	>10,000	>10,000
19	28.0	5,500	>10,000
20	42.5	4,800	>10,000
21	70.0	7,000	>10,000
22	>10,000	>10,000	>10,000
23	90.0	10,000	>10,000
24	23.5	4,500	>10,000
25	6.00	1,600	>10,000
26.	10.7	3,600	>10,000
27	6.40	1,600	>10,000
28	6.70	700	>10,000
29	4.00	445	>10,000
32	10.0	800	>10,000
33	20.0	4,500	>10,000
34	18.1	>10,000	>10,000
35	30.0	9,000	>10,000
36	15.8	2,100	>10,000

Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
. 37	30.0	1,750	>10,000
38	67.4	6,000	67.4
39	19.3	3,700	>10,000
40	26.8	900	>10,000
41	70.0	5,400	>10,000
42	82.5	>10,000	>10,000
43	17.9	5,000	>10,000
44	19.0	1,050	>10,000
45	360	5,000	>10,000
46	80.0	5,700	>10,000
47	11.4	6,000	>10,000
48	27.0	3,200	>10,000
49	20.0	6,500	>10,000
51	370	7,000	>10,000
52	90.0	1,900	>10,000
53	28.9	1,400	>10,000
54	40.0	5,700	>10,000
55	10.0	>10,000	>10,000
56	55.0	3,500	>10,000
57	80.0	2,700	>10,000
58	22.0	4,000	>10,000
59	4.00	530	>10,000
60	13.9	3,700	>10,000
61	7.00	1,500	>10,000
62	14.0	690	>10,000
63	20.0	2,900	>10,000
64	19.3	770	>10,000
65	5.00	195	>10,000
66	8.00	240	>10,000

Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
· 68	13.0	1,800	>10,000
69	18.1	3,500	>10,000
70	10.6	700	>10,000
71	7.70	270	>10,000
72	13.0	800	>10,000
73	15.4	2,000	>10,000
74	9.00	80.0	>10,000
75	11.5	500	>10,000
76	9.00	250	>10,000
77	75.0	3,400	>10,000
78	11.7	730	>10,000
79	20.0	2,000	>10,000
80	4.10	562	>10,000
81	60.0	158	>10,000
82	6.70	490	>10,000
83	2.70	21.1	3,100
84	28.6	1,400	>10,000
85	130	370	>10,000
86	0.6	12.1	>10,000
87	4.00	15.5	>10,000
88	9.00	40.0	>10,000
91	0.3	<0.1	>10,000
92	0.8	0.1	>10,000
95	0.3	<0.1	3,600
96	0.4	0.1	7,300
97	0.6	<0.1	>10,000
98	1.70	0.2	>10,000
99	1.00	<0.1	>10,000
100	0.5	<0.1	6,000

Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
101	1.10	0.8	>10,000
102	0.6	0.2	>10,000
103	1.80	0.3	>10,000
104	0.25	0.2	10,000
105	1.10	0.3	10,000
106	0.2	0.15	>10,000
106	0.1	<0.1	8,200
108	0.2	<0.1	5,000
109	0.3	<0.1	>10,000
110	0.6	0.2	>10,000
111	0.8	0.15	>10,000
112	0.5	<0.1	>10,000
113	0.3	<0.1	>10,000
114	0.4	<0.1	>10,000
115	0.1	<0.1	>10,000
116	0.3	<0.1	>10,000
117	0.2	0.1	>10,000
118	0.2	<0.1	>10,000
119	0.3	0.3	>10,000
120	0.4	0.1	>10,000
121	0.2	0.1	5,000
122	0.2	<0.1	3,000
123	0.7	<0.1	>10,000
124	<0.1	<0.1	>10,000
125	0.4	<0.1	>10,000
126	0.7	<0.1	>10,000
127	2.90	0.2	>10,000
128	0.1	<0.1	3,400
129	37.2	3.00	>10,000

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Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
130	0.5	0.3	1,600
131	0.2	<0.1	8,000
132	0.5	<0.1	>10,000
133	1.40	0.3	>10,000
134	1.80	0.3	>10,000
135	0.6	0.3	10,000
136	0.9	<0.1	>10,000
137	0.8	0.1	10,000
138	3.90	0.25	>10,000
140	11.4	0.8	>10,000
141	20.0	9.00	>10,000
142	12.6	10.0	>10,000
143	22.0	14.5	>10,000
144	0.4	0.2	10,000
145	0.4	0.2	3,700
146	0.2	0.3	3,000
147	0.4	0.2	7,700
148	2.50	3.70	>10,000
149	15.8	13.8	480
150	175	175	>10,000
151	270	290	>10,000
152	2.00	59.0	>10,000
153	50.0	5,000	>10,000
154	18.0	3,700	>10,000
155	130	240	>10,000
156	2.20	0.45	>10,000
157	0.5	0.2	>10,000
160	300	90.0	>10,000
161	32.6	900	>10,000

Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
162	27.8	7,000	>10,000
163	44.5	2,500	>10,000
164	3.50	440	>10,000
. 165	3.00	48.5	>10,000
166	32.7	240	>10,000
168	50.0	285	>10,000
169	20.0	175	>10,000
170	2.40	200	>10,000
171	5.40	186	>10,000
172	3.80	160	>10,000
173	6.70	330	3,400
174	23.5	800	>10,000
175	2.50	290	>10,000
176	4.00	250	>10,000
177	8.80	520	10,000
178	18.1	325	>10,000
179	20.6	170	>10,000
180	1.10	41.8	>10,000
181	190	2,300	>10,000
183	300	1,500	>10,000
184	480	1,500	>10,000
185	2.20	32.6	>10,000
187	10.0	600	>10,000
188	7.0	235	>10,000
189	7.00	235	>10,000
190	4.70	136	>10,000
191	3.50	25.1	>10,000
193	3.50	0.15	>10,000
194	0.3	<0.1	>7,300

Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
195	1.00	0.2	>10,000
196	1.60	<0.1	>10,000
197	2.70	<0.1	>10,000
198	0.375	0.25	7,300
199	0.2	<0.1	3,000
200	0.2	<0.1	3,000
201	0.3	0.2	>10,000
202	0.4	0.2	>10,000
207	28.8	900	>10,000
208	110	1,000	>10,000
209	50.0	130	>10,000
210	5.40	4.50	4,000
211	11.4	1,200	>10,000
212	160	240	>10,000
213	1,400	2,700	>10,000
214	4,900	3,500	>10,000
224	<0.1	<0.1	4,500
225	180	41.8	>10,000
227	28.8	21.7	>10,000
228	2,448	2,000	>10,000
229	0.18	0.1	>10,000
231	0.2	0.1	>10,000
233	43.5	2,050	>10,000
235	235	5,300	>10,000
236	9.00	400	>10,000
237	13.0	1,900	>10,000
238	80.0	10,000	>10,000
239	9.00	8,300	>10,000
240	76.9	10,000	>10,000

Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
241	4.80	>10,000	>10,000
242	42.5	1,500	>10,000
243	11.3	420	>10,000
244	67.4	4,400	>10,000
245	20.0	800	>10,000
246	32.7	2,700	>10,000
247	34.5	1,600	>10,000
248	2.29	270	>10,000
249	13.0	235	>10,000
251	<0.1	<0.1	5,840
252	<0.1	<0.1	3,933.33
253	<0.1, 0.15	3,400	<0.1
256	0.2	0.1	3,200
257	0.2	0.1	4,100
258	0.2	0.1	>10,000
260	0.1	<0.1	5,300
261	<0.1	<0.1	3,700
262	1.50	0.9	>10,000
264	0.2	<0.1	4,500
265	0.2	1.30, <0.1	>10,000
266	0.1	<0.1	5,500
267	0.2	0.15	10,000
268	<0.1, 0.2	4,000	<0.1
269	0.2	<0.1	>10,000
27 0	1.00	1.00	>10,000
27 1	0.3	0.17	>10,000
272	0.2	0.1	3,600
273	0.3	0.1	>10,000

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Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
274	160	>10,000	>10,000
275	70.0	>10,000	>10,000
276	37.3	>10,000	>10,000
277	70.0	>10,000	>10,000
278	19.3	>10,000	>10,000
279	20.0	7,300	>10,000
280	90.0	>10,000	>10,000
281	105	>10,000	>10,000
282	14.8	9,000	>10,000
283	13.8	>10,000	>10,000
284	130	>10,000	>10,000
285	19.3	9,000	>10,000
286	60.0	>10,000	>10,000
287	150	>10,000	>10,000
288	35.0	>10,000	>10,000
289	50.0	>10,000	>10,000
290	50.0	>10,000	>10,000
292	100	>10,000	>10,000
293	63.1	>10,000	>10,000
294	59.1	>10,000	>1,000
295	50.0	>10,000	>10,000
296	50.0	>10,,000	>10,000
297	34.9	>10,000	>10,000
298	40.0	>10,000	>10,000
299	30.6	9,000	>10,000
300	37.3	>10,000	>10,000
301	90.0	>10,000	>10,000
302	175	>10,000	>10,000
303	115	>10,000	>10,000

Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
304	30.6	7,000	>10,000
305	28.6	>10,000	>10,000
306	60.0	>10,000	>10,000
307	40.0	>10,000	>10,000
308	40.0	10,000	>10,000
309	48.5	>10,000	>10,000
310	60.0	10,000	>10,000
311	120	>10,000	>10,000
312	200	>10,000	>10,000
313	77.0	>10,000	>10,000
314	65.0	>10,000	>10,000
315	420	>10,000	>10,000
316	0.4	0.2	>10,000
317	1.40	0.4	>10,000
318	0.3	0.1	>10,000
319	0.5	0.2	>10,000
320	12.1	4.00	>10,000
321	0.5	0.3	>10,000
322	0.3	0.3	>10,000
323	1.30	0.4	>10,000
324	0.7	0.4	>10,000
325	0.9	0.2	>10,000
326	0.6	0.45	>10,000
327	0.9	0.3	>10,000
328	0.35	0.4	>10,000
329	0.9	0.4	>10,000
330	0.5	0.7	>10,000
331	0.7	0.2	>10,000
332	2.10	0.4	>10,000

Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
333	0.8	0.2	>10,000
334	0.7	0.3	>10,000
335	0.9	0.15	>10,000
336	1.00	<0.1	>10,000
337	2.70	0.2	>10,000
338	1.90	0.2	>10,000
339	1.00	0.3	>10,000
340	0.3	<0.1	>10,000
341	0.6	0.2	>10,000
342	4.00	0.3	>10,000
343	1.70	0.8	>10,000
344	2.90	0.65	>10,000
346	1.20	0.2	>10,000
347	3.00	0.7	>10,000
348	16.5	0.8	>10,000
349	0.2	<0.1	2600
350	0.1	<0.1	6000
351			
352	1.4	0.3	>10,000
353	0.3	<0.1	>10,000
354	1.6	15.4	
355	0.4	<0.1	7000
356	2.4	32.6	
357	0.3	0.1	>10,000
358			
359	34.9	12.2	>10,000
360	10.0	5.6	
361	0.5	<0.1	5000
362	2.7	<0.1	>10,000

Number IC ₅₀ (nM) IC ₅₀ (nM) IC ₅₀ (nM) 363 0.4 0.2 8800 364 1.0 0.2 >10,0 365 0.3 0.1 >10,0 366 13.0 2.5 >10,0 367	00
364 1.0 0.2 >10,0 365 0.3 0.1 >10,0 366 13.0 2.5 >10,0	00
365 0.3 0.1 >10,0 366 13.0 2.5 >10,0	00
366 13.0 2.5 >10,0	L
	00
367	
, 1 I	
368 0.5 7.0	
369 3.3 5.4	
370	
371 11.1 400	
372	
373 3.0 80.0	
374 3.3 4.0 >10,0	
375 16.9 15.6 >10,0	00
376 5.5 230	
378 1.7 0.3 200	
379 0.3 0.1 >10,0	00
380	
381	
382 11.4 260	
383 3.0 700 >10,0	00
384	
385	
386 0.4 0.2 2100)
387	
388 50.0 430	
389 1.7 16.1 >10,0	00
390	
391 0.1 <0.1 5400	
392 0.2 0.1 >10,0	00

Evennle	MMP-13	MMP-2	MMP-1
Example			
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
393	4.5	427	>10,000
394	0.5	. 8.0	
395	0.9	0.5	>10,000
396	4.8	330	>10,000
397	4.4	70.0	>10,000
398	7.0	70.0	>10,000
399	1.2	0.3	>10,000
400	23.5	520	
401	16.9	195	>10,000
402	15.8	340	>10,000
403	55.3	4.0	>10,000
404	0.5	0.25	>10,000
405			
406			
407	1.2	0.1	>10,000
408	25.1	800	>10,000
409	22.4	275	>10,000
410	0.6	0.25	>10,000
411	0.2	<0.1	>10,000
412	0.4	0.2	6400
413	1.1	0.3	8000
414	50.5	1500	>10,000
415	50.4	246	>10,000
416	0.4	0.2	3000
417	0.7	4.5	>10,000
418	7.0	1400	>10,000
4 19	4.2	400	>10,000
420			
421			
			L

Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
422			
423			
424	5.5	80.0	>10,000
425	20.0	1000	>10,000
426			
427			
428			
429			
430		· <u></u>	
431			
432	13.9	100	>10,000
433	450	3500	>10,000
434	190	3700	>10,000
435	5.9	1500	>10,000
436	1.8	330	>10,000
437	18.1	800	>10,000
438	1.4	160	>10,000
439	1070	1600	>10,000
440	26.8	240	>10,000
441	6.0	420	>10,000
442	10.0	211	>10,000
443	90.0	2200	>10,000
444			
445	90.0	1200	>10,000
446	270	7000	>10,000
447	23.9	155	>10,000
448	2.4	540	>10,000
449			
450			

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Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
451	0.3	0.1	3700
452	<0.1	<0.1	
453	0.4	35.0	>10,000
454	2.1	100	>10,000
455	6.3	26.8	>10,000
456			
457	1800	2700	>10,000
458	210	2100	>10,000
459	136	3100	>10,000
460	4.0	200	>10,000
461	20.0	145	>10,000
462	2.9	80.0	>10,000
463	16.9	210	>10,000
464	120	400	>10,000
465	80	370	>10,000
466	9.4	60	>10,000
467	27.0	140	>10,000
468			
469	0.8	12.0	>10,000
470	140	2000	>10,000
471	2400	>10,000	>10,000
472	4.0	200	>10,000
473	160	3300	>10,000
474	12.1	300	>10,000
475	27.1	500	>10,000
476	25.4	140	>10,000
477	11.3	160	>10,000
478	16.4	306	>10,000
479	5.0	60.0	>10,000

Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
480	18.6	155	>10,000
481	50.0	1400	>10,000
482	6.0	4.0	>10,000
483	32.6	10.6	>10,000
484	240	100	>10,000
485	8.0	4.2	>10,000
486	5400	4000	>10,000
487	140	800	>10,000
488	140	370	>10,000
489	770	1900	>10,000
490	61.0	3000	>10,000
491	>10,000	>10,000	>10,000
492	6100	>10,000	>10,000
493	>10,000	>10,000	>10,000
494	650	3300	>10,000
495	14.5	21.1	>10,000
496	30.7	200	>10,000
497	50.0	8000	>10,000
499	0.9	19.3	>10,000
500	3.0	22.0	>10,000
501	2.5	180	>10,000
502	14.0	63	>10,000
503	10.0	50.0	>10,000
504	6.3	220	>10,000
505	14.0	72.0	>10,000
506	5.0	400	>10,000
507	15.8	210	>10,000
508	19.3	210	>10,000
509	520	>10,000	>10,000

Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM).	IC ₅₀ (nM)	IC ₅₀ (nM)
510	7700	>10,000	>10,000
511	9000	6000	>10,000
512	7700	>10,000	>10,000
513	7700	>10,000	>10,000
514	1.0	0.6	5,000
515	8.0	27.0	>10,000
516	14.8	300	>10,000
517	14.0	1100	>10,000
518	11.4	350	>10,000
519	45.4	1300	>10,000
520	22.5	250	>10,000
521	3.5	50.0	>10,000
522	2.4	94.0	>10,000
523	2.4	190	>10,000
524	2700	6400	>10,000
525	290	700	>10,000
526	>10,000	>10,000	>10,000
527	6700	9000	>10,000
528	7700	>10,000	>10,000
529	8800	>10,000	>10,000
530	20.0	60.7	>10,000
531	13.0	10.0	>10,000
532	10.0	150	>10,000
533	60.0	150	>10,000
534	30.0	480	>10,000
535	1.9	35.0	>10,000
536	7.7	88.0	>10,000
537	70.0	55.0	5200
538	80.0	370	>10,000
		<u> </u>	<u> </u>

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Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
539	270	350	>10,000
540	11.4	500	>10,000
541	0.7	2.0	>10,000
542			
543	33.7	5400	>10,000
544	35.0	3100	>10,000
545	7.7	120	>10,000
546	2.7	18.6	>10,000
547	5.0	64.7	>10,000
548	40.0	800	>10,000
549	55.3	2900	>10,000
550	20.0	2000	>10,000
551	9.0	140	>10,000
552	12.8	140	>10,000
553	12.8	50.0	>10,000
554	3.7	140	>10,000
555	3.7	220	>10,000
556	4.5	170	>10,000
557	16.9	200	>10,000
558	4.5	66.4	>10,000
559	7.2	80.0	>10,000
560	4.5	306	>10,000
561	6.0	500	>10,000
562	1200	6300	>10,000
563	70.0	235	>10,000
564	150	550	>10,000
565	5.5	700	>10,000
566	15.8	57.1	>10,000
567	5.0	87.7	>10,000

Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
568	120	4600	>10,000
569	16.9	87.7	>10,000
570	290	>10,000	>10,000
571	28.6	140	>10,000
572	37.2	3000	>10,000
573	11.4	235	>10,000
574	10.6	220	>10,000
575	10.7	110	>10,000
576	8.8	78.0	>10,000
577	107	2200	>10,000
578	160	2000	>10,000
579	2.7	100	>10,000
580	37.2	700	>10,000
581	27.0	480	>10,000
582	30.0	1800	>10,000
583	70.0	4700	>10,000
584	2700	3500	>10,000
585	1400	3500	>10,000
586	>10,000	>10,000	>10,000
587	1.8	1.0	>10,000
588			
589	70.0	>10,000	>10,000
590	121	80.0	>10,000
591	70.0	730	>10,000
592	57.0	690	>10,000
593	420	650	>10,000
594	570.	650	>10,000
595	270	425	>10,000
596	1.1	10.6	>10,000

Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
597	670	700	>10,000
598	25.4	145	>10,000
600	9.0	600	>10,000
601	9.0	1300	>10,000
602	70.0	3000	>10,000
603	15.8	2300	>10,000
604	20.0	2500	>10,000
605	10.6	2000	>10,000
606	3.0	77.0	>10,000
607	2.9	220	>10,000
608	3.0	250	>10,000
609	30.6	2800	>10,000
610	425	1300	>10,000
611	139	1800	>10,000
612	290	2200	>10,000
613	8.0	30.7	>10,000
614	22.0	25.4	>10,000
615	3.1	11.0	>10,000
616	4.0	3.7	>10,000
617	7.0	5.7	>10,000
618			
619	4.3	5.7	>10,000
620	27.8	225	>10,000
621	120	1500	>10,000
622	500	1600	>10,000
623	350	1400	>10,000
624	120	940	>10,000
634	4.4	60.7	>10,000
635	13.9	260	>10,000

Number IC ₅₀ (nM) IC ₅₀ (nM) IC ₅₀ (nM) 636 3.0 8.0 >10,000 637 3.8 22 >10,000 638 639 1.5 1.5 9400 640 4.2 15.8 >10,000 641 4.0 13.7 >10,000 642 2.2 1.1 >10,000 643 1.8 1.2 6000 644 1.6 3.3 8800 645 370 1200 >10,000 646 7800 >10,000 647 6.0 160 >10,000 648 25.8 110 >10,000 650 14.7 1200 >10,000 651 13.7 60 >10,000 652 0.4 82.0 >10,000 653 0.8 160 >10,000 654 3.2 35.0 >10,000	Example	MMP-13	MMP-2	MMP-1
637 3.8 22 >10,000 638 639 1.5 1.5 9400 640 4.2 15.8 >10,000 641 4.0 13.7 >10,000 642 2.2 1.1 >10,000 643 1.8 1.2 6000 644 1.6 3.3 8800 645 370 1200 >10,000 646 7800 >10,000 647 6.0 160 >10,000 648 25.8 110 >10,000 649 130 1400 >10,000 650 14.7 1200 >10,000 651 13.7 60 >10,000 652 0.4 82.0 >10,000 653 0.8 160 >10,000 654 3.2 35.0 >10,000 655 37.3 1400 >10,000 656 3.1 120 >10,000 659 1.0 3.7	Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
638	636	3.0	8.0	>10,000
639 1.5 1.5 9400 640 4.2 15.8 >10,000 641 4.0 13.7 >10,000 642 2.2 1.1 >10,000 643 1.8 1.2 6000 644 1.6 3.3 8800 645 370 1200 >10,000 646 7800 >10,000 647 6.0 160 >10,000 648 25.8 110 >10,000 649 130 1400 >10,000 650 14.7 1200 >10,000 651 13.7 60 >10,000 652 0.4 82.0 >10,000 653 0.8 160 >10,000 654 3.2 35.0 >10,000 655 37.3 1400 >10,000 656 3.1 120 >10,000 659 1.0 3.7 >10,000	637	3.8	22	>10,000
640 4.2 15.8 >10,000 641 4.0 13.7 >10,000 642 2.2 1.1 >10,000 643 1.8 1.2 6000 644 1.6 3.3 8800 645 370 1200 >10,000 646 7800 >10,000 647 6.0 160 >10,000 648 25.8 110 >10,000 649 130 1400 >10,000 650 14.7 1200 >10,000 651 13.7 60 >10,000 652 0.4 82.0 >10,000 653 0.8 160 >10,000 654 3.2 35.0 >10,000 655 37.3 1400 >10,000 655 3.1 120 >10,000 658 12.2 1000 >10,000 659 1.0 3.7 >10,000	638			
641 4.0 13.7 >10,000 642 2.2 1.1 >10,000 643 1.8 1.2 6000 644 1.6 3.3 8800 645 370 1200 >10,000 646 7800 >10,000 647 6.0 160 >10,000 648 25.8 110 >10,000 649 130 1400 >10,000 650 14.7 1200 >10,000 651 13.7 60 >10,000 652 0.4 82.0 >10,000 653 0.8 160 >10,000 653 0.8 160 >10,000 655 37.3 1400 >10,000 655 37.3 1400 >10,000 658 12.2 1000 >10,000 659 1.0 3.7 >10,000 655 2.3 29.2 >10,000	639	1.5	1.5	9400
642 2.2 1.1 >10,000 643 1.8 1.2 6000 644 1.6 3.3 8800 645 370 1200 >10,000 646 7800 >10,000 647 6.0 160 >10,000 648 25.8 110 >10,000 649 130 1400 >10,000 650 14.7 1200 >10,000 651 13.7 60 >10,000 652 0.4 82.0 >10,000 653 0.8 160 >10,000 654 3.2 35.0 >10,000 655 37.3 1400 >10,000 656 3.1 120 >10,000 658 12.2 1000 >10,000 659 1.0 3.7 >10,000 655 2.3 29.2 >10,000 656 48.4 330 >10,000	640	4.2	15.8	>10,000
643 1.8 1.2 6000 644 1.6 3.3 8800 645 370 1200 >10,000 646 7800 >10,000 647 6.0 160 >10,000 648 25.8 110 >10,000 649 130 1400 >10,000 650 14.7 1200 >10,000 651 13.7 60 >10,000 652 0.4 82.0 >10,000 653 0.8 160 >10,000 654 3.2 35.0 >10,000 655 37.3 1400 >10,000 656 3.1 120 >10,000 659 1.0 3.7 >10,000 659 1.0 3.7 >10,000 666 48.4 330 >10,000 667 30 135 >10,000	641	4.0	13.7	>10,000
644 1.6 3.3 8800 645 370 1200 >10,000 646 7800 >10,000 647 6.0 160 >10,000 648 25.8 110 >10,000 649 130 1400 >10,000 650 14.7 1200 >10,000 651 13.7 60 >10,000 652 0.4 82.0 >10,000 653 0.8 160 >10,000 654 3.2 35.0 >10,000 655 37.3 1400 >10,000 656 3.1 120 >10,000 659 1.0 3.7 >10,000 659 1.0 3.7 >10,000 665 2.3 29.2 >10,000 666 48.4 330 >10,000 667 30 135 >10,000	642	2.2	1.1	>10,000
645 370 1200 >10,000 646 7800 >10,000 647 6.0 160 >10,000 648 25.8 110 >10,000 649 130 1400 >10,000 650 14.7 1200 >10,000 651 13.7 60 >10,000 652 0.4 82.0 >10,000 653 0.8 160 >10,000 654 3.2 35.0 >10,000 655 37.3 1400 >10,000 656 3.1 120 >10,000 659 1.0 3.7 >10,000 655 2.3 29.2 >10,000 665 2.3 29.2 >10,000 666 48.4 330 >10,000 667 30 135 >10,000	643	1.8	1.2	6000
646 7800 >10,000 647 6.0 160 >10,000 648 25.8 110 >10,000 649 130 1400 >10,000 650 14.7 1200 >10,000 651 13.7 60 >10,000 652 0.4 82.0 >10,000 653 0.8 160 >10,000 654 3.2 35.0 >10,000 655 37.3 1400 >10,000 656 3.1 120 >10,000 658 12.2 1000 >10,000 659 1.0 3.7 >10,000 665 2.3 29.2 >10,000 666 48.4 330 >10,000 667 30 135 >10,000	644	1.6	3.3	8800
647 6.0 160 >10,000 648 25.8 110 >10,000 649 130 1400 >10,000 650 14.7 1200 >10,000 651 13.7 60 >10,000 652 0.4 82.0 >10,000 653 0.8 160 >10,000 654 3.2 35.0 >10,000 655 37.3 1400 >10,000 656 3.1 120 >10,000 658 12.2 1000 >10,000 659 1.0 3.7 >10,000 655 2.3 29.2 >10,000 665 2.3 29.2 >10,000 666 48.4 330 >10,000 667 30 135 >10,000	645	370	1200	>10,000
648 25.8 110 >10,000 649 130 1400 >10,000 650 14.7 1200 >10,000 651 13.7 60 >10,000 652 0.4 82.0 >10,000 653 0.8 160 >10,000 654 3.2 35.0 >10,000 655 37.3 1400 >10,000 656 3.1 120 >10,000 658 12.2 1000 >10,000 659 1.0 3.7 >10,000 665 2.3 29.2 >10,000 666 48.4 330 >10,000 667 30 135 >10,000	646		7800	>10,000
649 130 1400 >10,000 650 14.7 1200 >10,000 651 13.7 60 >10,000 652 0.4 82.0 >10,000 653 0.8 160 >10,000 654 3.2 35.0 >10,000 655 37.3 1400 >10,000 656 3.1 120 >10,000 658 12.2 1000 >10,000 659 1.0 3.7 >10,000 665 2.3 29.2 >10,000 666 48.4 330 >10,000 667 30 135 >10,000	647	6.0	160	>10,000
650 14.7 1200 >10,000 651 13.7 60 >10,000 652 0.4 82.0 >10,000 653 0.8 160 >10,000 654 3.2 35.0 >10,000 655 37.3 1400 >10,000 656 3.1 120 >10,000 658 12.2 1000 >10,000 659 1.0 3.7 >10,000 665 2.3 29.2 >10,000 666 48.4 330 >10,000 667 30 135 >10,000	648	25.8	110	>10,000
651 13.7 60 >10,000 652 0.4 82.0 >10,000 653 0.8 160 >10,000 654 3.2 35.0 >10,000 655 37.3 1400 >10,000 656 3.1 120 >10,000 658 12.2 1000 >10,000 659 1.0 3.7 >10,000 665 2.3 29.2 >10,000 666 48.4 330 >10,000 667 30 135 >10,000	649	130	1400	>10,000
652 0.4 82.0 >10,000 653 0.8 160 >10,000 654 3.2 35.0 >10,000 655 37.3 1400 >10,000 656 3.1 120 >10,000 658 12.2 1000 >10,000 659 1.0 3.7 >10,000 665 2.3 29.2 >10,000 666 48.4 330 >10,000 667 30 135 >10,000	650	14.7	1200	>10,000
653 0.8 160 >10,000 654 3.2 35.0 >10,000 655 37.3 1400 >10,000 656 3.1 120 >10,000 658 12.2 1000 >10,000 659 1.0 3.7 >10,000 665 2.3 29.2 >10,000 666 48.4 330 >10,000 667 30 135 >10,000	651	13.7	60	>10,000
654 3.2 35.0 >10,000 655 37.3 1400 >10,000 656 3.1 120 >10,000 658 12.2 1000 >10,000 659 1.0 3.7 >10,000 665 2.3 29.2 >10,000 666 48.4 330 >10,000 667 30 135 >10,000	652	0.4	82.0	>10,000
655 37.3 1400 >10,000 656 3.1 120 >10,000 658 12.2 1000 >10,000 659 1.0 3.7 >10,000 665 2.3 29.2 >10,000 666 48.4 330 >10,000 667 30 135 >10,000	653	0.8	160	>10,000
656 3.1 120 >10,000 658 12.2 1000 >10,000 659 1.0 3.7 >10,000 665 2.3 29.2 >10,000 666 48.4 330 >10,000 667 30 135 >10,000	654	3.2	35.0	>10,000
658 12.2 1000 >10,000 659 1.0 3.7 >10,000 665 2.3 29.2 >10,000 666 48.4 330 >10,000 667 30 135 >10,000	655	37.3	1400	>10,000
659 1.0 3.7 >10,000 665 2.3 29.2 >10,000 666 48.4 330 >10,000 667 30 135 >10,000	656	3.1	120	>10,000
665 2.3 29.2 >10,000 666 48.4 330 >10,000 667 30 135 >10,000	658	12.2	1000	>10,000
666 48.4 330 >10,000 667 30 135 >10,000	659	1.0	3.7	>10,000
667 30 135 >10,000	665	2.3	29.2	>10,000
	656	48.4	330	>10,000
668 2.0 25.8 >10.000	667	30	135	>10,000
25.0	668	2.0	25.8	>10,000
669 4.3 22.7 >10,000	669	4.3	22.7	>10,000
670	670	-		

	100 12	100 O	1 10 1 1
Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
671	6.0	130	>10,000
672	6.7	60	>10,000
673	14.8	455	>10,000
674	8.0	110	>10,000
675	13.0	88	6000
676	7.7	90	>10,000
677	7.0	34.7	>10,000
678	5.0	50	>10,000
679			
680			
681			
682			
683	11.3	290	>10,000
684	60	1450	>10,000
685	3.0	34.7	>10,000
686	4200	3700	>10,000
688	17.6	110	>10,000
690	7.3	41.8	>10,000
691	10.0	130	>10,000
692	10.0	22.7	>10,000
693.	210	1900	>10,000
694	3.1	23.2	>10,000
695	2.0	22.7	>10,000
696	10.0	140	>10,000
697	18.1	1500	>10,000
698	16.9	700	>10,000
699	50.0	455	>10,000
705	44.5	1100	>10,000
706	4.3	40	>10,000
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Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
707	2.3	9.0	>10,000
708	114	3000	>10,000
714	28.8	420	>10,000
720	4.5	36.9	>10,000
724	28.6	300	>10,000
725	25.1	210	>10,000
726	15.8	250	>10,000
727	34.9	240	>10,000
₃ 728	9.4	106	>10,000
729	14.8	240	>10,000
730	37	3000	>10,000
731	1.9	35	>10,000
732	3.1	590	>10,000
733	1.6	270	>10,000
734	6.0	3300	>10,000
735	9.0	800	>10,000
736	0.9	145	>10,000
737	3.0	1280	>10,000
738	22.0	270	>10,000
740	61	175	>10,000
741	50	125	>10,000
752	14.8	271	>10,000
755	2.2	20	>10,000
756	7.0	28.8	>10,000
757	3.3	28.8	>10,000
758	5.0	34.7	>10,000
759	3.0	60.8	>10,000
760	6.0	25.4	>10,000
761	5.0	41.8	>10,000

Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
769	5.0	0.7	>10,000
770	270	485	>10,000
771	500	10,000	>10,000
772	350	4200	>10,000
773	6.0	2.7	>10,000
774			
775	120	270	>10,000
776	3.0	10.0	>10,000
777	2.5	6.5	>10,000
778	3.3	12	>10,000
779	40	210	>10,000
780	17.5	80	>10,000
781	800	5100	>10,000
782	21.1	100	>10,000
784	6.0	4500	>10,000
786	3.7	700	>10,000
787	1.2	175	>10,000
788	3.0	445	>10,000
789	12.2	3700	>10,000
790	4.5	700	>10,000
791	2.0	700	>10,000
793	4.0	23.5	>10,000
794	1500	2900	>10,000
796	5.7	130	>10,000
797	4.0	175	>10,000
798	20.0	210	>10,000
799	10.6	43.5	>10,000
802	2.3	10,000	>10,000
807	200	1400	>10,000

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Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
811	14.8	110	>10,000
815	140	1400	>10,000
816	1200	>10,000	>10,000
820	29.0	1400	>10,000
821	4.0	10.0	>10,000
822	10.0	210	>10,000
823	7.0	505	>10,000
825	11.3	70.0	>10,000
826	40.0	650	ND
827	10.0	540	>10,000
828	1.5	12.8	ND
829	6.0	22.0	ND
830	17.9	2100	>10,000
831	2.3	170	>10,000
832	18.1	2000	>10,000
833	11.0	1750	>10,000
834	150	780	ND
835	6.0	20.0	>10,000
836	135	4200	ND
838	3.0	70.0	>10,000
841	285	1900	ND
842	5.5	45.4	>10,000
844	5.0	4700	>10,000
845	28.6	2000	ND
846	4.5	186	>10,000
847	20.0	1800	ND
848			ND
850	4.5	150	>10,000
851	3.7	42.5	ND

Example	MMP-13	MMP-2	MMP-1
Number	IC ₅₀ (nM)	IC ₅₀ (nM)	IC ₅₀ (nM)
852	25.0	3000	ND
853	15.8	120	ND
854	40.0	3300	ND
856	1.2	250	ND
857	1.3	120	ND
858	3.7	600	>10,000
859	5.5	440	ND
860	2.7	1500	>10,000
861	2.0	34.9	ND
862	1.7	40.0	ND
863			ND
864			ND
867	16.5	10,000	>10,000
868			ND
869	2.0	76.9	ND
870	305	6000	, ND

Example 944: *In Vivo* Angiogenesis Assay

The study of angiogenesis depends on a

5 reliable and reproducible model for the stimulation
and inhibition of a neovascular response. The
corneal micropocket assay provides such a model of
angiogenesis in the cornea of a mouse. See, A Model
of Angiogenesis in the Mouse Cornea; Kenyon, BM,

10 et al., Investigative Ophthalmology & Visual Science,
July 1996, Vol. 37, No. 8.

In this assay, uniformLy sized Hydron pellets containing bFGF and sucralfate were prepared

and surgically implanted into the stroma mouse cornea adjacent to the temporal limbus. The pellets were formed by making a suspension of 20 μ L sterile saline containing 10 μ g recombinant bFGF, 10 mg of sucralfate and 10 μ L of 12 percent Hydron in ethanol. The slurry was then deposited on a 10 x 10 mm piece of sterile nylon mesh. After drying, the nylon fibers of the mesh were separated to release the pellets.

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10 The corneal pocket is made by anesthetizing a 7 week old C57Bl/6 female mouse, then proptosing the eye with a jeweler's forceps. Using a dissecting microscope, a central, intrastromal linear keratotomy of approximately 0.6 mm in length is performed with a #15 surgical blade, parallel to the insertion of the lateral rectus muscle. Using a modified cataract knife, a lamellar micropocket is dissected toward the temporal limbus. The pocket is extended to within 1.0 mm of the temporal limbus. A single pellet was placed on the corneal surface at the base of the 20 pocket with a jeweler's forceps. The pellet was then advanced to the temporal end of the pocket. Antibiotic ointment was then applied to the eye.

Mice were dosed on a daily basis for the

25 duration of the assay. Dosing of the animals was
based on bioavailability and overall potency of the
compound. an exemplary dose was 10 or 50 mg/kg (mpk)
bid, po. Neovascularization of the corneal stroma
begins at about day three and was permitted to

30 continue under the influence of the assayed compound
until day five. At day five, the degree of

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angiogenic inhibition was scored by viewing the neovascular progression with a slit lamp microscope.

The mice were anesthetized and the studied eye was once again proptosed. The maximum vessel 5 length of neovascularization, extending from the limbal vascular plexus toward the pellet was measured. In addition, the contiguous circumferential zone of neovascularization was measured as clock hours, where 30 degrees of arc equals one clock hour. The area of angiogenesis was calculated as follows.

$area = \frac{(0.4 \times clock\ hours \times 3.14 \times vessel\ length\ (in\ mm))}{}$

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Five to six mice were utilized for each 15 compound in each study. The studied mice were thereafter compared to control mice and the difference in the area of neovascularization was recorded as an averaged value. Each group of mice so studied constitutes an "n" value of one, so that "n" 20 values greater than one represent multiple studies whose averaged result is provided in the table. A contemplated compound typically exhibits about 25 to about 75 percent inhibition, whereas the vehicle control exhibits zero percent inhibition. 25

Example 350: In Vivo PC-3 Tumor Reduction PC-3 human pancreatic cancer eclls (ATCC CRL 1435) were grown to 90% confluence in F12/MEM (Gibco) containing 7% FBS (Gibco). Cells were

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mechanically harvested using a rubber scraper, and then washed twice with cold medium. The resulting cells were resuspended in cold medium with 30% matrigel (Collaborative Research) and the cell-containing medium was maintained on ice until used.

Balb/c nu/nu mice at 7-9 weeks of age were anesthetized with avertin [2,2,2-tribromethanol/t-amyl alcohol (1 g/1 mL) diluted 1:60 into phosphate-buffered sline] and 3-5x10⁶ of the above cells in 0.2 mL of medium were injected into the left flank of each mouse. Cells were injected in the morning, whereas dosing with an inhibitor began at 6 PM. The animals were gavaged BID from day zero (cell injection day) to day 25-30, at which time the animals were euthanized and tumors weighed.

Compounds were dosed at 10 mg/mL in 0.5% methylcellulose/0.1% polysorbate 80 to provide a 50 mg/kg (mpk) dose twice each day, or diluted to provide a 10 mg/kg (mpk) dose twice each day. Tumor measurements began on day 7 and continued every third or fourth day until completion of the study. Groups of ten mice were used in each study and nine to ten survived. Each group of mice so studied constitutes an "n" value of one, so that "n" values greater than one represent multiple studies whose averaged result is provided in the table.

Example 945: Tumor Necrosis Factor Assays

30 Cell Culture.

The cells used in the assay are the human moncytic line U-937 (ATCC CRL-1593). The cells are grown in RPMI w/10% FCS and PSG supplement (R-10) and

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are not permitted to overgrow. The assay is carried out as follows:

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- 1. Count, then harvest cells by 5 centrifugation. Resuspend the pellet in R-10 supplement to a concentration of 1.540 x 106 cells/mL.
- 2. Add test compound in 65 uL R-10 to the appropriate wells of a 96-well flat bottom tissue 10 culture plate. The initial dilution from a DMSO stock (100 mM compound) provides a 400 uM solution, from which five additional three-fold serial dilutions are made. Each dilution of 65 ul (in triplicate) yields final compound test concentrations of 100 µM, 33.3 µM, 11.1 µM, 3.7 µM, 1.2 µM and 0.4 15 μM.
 - 3. The counted, washed and resuspended cells (200,000 cells/well) in 130 μL are added to the wells.
- 4. Incubation is for 45 minutes to one 20 hour at 37°C in 5% CO2 in a water saturated container.
 - 5. R-10 (65 uL)containing 160 ng/mL PMA (Sigma) is added to each well.
- The test system is incubated at 37°C in 5% CO2 overnight (18-20 hours) under 100% humidity. 25
 - 7. Supernatant, 150 μL, is carefully removed from each well for use in the ELISA assay.
 - 8. For toxicity, a 50 μL aliquot of working solution containg 5 mL R-10, 5 mL MTS solution [CellTiter 96 AQueous One Solution Cell Proliferation Assay Cat.#G358/0,1 (Promega Biotech)] and 250 ul PMS solution are added to each well containing the

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remaining supernatant and cells and the cells incubated at 37°C in 5% CO₂ until the color develops. The system is excited at 570 nm and read at 630 nm.

5 TNF Receptor II ELISA Assay

- 1. Plate 100 μ L/well 2 ug/mL mouse antihuman TNFrII antibody (R&D Systems #MAB226) in 1 x PBS (pH 7.1, Gibco) on NUNC-Immuno Maxisorb plate. Incubate the plate at 4°C overnight (about 18-20 hours).
- 2. Wash the plate with PBS-Tween (1 x PBS w/ 0.05% Tween).
- $3.\ Add\ 200\ \mu L$ 5% BSA in PBS and block at $37^{o}C$ in a water saturated atmosphere for 2 hours.
 - 4. Wash the plate with PBS-Tween.
- 5. Add sample and controls (100 ul of each) to each well. The standards are 0, 50, 100, 200, 300 and 500 pg recombinant human TNFrII (R&D Systems #226-B2) in 100 μ L 0.5% BSA in PBS. The assay is linear to between 400-500 pg of standard.
- 6. Incubate at 37° C in a saturated atmosphere for 1.5 hours.
 - 7. Wash the plate with PBS-Tween.
 - 8. Add 100 μL goat anti-human TNFrII
- 25 polyclonal (1.5 μg/mL R&D Systems #AB226-PB in 0.5% BSA in PBS).
 - 9. Incubate at 37°C in a saturated atmosphere for 1 hour.
 - 10. Wash the plate with PBS-Tween.
- 30 11. Add 100 μ L anti-goat IgG-peroxidase (1:50,000 in 0.5% BSA in PBS, Sigma #A5420).

- 11. Incubate at 37° C in a saturated atmosphere for 1 hour.
 - 12. Wash the plate with PBS-Tween.
- 13. Add 10 μ L KPL TMB developer, develop at room temperature (usually about 10 minutes), then terminate with phosphoric acid and excite at 450 nm and read at 570 nm.

TNFα ELISA Assay

Coat Immulon® 2 plates with 0.1 mL/well of lug/mL Genzyme mAb in 0.1 M NaHCO3 pH 8.0 buffer overnight (about 18-20 hours) at 4°C, wrapped tightly in Saran® wrap.

Flick out coating solution and block plates

15 with 0.3 mL/well blocking buffer overnight at 4°C,

wrapped in Saran® wrap.

Wash wells thoroughly 4X with wash buffer and completely remove all wash buffer. Add 0.1 mL/well of either samples or rhTNFa standards.

20 Dilute samples if necessary in appropriate diluant (e.g. tissue culture medium). Dilute standard in same diluant. Standards and samples should be in triplicates.

Incubate at 37°C for 1 hour in humified 25 container.

Wash plates as above. Add 0.1 mL/well of 1:200 dilution of Genzyme rabbit anti-hTNF.

Repeat incubation.

Repeat wash. Add 0.1 mL/well of 1 µg/mL 30 Jackson goat anti-rabbit IgG (H+L)-peroxidase.

Incubate at 37°C for 30 minutes.

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Repeat wash. Add 0.1 mL/well of peroxide-ABTS solution.

Incubate at room temperature for 5-20 minutes.

5 Read OD at 405 nm.

12 Reagents are:

Genzyme mouse anti-human TNF? monoclonal (Cat.# 80-3399-01)

10 Genzyme rabbit anti-human TNF? polyclonal (Cat.#IP-300)

Genzyme recombinant human TNF? (Cat. #TNF-H).

Jackson Immunoresearch peroxide-conjugated goat anti-rabbit IgG (H+L) (Cat.#111-035-144).

15 Kirkegaard/Perry peroxide ABTS solution (Cat#50-66-01).

Immulon 2 96-well microtiter plates.

Blocking solution is 1 mg/mL gelatin in PBS with 1X thimerasol.

20 Wash buffer is 0.5 mL Tween[®] 20 in 1 liter of PBS.

that numerous modifications and variations can be effectuated without departing from the true spirit and scope of the novel concepts of the present invention. It is to be understood that no limitation with respect to the specific example presented is intended or should be inferred. The disclosure is intended to cover by the appended claims all such modifications as fall within the scope of the claims.

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MISSING AT THE TIME OF PUBLICATION

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WHAT IS CLAIMED:

1. A process for treating a host mammal having a condition associated with pathological matrix metalloprotease (MMP) activity that comprises administering a metalloprotease inhibitor compound or a pharmaceutically acceptable salt thereof in an effective amount to a mammalian host having such a condition, said metalloprotease inhibitor inhibiting the activity of one or more of MMP-2, MMP-9 and MMP-13, while exhibiting substantially less inhibitory activity against MMP-1, said compound corresponding in structure to formula B, below

$$\begin{array}{c}
(CH_2)_n - Z \\
X \\
R_{20} (CH_2)_p (CH_2)_p \\
S(O)_g
\end{array}$$

$$\begin{array}{c}
Q \\
A \\
B
\end{array}$$

15

5

10

wherein

R²⁰ is -NH-O-R¹⁴, where R¹⁴ is hydrido, a pharmaceutically acceptable cation or C(W)R²⁵ where W is O or S and R²⁵ is selected from the group consisting of an C₁-C₆-alkyl, aryl, C₁-C₆-alkoxy, heteroaryl-C₁-C₆-alkyl, C₃-C₈-cycloalkyl-C₁-C₆-alkyl, aryloxy, ar-C₁-C₆-alkoxy, ar-C₁-C₆-alkyl, heteroaryl and amino C₁-C₆-alkyl group wherein the amino C₁-C₆-alkyl nitrogen is (i) unsubstituted or (ii) substituted with one or two substituents independently selected from the group consisting of

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an C_1 - C_6 -alkyl, aryl, ar- C_1 - C_6 -alkyl, C_3 - C_8 -cycloalkyl- C_1 - C_6 -alkyl, ar- C_1 - C_6 -alkoxycarbonyl, C_1 - C_6 -alkoxycarbonyl, and C_1 - C_6 -alkanoyl radical, or (iii) wherein the amino C_1 - C_6 -alkyl nitrogen and two substituents attached thereto form a 5- to 8-membered heterocyclo or heteroaryl ring;

g is 2;

m is zero, 1 or 2;

n is zero, 1 or 2;

10 p is zero, 1 or 2;

the sum of m + n + p = 1, 2, 3 or 4;

- (a) one of X, Y and Z is selected from the group consisting of C(0), NR^6 , O, S, S(0), $S(0)_2$ and $NS(0)_2R^7$, and the remaining two of X, Y and Z are
- 15 CR^8R^9 , and $CR^{10}R^{11}$, or

20

- (b) X and Z or Z and Y together constitute a moiety that is selected from the group consisting of $NR^6C(0)$, $NR^6S(0)$, $NR^6S(0)_2$, NR^6S , NR^6O , SS, NR^6NR^6 and OC(0), with the remaining one of X, Y and Z being CR^8R^9 , or
- (c) n is zero and X, Y and Z together constitute a moiety selected from the group consisting of

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wherein wavy lines are bonds to the atoms of the depicted ring;

 $_{\rm R}^6$ and $_{\rm R}^6$ ' are independently selected from the group consisting of hydrido, formyl, sulfonic-C $_{\rm C}^{\rm C}_{\rm C}$ c $_{\rm C}^{\rm C}_{\rm C}$ c $_{\rm C}^{\rm C}_{\rm C}$ c $_{\rm C}^{\rm C}_{\rm C}$

10 hydroxycarbonyl- C_1 - C_6 -alkyl, C_1 - C_6 -alkylcarbonyl- C_1 - C_6 -alkyl, R^8R^9 -aminocarbonyl- C_1 - C_6 -alkyl, C_1 - C_6 -

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 $alkoxycarbonyl-C_1-C_6-alkylcarbonyl$, hydroxycarbonyl-C₁-C₆-alkylcarbonyl, C₁-C₆-alkylcarbonyl-C₁-C₆alkylcarbonyl, C₁-C₆-alkoxycarbonylcarbonyl, hydroxycarbonylcarbonyl, C1-C6-alkylcarbonylcarbonyl, R^8R^9 -aminocarbonylcarbonyl, C_1 - C_6 -alkanoyl, aryl- C_1 - C_6 -alkyl, aroyl, bis(C_1 - C_6 -alkoxy- C_1 - C_6 -alkyl)- C_1 - C_6 alkyl, C_1-C_6 -alkyl, C_1-C_6 -haloalkyl, C_1-C_6 perfluoroalkyl, C₁-C₆-trifluoromethylalkyl, C₁-C₆perfluoroalkoxy-C₁-C₆-alkyl, C₁-C₆-alkoxy-C₁-C₆-10 alkyl, C3-C6-cycloalkyl, heteroarycarbonyl, heterocyclocarbonyl, C3-C8-heterocycloalkyl, C3-C8heterocycloalkylcarbonyl, aryl, C5-C6-heterocyclo, C_5-C_6 -heteroaryl, C_3-C_8 -cycloalkyl- C_1-C_6 -alkyl, aryloxy-C₁-C₆-alkyl, heteroaryloxy-C₁-C₆-alkyl, heteroaryl-C₁-C₆-alkoxy-C₁-C₆-alkyl, heteroarylthio-15 C_1-C_6 -alkyl, arylsulfonyl, C_1-C_6 -alkylsulfonyl, C_5 -C₆-heteroarylsulfonyl, carboxy-C₁-C₆-alkyl, C₁-C₄alkoxycarbonyl- C_1 - C_6 -alkyl, aminocarbonyl, C_1 - C_6 alkyl(R⁸N)iminocarbonyl, aryl(R⁸N)iminocarbonyl, C₅-C₆-heterocyclo(R⁸N)iminocarbonyl, arylthio-C₁-C₆-20 alkyl, C₁-C₆-alkylthio-C₁-C₆-alkyl, arylthio-C₃-C₆alkenyl, C₁-C₄-alkylthio-C₃-C₆-alkenyl, C₅-C₆heteroaryl-C₁-C₆-alkyl, halo-C₁-C₆-alkanoyl, hydroxy- C_1-C_6 -alkanoyl, thiol- C_1-C_6 -alkanoyl, C_3-C_6 -alkenyl, $C_3-C_6-alkynyl$, $C_1-C_4-alkoxy-C_1-C_4-alkyl$, $C_1-C_5-alkyl$ alkoxycarbonyl, aryloxycarbonyl, NR⁸R⁹-(R⁸)iminomethyl, NR⁸R⁹-C₁-C₅-alkylcarbonyl, hydroxy-

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 C_1 - C_5 -alkyl, R^8R^9 -aminocarbonyl, R^8R^9 -aminocarbonyl- C_1 - C_6 -alkylcarbonyl, hydroxyaminocarbonyl, R^8R^9 -aminosulfonyl, R^8R^9 -aminosulfon- C_1 - C_6 -alkyl, R^8R^9 -amino- C_1 - C_6 -alkylsulfonyl and an R^8R^9 -amino- C_1 - C_6 -alkyl group;

 R^7 is selected from the group consisting of a arylalkyl, aryl, heteroaryl, heterocyclo, C_1 - C_6 -alkyl, C_3 - C_6 -alkynyl, C_3 - C_6 -alkenyl, C_1 - C_6 -carboxyalkyl and a C_1 - C_6 -hydroxyalkyl group;

 R^8 and R^9 and R^{10} and R^{11} are independently 10 selected from the group consisting of a hydrido, hydroxy, C_1 - C_6 -alkyl, C_1 - C_6 -alkanoyl, aroyl, aryl, ar-C1-C6-alkyl, heteroaryl, heteroar-C1-C6-alkyl, C2- C_6 -alkynyl, C_2 - C_6 -alkenyl, thiol- C_1 - C_6 -alkyl, C_1 - C_6 alkylthio-C₁-C₆-alkyl, cycloalkyl, cycloalkyl-C₁-C₆-15 alkyl, heterocycloalkyl- C_1 - C_6 -alkyl, C_1 - C_6 -alkoxy- C_1 - C_6 -alkyl, aralkoxy- C_1 - C_6 -alkyl, C_1 - C_6 -alkoxy- C_1 - C_6 alkoxy- C_1 - C_6 -alkyl, hydroxy- C_1 - C_6 -alkyl, hydroxycarbonyl-C₁-C₆-alkyl, hydroxycarbonylar-C₁-C₆alkyl, aminocarbonyl- C_1 - C_6 -alkyl, aryloxy- C_1 - C_6 -20 alkyl, heteroaryloxy-C₁-C₆-alkyl, arylthio-C₁-C₆alkyl, heteroarylthio-C₁-C₆-alkyl, the sulfoxide or sulfone of any said thio substituents, perfluoro-C1-C₆-alkyl, trifluoromethyl-C₁-C₆-alkyl, halo-C₁-C₆alkyl, alkoxycarbonylamino-C₁-C₆-alkyl and an amino-25 C_1-C_6 -alkyl group wherein the aminoalkyl nitrogen is (i) unsubstituted or (ii) substituted with one or two 10

radicals independently selected from the group consisting of C_1 - C_6 -alkyl, ar- C_1 - C_6 -alkyl, cycloalkyl and C_1-C_6 -alkanoyl, or wherein \mathbb{R}^8 and \mathbb{R}^9 or \mathbb{R}^{10} and R^{11} and the carbon to which they are bonded form a 5 carbonyl group, or wherein R^8 and R^9 or R^{10} and R^{11} , or \mathbb{R}^8 and \mathbb{R}^{10} together with the atoms to which they are bonded form a 5- to 8-membered carbocyclic ring, or a 5- to 8-membered heterocyclic or heteroaryl ring containing one or two heteroatoms that are nitrogen, oxygen, or sulfur, with the proviso that only one of R^8 and R^9 or R^{10} and R^{11} is hydroxy;

 R^{12} and R^{12} are independently selected from the group consisting of a hydrido, C_1-C_6 -alkyl, aryl, ar-C₁-C₆-alkyl, heteroaryl, heteroaralkyl, C₂-15 C₆-alkynyl, C₂-C₆-alkenyl, thiol-C₁-C₆-alkyl, cycloalkyl, cycloalkyl-C1-C6-alkyl, heterocycloalkyl- $C_1-C_6-alkyl$, $C_1-C_6-alkoxy-C_1-C_6-alkyl$, $aryloxy-C_1-C_6-alkyl$ alkyl, amino- C_1 - C_6 -alkyl, C_1 - C_6 -alkoxy- C_1 - C_6 -alkoxy- $C_1-C_6-alkyl$, hydroxy- $C_1-C_6-alkyl$, hydroxycarbonyl- $C_1-alkyl$ C₆-alkyl, hydroxycarbonylar-C₁-C₆-alkyl, 20 aminocarbonyl-C₁-C₆-alkyl, aryloxy-C₁-C₆-alkyl, heteroaryloxy-C₁-C₆-alkyl, C₁-C₆-alkylthio-C₁-C₆alkyl, arylthio- C_1 - C_6 -alkyl, heteroarylthio- C_1 - C_6 alkyl, the sulfoxide or sulfone of any said thio substituents, perfluoro-C₁-C₆-alkyl, trifluoromethyl-25 C₁-C₆-alkyl, halo-C₁-C₆-alkyl, alkoxycarbonylamino- C_1-C_6 -alkyl and an amino- C_1-C_6 -alkyl group wherein the aminoalkyl nitrogen is (i) unsubstituted or (ii)

substituted with one or two radicals independently selected from the group consisting of C_1 - C_6 -alkyl, ar- C_1 - C_6 -alkyl, cycloalkyl and C_1 - C_6 -alkanoyl;

 R^{13} is selected from the group consisting of a hydrido, benzyl, phenyl, C_1 - C_6 -alkyl, C_2 - C_6 -alkynyl, C_2 - C_6 -alkenyl and a C_1 - C_6 -hydroxyalkyl group;

-Q-A-R-E-Y is a substituent in which the moiety Q is a 5- to 7-membered heterocyclic ring

10 containing one or two nitrogen atoms one of which is bonded the depicted phenyl group, and whose remaining members (A-R-E-Y) are bonded at the 4-position relative to said phenyl-bonded nitrogen atom when Q is a 6- or 7-membered ring and at the 3- or 4
15 position relative to that nitrogen when Q is a 5-membered ring;

A is selected from the group consisting of

- (1) -0-;
- (2) -S-;
- 20
- $(3) NR^{17} ;$
- (4) $-CO-N(R^{17})$ or $-N(R^{17})-CO-$, wherein R^{17} is hydrogen, C_1-C_4 -alkyl, or phenyl;
- (5) -CO-O- or -O-CO-;
- (6) -O-CO-O-;
- 25 (7) -HC=CH-;
 - (8) —NH-CO-NH-;
 - (9) -C≡C-;
 - (10) -NH-CO-O- or -O-CO-NH-;
 - (11) -N=N-;
- 30 (12) -NH-NH-; and

5

(13) $-CS-N(R^{18})-$ or $-N(R^{18})-CS-$, wherein R^{18} is hydrogen C_1-C_4 -alkyl, or phenyl; or

(14) A is absent and Q is bonded directly to R:

R is a moiety selected from the group consisting of alkyl, alkoxyalkyl, aryl, heteroaryl, cycloalkyl, heterocycloalkyl, aralkyl, heteroaralkyl, heterocycloalkylalkyl, cycloalkylalkyl,

- 10 cycloalkoxyalkyl, heterocycloalkoxyalkyl, aryloxyalkyl, heteroaryloxyalkyl, arylthioalkyl, heteroarylthioalkyl, cycloalkylthioalkyl, and a heterocycloalkylthioalkyl group wherein the aryl or heteroaryl or cycloalkyl or heterocycloalkyl
- substituent is (i) unsubstituted or (ii) substituted with one or two radicals selected from the group consisting of a halo, alkyl, perfluoroalkyl, perfluoroalkoxy, perfluoroalkylthio, trifluoromethylalkyl, amino, alkoxycarbonylalkyl,
- alkoxy, C₁-C₂-alkylene-dioxy, hydroxycarbonylalkyl, hydroxycarbonylalkylamino, nitro, hydroxy, hydroxyalkyl, alkanoylamino, and a alkoxycarbonyl group, and R is other than alkyl or alkoxyalkyl when A is -O- or -S-;
- 25 the moiety E is selected from the group consisting of
 - (1) $-CO(R^{19})$ or $-(R^{19})CO$ -, wherein R^{19} is a heterocycloalkyl, or a cycloalkyl group;
- 30 (2) -CONH- or -HNCO-; and
 - (3) -CO-;

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(4) $-SO_2-R^{19}$ or $-R^{19}-SO_2$;

- $(5) -SO_2-;$
- (6) $-NH-SO_2- \text{ or } -SO_2-NH-;$
- (7) -S-;
- (8) -NH-CO-O- or -O-CO-NH-; or
- (9) E is absent and R is bonded directly to Y; and

the moiety Y is absent or is selected from the group consisting of a hydrido, alkyl, alkoxy, 10 haloalkyl, aryl, aralkyl, cycloalkyl, heteroaryl, hydroxy, aryloxy, aralkoxy, heteroaryloxy, heteroaralkyl, perfluoroalkoxy, perfluoroalkylthio, trifluoromethylalkyl, alkenyl, heterocycloalkyl, cycloalkyl, trifluoromethyl, alkoxycarbonyl, and a aminoalkyl group, wherein the aryl, heteroaryl, 15 aralkyl or heterocycloalkyl group is (i) unsubstituted or (ii) substituted with one or two radicals independently selected from the group consisting of an alkanoyl, halo, nitro, aralkyl, aryl, alkoxy, trifluoroalkyl, trifluoroalkoxy and an 20 amino group wherein the amino nitrogen is (i)

25

and an aralkyl group.

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2. The process according to claim 1 wherein said compound corresponds in structure to formula B-1

unsubstituted or (ii) substituted with one or two groups independently selected from hydrido, alkyl,

$$\begin{array}{c|c}
(CH_2)_{n} & Z \\
X & Q \\
R^{20} & (CH_2)_{m} & (CH_2)_{p} \\
\hline
S(O)_{g} & B-1
\end{array}$$

3. The process according to claim 1 wherein said compound corresponds in structure to formula B-2

- 4. The process according to claim 1 10 wherein the sum of m + n + p = 1 or 2.
 - 5. The process according to claim 1 wherein said compound or salt is administered a plurality of times.

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6. A process for treating a host mammal having a condition associated with pathological matrix metalloprotease (MMP) activity that comprises administering a metalloprotease inhibitor compound or a pharmaceutically acceptable salt thereof in an effective amount to a mammalian host having such a condition, said metalloprotease inhibitor inhibiting the activity of one or more of MMP-2, MMP-9 and MMP-13, while exhibiting substantially less inhibitory

activity against MMP-1, said compound corresponding in structure to formula B-3, below

$$\begin{array}{c} (CH_2)_{n} & Z \\ X \\ X \\ CH_2)_{m} & (CH_2)_{p} \\ S(O)_{g} \\ \end{array}$$

5

wherein

 R^{20} is $-NH-O-R^{14}$, where R^{14} is hydrido, a pharmaceutically acceptable cation or C(W)R²⁵ where W is O or S and \mathbb{R}^{25} is selected from the group consisting of an C_1 - C_6 -alkyl, aryl, C_1 - C_6 -alkoxy, 10 heteroaryl-C₁-C₆-alkyl, C₃-C₈-cycloalkyl-C₁-C₆-alkyl, aryloxy, $ar-C_1-C_6-alkoxy$, $ar-C_1-C_6-alkyl$, heteroaryl and amino C_1-C_6 -alkyl group wherein the amino C_1-C_6 alkyl nitrogen is (i) unsubstituted or (ii) 15 substituted with one or two substituents independently selected from the group consisting of an C_1 - C_6 -alkyl, aryl, ar- C_1 - C_6 -alkyl, C_3 - C_8 cycloalkyl- C_1 - C_6 -alkyl, ar- C_1 - C_6 -alkoxycarbonyl, C_1 - C_6 -alkoxycarbonyl, and C_1 - C_6 -alkanoyl radical, or 20 (iii) wherein the amino C_1 - C_6 -alkyl nitrogen and two substituents attached thereto form a 5- to 8-membered heterocyclo or heteroaryl ring;

g is 2; m is zero, 1 or 2; 25 n is zero, 1 or 2; p is zero, 1 or 2;

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the sum of m + n + p = 1, 2, 3 or 4;
(a) one of X, Y and Z is selected from the group consisting of C(0), NR^6 , O, S, S(0), $S(0)_2$ and $NS(0)_2R^7$, and the remaining two of X, Y and Z are

5 CR^8R^9 , and $CR^{10}R^{11}$, or

(b) X and Z or Z and Y together constitute a moiety that is selected from the group consisting of $NR^6C(0)$, $NR^6S(0)$, $NR^6S(0)_2$, NR^6S , NR^6O , SS, NR^6NR^6 and OC(0), with the remaining one of X, Y and Z being CR^8R^9 , or

(c) n is zero and X, Y and Z together constitute a moiety selected from the group consisting of

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wherein wavy lines are bonds to the atoms of the depicted ring;

 R^6 and $R^{6'}$ are independently selected from 5 the group consisting of hydrido, formyl, sulfonic-C₁-C6-alkyl, C1-C6-alkoxycarbonyl-C1-C6-alkyl, $hydroxycarbonyl-C_1-C_6-alkyl, C_1-C_6-alkylcarbonyl-C_1-alkylca$ C_6 -alkyl, R^8R^9 -aminocarbonyl- C_1 - C_6 -alkyl, C_1 - C_6 alkoxycarbonyl-C₁-C₆-alkylcarbonyl, hydroxycarbonyl-10 C_1-C_6 -alkylcarbonyl, C_1-C_6 -alkylcarbonyl- C_1-C_6 alkylcarbonyl, C_1-C_6 -alkoxycarbonylcarbonyl, hydroxycarbonylcarbonyl, C₁-C₆-alkylcarbonylcarbonyl, R^8R^9 -aminocarbonylcarbonyl, C_1 - C_6 -alkanoyl, aryl- C_1 - C_6 -alkyl, aroyl, bis(C_1 - C_6 -alkoxy- C_1 - C_6 -alkyl)- C_1 - C_6 alkyl, C_1-C_6 -alkyl, C_1-C_6 -haloalkyl, C_1-C_6 perfluoroalkyl, C1-C6-trifluoromethylalkyl, C1-C6perfluoroalkoxy-C₁-C₆-alkyl, C₁-C₆-alkoxy-C₁-C₆alkyl, C3-C6-cycloalkyl, heteroarycarbonyl,

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heterocyclocarbonyl, C3-C8-heterocycloalkyl, C3-C8heterocycloalkylcarbonyl, aryl, C5-C6-heterocyclo, C₅-C₆-heteroaryl, C₃-C₈-cycloalkyl-C₁-C₆-alkyl, aryloxy-C₁-C₆-alkyl, heteroaryloxy-C₁-C₆-alkyl, 5 heteroary1-C₁-C₆-alkoxy-C₁-C₆-alky1, heteroary1thio-C₁-C₆-alkyl, arylsulfonyl, C₁-C₆-alkylsulfonyl, C₅-

C6-heteroarylsulfonyl, carboxy-C1-C6-alkyl, C1-C4alkoxycarbonyl-C₁-C₆-alkyl, aminocarbonyl, C₁-C₆alkyl(R8N)iminocarbonyl, aryl(R8N)iminocarbonyl, C5-

C6-heterocyclo(R8N)iminocarbonyl, arylthio-C1-C6alkyl, C₁-C₆-alkylthio-C₁-C₆-alkyl, arylthio-C₃-C₆alkenyl, C₁-C₄-alkylthio-C₃-C₆-alkenyl, C₅-C₆heteroaryl-C₁-C₆-alkyl, halo-C₁-C₆-alkanoyl, hydroxy-C₁-C₆-alkanoyl, thiol-C₁-C₆-alkanoyl, C₃-C₆-alkenyl,

 c_3-c_6 -alkynyl, c_1-c_4 -alkoxy- c_1-c_4 -alkyl, c_1-c_5 -15 alkoxycarbonyl, aryloxycarbonyl, NR⁸R⁹-(R⁸)iminomethyl, NR⁸R⁹-C₁-C₅-alkylcarbonyl, hydroxy-C₁-C₅-alkyl, R⁸R⁹-aminocarbonyl, R⁸R⁹-aminocarbonyl c_1-c_6 -alkylcarbonyl, hydroxyaminocarbonyl, R^8R^9 -

aminosulfonyl, R⁸R⁹-aminosulfon-C₁-C₆-alkyl, R⁸R⁹-20 ${\tt amino-C_1-C_6-alkylsulfonyl\ and\ an\ R^8R^9-amino-C_1-C_6-alkylsulfonyl\ an\ Alkylsulfonyl\ an\$ alkyl group;

25

 R^7 is selected from the group consisting of a arylalkyl, aryl, heteroaryl, heterocyclo, C₁-C₆alkyl, C₃-C₆-alkynyl, C₃-C₆-alkenyl, C₁-C₆carboxyalkyl and a C1-C6-hydroxyalkyl group;

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 R^8 and R^9 and R^{10} and R^{11} are independently selected from the group consisting of a hydrido, hydroxy, C₁-C₆-alkyl, C₁-C₆-alkanoyl, aroyl, aryl, ar-C₁-C₆-alkyl, heteroaryl, heteroar-C₁-C₆-alkyl, C₂- C_6 -alkynyl, C_2 - C_6 -alkenyl, thiol- C_1 - C_6 -alkyl, C_1 - C_6 alkylthio-C₁-C₆-alkyl, cycloalkyl, cycloalkyl-C₁-C₆alkyl, heterocycloalkyl-C₁-C₆-alkyl, C₁-C₆-alkoxy-C₁- C_6 -alkyl, aralkoxy- C_1 - C_6 -alkyl, C_1 - C_6 -alkoxy- C_1 - C_6 alkoxy-C₁-C₆-alkyl, hydroxy-C₁-C₆-alkyl, hydroxycarbonyl-C₁-C₆-alkyl, hydroxycarbonylar-C₁-C₆-10 alkyl, aminocarbonyl-C₁-C₆-alkyl, aryloxy-C₁-C₆alkyl, heteroaryloxy-C₁-C₆-alkyl, arylthio-C₁-C₆alkyl, heteroarylthio-C1-C6-alkyl, the sulfoxide or sulfone of any said thio substituents, perfluoro-C1-C6-alkyl, trifluoromethyl-C1-C6-alkyl, halo-C1-C6alkyl, alkoxycarbonylamino-C1-C6-alkyl and an amino-C1-C6-alkyl group wherein the aminoalkyl nitrogen is (i) unsubstituted or (ii) substituted with one or two radicals independently selected from the group consisting of C_1-C_6 -alkyl, ar- C_1-C_6 -alkyl, cycloalkyl 20 and C_1-C_6 -alkanoyl, or wherein \mathbb{R}^8 and \mathbb{R}^9 or \mathbb{R}^{10} and R^{11} and the carbon to which they are bonded form a carbonyl group, or wherein ${\bf R}^{\bf 8}$ and ${\bf R}^{\bf 9}$ or ${\bf R}^{\bf 10}$ and ${\bf R}^{\bf 11}$, or R⁸ and R¹⁰ together with the atoms to which they are bonded form a 5- to 8-membered carbocyclic ring, 25 or a 5- to 8-membered heterocyclic or heteroaryl ring

containing one or two heteroatoms that are nitrogen,

oxygen, or sulfur, with the proviso that only one of \mathbb{R}^8 and \mathbb{R}^9 or \mathbb{R}^{10} and \mathbb{R}^{11} is hydroxy;

 R^{12} and R^{12} are independently selected from the group consisting of a hydrido, C_1-C_6 -alkyl, 5 aryl, ar-C₁-C₆-alkyl, heteroaryl, heteroaralkyl, C₂-C6-alkynyl, C2-C6-alkenyl, thiol-C1-C6-alkyl, cycloalkyl, cycloalkyl-C1-C6-alkyl, heterocycloalkyl- C_1-C_6 -alkyl, C_1-C_6 -alkoxy- C_1-C_6 -alkyl, aryloxy- C_1-C_6 alkyl, amino-C₁-C₆-alkyl, C₁-C₆-alkoxy-C₁-C₆-alkoxy-C₁-C₆-alkyl, hydroxy-C₁-C₆-alkyl, hydroxycarbonyl-C₁-10 C_6 -alkyl, hydroxycarbonylar- C_1 - C_6 -alkyl, aminocarbonyl-C₁-C₆-alkyl, aryloxy-C₁-C₆-alkyl, heteroaryloxy-C₁-C₆-alkyl, C₁-C₆-alkylthio-C₁-C₆alkyl, arylthio-C₁-C₆-alkyl, heteroarylthio-C₁-C₆alkyl, the sulfoxide or sulfone of any said thio 15 substituents, perfluoro-C₁-C₆-alkyl, trifluoromethyl-C₁-C₆-alkyl, halo-C₁-C₆-alkyl, alkoxycarbonylamino- C_1-C_6 -alkyl and an amino- C_1-C_6 -alkyl group wherein the aminoalkyl nitrogen is (i) unsubstituted or (ii) substituted with one or two radicals independently 20 selected from the group consisting of C1-C6-alkyl, $ar-C_1-C_6-alkyl$, cycloalkyl and $C_1-C_6-alkanoyl$;

 R^{13} is selected from the group consisting of a hydrido, benzyl, phenyl, C_1 - C_6 -alkyl, C_2 - C_6 -alkynyl, C_2 - C_6 -alkenyl and a C_1 - C_6 -hydroxyalkyl group;

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Q is a 5- to 7-membered heterocyclic ring containing one or two nitrogen atoms one of which is bonded the depicted phenyl group, and whose remaining

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members (A-R-E-Y) are bonded at the 4-position relative to said phenyl-bonded nitrogen atom when Q is a 6- or 7-membered ring and at the 3- or 4-position relative to that nitrogen when Q is a 5-membered ring;

 $\label{eq:themoiety} \ \mbox{te is selected from the group} \\ \mbox{consisting of}$

(1) $-CO(R^{19})$ or $-(R^{19})CO$, wherein R^{19} is a heterocycloalkyl, or a cycloalkyl group;

(2) -CONH- or -HNCO-; and

(3) -CO-;

(4) $-SO_2-R^{19}$ or $-R^{19}-SO_2$;

 $(5) -SO_2 -;$

(6) $-NH-SO_2- \text{ or } -SO_2-NH-;$

(7) -S-;

(8) -NH-CO-O- or -O-CO-NH-; or

(9) E is absent and Y is bonded directly to the Q ring; and

the moiety Y is absent or is selected from 20 the group consisting of a hydrido, alkyl, alkoxy, haloalkyl, aryl, aralkyl, cycloalkyl, heteroaryl, hydroxy, aryloxy, aralkoxy, heteroaryloxy, heteroaralkyl, perfluoroalkoxy, perfluoroalkylthio, trifluoromethylalkyl, alkenyl, heterocycloalkyl, 25 cycloalkyl, trifluoromethyl, alkoxycarbonyl, and a aminoalkyl group, wherein the aryl, heteroaryl, aralkyl or heterocycloalkyl group is (i) unsubstituted or (ii) substituted with one or two radicals independently selected from the group 30 consisting of an alkanoyl, halo, nitro, aralkyl, aryl, alkoxy, trifluoroalkyl, trifluoroalkoxy and an

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amino group wherein the amino nitrogen is (i) unsubstituted or (ii) substituted with one or two groups independently selected from hydrido, alkyl, and an aralkyl group.

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7. The process according to claim 6 wherein said compound corresponds in structure to formula VIC

$$\begin{array}{c|c} & (CH_2)_{\overline{n}} & Z \\ & X \\$$

10

- 8. The process according to claim 6 wherein the sum of m + n + p = 1.
- The process according to claim 6
 wherein said compound corresponds in structure to formula IX

$$R^{20}$$
 SO_2
 IX

wherein Z is selected group the group consisting of O, S, NR 6 , SO, SO $_2$, and NSO $_2$ R 7 , and R 6 and R 7 are defined before.

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10. The process according to claim 9 wherein Z is $\ensuremath{\mathsf{NR}^6}\xspace$.

- 11. The process according to claim 10
 5 wherein Z is O.
 - 12. The process according to claim 6 wherein said compound corresponds in structure to formula VIII

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$$\begin{array}{c|c} (CH_2)_{n} & & & \\ X & & \\ X & & \\ X & & \\ X & & \\ X & & \\ X & & \\ X & & \\ X & & \\ X & & \\ X & & \\ X & & \\$$

13. The process according to claim 6 wherein said compound corresponds in structure to formula X

$$R^{20}$$
 SO_2 X

wherein Z is selected group the group consisting of O, S, NR 6 , SO, SO $_2$, and NSO $_2$ R 7 , and R 6 and R 7 are defined before.

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14. The process according to claim 13 wherein z is NR^6 .

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15. The process according to claim 13 wherein ${\bf Z}$ is ${\bf O}$.

- 16. The process according to claim 6 wherein said compound or salt is administered a plurality of times.
- having a condition associated with pathological
 matrix metalloprotease (MMP) activity that comprises
 administering a metalloprotease inhibitor compound or
 a pharmaceutically acceptable salt thereof in an
 effective amount to a mammalian host having such a
 condition, said metalloprotease inhibitor inhibiting
 the activity of one or more of MMP-2, MMP-9 and MMP13, while exhibiting substantially less inhibitory
 activity against MMP-1, said compound corresponding
 in structure to formula VIC, below

$$R^{20}$$
 $(CH_2)_n$
 $(CH_2)_p$
 20

wherein

 R^{20} is $-NH-O-R^{14}$, where R^{14} is hydrido, a pharmaceutically acceptable cation or $C(W)R^{25}$ where W is 0 or S and R^{25} is selected from the group consisting of an C_1-C_6 -alkyl, aryl, C_1-C_6 -alkoxy, heteroaryl- C_1-C_6 -alkyl, C_3-C_8 -cycloalkyl- C_1-C_6 -alkyl,

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aryloxy, ar-C₁-C₆-alkoxy, ar-C₁-C₆-alkyl, heteroaryl and amino C₁-C₆-alkyl group wherein the amino C₁-C₆-alkyl nitrogen is (i) unsubstituted or (ii) substituted with one or two substituents independently selected from the group consisting of an C₁-C₆-alkyl, aryl, ar-C₁-C₆-alkyl, C₃-C₈-cycloalkyl-C₁-C₆-alkyl, ar-C₁-C₆-alkoxycarbonyl, C₁-C₆-alkoxycarbonyl, and C₁-C₆-alkanoyl radical, or (iii) wherein the amino C₁-C₆-alkyl nitrogen and two substituents attached thereto form a 5- to 8-membered heterocyclo or heteroaryl ring;

g is 2;

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m is zero, 1 or 2;

n is zero, 1 or 2;

p is zero, 1 or 2;

the sum of m + n + p = 1, 2, 3 or 4;

- (a) one of X, Y and Z is selected from the group consisting of C(0), NR^6 , O, S, S(0), $S(0)_2$ and $NS(0)_2R^7$, and the remaining two of X, Y and Z are CR^8R^9 , and $CR^{10}R^{11}$, or
- (b) X and Z or Z and Y together constitute a moiety that is selected from the group consisting of $NR^6C(0)$, $NR^6S(0)$, $NR^6S(0)_2$, NR^6S , NR^6O , SS, NR^6NR^6 and OC(0), with the remaining one of X, Y and Z being CR^8R^9 , or
- (c) n is zero and X, Y and Z together constitute a moiety selected from the group consisting of

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wherein wavy lines are bonds to the atoms of the depicted ring;

 $$\rm R^6$$ and $\rm R^6"$ are independently selected from the group consisting of hydrido, formyl, sulfonic-C_1-C_6-alkyl, C_1-C_6-alkoxycarbonyl-C_1-C_6-alkyl,

10 hydroxycarbonyl- C_1 - C_6 -alkyl, C_1 - C_6 -alkyl, C_1 - C_6 -alkyl, R^8R^9 -aminocarbonyl- C_1 - C_6 -alkyl, C_1 - C_6 -

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alkoxycarbonyl-C₁-C₆-alkylcarbonyl, hydroxycarbonyl-C₁-C₆-alkylcarbonyl, C₁-C₆-alkylcarbonyl-C₁-C₆alkylcarbonyl, C₁-C₆-alkoxycarbonylcarbonyl, hydroxycarbonylcarbonyl, C₁-C₆-alkylcarbonylcarbonyl, 5 R^8R^9 -aminocarbonylcarbonyl, C_1 - C_6 -alkanoyl, aryl- C_1 - C_6 -alkyl, aroyl, bis(C_1 - C_6 -alkoxy- C_1 - C_6 -alkyl)- C_1 - C_6 alkyl, C_1 - C_6 -alkyl, C_1 - C_6 -haloalkyl, C_1 - C_6 perfluoroalkyl, C₁-C₆-trifluoromethylalkyl, C₁-C₆perfluoroalkoxy-C₁-C₆-alkyl, C₁-C₆-alkoxy-C₁-C₆alkyl, C3-C6-cycloalkyl, heteroarycarbonyl, 10 heterocyclocarbonyl, C3-C8-heterocycloalkyl, C3-C8heterocycloalkylcarbonyl, aryl, C5-C6-heterocyclo, C₅-C₆-heteroaryl, C₃-C₈-cycloalkyl-C₁-C₆-alkyl, $aryloxy-C_1-C_6-alkyl$, heteroaryloxy- $C_1-C_6-alkyl$, 15 C₁-C₆-alkyl, arylsulfonyl, C₁-C₆-alkylsulfonyl, C₅-C₆-heteroarylsulfonyl, carboxy-C₁-C₆-alkyl, C₁-C₄alkoxycarbonyl-C₁-C₆-alkyl, aminocarbonyl, C₁-C₆alkyl(R8N)iminocarbonyl, aryl(R8N)iminocarbonyl, C5-C₆-heterocyclo(R⁸N)iminocarbonyl, arylthio-C₁-C₆-20 alkyl, C_1-C_6 -alkylthio- C_1-C_6 -alkyl, arylthio- C_3-C_6 alkenyl, C₁-C₄-alkylthio-C₃-C₆-alkenyl, C₅-C₆heteroaryl-C₁-C₆-alkyl, halo-C₁-C₆-alkanoyl, hydroxy- C_1 - C_6 -alkanoyl, thiol- C_1 - C_6 -alkanoyl, C_3 - C_6 -alkenyl, C_3 - C_6 -alkynyl, C_1 - C_4 -alkoxy- C_1 - C_4 -alkyl, C_1 - C_5 -25 alkoxycarbonyl, aryloxycarbonyl, NR⁸R⁹- (R^8) iminomethyl, $NR^8R^9-C_1-C_5$ -alkylcarbonyl, hydroxy5

 C_1 - C_5 -alkyl, R^8R^9 -aminocarbonyl, R^8R^9 -aminocarbonyl- C_1 - C_6 -alkylcarbonyl, hydroxyaminocarbonyl, R^8R^9 -aminosulfonyl, R^8R^9 -aminosulfon- C_1 - C_6 -alkyl, R^8R^9 -amino- C_1 - C_6 -alkylsulfonyl and an R^8R^9 -amino- C_1 - C_6 -alkyl group;

 R^7 is selected from the group consisting of a arylalkyl, aryl, heteroaryl, heterocyclo, C_1 - C_6 -alkyl, C_3 - C_6 -alkynyl, C_3 - C_6 -alkenyl, C_1 - C_6 -carboxyalkyl and a C_1 - C_6 -hydroxyalkyl group;

 R^8 and R^9 and R^{10} and R^{11} are independently 10 selected from the group consisting of a hydrido, hydroxy, C₁-C₆-alkyl, C₁-C₆-alkanoyl, aroyl, aryl, ar-C₁-C₆-alkyl, heteroaryl, heteroar-C₁-C₆-alkyl, C₂- C_6 -alkynyl, C_2 - C_6 -alkenyl, thiol- C_1 - C_6 -alkyl, C_1 - C_6 alkylthio- C_1 - C_6 -alkyl, cycloalkyl, cycloalkyl- C_1 - C_6 -15 alkyl, heterocycloalkyl-C₁-C₆-alkyl, C₁-C₆-alkoxy-C₁- C_6 -alkyl, aralkoxy- C_1 - C_6 -alkyl, C_1 - C_6 -alkoxy- C_1 - C_6 alkoxy-C₁-C₆-alkyl, hydroxy-C₁-C₆-alkyl, $hydroxycarbonyl-C_1-C_6-alkyl$, $hydroxycarbonylar-C_1-C_6-alkyl$ 20 alkyl, aminocarbonyl-C₁-C₆-alkyl, aryloxy-C₁-C₆alkyl, heteroaryloxy-C₁-C₆-alkyl, arylthio-C₁-C₆alkyl, heteroarylthio-C₁-C₆-alkyl, the sulfoxide or sulfone of any said thio substituents, perfluoro-C1-C₆-alkyl, trifluoromethyl-C₁-C₆-alkyl, halo-C₁-C₆alkyl, alkoxycarbonylamino-C₁-C₆-alkyl and an amino-25 C_1-C_6 -alkyl group wherein the aminoalkyl nitrogen is

(i) unsubstituted or (ii) substituted with one or two

radicals independently selected from the group consisting of C₁-C₆-alkyl, ar-C₁-C₆-alkyl, cycloalkyl and C₁-C₆-alkanoyl, or wherein R⁸ and R⁹ or R¹⁰ and R¹¹ and the carbon to which they are bonded form a carbonyl group, or wherein R⁸ and R⁹ or R¹⁰ and R¹¹, or R⁸ and R¹⁰ together with the atoms to which they are bonded form a 5- to 8-membered carbocyclic ring, or a 5- to 8-membered heterocyclic or heteroaryl ring containing one or two heteroatoms that are nitrogen, oxygen, or sulfur, with the proviso that only one of R⁸ and R⁹ or R¹⁰ and R¹¹ is hydroxy;

 R^{12} and R^{12} ' are independently selected from the group consisting of a hydrido, C₁-C₆-alkyl, aryl, ar-C1-C6-alkyl, heteroaryl, heteroaralkyl, C2-C6-alkynyl, C2-C6-alkenyl, thiol-C1-C6-alkyl, 15 cycloalkyl, cycloalkyl-C1-C6-alkyl, heterocycloalkyl c_1-c_6 -alkyl, c_1-c_6 -alkoxy- c_1-c_6 -alkyl, aryloxy- c_1-c_6 alkyl, amino- C_1 - C_6 -alkyl, C_1 - C_6 -alkoxy- C_1 - C_6 -alkoxy- $C_1-C_6-alkyl$, hydroxy- $C_1-C_6-alkyl$, hydroxycarbonyl- $C_1-alkyl$ C6-alkyl, hydroxycarbonylar-C1-C6-alkyl, 20 aminocarbonyl-C₁-C₆-alkyl, aryloxy-C₁-C₆-alkyl, heteroaryloxy-C₁-C₆-alkyl, C₁-C₆-alkylthio-C₁-C₆alkyl, arylthio-C₁-C₆-alkyl, heteroarylthio-C₁-C₆alkyl, the sulfoxide or sulfone of any said thio 25 substituents, perfluoro-C1-C6-alkyl, trifluoromethyl- C_1 - C_6 -alkyl, halo- C_1 - C_6 -alkyl, alkoxycarbonylamino- C_1 - C_6 -alkyl and an amino- C_1 - C_6 -alkyl group wherein the aminoalkyl nitrogen is (i) unsubstituted or (ii)

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substituted with one or two radicals independently selected from the group consisting of C_1 - C_6 -alkyl, ar- C_1 - C_6 -alkyl, cycloalkyl and C_1 - C_6 -alkanoyl;

 R^{13} is selected from the group consisting of a hydrido, benzyl, phenyl, C_1 - C_6 -alkyl, C_2 - C_6 -alkynyl, C_2 - C_6 -alkenyl and a C_1 - C_6 -hydroxyalkyl group;

-E-Y is a substituent of whose members, the moiety E is selected from the group consisting of

10 (1) $-CO(R^{19})$ or $-(R^{19})CO$, wherein R^{19} is a heterocycloalkyl, or a cycloalkyl group;

- (2) -CONH- or -HNCO-; and
- (3) -CO-;

15 (4) $-SO_2-R^{19}- \text{ or } -R^{19}-SO_2-;$

- $(5) -SO_2 -;$
- (6) $-NH-SO_2- or -SO_2-NH-;$
- (7) -S-;
- (8) -NH-CO-O- or -O-CO-NH-; or

20 (9) E is absent and Y is bonded directly to the depicted Q ring; and

the moiety Y is absent or is selected from the group consisting of a hydrido, alkyl, alkoxy, haloalkyl, aryl, aralkyl, cycloalkyl, heteroaryl,

- 25 hydroxy, aryloxy, aralkoxy, heteroaryloxy, heteroaralkyl, perfluoroalkoxy, perfluoroalkylthio, trifluoromethylalkyl, alkenyl, heterocycloalkyl, cycloalkyl, trifluoromethyl, alkoxycarbonyl, and a aminoalkyl group, wherein the aryl, heteroaryl,
- 30 aralkyl or heterocycloalkyl group is (i) unsubstituted or (ii) substituted with one or two

radicals independently selected from the group consisting of an alkanoyl, halo, nitro, aralkyl, aryl, alkoxy, trifluoroalkyl, trifluoroalkoxy and an amino group wherein the amino nitrogen is (i) unsubstituted or (ii) substituted with one or two groups independently selected from hydrido, alkyl, and an aralkyl group.

- 18. The process according to claim 17 wherein Z is O, S or NR^6 .
 - 19. The process according to claim 17 wherein m = zero, n = 1, p = 1, and $z = NR^6$.
- 15 20. The process according to claim 17 wherein m = zero, n = 1, p = 1, and Z is O.
- 21. The process according to claim 17

 wherein said compound corresponds in structure to

 20 formula VIC-1

$$\begin{array}{c|c} (CH_2)_{\overline{n}} & Z \\ X \\ X \\ X \\ X \\ CCH_2)_{\overline{p}} & CCH_2)_{\overline{p}} \\ S(O)_g \\ VIC-1 \end{array}$$

22. The process according to claim 17
25 wherein said compound corresponds in structure to
Formula VIC-2

$$(CH_2)_n$$
 Z N E Y $S(O)_g$ $VIC-2$

23. The process according to claim 17 wherein said compound corresponds in structure to formula IX-1

24. The process according to claim 13 wherein said compound corresponds in structure to 10 formula IX-2

25. A compound corresponding in structure to formula B, below, or a pharmaceutically acceptable salt thereof:

wherein

substituent R^{20} is (a) $-0-R^{21}$, where R^{21} is selected from the group consisting of a hydrido, C1-C6-alkyl, aryl, ar-C1-C6-alkyl group and a pharmaceutically acceptable cation, (b) $-NH-O-R^{22}$ wherein R²² is a selectively removable protecting group, (c) $-NH-O-R^{14}$, where R^{14} is hydrido, a pharmaceutically acceptable cation or $C(W)R^{25}$ where W10 is 0 or S and R^{25} is selected from the group consisting of an C_1 - C_6 -alkyl, aryl, C_1 - C_6 -alkoxy, heteroaryl-C₁-C₆-alkyl, C₃-C₈-cycloalkyl-C₁-C₆-alkyl, aryloxy, ar-C₁-C₆-alkoxy, ar-C₁-C₆-alkyl, heteroaryl and amino C_1-C_6 -alkyl group wherein the amino C_1-C_6 -15 alkyl nitrogen is (i) unsubstituted or (ii) substituted with one or two substituents independently selected from the group consisting of an C_1-C_6 -alkyl, aryl, ar- C_1-C_6 -alkyl, C_3-C_8 cycloalkyl-C₁-C₆-alkyl, ar-C₁-C₆-alkoxycarbonyl, C₁-C₆-alkoxycarbonyl, and C₁-C₆-alkanoyl radical, or (iii) wherein the amino C₁-C₆-alkyl nitrogen and two substituents attached thereto form a 5- to 8-membered heterocyclo or heteroaryl ring, or (d) -NR²⁶R²⁷, where R^{26} and R^{27} are independently selected from the 25

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group consisting of a hydrido, C_1 - C_6 -alkyl, amino C_1 - C_6 -alkyl, hydroxy C_1 - C_6 -alkyl, aryl, ar- C_1 - C_6 -alkyl group, or R^{26} and R^{27} together with the depicted nitrogen atom form a 5- to 8-membered ring containing zero or one additional heteroatom that is oxygen, nitrogen or sulfur;

g is zero, 1 or 2;

m is zero, 1 or 2;

n is zero, 1 or 2;

10 p is zero, 1 or 2;

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the sum of m + n + p = 1, 2, 3 or 4;

- (a) one of X, Y and Z is selected from the group consisting of C(0), NR^6 , O, S, S(0), $S(0)_2$ and $NS(0)_2R^7$, and the remaining two of X, Y and Z are CR^8R^9 , and $CR^{10}R^{11}$, or
- (b) X and Z or Z and Y together constitute a moiety that is selected from the group consisting of $NR^6C(0)$, $NR^6S(0)$, $NR^6S(0)_2$, NR^6S , NR^6O , SS, NR^6NR^6 and OC(0), with the remaining one of X, Y and Z being CR^8R^9 , or
- (c) n is zero and X, Y and Z together constitute a moiety selected from the group consisting of

$$R^{6}$$
 R^{6}
 R^{6

5 wherein wavy lines are bonds to the atoms of the depicted ring;

 $_{\rm R}^6$ and $_{\rm R}^6$ are independently selected from the group consisting of hydrido, formyl, sulfonic-C $_1$ -C $_6$ -alkyl, C $_1$ -C $_6$ -alkoxycarbonyl-C $_1$ -C $_6$ -alkyl,

hydroxycarbonyl- C_1 - C_6 -alkyl, C_1 - C_6 -alkylcarbonyl- C_1 - C_6 -alkyl, R^8R^9 -aminocarbonyl- C_1 - C_6 -alkyl, C_1 - C_6 -

alkoxycarbonyl-C₁-C₆-alkylcarbonyl, hydroxycarbonyl-C₁-C₆-alkylcarbonyl, C₁-C₆-alkylcarbonyl-C₁-C₆alkylcarbonyl, C₁-C₆-alkoxycarbonylcarbonyl, hydroxycarbonylcarbonyl, C₁-C₆-alkylcarbonylcarbonyl, R^8R^9 -aminocarbonylcarbonyl, C_1 - C_6 -alkanoyl, aryl- C_1 - C_6 -alkyl, aroyl, bis(C_1 - C_6 -alkoxy- C_1 - C_6 -alkyl)- C_1 - C_6 alkyl, C_1 - C_6 -alkyl, C_1 - C_6 -haloalkyl, C_1 - C_6 perfluoroalkyl, C₁-C₆-trifluoromethylalkyl, C₁-C₆perfluoroalkoxy-C₁-C₆-alkyl, C₁-C₆-alkoxy-C₁-C₆alkyl, C3-C6-cycloalkyl, heteroarycarbonyl, 10 heterocyclocarbonyl, C3-C8-heterocycloalkyl, C3-C8heterocycloalkylcarbonyl, aryl, C5-C6-heterocyclo, C_5-C_6 -heteroaryl, C_3-C_8 -cycloalkyl- C_1-C_6 -alkyl, $aryloxy-C_1-C_6-alkyl$, heteroaryloxy- $C_1-C_6-alkyl$, $heteroaryl-C_1-C_6-alkoxy-C_1-C_6-alkyl$, heteroarylthio-15 C₁-C₆-alkyl, arylsulfonyl, C₁-C₆-alkylsulfonyl, C₅-C6-heteroarylsulfonyl, carboxy-C1-C6-alkyl, C1-C4alkoxycarbonyl-C₁-C₆-alkyl, aminocarbonyl, C₁-C₆alkyl(R8N)iminocarbonyl, aryl(R8N)iminocarbonyl, C5-C₆-heterocyclo(R⁸N)iminocarbonyl, arylthio-C₁-C₆-20 alkyl, c_1-c_6 -alkylthio- c_1-c_6 -alkyl, arylthio- c_3-c_6 alkenyl, C_1-C_4 -alkylthio- C_3-C_6 -alkenyl, C_5-C_6 heteroaryl-C₁-C₆-alkyl, halo-C₁-C₆-alkanoyl, hydroxy c_1-c_6 -alkanoyl, thiol- c_1-c_6 -alkanoyl, c_3-c_6 -alkenyl, C_3-C_6 -alkynyl, C_1-C_4 -alkoxy- C_1-C_4 -alkyl, C_1-C_5 -25 alkoxycarbonyl, aryloxycarbonyl, NR⁸R⁹-(R⁸)iminomethyl, NR⁸R⁹-C₁-C₅-alkylcarbonyl, hydroxy-

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 C_1 - C_5 -alkyl, R^8R^9 -aminocarbonyl, R^8R^9 -aminocarbonyl- C_1 - C_6 -alkylcarbonyl, hydroxyaminocarbonyl, R^8R^9 -aminosulfonyl, R^8R^9 -aminosulfon- C_1 - C_6 -alkyl, R^8R^9 -amino- C_1 - C_6 -alkylsulfonyl and an R^8R^9 -amino- C_1 - C_6 -alkyl group;

 R^7 is selected from the group consisting of a arylalkyl, aryl, heteroaryl, heterocyclo, C_1 - C_6 -alkyl, C_3 - C_6 -alkynyl, C_3 - C_6 -alkenyl, C_1 - C_6 -carboxyalkyl and a C_1 - C_6 -hydroxyalkyl group;

 R^8 and R^9 and R^{10} and R^{11} are independently 10 selected from the group consisting of a hydrido, hydroxy, C₁-C₆-alkyl, C₁-C₆-alkanoyl, aroyl, aryl, ar-C₁-C₆-alkyl, heteroaryl, heteroar-C₁-C₆-alkyl, C₂-C6-alkynyl, C2-C6-alkenyl, thiol-C1-C6-alkyl, C1-C6alkylthio- C_1 - C_6 -alkyl, cycloalkyl, cycloalkyl- C_1 - C_6 -15 alkyl, heterocycloalkyl- C_1 - C_6 -alkyl, C_1 - C_6 -alkoxy- C_1 - C_6 -alkyl, aralkoxy- C_1 - C_6 -alkyl, C_1 - C_6 -alkoxy- C_1 - C_6 alkoxy-C₁-C₆-alkyl, hydroxy-C₁-C₆-alkyl, hydroxycarbonyl-C₁-C₆-alkyl, hydroxycarbonylar-C₁-C₆alkyl, aminocarbonyl-C₁-C₆-alkyl, aryloxy-C₁-C₆-20 alkyl, heteroaryloxy-C₁-C₆-alkyl, arylthio-C₁-C₆alkyl, heteroarylthio-C1-C6-alkyl, the sulfoxide or sulfone of any said thio substituents, perfluoro-C1-C₆-alkyl, trifluoromethyl-C₁-C₆-alkyl, halo-C₁-C₆alkyl, alkoxycarbonylamino-C1-C6-alkyl and an amino-25 C₁-C₆-alkyl group wherein the aminoalkyl nitrogen is (i) unsubstituted or (ii) substituted with one or two 10

radicals independently selected from the group consisting of C_1 - C_6 -alkyl, ar- C_1 - C_6 -alkyl, cycloalkyl and C_1-C_6 -alkanoyl, or wherein \mathbf{R}^8 and \mathbf{R}^9 or \mathbf{R}^{10} and R¹¹ and the carbon to which they are bonded form a 5 carbonyl group, or wherein R^8 and R^9 or R^{10} and R^{11} , or R^8 and R^{10} together with the atoms to which they are bonded form a 5- to 8-membered carbocyclic ring, or a 5- to 8-membered heterocyclic or heteroaryl ring containing one or two heteroatoms that are nitrogen, oxygen, or sulfur, with the proviso that only one of R^8 and R^9 or R^{10} and R^{11} is hydroxy;

 ${\tt R}^{12}$ and ${\tt R}^{12}$ ' are independently selected from the group consisting of a hydrido, C1-C6-alkyl, aryl, ar-C₁-C₆-alkyl, heteroaryl, heteroaralkyl, C₂- C_6 -alkynyl, C_2 - C_6 -alkenyl, thiol- C_1 - C_6 -alkyl, cycloalkyl, cycloalkyl-C1-C6-alkyl, heterocycloalkyl- C_1-C_6 -alkyl, C_1-C_6 -alkoxy- C_1-C_6 -alkyl, aryloxy- C_1-C_6 alkyl, amino-C₁-C₆-alkyl, C₁-C₆-alkoxy-C₁-C₆-alkoxy-C₁-C₆-alkyl, hydroxy-C₁-C₆-alkyl, hydroxycarbonyl-C₁-C6-alkyl, hydroxycarbonylar-C1-C6-alkyl, aminocarbonyl-C₁-C₆-alkyl, aryloxy-C₁-C₆-alkyl, heteroaryloxy-C₁-C₆-alkyl, C₁-C₆-alkylthio-C₁-C₆alkyl, arylthio-C₁-C₆-alkyl, heteroarylthio-C₁-C₆alkyl, the sulfoxide or sulfone of any said thio 25 substituents, perfluoro-C₁-C₆-alkyl, trifluoromethyl-C1-C6-alkyl, halo-C1-C6-alkyl, alkoxycarbonylamino- C_1-C_6 -alkyl and an amino- C_1-C_6 -alkyl group wherein the aminoalkyl nitrogen is (i) unsubstituted or (ii)

substituted with one or two radicals independently selected from the group consisting of C_1 - C_6 -alkyl, ar- C_1 - C_6 -alkyl, cycloalkyl and C_1 - C_6 -alkanoyl;

 R^{13} is selected from the group consisting of a hydrido, benzyl, phenyl, C_1 - C_6 -alkyl, C_2 - C_6 -alkynyl, C_2 - C_6 -alkenyl and a C_1 - C_6 -hydroxyalkyl group; and

-Q-A-R-E-Y is a substituent in which the moiety Q is a 5- to 7-membered heterocyclic ring

10 containing one or two nitrogen atoms one of which is bonded the depicted phenyl group, and whose remaining members (A-R-E-Y) are bonded at the 4-position relative to said phenyl-bonded nitrogen atom when Q is a 6- or 7-membered ring and at the 3- or 4
15 position relative to that nitrogen when Q is a 5-membered ring;

A is selected from the group consisting of

- (1) -0-;
- (2) -S-;
- 20 (3) $-NR^{17}$ -;
 - (4) $-CO-N(R^{17})$ or $-N(R^{17})-CO-$, wherein R^{17} is hydrogen, C_1-C_4 -alkyl, or phenyl;
 - (5) -CO-O- or -O-CO-;
 - (6) -O-CO-O-;
- 25 (7) -HC=CH-;
 - (8) —NH-CO-NH-;
 - (9) -C≡C-;
 - (10) —NH-CO-O- or -O-CO-NH-;
 - (11) -N=N-;
- 30 (12) -NH-NH-; and

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(13) $-CS-N(R^{18})-$ or $-N(R^{18})-CS-$, wherein R^{18} is hydrogen C_1-C_4 -alkyl, or phenyl; or

(14) A is absent and Q is bonded directly to R;

R is a moiety selected from the group consisting of alkyl, alkoxyalkyl, aryl, heteroaryl, cycloalkyl, heterocycloalkyl, aralkyl, heteroaralkyl, heterocycloalkylalkyl, cycloalkylalkyl,

10 cycloalkoxyalkyl, heterocycloalkoxyalkyl, aryloxyalkyl, heteroaryloxyalkyl, arylthioalkyl, heteroarylthioalkyl, cycloalkylthioalkyl, and a heterocycloalkylthioalkyl group wherein the aryl or heteroaryl or cycloalkyl or heterocycloalkyl

- substituent is (i) unsubstituted or (ii) substituted with one or two radicals selected from the group consisting of a halo, alkyl, perfluoroalkyl, perfluoroalkoxy, perfluoroalkylthio, trifluoromethylalkyl, amino, alkoxycarbonylalkyl,
- alkoxy, C₁-C₂-alkylene-dioxy, hydroxycarbonylalkyl, hydroxycarbonylalkylamino, nitro, hydroxy, hydroxyalkyl, alkanoylamino, and a alkoxycarbonyl group, and R is other than alkyl or alkoxyalkyl when A is -O- or -S-;
- 25 the moiety E is selected from the group consisting of
 - (1) $-CO(R^{19})$ or $-(R^{19})CO$ -, wherein R^{19} is a heterocycloalkyl, or a cycloalkyl group;
- 30 (2) -CONH- or -HNCO-; and
 - (3) -CO-;

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(4) $-SO_2-R^{19}- \text{ or } -R^{19}-SO_2-;$

- $(5) -SO_2 -;$
- (6) $-NH-SO_2- \text{ or } -SO_2-NH-;$
- (7) -S-;
- (8) -NH-CO-O- or -O-CO-NH-; or
- (9) E is absent and R is bonded directly to Y; and

the moiety Y is absent or is selected from the group consisting of a hydrido, alkyl, alkoxy, 10 haloalkyl, aryl, aralkyl, cycloalkyl, heteroaryl, hydroxy, aryloxy, aralkoxy, heteroaryloxy, heteroaralkyl, perfluoroalkoxy, perfluoroalkylthio, trifluoromethylalkyl, alkenyl, heterocycloalkyl, cycloalkyl, trifluoromethyl, alkoxycarbonyl, and a 15 aminoalkyl group, wherein the aryl, heteroaryl, aralkyl or heterocycloalkyl group is (i) unsubstituted or (ii) substituted with one or two radicals independently selected from the group consisting of an alkanoyl, halo, nitro, aralkyl, 20 aryl, alkoxy, trifluoroalkyl, trifluoroalkoxy and an amino group wherein the amino nitrogen is (i) unsubstituted or (ii) substituted with one or two groups independently selected from hydrido, alkyl, and an aralkyl group.

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- 26. The compound or salt according to claim 25 wherein A is -O- or -S-.
- 27. The compound or salt according to 30 claim 25 wherein A is absent.

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- 28. The compound or salt according to claim 25 wherein R is an aryl, heteroaryl, cycloalkyl or heterocycloalkyl group.
- 5 29. The compound or salt according to claim 25 wherein \mathbb{R}^{14} is hydrido.
- 30. The compound or salt according to claim 25 wherein W of the $C(W)R^{15}$ is 0 and R^{15} is a 10 C_1 - C_6 -alkyl, aryl, C_1 - C_6 -alkoxy, heteroaryl- C_1 - C_6 -alkyl, C_3 - C_8 -cycloalkyl- C_1 - C_6 -alkyl, or aryloxy group.
- 31. The compound or salt according to 15 claim 25 wherein the sum of m + n + p = 1 or 2.
 - 32. The compound or salt according to claim 25 wherein the sum of m + n + p = 1.
- 20 33. The compound or salt according to claim 25 wherein the moiety Q is a 5-membered ring.
 - 34. The compound or salt according to claim 25 wherein the moiety Q is a 7-membered ring.
 - 35. The compound or salt according to claim 25 wherein the moiety Q is a 6-membered ring.
- 36. The compound or salt according to 30 claim 35 wherein said compound corresponds in structure to formula B-A

wherein Z is selected group the group consisting of O, S, NR 6 , SO, SO $_2$, and NSO $_2$ R 7 , and R 6 and R 7 are defined before.

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- 37. The compound or salt according to claim 36 wherein Z is NR^6 .
- 38. The compound or salt according to 10 claim 36 wherein Z is O.
 - 39. The compound or salt according to claim 25 wherein the moiety Q contains two nitrogen atoms in the ring.

- 40. The compound or salt according to claim 25 wherein R^{20} is $-NH-O-R^{22}$.
- 41. The compound or salt according to 20 claim 25 wherein R^{20} is $-NH-O-R^{14}$.
 - 42. A compound corresponding in structure to formula B-A, below, or a pharmaceutically acceptable salt thereof

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HONH
$$S(0)_2$$
 $B-A$ $B-A$

wherein

Z is selected from the group consisting of C(0), NR^6 , O, S, S(0), $S(0)_2$ and $NS(0)_2R^7$;

 R^6 is selected from the group consisting of hydrido, formyl, sulfonic-C₁-C₆-alkyl, C₁-C₆alkoxycarbonyl-C₁-C₆-alkyl, hydroxycarbonyl-C₁-C₆alkyl, C₁-C₆-alkylcarbonyl-C₁-C₆-alkyl, R⁸R⁹- ${\tt aminocarbonyl-C_1-C_6-alkyl,\ C_1-C_6-alkoxycarbonyl-C_1-}$ 10 C₆-alkylcarbonyl, hydroxycarbonyl-C₁-C₆alkylcarbonyl, C₁-C₆-alkylcarbonyl-C₁-C₆alkylcarbonyl, C₁-C₆-alkoxycarbonylcarbonyl, hydroxycarbonylcarbonyl, C₁-C₆-alkylcarbonylcarbonyl, ${\tt R^8R^9-aminocarbonylcarbonyl}$, ${\tt C_1-C_6-alkanoyl}$, aryl-C_1-15 C_6 -alkyl, aroyl, bis(C_1 - C_6 -alkoxy- C_1 - C_6 -alkyl)- C_1 - C_6 alkyl, C_1 - C_6 -alkyl, C_1 - C_6 -haloalkyl, C_1 - C_6 perfluoroalkyl, C₁-C₆-trifluoromethylalkyl, C₁-C₆perfluoroalkoxy-C₁-C₆-alkyl, C₁-C₆-alkoxy-C₁-C₆alkyl, C3-C6-cycloalkyl, heteroarycarbonyl, 20 heterocyclocarbonyl, C3-C8-heterocycloalkyl, C3-C8heterocycloalkylcarbonyl, aryl, C5-C6-heterocyclo,

C5-C6-heteroaryl, C3-C8-cycloalkyl-C1-C6-alkyl,

aryloxy-C₁-C₆-alkyl, heteroaryloxy-C₁-C₆-alkyl,

heteroaryl-C₁-C₆-alkoxy-C₁-C₆-alkyl, het roarylthio-C₁-C₆-alkyl, arylsulfonyl, C₁-C₆-alkylsulfonyl, C₅-C6-heteroarylsulfonyl, carboxy-C1-C6-alkyl, C1-C4alkoxycarbonyl-C₁-C₆-alkyl, aminocarbonyl, C₁-C₆-5 alkyl(R⁸N)iminocarbonyl, aryl(R⁸N)iminocarbonyl, C₅-C6-heterocyclo(R8N)iminocarbonyl, arylthio-C1-C6alkyl, C₁-C₆-alkylthio-C₁-C₆-alkyl, arylthio-C₃-C₆alkenyl, C₁-C₄-alkylthio-C₃-C₆-alkenyl, C₅-C₆heteroaryl-C₁-C₆-alkyl, halo-C₁-C₆-alkanoyl, hydroxy-C₁-C₆-alkanoyl, thiol-C₁-C₆-alkanoyl, C₃-C₆-alkenyl, C_3-C_6 -alkynyl, C_1-C_4 -alkoxy- C_1-C_4 -alkyl, C_1-C_5 alkoxycarbonyl, aryloxycarbonyl, NR⁸R⁹-(R⁸)iminomethyl, NR⁸R⁹-C₁-C₅-alkylcarbonyl, hydroxy-C₁-C₅-alkyl, R⁸R⁹-aminocarbonyl, R⁸R⁹-aminocarbonyl- C_1-C_6 -alkylcarbonyl, hydroxyaminocarbonyl, R^8R^9 -15 aminosulfonyl, R⁸R⁹-aminosulfon-C₁-C₆-alkyl, R⁸R⁹amino-C₁-C₆-alkylsulfonyl and an R⁸R⁹-amino-C₁-C₆alkyl group;

R⁷ is selected from the group consisting of a arylalkyl, aryl, heteroaryl, heterocyclo, C₁-C₆-alkyl, C₃-C₆-alkyl, C₃-C₆-alkelyl, C₁-C₆-carboxyalkyl and a C₁-C₆-hydroxyalkyl group;

R⁸ and R⁹ are independently selected from the group consisting of a hydrido, hydroxy, C₁-C₆-25 alkyl, C₁-C₆-alkanoyl, aroyl, aryl, ar-C₁-C₆-alkyl, heteroaryl, heteroar-C₁-C₆-alkyl, C₂-C₆-alkynyl, C₂-

C6-alkenyl, thiol-C1-C6-alkyl, C1-C6-alkylthio-C1-C6alkyl, cycloalkyl, cycloalkyl-C1-C6-alkyl, heterocycloalkyl-C₁-C₆-alkyl, C₁-C₆-alkoxy-C₁-C₆alkyl, aralkoxy-C₁-C₆-alkyl, C₁-C₆-alkoxy-C₁-C₆alkoxy-C₁-C₆-alkyl, hydroxy-C₁-C₆-alkyl, hydroxycarbonyl-C₁-C₆-alkyl, hydroxycarbonylar-C₁-C₆alkyl, aminocarbonyl-C₁-C₆-alkyl, aryloxy-C₁-C₆alkyl, heteroaryloxy-C₁-C₆-alkyl, arylthio-C₁-C₆alkyl, heteroarylthio-C1-C6-alkyl, the sulfoxide or sulfone of any said thio substituents, perfluoro- C_1 -10 C_6 -alkyl, trifluoromethyl- C_1 - C_6 -alkyl, halo- C_1 - C_6 alkyl, alkoxycarbonylamino-C1-C6-alkyl and an amino-C₁-C₆-alkyl group wherein the aminoalkyl nitrogen is (i) unsubstituted or (ii) substituted with one or two radicals independently selected from the group consisting of C_1 - C_6 -alkyl, ar- C_1 - C_6 -alkyl, cycloalkyl and C_1-C_6 -alkanoyl, or wherein R^8 and R^9 and the carbon to which they are bonded form a carbonyl group, or wherein R^8 and R^9 together with the atoms to which they are bonded form a 5- to 8-membered 20 carbocyclic ring, or a 5- to 8-membered heterocyclic or heteroaryl ring containing one or two heteroatoms that are nitrogen, oxygen, or sulfur, with the proviso that only one of R⁸ and R⁹ is hydroxy; -Q-A-R-E-Y is a substituent in which the 25 moiety Q is a 6-membered heterocyclic ring containing one or two nitrogen atoms one of which is bonded the

depicted phenyl group, and whose remaining members

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(A-R-E-Y) are bonded at the 4-position relative to said phenyl-bonded nitrogen;

A is selected from the group consisting of

(1) -0-;

(2) —S-;

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20

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 $(3) -NR^{17}-;$

(4) $-CO-N(R^{17})$ or $-N(R^{17})-CO-$, wherein R^{17} is hydrogen, C_1-C_4 -alkyl, or phenyl;

(5) —CO-O- or —O-CO-;

10 (6) -0-C0-0-;

(7) -HC=CH-;

(8) -NH-CO-NH-;

(9) —C≡C-;

(10) -NH-CO-O- or -O-CO-NH-;

15 (11) -N=N-;

(12) -NH-NH-; and

(13) $-CS-N(R^{18})-$ or $-N(R^{18})-CS-$, wherein R^{18} is hydrogen C_1-C_4 -alkyl, or phenyl; or

(14) A is absent and Q is bonded directly to R;

R is a moiety selected from the group consisting of alkyl, alkoxyalkyl, aryl, heteroaryl, cycloalkyl, heterocycloalkyl, aralkyl, heteroaralkyl, heterocycloalkylalkyl, cycloalkylalkyl,

cycloalkoxyalkyl, heterocycloalkoxyalkyl, aryloxyalkyl, heteroaryloxyalkyl, arylthioalkyl, heteroarylthioalkyl, cycloalkylthioalkyl, and a heterocycloalkylthioalkyl group wherein the aryl or

30 heteroaryl or cycloalkyl or heterocycloalkyl substituent is (i) unsubstituted or (ii) substituted

with one or two radicals selected from the group consisting of a halo, alkyl, perfluoroalkyl, perfluoroalkoxy, perfluoroalkylthio, trifluoromethylalkyl, amino, alkoxycarbonylalkyl, alkoxy, C₁-C₂-alkylene-dioxy, hydroxycarbonylalkyl, hydroxycarbonylalkylamino, nitro, hydroxy, hydroxyalkyl, alkanoylamino, and a alkoxycarbonyl group, and R is other than alkyl or alkoxyalkyl when A is -O- or -S-;

10 the moiety E is selected from the group consisting of

- (1) $-CO(R^{19})$ or $-(R^{19})CO$, wherein R^{19} is a heterocycloalkyl, or a cycloalkyl group;
- 15 (2) -CONH- or -HNCO-; and
 - (3) -CO-;
 - (4) $-SO_2-R^{19}$ or $-R^{19}-SO_2$;
 - $(5) -SO_2-;$
 - (6) $-NH-SO_2-$ or $-SO_2-NH-$;
- 20 (7) -S-;
 - (8) -NH-CO-O- or -O-CO-NH-; or
 - (9) E is absent and R is bonded directly to Y; and

the moiety Y is absent or is selected from
the group consisting of a hydrido, alkyl, alkoxy,
haloalkyl, aryl, aralkyl, cycloalkyl, heteroaryl,
hydroxy, aryloxy, aralkoxy, heteroaryloxy,
heteroaralkyl, perfluoroalkoxy, perfluoroalkylthio,
trifluoromethylalkyl, alkenyl, heterocycloalkyl,
cycloalkyl, trifluoromethyl, alkoxycarbonyl, and a
aminoalkyl group, wherein the aryl, heteroaryl,
aralkyl or heterocycloalkyl group is (i)

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unsubstituted or (ii) substituted with one or two radicals independently selected from the group consisting of an alkanoyl, halo, nitro, aralkyl, aryl, alkoxy, trifluoroalkyl, trifluoroalkoxy and an amino group wherein the amino nitrogen is (i) unsubstituted or (ii) substituted with one or two groups independently selected from hydrido, alkyl, and an aralkyl group.

- 10 43. The compound or salt according to claim 42 wherein Z is O, S or NR⁶.
- 44. The compound or salt according to claim 42 wherein Z is NR⁶, and R⁶ is selected from the group consisting of C₃-C₆-cycloalkyl, C₁-C₆-alkyl, C₃-C₆-alkenyl, C₃-C₆-alkynyl, C₁-C₆-alkoxy-C₁-C₆-alkyl, amino-C₁-C₆-alkyl, aminosulfonyl, heteroaryl-C₁-C₆-alkyl, aryloxycarbonyl, and C₁-C₆-alkoxycarbonyl.

- 45. The compound or salt according to claim 42 wherein Z is O.
- 46. The compound or salt according to 25 claim 42 wherein A is absent.
 - 47. The compound or salt according to claim 46 wherein said compound corresponds in structure to formula B-2A

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HONH
$$S(0)_2$$
 $B-2A$

48. The compound or salt according to claim 47 wherein said heterocyclic ring Q contains one nitrogen atom.

49. The compound or salt according to claim 48 wherein said compound corresponds in structure to the formula

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50. A compound corresponding in structure to formula B-3A, below, or a pharmaceutically acceptable salt thereof

HONH
$$S(0)_2$$
 $B-3A$

15

wherein

Z is selected from the group consisting of C(0), NR^6 , O, S, S(0), $S(0)_2$ and $NS(0)_2R^7$;

R⁶ is selected from the group consisting of hydrido, formyl, sulfonic-C₁-C₆-alkyl, C₁-C₆alkoxycarbonyl-C₁-C₆-alkyl, hydroxycarbonyl-C₁-C₆alkyl, C₁-C₆-alkylcarbonyl-C₁-C₆-alkyl, R⁸R⁹aminocarbonyl-C₁-C₆-alkyl, C₁-C₆-alkoxycarbonyl-C₁-C₆-alkylcarbonyl, hydroxycarbonyl-C₁-C₆alkylcarbonyl, C₁-C₆-alkylcarbonyl-C₁-C₆alkylcarbonyl, C₁-C₆-alkoxycarbonylcarbonyl, 10 hydroxycarbonylcarbonyl, C₁-C₆-alkylcarbonylcarbonyl, R^8R^9 -aminocarbonylcarbonyl, C_1-C_6 -alkanoyl, aryl- C_1 - C_6 -alkyl, aroyl, bis(C_1 - C_6 -alkoxy- C_1 - C_6 -alkyl)- C_1 - C_6 alkyl, C_1-C_6 -alkyl, C_1-C_6 -haloalkyl, C_1-C_6 perfluoroalkyl, C₁-C₆-trifluoromethylalkyl, C₁-C₆-15 perfluoroalkoxy-C₁-C₆-alkyl, C₁-C₆-alkoxy-C₁-C₆alkyl, C3-C6-cycloalkyl, heteroarycarbonyl, heterocyclocarbonyl, C3-C8-heterocycloalkyl, C3-C8heterocycloalkylcarbonyl, aryl, C5-C6-heterocyclo, C5-C6-heteroaryl, C3-C8-cycloalkyl-C1-C6-alkyl, 20 aryloxy-C₁-C₆-alkyl, heteroaryloxy-C₁-C₆-alkyl, heteroaryl-C₁-C₆-alkoxy-C₁-C₆-alkyl, heteroarylthio-C₁-C₆-alkyl, arylsulfonyl, C₁-C₆-alkylsulfonyl, C₅-C₆-heteroarylsulfonyl, carboxy-C₁-C₆-alkyl, C₁-C₄alkoxycarbonyl-C₁-C₆-alkyl, aminocarbonyl, C₁-C₆alkyl(R8N)iminocarbonyl, aryl(R8N)iminocarbonyl, C5-C6-heterocyclo(R8N)iminocarbonyl, arylthio-C1-C6-

alkyl, C_1 - C_6 -alkylthio- C_1 - C_6 -alkyl, arylthio- C_3 - C_6 -alkenyl, C_1 - C_4 -alkylthio- C_3 - C_6 -alkenyl, C_5 - C_6 -heteroaryl- C_1 - C_6 -alkyl, halo- C_1 - C_6 -alkanoyl, hydroxy- C_1 - C_6 -alkanoyl, thiol- C_1 - C_6 -alkanoyl, C_3 - C_6 -alkenyl,

5 C_3-C_6 -alkynyl, C_1-C_4 -alkoxy- C_1-C_4 -alkyl, C_1-C_5 -alkoxycarbonyl, aryloxycarbonyl, NR^8R^9 - (R^8) iminomethyl, NR^8R^9 - C_1 - C_5 -alkylcarbonyl, hydroxy- C_1 - C_5 -alkyl, R^8R^9 -aminocarbonyl, R^8R^9 -aminocarbonyl- C_1 - C_6 -alkylcarbonyl, hydroxyaminocarbonyl, R^8R^9 -

aminosulfonyl, R⁸R⁹-aminosulfon-C₁-C₆-alkyl, R⁸R⁹-amino-C₁-C₆-alkylsulfonyl and an R⁸R⁹-amino-C₁-C₆-alkyl group;

 R^7 is selected from the group consisting of a arylalkyl, aryl, heteroaryl, heterocyclo, C_1 - C_6 -alkyl, C_3 - C_6 -alkynyl, C_3 - C_6 -alkenyl, C_1 - C_6 -carboxyalkyl and a C_1 - C_6 -hydroxyalkyl group;

R⁸ and R⁹ are independently selected from the group consisting of a hydrido, hydroxy, C₁-C₆-alkyl, C₁-C₆-alkanoyl, aroyl, aryl, ar-C₁-C₆-alkyl,

20 heteroaryl, heteroar-C₁-C₆-alkyl, C₂-C₆-alkynyl, C₂-C₆-alkenyl, thiol-C₁-C₆-alkyl, C₁-C₆-alkylthio-C₁-C₆-alkyl, cycloalkyl, cycloalkyl-C₁-C₆-alkyl,

heterocycloalkyl-C₁-C₆-alkyl, C₁-C₆-alkoxy-C₁-C₆-alkyl, aralkoxy-C₁-C₆-alkyl, C₁-C₆-alkoxy-C₁-C₆-alkyl, hydroxy-C₁-C₆-alkyl,

hydroxycarbonyl-C₁-C₆-alkyl, hydroxycarbonylar-C₁-C₆-alkyl,

alkyl, aminocarbonyl-C₁-C₆-alkyl, aryloxy-C₁-C₆-

alkyl, heteroaryloxy-C₁-C₆-alkyl, arylthio-C₁-C₆-alkyl, heteroarylthio-C₁-C₆-alkyl, the sulfoxide or sulfone of any said thio substituents, perfluoro-C₁-C₆-alkyl, trifluoromethyl-C₁-C₆-alkyl, halo-C₁-C₆-alkyl, alkoxycarbonylamino-C₁-C₆-alkyl and an amino-C₁-C₆-alkyl group wherein the aminoalkyl nitrogen is (i) unsubstituted or (ii) substituted with one or two radicals independently selected from the group consisting of C₁-C₆-alkyl, ar-C₁-C₆-alkyl, cycloalkyl

and C₁-C₆-alkanoyl, or wherein R⁸ and R⁹ and the carbon to which they are bonded form a carbonyl group, or wherein R⁸ and R⁹ together with the atoms to which they are bonded form a 5- to 8-membered carbocyclic ring, or a 5- to 8-membered heterocyclic or heteroaryl ring containing one or two heteroatoms that are nitrogen, oxygen, or sulfur, with the proviso that only one of R⁸ and R⁹ is hydroxy;

-Q-E-Y is a substituent in which the moiety Q is a 6-membered heterocyclic ring containing one or two nitrogen atoms one of which is bonded the depicted phenyl group, and whose remaining members (E-Y) are bonded at the 4-position relative to said phenyl-bonded nitrogen atom;

in the substituent -E-Y, the moiety E is selected from the group consisting of

- (1) $-CO(R^{19})$ or $-(R^{19})CO$ -, wherein R^{19} is a heterocycloalkyl, or a cycloalkyl group;
- (2) -CONH- or -HNCO-; and

30 (3) -CO-;

10

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(4) $-SO_2-R^{19}$ or $-R^{19}-SO_2$;

- $(5) -SO_2-;$
- (6) $-NH-SO_2- or -SO_2-NH-;$
- (7) -S-;

(8) -NH-CO-O- or -O-CO-NH-; or

(9) E is absent and Y is bonded directly to the ring Q; and

the moiety Y is absent or is selected from the group consisting of a hydrido, alkyl, alkoxy, haloalkyl, aryl, aralkyl, cycloalkyl, heteroaryl, 10 hydroxy, aryloxy, aralkoxy, heteroaryloxy, heteroaralkyl, perfluoroalkoxy, perfluoroalkylthio, trifluoromethylalkyl, alkenyl, heterocycloalkyl, cycloalkyl, trifluoromethyl, alkoxycarbonyl, and a 15 aminoalkyl group, wherein the aryl, heteroaryl, aralkyl or heterocycloalkyl group is (i) unsubstituted or (ii) substituted with one or two radicals independently selected from the group consisting of an alkanoyl, halo, nitro, aralkyl, aryl, alkoxy, trifluoroalkyl, trifluoroalkoxy and an 20 amino group wherein the amino nitrogen is (i) unsubstituted or (ii) substituted with one or two groups independently selected from hydrido, alkyl, and an aralkyl group.

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- 51. The compound or salt according to claim 50 wherein said heterocyclic ring Q contains two nitrogen atoms.
- 30 52. The compound or salt according to claim 51 wherein compound corresponds in structure to formula X

- 53. The compound or salt according toclaim 50 wherein said heterocyclic ring Q containsone nitrogen atom.
 - 54. The compound or salt according to claim 53 wherein said compound corresponds in structure to formula IX-1

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55. The compound or salt according to claim 53 wherein said compound corresponds in structure to formula IX-2

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56. The compound or salt according to claim 50 wherein z is O, S or NR^6 .

- 57. The compound or salt according to claim 56 wherein Z is NR⁶, and R⁶ is selected from the group consisting of C₃-C₆-cycloalkyl, C₁-C₆
 5 alkyl, C₃-C₆-alkenyl, C₃-C₆-alkynyl, C₁-C₆-alkoxy-C₁-C₆-alkyl, amino-C₁-C₆-alkyl, aminosulfonyl, heteroaryl-C₁-C₆-alkyl, aryloxycarbonyl, and C₁-C₆-alkoxycarbonyl.
- 10 58. The compound or salt according to claim 57 wherein said compound corresponds in structure to the formula

15 59. The compound or salt according to claim 57 wherein said compound corresponds in structure to the formula

20 60. The compound or salt according to claim 56 wherein Z is O.

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The compound or salt according to claim 60 wherein said compound corresponds in structure to the formula

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- 62. A pharmaceutical composition that comprises a compound or salt according to claim 25 dissolved or dispersed in a pharmaceutically acceptable carrier.
- 63. A pharmaceutical composition that comprises a compound according to claim 42 dissolved or dispersed in a pharmaceutically acceptable
- 15 carrier.
 - 64. A pharmaceutical composition that comprises a compound according to claim 47 dissolved or dispersed in a pharmaceutically acceptable carrier.
 - 65. A pharmaceutical composition that comprises a compound according to claim 50 dissolved or dispersed in a pharmaceutically acceptable carrier.
 - A pharmaceutical composition that comprises a compound according to claim 56 dissolved

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or dispersed in a pharmaceutically acceptable carrier.

INTERNATIONAL SEARCH REPORT

Intimilional Application No PCT/US 00/06719

A CLASSIFICATION OF SUBJECT MATTER IPC 7 C07D211/66 C07D405/12 C07D405/14 C07D309/08 C07D335/02 A61K31/445 C07D409/12 C07D401/06 C07D401/12 A61K31/4523 A61P9/00 According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

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C. DOCUM	ENTS CONSIDERED TO BE RELEVANT			
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"A" docume consider adjusted to the consider adjusted to the constant of the c	ent defining the general state of the art which is not defend to be of particular relevance document but published on or after the international date ent which may throw doubts on priority claim(s) or is cited to establish the publication date of another no or other special reason (as specified) ent referring to an oral disclosure, use, exhibition or means ent published prior to the international filing date but han the priority date claimed	"T" later document published after the interest or priority date and not in conflict with cited to understand the principle or the invention of particular relevance; the cannot be considered novel or cannot involve an inventive step when the document of particular relevance; the cannot be considered to involve an indocument is combined with one or ments, such combination being obvious in the art. "&" document member of the same patent	the application but every underlying the stained invention to considered to coursent is taken alone chaimed invention ventive step when the ore other such docuus to a person skilled	
2	1 August 2000	06/09/2000		
Name and	mailing address of the ISA European Patent Office, P.B. 5818 Patentlaan 2 NL – 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nł, Fax: (+31-70) 340-3016	Authorized officer De Jong, B		

INTERNATIONAL SEARCH REPORT

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C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT									
Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.							
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